Interactive comment on “Chemical characteristics and causes of airborne particulate pollution in warm seasons in Wuhan, central China” by X. P. Lyu et al.

Anonymous Referee #3

The manuscript by Lyu et al. offers interesting results on the chemical compositions of PM and precursors in Wuhan China, in specific campaigns in 2014. In my opinion, results are noteworthy but several revisions are necessary before publication in ACP.

Reply: We thank the reviewer for his/her positive and encouraging comments. Detailed responses to the specific comments were listed point by point below.

1. Abstract needs significant improvement. It should be self-explanatory. For examples cases explanation is missing. K is 47% of what? etc.

Reply: Thanks for the suggestion. The abstract was substantially revised, including the points raised by the reviewer.

For details, please refer to lines 15-33.

2. Introduction. The first sentence is not needed.

Reply: Accepted with thanks. In addition, we also revised the second sentence as follows.

Airborne particulate pollution, also called “haze,” has swept across China in recent years, particularly over its northern, central, and eastern parts (Cheng et al., 2014; Kang et al., 2013; Wang et al., 2013).

For details, please refer to lines 37-39.

3. English should be significantly improved. Some sentences are too simple for a scientific journal. Some terms need corrections (ammonia in page 2; the use of past tense should be avoided for general sentences like EC was the typical tracer of uncomplete combustion; aromatically instead of automatically etc....). Syntaxis also needs improvement.

Reply: Thanks for pointing out the errors. The grammatical errors and typos pointed out by the reviewer were corrected. To thoroughly improve the quality of the manuscript, we have asked an editing company to correct grammars and syntaxes before the submission of the revised version.

4. I advice the use of SIA instead of SIOA.
Reply: Accepted with thanks. Replacements were made throughout the whole manuscript.

5. Do authors mean wildfires with the term "fire spot"?

Reply: Exactly. The use of “fire spot” was replaced with “wildfires” throughout the whole manuscript.

6. Information on traffic volume next to the monitoring site is lacking.

Reply: Thanks for the useful comment. More details about the sampling site including traffic volume were provided as follows.

The traffic volume of the road was around 200 vehicles per hour. However, a wall (~2 m high) and several rows of trees (7 to 8 m high) were located between the road and the sampling site.

For details, please refer to lines 123-125.

7. The TEOM model does not include FDMS. Authors should discuss what uncertainty does this add to the conclusions.

Reply: Thanks for the comments. In this study, a filter dynamics measurement system (FDMS) was integrated into the TEOM system to correct the measurement deviation of the TEOM system caused by mass loss of semi-volatile particulate matters. Revisions were made as follows:

PM$_{10}$ and PM$_{2.5}$ were measured with a continuous ambient particulate monitor (Thermo Fisher-1405D, USA) integrated with a filter dynamics measurement system to minimize the loss of semivolatile particulate matter.

For details, please refer to lines 126-128.


Reply: Accepted with thanks. The sentences were rephrased as follows.

Bearing in mind that the sampling site, period, method, and instrument all interfere with comparisons, the ambient particulate pollution in Wuhan was severe.

For details, please refer to lines 222-224.
Because smaller particles tend to pose more harm to human health and to the atmosphere due to their larger specific surface areas (Yang et al., 2012; Goldberg et al., 2001), and because the chemical compositions in PM$_{10}$ were not analyzed, this study focused mainly on PM$_{2.5}$.

For details, please refer to lines 243-245.

9. What do you mean with "completely" in row 225?

Reply: Thanks for pointing out the inappropriate use of “completely”. Since this sentence was rewritten, this word was not used. Please refer to lines 243-245 for details.

10. Row 233: not fully true, also primary OC and EC are fine particles.

Reply: Thanks for the valuable comments. We fully agreed to the reviewer that primary OC and EC also constitute fine particles. The discussion in this part was revised as follows.

The PM$_{2.5}$/PM$_{10}$ value also increased remarkably on episode days compared to that on non-episode days, except for episode 2 (45.9% ± 2.5%), which suggests that more secondary species and/or primary fine particles (e.g., primary OC and EC generated from combustion) were generated or released during the episodes. In contrast, the lower PM$_{2.5}$/PM$_{10}$ value during episode 2 might imply a strong source of coarse particles. Indeed, this inference was confirmed by the source apportionment analysis in section 3.3.3.

For details, please refer to lines 253-258.

11. I would rather use the term episode instead of case.

Reply: Accepted with thanks. Changes were made throughout the whole manuscript.

12. Row 240: the contribution of fugitive dust is estimated in 5 ug/m$^3$, well before the Source apportionment section. Please reorder.

Reply: Thanks for the suggestion. The sentences were revised as follows.

In contrast, the lower PM$_{2.5}$/PM$_{10}$ value during episode 2 might imply a strong source of coarse particles. Indeed, this inference was confirmed by the source apportionment analysis in section 3.3.3.

For details, please refer to lines 256-258.
13. Relating sulfate to point source and nitrate to mobile ones, is too simplistic. Traffic also emit primary particles, and no3 can also come from industries.

**Reply:** The comment is highly appreciated. We fully agree with the reviewer. Since this short discussion was not closely related to the aims of this manuscript, it was removed.

14. How OCnon-comb was estimated?? the reference Cabada et al., 2004 is missing in the Bivliography. More clarification is needed here. Do they mean that biogenic OC is all primary??

**Reply:** Thanks for the question. OC non-comb means primary OC that is not related to combustion. As the intercept in the regression (Figure 5), OC non-comb did not vary with EC (the independent variable) which was primarily emitted from combustion activities. OC non-comb might include primary biogenic OC, however it did not mean that biogenic OC is all primary. In fact, this method has been extensively used to estimate POC and SOC (e.g., Chu et al., 2005; Saylor et al., 2006). The method to estimate OC non-comb was further clarified, and revisions were made in the manuscript.

where (OC/EC) prim was the ratio of primary OC to EC, obtained from the pairs of OC and EC with the OC/EC ratios among the 10% lowest; and OC non-comb was the primary OC that was not related to combustion activities. These values were determined by the slope and intercept of the linear regression between primary OC and EC, respectively (Figure 5).

For details, please refer to lines 305-308.

The missing reference was added. For details, please refer to lines 603-605.

15. Row 295: p value is missing.

**Reply:** Sorry for the carelessness. “p<0.05” was added.

For detail, please refer to line 313.

16. Row 280: provide references for value of 2.

**Reply:** Accepted with thanks.

For details, please refer to line 298.

17. Figure 6: how authors interpret daily variation of HO2.
Reply: Excellent question. The diurnal variation of HO₂ presents bimodal – one at daytime hours and another at night. The daytime pattern of HO₂ is typically bell-shaped peaked at noon or in early afternoon, consistent with the variation of atmospheric oxidative capacity (i.e., O₃ generally peaks at noon or in early afternoon). Solar radiation is the main driving factor of this pattern. Namely, when the radiation is the strongest at noon, the photochemical reactions are the most intensive, which results in the highest production of O₃ and oxidative radicals including HO₂. Another peak of HO₂ is usually observed in the evening because reactions among alkenes and O₃ and NO₃ lead to the formation of HO₂. Since concentrations of alkenes increase rapidly from late afternoon to the evening due to the decreasing photochemical consumption, increasing emissions from vehicles at rush hours in the evening, and reduction of boundary layer, the HO₂ production subsequently increases. With the consumption of O₃ and NO₃, the reactions with alkenes weaken in late evening. In addition, reactions among radicals consume HO₂. Hence, HO₂ decreases from late evening to early morning.

Overall, the mechanisms of HO₂ production and loss are complicated, and it is difficult to comprehensively discuss them in this paper. Therefore, we briefly explained the diurnal pattern of HO₂ in the revised manuscript.

Two peaks were found for the simulated diurnal pattern of HO₂, which might be caused by strong solar radiation at noon and in the early afternoon and by reactions among alkenes and O₃ and NO₃ at night (Emmerson et al., 2005; Kanaya et al., 1999).

For details, please refer to lines 318-320.

18. Section 3.2.2. there is a constrast between the conclusion that all PM components increase during episodes (row 324) and theta the OC decreases (row 329).

Reply: Sorry for the confusion. They results are not contradictory because the absolute concentrations of PM₂.₅ components increased while the fraction (percentage) of OC in PM₂.₅ decreased during the episodes.

To avoid confusion, both concentration and percentage were provided in Table 3 in the revised manuscript. Subsequently, the discussion was revised as follows.
Table 3 summarizes the mass concentrations and percentages of the main components in PM$_{2.5}$. The mass concentrations of PM$_{2.5}$ components significantly increased from nonEpisode days to episode days ($p < 0.05$). In contrast, the percentages of the chemical components in PM$_{2.5}$ varied by species. In summer, the fractions of EC and K in PM$_{2.5}$ experienced significant increases from non-episode 1 (EC, 4.8% ± 0.2%; K, 2.0% ± 0.1%) to episode 1 (EC, 5.7% ± 0.5%; K, 4.4% ± 0.3%) and episode 3 (EC, 5.3% ± 0.2%; K, 3.0% ± 0.2%). Because EC is the tracer of incomplete combustion (Chow et al., 1996) and K is the indicator of biomass burning (Saarikoski et al., 2007; Echalar et al., 1995), the higher percentages of EC and K in episodes 1 and 3 imply the outstanding contribution of biomass burning. In contrast, the fraction of OC in PM$_{2.5}$ remained stable on both episode and non-episode days ($p > 0.05$), possibly because the high temperatures in summer hindered the gas-to-particle partitioning of semivolatile organics (Takekawa et al., 2003). Furthermore, the percentages of Ca (2.9% ± 0.4%) and Fe (2.7% ± 0.3%) significantly increased during episode 2 ($p < 0.05$) compared to those in non-episode 1 (Ca, 1.1% ± 0.1%; Fe, 1.5% ± 0.1%), which shows that fugitive dust made a considerable contribution to PM$_{2.5}$ in episode 2. In addition, biomass burning might also have contributed to PM$_{2.5}$, in view of the increase in the percentage of K (non-episode 1, 2.0% ± 0.1%; episode 2, 3.2% ± 0.2%).

In autumn, the percentage of K significantly ($p < 0.05$) increased during episode 4 (3.1% ± 0.1% vs. 2.1% ± 0.1% in non-episode 2), as did that of OC (27.3% ± 0.7% vs. 20.9% ± 0.8% in non-episode 2), suggesting the dominant role of biomass burning in episode 4. Furthermore, the fractions of OC in episode 5 (23.8% ± 1.5%) and NO$_3^-$ in episode 6 (26.1% ± 1.0%) were obviously higher than those in non-episode 2 (OC, 20.9% ± 0.8%; NO$_3^-$, 19.8% ± 0.9%). Due to the complexity of the sources of OC and NO$_3^-$, the causes of episodes 5 and 6 are further explored in the following sections.

In summary, episodes 1, 3, and 4 were greatly affected by biomass burning. This finding was further confirmed by the significant increases in the gaseous tracers of biomass burning such as ethyne (C$_2$H$_2$) and methyl chloride (CH$_3$Cl) (Guo et al., 2011b; Simoneit et al., 2002) during these episodes ($p < 0.05$; see Figure S8 in the Supplement).

As this part was more related to the causes of PM$_{2.5}$ episodes, it was moved to section 3.3.2 “Chemical signatures” under section 3.3 “Causes of PM$_{2.5}$ episodes”. For details, please refer to lines 378-404, Table 3 and Figure S8 in the supplement.
19. If cases 1 and 3 are attributed to Biomass burning, why OC decreases? BB is the largest source of OC, as shown in Figures 10 and 11.

**Reply:** Thanks for the question. As explained above, the absolute concentration of OC increased remarkably in episodes 1 and 3, while the percentage of OC in PM$_{2.5}$ decreased. Since the decrease in percentage was not significant ($p>0.05$), we revised the wording to “the fraction of OC in PM$_{2.5}$ remained stable ……”. Furthermore, the possible reason for the stable percentage of OC was given.

In contrast, the fraction of OC in PM$_{2.5}$ remained stable on both episode and non-episode days ($p > 0.05$), possibly because the high temperatures in summer hindered the gas-to-particle partitioning of semivolatile organics (Takekawa et al., 2003).

For details, please refer to lines 386-389.

20. Ca and Fe also come from traffic and construction/demolition works. And K in case 2 can also be emitted by mineral sources.

**Reply:** We agreed with the reviewer. In this study, fugitive dust included dust from traffic, construction/demolition works, yard and bare soil. The following was added into the revised manuscript:

Because Fe and Ca are typical crustal elements, fugitive dust (e.g., dust from traffic, construction and demolition works, yards, and bare soil) was their most likely source.

For details, please refer to lines 342-343.

In addition, correlation analyses among the main elements were carried out. Weak correlations of K with Fe and Ca were found. Thus, the possibility of K emitted from mineral source was eliminated in this study. The following was added into the revised manuscript.

Correlation analysis indicated that Fe had good correlation with Ca ($R^2 = 0.66$; Figure S7 in the Supplement), whereas weak correlations of K with Fe ($R^2 = 0.14$) and Ca ($R^2 = 0.09$) were found, suggesting that Fe and Ca shared common sources that were different from the sources of K. Because Fe and Ca are typical crustal elements, fugitive dust (e.g., dust from traffic, construction and demolition works, yards, and bare soil) was their most likely source. In contrast, apart from emissions from mineral sources, K is also emitted from biomass burning. As such, K was
believed to be mainly emitted from biomass burning in this study, which is further supported by the moderate correlations of K with OC ($R^2 = 0.52$) and EC ($R^2 = 0.48$) because biomass burning also emits OC and EC (Saarikoski et al., 2007; Echalar et al., 1995).

For details, please refer to lines 338-347.

21. The source apportionment section lacks of many details which are needed to ensure that the solution is the omore realistic one. Why authors decide to perform separate PMFs for different cases? They at least present aldso the total (asembled) PMF, which will certainly imporve statistically signifance and redice random and rotational errors of the solution.

**Reply:** The excellent comments are highly appreciated. The source apportionment part was revised substantially according to the reviewer’s suggestions. The revisions included:

(1) Source apportionment was performed together with all case data collected in both episodes and non-episodes. However, since summer data of water soluble ions (WSIs) were not available, the source apportionments in summer (including episodes 1, 2, 3 and non-episode 1) and autumn (including episodes 4, 5, 6 and non-episode 2) were still separately conducted.

For details, please refer to section 3.3.3 “source apportionment”.

(2) The selection criteria for best solution and the evidences (Q value, residual, G-space plot and etc.) were provided.

The selection of the factor number and the best solution depended upon the following criteria. (1) A lower Q value (Equation 6; a function to evaluate the model runs) was preferable. (2) The ratio between $Q_{\text{robust}}$ and $Q_{\text{true}}$ was lower than 1.5. In this study, the ratios were 0.8 and 0.9 for the summer and autumn data simulation, respectively. (3) Good agreement was shown between the predicted and observed PM$_{2.5}$. The slope and correlation coefficient ($R^2$) for the linear regression were 0.91 and 0.86 in summer and 0.95 and 0.98 in autumn, respectively, as shown in Figure S2 in the Supplement. The lower $R^2$ value seen during the summer might be due to the lack of WSI data. (4) The residuals were normally distributed between $-3$ and 3. Table S2 summarizes the percentage of samples with residuals between $-3$ and 3 for each species; the lowest percentages were 92.9% and 96.0% for Ni in summer and autumn, respectively. The scaled residuals for PM$_{2.5}$ are shown in Figure S3 in the Supplement. The percentage of residuals between $-3$ and 3
was comparable between summer (97.5%) and autumn (98.1%). Finally (5), no correlation was found between the factors, which was achieved by examining the G-space plots and controlled by the FPEAK model runs. Figures S4 and S5 in the Supplement present the G-space plots in summer and autumn, respectively. The low factor contributions and poor correlations indicated that rotational ambiguity was effectively controlled.

For details, please refer to lines 189-204.

22. The lack of SIA in summer is a critical issue. They are a major contributor to the mass, so the PM source apportionment has certainly larger errors than in autumn. This needs to be discussed, showing residuals of PM and performing error estimate tools, such as BS, DISP and BS-DISP which are implemented in EPA PMFv5.

Reply: Many thanks for the excellent comment. The scaled residuals of PM$_{2.5}$ are provided in Figure S3 in the Supplement. It was found that the residuals were normally distributed between -3 and 3. However, the percentage of samples with residuals between -3 and 3 was comparable between summer (97.5%) and autumn (98.1%). In addition, bootstrap method was used to estimate the errors. According to the standard deviations estimated by bootstrap, we calculated 95% confidence intervals, which are shown as error bars in Figures 10 and 11. The error was even smaller in summer (0.6 μg/m$^3$; 0.7%) than that in autumn (2.6 μg/m$^3$; 3.2%).

The agreement between the predicted and observed PM$_{2.5}$ was also used to evaluate the simulation, as presented in Figure S2. The agreement was indeed lower in summer (slope=0.91; R$^2=0.86$) than that in autumn (slope=0.95; R$^2=0.98$). Relevant discussion was provided as follows.

(3) Good agreement was shown between the predicted and observed PM$_{2.5}$. The slope and correlation coefficient (R$^2$) for the linear regression were 0.91 and 0.86 in summer and 0.95 and 0.98 in autumn, respectively, as shown in Figure S2 in the Supplement. The lower R$^2$ value seen during the summer might be due to the lack of WSI data. (4) The residuals were normally distributed between −3 and 3. Table S2 summarizes the percentage of samples with residuals between −3 and 3 for each species; the lowest percentages were 92.9% and 96.0% for Ni in summer and autumn, respectively. The scaled residuals for PM$_{2.5}$ are shown in Figure S3 in the
Supplement. The percentage of residuals between −3 and 3 was comparable between summer (97.5%) and autumn (98.1%).

For details, please refer to lines 192-200, Table S2, Figure S2 and Figure S3 in the Supplement.

A bootstrap method was used to estimate the model errors, according to which 95% confidence intervals (CIs) were calculated. The 95% CI for PM$_{2.5}$ was 0.6 μg/m$^3$ (0.7% of predicted PM$_{2.5}$) in summer and 2.6 μg/m$^3$ (3.2% of predicted PM$_{2.5}$) in autumn.

For details, please refer to lines 205-207.

23. By the way which software have been used? What uncertainty of the data have been used as input? Was the Q-values the only criterion used for number of factor selection? What about distribution of residuals, G space plot, factor profiles in g/g (which are missing and need to be shown)....?

Reply: Thanks a lot for the questions about the PMF model. In this study, EPA PMF v5.0 was used to conduct the source apportionment analysis. The uncertainties were set as follows.

The uncertainties were $\sqrt{(10\% \times \text{concentration})^2 + DL^2}$ and $5/6 \times DL$ for the samples with concentrations higher and lower than DL, respectively.

For details, please refer to line 175 and lines 185-186.

The criteria used for the selection of factor number and the best solution were provided in the answer to comment 21, which involved Q-value, the ratio between $Q_{\text{robust}}$ and $Q_{\text{true}}$, agreement between the predicted and observed values, residuals and G-space plot. The factor profiles are provided in Figures 10 and 11 in the revised manuscript.

For details, please refer to lines 189-204 and Figures 10 and 11.

24. Traffic is missing among the sources. This is hard to believe for a mega city. There must be a mix of sources, so that solutions with 5, 6...factors should be explored. Residuals of OC, EC, Cu should be provided.

Reply: Thank you very much for the great comments and suggestions. In the revised manuscript, 5 and 6 sources including traffic emissions were identified in summer and autumn, respectively.
Table S2 in the supplement lists the percentage of samples with residuals between -3 and 3. Also, the residuals of OC, EC and Cu were provided here for the reviewer’s reference.

For details, please refer to section 3.3.3 “Source apportionment”.

Residuals of OC, EC and Cu in summer:
Residuals of OC, EC and Cu in autumn:

25. Back trajectories are presented, but please specify which day did you select for each case? Cases span overs several days... and the selection should be supported by some discussion.
Reply: Thanks for the suggestion. Yes, the backward trajectories were simulated during the entire period of each episode. The reason for this selection was also provided as follows.

Because the concentrations, compositions and source contributions of PM$_{2.5}$ were averaged over the entire period of each episode, the wildfire distribution and backward trajectories were also averaged for the entire period of each episode.

For details, please refer to lines 434-437.

26. Row 546 "between" should be replaced by "with".

Reply: Thanks for the correction. Replaced as suggested.

For details, please refer to line 519.

27. The conclusion in row 608, should be revisited. The lack of SIA data in summer does not allow to draw comprehensive knowledge of PM$_{2.5}$.

Reply: Thanks for the suggestion. This sentence was revised as follows.

This study advances our understanding of the chemical characteristics of PM$_{2.5}$ in warm seasons in Wuhan and for the first time quantifies the contribution of biomass burning to PM$_{2.5}$.

For details, please refer to lines 581-582.

References:


Anonymous Referee #4

The paper on chemical composition and source analysis by Lyu et al. presents interesting results for Wuhan, China and would be a good addition to the literature on the topic. The paper discusses chemical composition of PM2.5, possible sources and formation mechanisms for SIA and SOA. However, I have a few questions and suggestions for the authors before the paper is accepted for publication. Some of the sections of the paper present conflicting information, and the introduction section needs to be thoroughly edited (e.g. Lines 39-40, 46, 51-52, 60-62).

Reply: The great comments and suggestions are highly appreciated, which helped us to improve the manuscript substantially. The introduction section was revised according to the reviewer’s comments. In addition, the grammatical errors and typos in the manuscript were corrected by an editing company before the revised manuscript was submitted. Detailed responses to the specific comments were provided point by point below.

Airborne particulate pollution, also called “haze,” has swept across China in recent years, particularly over its northern, central, and eastern parts (Cheng et al., 2014; Kang et al., 2013; Wang et al., 2013).

Numerous studies have been conducted in China to understand the spatiotemporal variations in particle concentrations, the chemical compositions, and the causes of haze events (Cheng et al., 2014; Cao et al., 2012; Zheng et al., 2005; Yao et al., 2002).

For details, please refer to lines 37-39 and lines 48-50.

Specific Comments:

1. Lines 204-205: Please clarify “. . . related to variations in number of construction sites. “ Is there a considerable variation in the number of active construction sites? It is surprising that the contribution from carbonaceous aerosols to total PM is very low. It is somewhat counterintuitive that OC did not increase during the pollution episodes (Lines 328-338), since the increase is being attributed to biomass burning. In autumn, however, an increase in OC is associated with biomass burning. This is not coherent.

Reply: Thank you very much for the excellent comments. Since it is difficult to get the exact number of construction sites, we explained the variations of PM10 as follows.
From summer to autumn, PM$_{10}$ levels declined considerably from $135.1 \pm 4.4$ to $118.9 \pm 3.7$ μg/m$^3$, whereas PM$_{2.5}$ remained statistically stable ($p > 0.05$). The higher summer PM$_{10}$ concentration was probably related to a higher load of fugitive dust. In Wuhan, the temperature ($25.6^\circ C \pm 0.2^\circ C$) in summer was considerably higher than that ($17.5^\circ C \pm 0.3^\circ C$) in autumn ($p < 0.05$), which led to lower water content in the soil and a higher tendency of dust suspension. In addition, the average wind speed in summer ($1.2 \pm 0.04$ vs. $0.8 \pm 0.03$ m/s in autumn) was also higher ($p < 0.05$), which could also have favored the generation of fugitive dust.

For details, please refer to lines 225-231.

In this study, carbonaceous aerosols constituted $23.2\pm0.5\%$ of PM$_{2.5}$, which was comparable to those ($17.7\%$ at an urban site and $17.8\%$ at a suburban site) reported by Zhang et al. (2015) in Wuhan.

For the variations of OC, we are sorry for the confusion about absolute concentration and percentage. In this study, the absolute concentrations of OC and EC greatly increased in all the episodes. However, the percentage (or fraction) of OC in PM$_{2.5}$ remained stable in episode 1 and episode 3 compared to non-episode 1, while the percentage of OC increased in all episodes in autumn compared to non-episode 2. This discrepancy might be due to the fact that high temperature suppressed the gas-to-particle partitioning of semi-volatile organics in summer. To avoid confusion, the concentrations and percentages of main PM$_{2.5}$ components including OC were provided in Table 3 in the revised manuscript. And the explanation for the stable percentage of OC in summer episodes was provided as follows.

In contrast, the fraction of OC in PM$_{2.5}$ remained stable on both episode and non-episode days ($p > 0.05$), possibly because the high temperatures in summer hindered the gas-to-particle partitioning of semivolatile organics (Takekawa et al., 2003).

For details, please refer to lines 386-389 of section 3.3.2 “Chemical signatures” and Table 3.

2. Lines 268-273: This estimation can help interpret if the ions are neutralized. Consider expanding this section to include a discussion on whether or not the aerosols were neutralized during autumn season.

Reply: Accepted with thanks. The discussion was supplemented as follows.
When extending NH$_4^+$ to total cations (NH$_4^+$, Ca$^{2+}$, Mg$^{2+}$, Na$^+$, and K$^+$) and NO$_3^-$ and SO$_4^{2-}$ to total anions (NO$_3^-$, SO$_4^{2-}$, and Cl$^-$), the molar charges of the cations and anions were balanced (slope, 0.98; $R^2 = 0.98$), as shown in Figure S6 in the Supplement, indicating that PM$_{2.5}$ was neutralized during autumn in Wuhan.

For details, please refer to lines 288-291 and Figure S6.

3. Were the correlations between different metals analysed? Some of the K can also come from sources other than biomass burning (especially crustal material), and it would be useful to understand if that is the case. This is particularly relevant with reference to the earlier comment about construction sites. Also, was K found to be correlated with OC/EC?

Reply: Excellent comments. The correlation analyses were supplemented in the revised manuscript as follows.

Correlation analysis indicated that Fe had good correlation with Ca ($R^2 = 0.66$; Figure S7 in the Supplement), whereas weak correlations of K with Fe ($R^2 = 0.14$) and Ca ($R^2 = 0.09$) were found, suggesting that Fe and Ca shared common sources that were different from the sources of K. Because Fe and Ca are typical crustal elements, fugitive dust (e.g., dust from traffic, construction and demolition works, yards, and bare soil) was their most likely source. In contrast, apart from emissions from mineral sources, K is also emitted from biomass burning. As such, K was believed to be mainly emitted from biomass burning in this study, which is further supported by the moderate correlations of K with OC ($R^2 = 0.52$) and EC ($R^2 = 0.48$) because biomass burning also emits OC and EC (Saarikoski et al., 2007; Echalar et al., 1995).

For details, please refer to lines 338-347.

4. PMF: The authors do not provide any details about the outputs, and the process used for determining the factor number, or the stability of the factors. How was the uncertainty estimated? In the current analysis, the fourth factor (biomass burning) seems to include SIA, SOA and the typical biomass burning tracers. Is it possible to tease out secondary aerosol factors if 5-7 factor solutions are used? What was the total sample size used for PMF analysis? Is traffic not a source for PM in the sampling region?
Reply: Many thanks for the great comments. The same comments are also given by another reviewer. In the revised manuscript, new source apportionment simulations were performed, which separated the source of SIA and traffic. Details about the PMF outputs were also provided. Source profiles are shown in Figures 10 and 11. Source contributions are listed in Table 4. Besides, section 2.3 provided other details about the PMF inputs and outputs, i.e., estimates of uncertainties, sample sizes, Q values, agreement between the predicted and observed values, the residuals and G-space plots. The model errors were estimated with the bootstrap method in PMF.

For details, please refer to lines 183-204, Figures 10 and 11, Table 4 and section 3.3.3 “Source apportionment”.

Technical comments.

5. Please proofread the manuscript carefully, and edit for language. Line 24: Contribution for PM10 or PM2.5?

Reply: Thanks for the suggestion. To ensure the quality of the manuscript, the grammatical errors and typos in this manuscript were corrected by an editing company before the revised manuscript was submitted.

The contribution of $47.0 \pm 2.2\%$ was related to the total metal elements in PM$_{2.5}$. Since the abstract was substantially revised, the wording did not exist in the revised manuscript.

6. Methods: Were these hourly measurements? Which method was used for OC/EC analysis? Please include 1-2 lines about the custom element analyser.

Reply: Thanks for the questions and suggestion. Yes, PM and the chemical compositions were all hourly measurements. NIOSH thermal-optical transmission (TOT) method was used for OC/EC analysis. More detailed introduction about the customized metal analyzer was provided.

Hourly data were obtained for each species.

… and an aerosol OC/EC online analyzer (Sunset-RT-4, USA); the NIOSH thermal-optical transmission method was used to resolve the carbonaceous aerosols (OC and EC).
This instrument used a PM$_{2.5}$ impactor to collect the airborne particulate samples, which were analyzed by the β-ray in terms of mass concentrations. The filters loaded with particles were then sent to an x-ray fluorescence analysis system for quantitative analysis.

For details, please refer to lines 118-119, lines 130-132 and lines 133-135.

7. Table 1: For some cities (e.g. Beijing), is the reported value the arithmetic mean, or some other statistic? Please clarify. Also, change Tai Wan to Taiwan.

Reply: Thanks for the question and suggestion. Yes, the values cited from Liu et al. (2014) (Beijing) and Deng et al. (2015) (Guangzhou) were arithmetic means, which were specified in the revised manuscript. Tai Wan was revised to Taiwan.

For details, please refer to Table 1 and line 235.


Reply: Accepted with thanks. The sentence was rephrased as “In general, secondary species and mineral or sea salt components are prone to be apportioned in fine and coarse particles (Zhang et al., 2013; Theodosi et al., 2011)”.

For details, please refer to lines 57-58.

9. Lines 194-195: Either specify when these statistical parameters are reported in Table 1, or remove this sentence.

Reply: Thanks for the suggestion. The statistical values in Table 1 were specified.

For details, please refer to Table 1 and line 235.

10. Lines 202-207: How strong was the correlation between the two fractions of PM? Was there a difference in this relationship during the episode compared to non-episode days?

Reply: Thanks for the great questions. Overall, PM$_{2.5}$ had fair correlation with PM$_{10}$ ($R^2=0.59$). The correlation was better in autumn ($R^2=0.70$) than that in summer ($R^2=0.54$). In addition, their correlation (PM$_{2.5}$ vs. PM$_{10}$) was comparable between episodes ($R^2=0.44$) and non-episodes ($R^2=0.42$). However, all these correlations cannot explain why PM$_{10}$ decreased in autumn while PM$_{2.5}$ remained stable. Combining your first specific comment, we revised this paragraph as follows.
From summer to autumn, PM$_{10}$ levels declined considerably from 135.1 ± 4.4 to 118.9 ± 3.7 μg/m$^3$, whereas PM$_{2.5}$ remained statistically stable ($p > 0.05$). The higher summer PM$_{10}$ concentration was probably related to a higher load of fugitive dust. In Wuhan, the temperature (25.6°C ± 0.2°C) in summer was considerably higher than that (17.5°C ± 0.3°C) in autumn ($p < 0.05$), which led to lower water content in the soil and a higher tendency of dust suspension. In addition, the average wind speed in summer (1.2 ± 0.04 vs. 0.8 ± 0.03 m/s in autumn) was also higher ($p < 0.05$), which could also have favored the generation of fugitive dust.

For details, please refer to lines 225-231.

11. Lines 223-224: What does this sentence mean- “. . . Due to the fact that the chemical, optical and toxic properties tend to be more apparent in smaller particles..”

Reply: Sorry for the confusion. This sentence was rephrased as follows.

Because smaller particles tend to pose more harm to human health and to the atmosphere due to their larger specific surface areas (Yang et al., 2012; Goldberg et al., 2001), and because the chemical compositions in PM$_{10}$ were not analyzed, this study focused mainly on PM$_{2.5}$.

For details, please refer to lines 243-245.

12. Lines 233-234: PM2.5 isn’t entirely composed of secondary particles. Please edit the statement.

Reply: Thanks for the excellent comment. The statement “PM$_{2.5}$/PM$_{10}$ was a measure of the proportion of secondary species in particles.” was deleted. Instead, the discussion was revised as follows.

The PM$_{2.5}$/PM$_{10}$ value also increased remarkably on episode days compared to that on non-episode days, except for episode 2 (45.9% ± 2.5%), which suggests that more secondary species and/or primary fine particles (e.g., primary OC and EC generated from combustion) were generated or released during the episodes. In contrast, the lower PM$_{2.5}$/PM$_{10}$ value during episode 2 might imply a strong source of coarse particles. Indeed, this inference was confirmed by the source apportionment analysis in section 3.3.3.

For details, please refer to lines 253-258.
13. Line 295: p-value is missing.

**Reply:** Sorry for the carelessness. It was added. For details, please refer to line 313.

14. Secondary inorganic aerosol is typically referred to as SIA in the literature. Consider using the same nomenclature.

**Reply:** Accepted with thanks. Revisions were made throughout the manuscript.

14. Section 3.4.1: What is the correlation coefficient for the observed/modelled ozone?

**Reply:** Thanks for the question. The slope and R² of linear regression between the observed and modelled ozone was 0.87 and 0.60, respectively. Since index of agreement (IOA) is often used to evaluate the model performance, we used IOA in the manuscript.

15. Figure 11: For reporting PMF profiles, it would be easier if the authors split the profiles by episode-non-episode. The current plots are difficult to interpret since there is a lot of information on the same plot.

**Reply:** Thanks for the suggestion. Since the data collected in episodes and non-episodes were combined to conduct source apportionment analysis in the revised manuscript, the average profiles in summer and autumn were given separately (Figures 10 and 11), which looked better.

For details, please refer to Figures 10 and 11.

References:

Chemical characteristics and causes of airborne particulate pollution in warm seasons in Wuhan, central China

X.P. Lyu, N. Chen, H. Guo, L.W. Zeng, F. Shen, J.H. Quan, N. Wang

1 Department of Civil and Environmental Engineering, The Hong Kong Polytechnic University, Hong Kong
2 Hubei Provincial Environment Monitoring Center, Wuhan, China
3 Department of Environmental Sciences, School of Resource and Environmental Sciences, Wuhan University, Wuhan, China
4 Guangdong Provincial Key Laboratory of Regional Numerical Weather Prediction, Institute of Tropical and Marine Meteorology, Guangzhou, China

*Corresponding author. Tel: +852 3400 3962; Fax: +852 2334 6389; Email: ceguohai@polyu.edu.hk

Abstract: Continuous measurements of airborne particles and their chemical compositions were conducted in May, June, October, and November 2014 at an urban site in Wuhan, central China. The results indicate that particle concentrations remained at a relatively high level in Wuhan, with averages of 135.1 ± 4.4 (mean ± 95% confidence interval) and 118.9 ± 3.7 μg/m³ for PM₁₀ and 81.2 ± 2.6 and 85.3 ± 2.6 μg/m³ for PM₂.₅ in summer and autumn, respectively. Moreover, PM₂.₅ levels frequently exceeded the National Standard Level II (i.e., daily average of 75 μg/m³), and six PM₂.₅ episodes (i.e., daily PM₂.₅ averages above 75 μg/m³ for 3 or more consecutive days) were captured during the sampling campaign. Potassium was the most abundant element in PM₂.₅, with an average concentration of 2060.7 ± 82.3 ng/m³; this finding indicates intensive biomass burning in and around Wuhan during the study period, because almost no correlation was found between potassium and mineral elements (iron and calcium). The source apportionment results confirm that biomass burning was the main cause of episodes 1, 3, and 4, with contributions to PM₂.₅ of 46.6% ± 3.0%, 50.8% ± 1.2%, and 44.8% ± 2.6%, respectively, whereas fugitive dust was the leading factor in episode 2. Episodes 5 and 6 resulted mainly from increases in vehicular emissions and secondary inorganic aerosols, and the mass and proportion of NO₃⁻ both peaked during episode 6. The high levels of NOₓ and NH₃ and the low temperature during episode 6 were responsible for the increase of NO₃⁻. Moreover, the formation of secondary organic carbon
was found to be dominated by aromatics and isoprene in autumn, and the contribution of aromatics to secondary organic carbon increased during the episodes.

**Keywords:** PM$_{2.5}$; NO$_3^-$; SOA; biomass burning; formation mechanism

1. **Introduction**

Airborne particulate pollution, also called “haze,” has swept across China in recent years, particularly over its northern, central, and eastern parts (Cheng et al., 2014; Kang et al., 2013; Wang et al., 2013). Due to its detrimental effects on human health (Anderson et al., 2012; Goldberg et al., 2001), the atmosphere (Yang et al., 2012; White and Roberts, 1977), acid precipitation (Zhang et al., 2007; Kerminen et al., 2001), and climate change (Ramanathan et al., 2001; Nemesure et al., 1995), particulate pollution has become a major concern of scientific communities and local governments. China’s national ambient air quality standards issued in 2012 regulate the annual upper limit of PM$_{10}$ (i.e., particulate matter with an aerodynamic diameter of less than 10 μm) and PM$_{2.5}$ (i.e., particulate matter with an aerodynamic diameter of less than 2.5 μm) as 70 μg/m$^3$ and 35 μg/m$^3$ and 24-h averages as 150 μg/m$^3$ and 75 μg/m$^3$, respectively (GB 3095-2012).

Numerous studies have been conducted in China to understand the spatiotemporal variations in particle concentrations, the chemical compositions, and the causes of haze events (Cheng et al., 2014; Cao et al., 2012; Zheng et al., 2005; Yao et al., 2002). In general, particulate pollution is more severe in winter due to additional emissions (e.g., coal burning) and unfavorable dispersion conditions (Lyu et al., 2015a; Zheng et al., 2005). Northern China often suffers heavier, longer, and more frequent haze pollution than southern China (Cao et al., 2012). Chemical analysis indicates that secondary inorganic aerosol (SIA; i.e., sulfate [SO$_4^{2-}$], nitrate [NO$_3^-$], and ammonium [NH$_4^+$]) and secondary organic aerosol (SOA) dominate the total mass of airborne particles (Zhang et al., 2014; Zhang et al., 2012). However, the composition differs among the size-segregated particles. In general, secondary species and mineral or sea salt components are prone to be apportioned in fine and coarse particles (Zhang et al., 2013; Theodosi et al., 2011). Indeed, the general characteristics of particles (e.g., toxicity, radiative forcing, acidity) are all tightly associated with their chemical compositions and physical sizes, which therefore have been extensively studied in the field of aerosols. To better understand and control airborne particulate pollution, the causes and formation mechanisms have often been investigated (Wang et al., 2013; Theodosi et al., 2011).
et al., 2014a and b; Kang et al., 2013; Oanh and Leelasakultum, 2011). Apart from the unfavorable meteorological conditions, emission enhancement was often the major culprit. There is little doubt that industrial and vehicular emissions contributed greatly to the particle mass via direct emission and secondary formation of particles from gaseous precursors, such as sulfur dioxide (SO$_2$), nitrogen oxides (NO$_x$), and volatile organic compounds (VOCs; Guo et al., 2011a). In addition, some other sources in specific regions or during specific time periods have also built up the particle concentrations to a remarkable degree, e.g., coal combustion in north China (Cao et al., 2005; Zheng et al., 2005) and biomass burning in Southeast Asia (Deng et al., 2008; Koe et al., 2001). Furthermore, some studies have explored the possible formation mechanisms of the main particle components (SIA and SOA) and distinguished the contributions of different formation pathways. For example, Wang et al. (2014) demonstrated that heterogeneous oxidation of SO$_2$ on aerosol surfaces was an important supplementary pathway to particle-bound SO$_4^{2-}$ in addition to gas phase oxidation and reactions in clouds. In contrast, it was reported that homogeneous and heterogeneous reactions dominated the formation of NO$_3^-$ during the day and night, respectively (Pathak et al., 2011; Lin et al., 2007; Seinfeld and Pandis, 1998). Furthermore, biogenic VOCs and aromatics were shown to be the main precursors of SOA (Kanakidou et al., 2005; Forstner et al., 1997).

Despite numerous studies, the full components of airborne particles have seldom been reported due to the cost of sampling and chemical analysis, resulting in a gap in our understanding of the chemical characteristics of particles. In addition, although the causes of particle episodes have been discussed in many case studies (Wang et al., 2014; Deng et al., 2008), the contributions have rarely been quantified. Furthermore, the formation mechanisms might differ in various circumstances. Therefore, an overall understanding of the chemical characteristics of airborne particles, the causes of the particle episodes, and the formation mechanisms of the enhanced species would be of great value. In addition, the frequent occurrence of haze pollution has become a regular phenomenon in central China during warm seasons, but the causes have not been identified and the contributions have not been quantified. Wuhan is the largest megacity in central China and has suffered from severe particulate pollution in recent years. The data indicate that the frequency of days in which PM$_{2.5}$ exceeded the national standard level II (i.e., a daily average of 75 μg/m$^3$) in Wuhan reached 55.1% in 2014 (Wuhan Environmental Bulletin, 2014). In the warm seasons of 2014, the hourly maximum PM$_{2.5}$ (564 μg/m$^3$) was even higher than that
in winter (383 μg/m$^3$), as shown in Figure S1 in the Supplementary Material. Moreover, because the air quality in Wuhan is strongly influenced by the surrounding cities, the pollution level in Wuhan also reflects the status of the city clusters in central China. However, previous studies (Lyu et al., 2015a; Cheng et al., 2014) did not allow a complete understanding of the properties of airborne particles in this region, particularly during the warm seasons, nor could they guide control strategies. It is therefore urgent to understand the chemical characteristics of airborne particles and to explore the causes and formation mechanisms of the particle episodes in Wuhan. This study deeply analyzed the chemical characteristics of PM$_{2.5}$ in Wuhan from a full suite of component measurement data: SO$_4^{2-}$, NO$_3^-$, NH$_4^+$, organic carbon (OC), including primary organic carbon (POC) and secondary organic carbon (SOC), elemental carbon (EC), and metal elements. Furthermore, based on the analysis of meteorological conditions, chemical signatures, source apportionment, and distribution of wildfires, the causes of the PM$_{2.5}$ episodes are identified and their contributions quantified. Finally, this study used a photochemical box model incorporating a master chemical mechanism (PBM-MCM) and theoretical calculation to investigate the formation processes of NO$_3^-$ and SOC. Ours is the first study to quantify the contribution of biomass burning to PM$_{2.5}$ and examine the formation mechanisms of both inorganic and organic components in PM$_{2.5}$ in central China.

2. Methods
2.1 Data collection
The whole set of air pollutants were continuously monitored at an urban site in the largest megacity of central China, i.e., Wuhan. The measurement covered two periods: May and June in summer and October and November in autumn of 2014. The measured species included particle-phase pollutants such as PM$_{10}$, PM$_{2.5}$, and particle-bound components and gas-phase pollutants, including VOCs, SO$_2$, CO, NO, NO$_2$, O$_3$, HNO$_3$ (g), NH$_3$ (g), and HCl (g). Hourly data were obtained for each species. The sampling site (30.54°N, 114.37°E) was located in the Hubei Environmental Monitoring Center Station, as shown in Figure 1, located in a mixed commercial and residential area in which industries are seldom permitted. The instruments were housed in a room in a six-story building (~18 m above ground level) adjacent to a main road at a straight-line distance of ~15 m. The traffic volume of the road was around 200 vehicles per hour. However, a wall (~2 m high) and several rows of trees (7 to 8 m high) were located between the road and the
sampling site. PM$_{10}$ and PM$_{2.5}$ were measured with a continuous ambient particulate monitor (Thermo Fisher-1405D, USA) integrated with a filter dynamics measurement system to minimize the loss of semivolatile particulate matter. The water-soluble ions (WSIs) in PM$_{2.5}$ and gases including HNO$_3$, HCl, and NH$_3$ were detected with an online ion chromatography monitor (Metrohm-MARGA 1S, Switzerland) and an aerosol OC/EC online analyzer (Sunset-RT-4, USA); the NIOSH thermal-optical transmission method was used to resolve the carbonaceous aerosols (OC and EC). In addition, the elements in PM$_{2.5}$ were measured with a customized metal analyzer. This instrument used a PM$_{2.5}$ impactor to collect the airborne particulate samples, which were analyzed by the β-ray in terms of mass concentrations. The filters loaded with particles were then sent to an x-ray fluorescence analysis system for quantitative analysis. For the analysis of trace gases (SO$_2$, CO, NO, NO$_2$, and O$_3$), we used a suite of commercial analyzers developed by Thermo Environmental Instruments Inc., which have been described in detail (Lyu et al. 2016; Geng et al., 2009). Furthermore, a gas chromatography–flame ionization detector–mass spectrometry system (TH_PKU-300) was used to resolve the real time data of the ambient VOCs. The details of the analysis techniques, resolution, detection limits, and the protocol of quality assurance/control were provided by Lyu et al. (2016) and Wang et al. (2014).

Figure 1. Geographic location of the sampling site. White blocks in left panel represent total distribution of wildfires in autumn 2014, and urban area in Wuhan is highlighted in gray in the right panel.
2.2 Theoretical calculation and model simulation

Theoretical calculation and model simulation were applied in this study to examine the formation mechanisms of NO$_3^-$ and SOC. The particle-bound NO$_3^-$ was generally combined with NH$_3$ or presented as HNO$_3$ in the ammonia-deficient environment, following the processes described in R1 through R3 after HNO$_3$ was formed by the oxidation of NO$_x$ (Pathak et al., 2011, Lin et al., 2010). The production of NO$_3^-$ can be calculated with Equations 1-4.

\[
\begin{align*}
\text{NH}_3(g) + \text{HNO}_3(g) &\leftrightarrow \text{NH}_4\text{NO}_3(s) \quad k_1 = \exp \left[118.87 - \frac{24084}{T} - 6.025\ln(T)\right] \text{(ppb)}^2 \quad \text{(R1)} \\
\text{NH}_3(g) + \text{HNO}_3(g) &\leftrightarrow \text{NH}_4^+ + \text{NO}_3^- \quad k_2 = (P_1 - P_2(1 - a_w) + P_3(1 - a_w)^2) \times (1 - a_w) (1 - a_w)^{1.75} \quad \text{(R2)} \\
\text{N}_2\text{O}_5 + \text{H}_2\text{O} &\rightarrow 2\text{HNO}_3 \quad k_3 = \frac{\gamma}{4(8kT/\pi m_{\text{N}_2\text{O}_5})^{0.5}} \quad \text{Ap} \text{(s)}^{-1} \quad \text{(R3)} \\
\ln(P_1) &=-135.94 + 8763/T + 19.12 \ln(T) \quad \text{(Eq.1)} \\
\ln(P_2) &=-122.65 + 9969/T + 16.22 \ln(T) \quad \text{(Eq.2)} \\
\ln(P_3) &=-182.61 + 13875/T + 24.46 \ln(T) \quad \text{(Eq.3)} \\
\left[\text{NO}_3^-\right] &\approx 0.775 \left(\frac{[\text{NH}_3] + [\text{HNO}_3] - \sqrt{([\text{NH}_3] + [\text{HNO}_3])^2 - 4([\text{NH}_3][\text{HNO}_3] - k_2(k_3))}}{2} \right) \quad \text{(Eq.4)}
\end{align*}
\]

where R1 and R2 describe the homogeneous formation of NO$_3^-$ in humidity conditions lower and higher than the deliquescence relative humidity of NH$_4$NO$_3$ (i.e., 62%; Tang and Munkelwitz, 1993), respectively. R3 presents the heterogeneous reaction of N$_2$O$_5$ on the preexisting aerosol surfaces. $k_{1,3}$ represents the rate of reactions R1-3. $T$, $a_w$, and $P$ are the temperature, the relative humidity, and the temperature-related coefficient, respectively. In R3, $\gamma$ is the reaction probability of N$_2$O$_5$ on aerosol surfaces, assigned as 0.05 and 0.035 on the surface of sulfate ammonia and element carbon, respectively (Aumont et al., 1999; Hu and Abbatt, 1997). $k$ is the Boltzmann constant ($1.38 \times 10^{-23}$), $m_{\text{N}_2\text{O}_5}$ is the molecular mass of N$_2$O$_5$ ($1.79 \times 10^{-22}$ g), and Ap is the aerosol specific surface area ($\text{cm}^2/\text{cm}^3$).

Furthermore, the PBM-MCM model was used to simulate the oxidation products in this study, i.e., O$_3$, N$_2$O$_5$, the semi-volatile oxidation products of VOCs (SVOCs), and radicals such as OH, HO$_2$, and RO$_2$. With full consideration of photochemical mechanisms and real meteorological conditions, the model has been successfully applied in the study of photochemistry. Details about the model construction and application were published by Lyu et al. (2015b), Ling et al. (2014), and Lam et al. (2013).

2.3 Source apportionment model
The positive matrix factorization (PMF) model (EPA PMF v5.0) was used to resolve the sources of PM$_{2.5}$. As a receptor model, PMF has been extensively used in the source apportionment of airborne particles and VOCs (Brown et al., 2007; Lee et al., 1999). Detailed introductions of the model can be found in Paatero (1997) and Paatero and Tapper (1994). Briefly, it decomposes the input matrix ($X$) into matrices of factor contribution ($G$) and factor profile ($F$) in $p$ sources, as shown in Equation 5. The hourly concentrations of PM$_{2.5}$ components were included in the input matrix. Values below the detection limit (DL; see Table S1 in the Supplement) were replaced with DL/2. The uncertainties were $\sqrt{(10\% \times \text{concentration})^2 + DL^2}$ and $5/6\times DL$ for the samples with concentrations higher and lower than DL, respectively. Samples with any missing values were excluded. In total, 807 and 806 samples were applied for source apportionment in summer and autumn, respectively.

The selection of the factor number and the best solution depended upon the following criteria. (1) A lower Q value (Equation 6; a function to evaluate the model runs) was preferable. (2) The ratio between $Q_{\text{robust}}$ and $Q_{\text{true}}$ was lower than 1.5. In this study, the ratios were 0.8 and 0.9 for the summer and autumn data simulation, respectively. (3) Good agreement was shown between the predicted and observed PM$_{2.5}$. The slope and correlation coefficient ($R^2$) for the linear regression were 0.91 and 0.86 in summer and 0.95 and 0.98 in autumn, respectively, as shown in Figure S2 in the Supplement. The lower $R^2$ value seen during the summer might be due to the lack of WSI data. (4) The residuals were normally distributed between $-3$ and $3$. Table S2 summarizes the percentage of samples with residuals between $-3$ and $3$ for each species; the lowest percentages were 92.9% and 96.0% for Ni in summer and autumn, respectively. The scaled residuals for PM$_{2.5}$ are shown in Figure S3 in the Supplement. The percentage of residuals between $-3$ and $3$ was comparable between summer (97.5%) and autumn (98.1%). Finally (5), no correlation was found between the factors, which was achieved by examining the G-space plots and controlled by the FPEAK model runs. Figures S4 and S5 in the Supplement present the G-space plots in summer and autumn, respectively. The low factor contributions and poor correlations indicated that rotational ambiguity was effectively controlled.

A bootstrap method was used to estimate the model errors, according to which 95% confidence intervals (CIs) were calculated. The 95% CI for PM$_{2.5}$ was 0.6 $\mu$g/m$^3$ (0.7% of predicted PM$_{2.5}$) in summer and 2.6 $\mu$g/m$^3$ (3.2% of predicted PM$_{2.5}$) in autumn.

$$x_{ij} = \sum_{k=1}^{p} g_{ik}f_{kj} + e_{ij} \quad \text{(Eq.5)}$$
\[ Q = \sum_{i=1}^{n} \sum_{j=1}^{m} \left[ \frac{x_{ij} - \sum_{k=1}^{p} g_{ik} f_{kj}}{u_{ij}} \right]^2 \]

(Eq.6)

where \( x_{ij} \) and \( u_{ij} \) are the concentration and uncertainty of \( j \) species (total of \( m \)) in \( i \) sample (total of \( n \)), \( g_{ik} \) represents the contribution of \( k_{th} \) source to \( i \) sample, \( f_{kj} \) indicates the fraction of \( j \) species in \( k_{th} \) source, and \( e_{ij} \) is the residual for \( j \) species in the \( i \) sample.

3. Results and discussion

3.1 Concentrations of PM\(_{10}\) and PM\(_{2.5}\)

Table 1 shows the mean concentrations of PM\(_{10}\) and PM\(_{2.5}\) in Wuhan and other Chinese cities and regions. The mean, maximum and minimum values, and standard deviation or 95% CI were provided if available. In general, the concentrations of airborne particles in Wuhan (135.1 ± 4.4 and 81.2 ± 2.6 and 85.3 ± 2.6 μg/m\(^3\) for PM\(_{10}\); 118.9 ± 3.7 μg/m\(^3\) for PM\(_{2.5}\) in summer and autumn, respectively) were lower than those in northern China (i.e., Beijing and Xi’an), comparable to those in eastern China (i.e., Shanghai and Nanjing), and higher than those in southern China (i.e., Guangzhou and Hong Kong) and Taiwan. Bearing in mind that the sampling site, period, method, and instrument all interfere with comparisons, the ambient particulate pollution in Wuhan was severe.

From summer to autumn, PM\(_{10}\) levels declined considerably from 135.1 ± 4.4 to 118.9 ± 3.7 μg/m\(^3\), whereas PM\(_{2.5}\) remained statistically stable (\( p > 0.05 \)). The higher summer PM\(_{10}\) concentration was probably related to a higher load of fugitive dust. In Wuhan, the temperature (25.6°C ± 0.2°C) in summer was considerably higher than that (17.5°C ± 0.3°C) in autumn (\( p < 0.05 \)), which led to lower water content in the soil and a higher tendency of dust suspension. In addition, the average wind speed in summer (1.2 ± 0.04 vs. 0.8 ± 0.03 m/s in autumn) was also higher (\( p < 0.05 \)), which could also have favored the generation of fugitive dust.

<table>
<thead>
<tr>
<th></th>
<th>PM(_{10})</th>
<th>PM(_{2.5})</th>
<th>Sampling period</th>
</tr>
</thead>
<tbody>
<tr>
<td>Wuhan</td>
<td>135.1 ± 4.4 (^1)</td>
<td>81.2 ± 2.6 (^1)</td>
<td>May-Jun. 2014 (this study)</td>
</tr>
<tr>
<td></td>
<td>118.9 ± 3.7 (^1)</td>
<td>85.3 ± 2.6 (^1)</td>
<td>Oct.-Nov. 2014 (this study)</td>
</tr>
<tr>
<td>Beijing</td>
<td>155.9 (^II)</td>
<td>73.8 (^II)</td>
<td>Jun.-Aug. 2009 (^a)</td>
</tr>
</tbody>
</table>

Table 1. Comparisons of PM\(_{10}\) and PM\(_{2.5}\) (in μg/m\(^3\)) between Wuhan and other Chinese cities and regions.
Table 2 summarizes the concentrations of PM$_{10}$ and PM$_{2.5}$ and the percentage of PM$_{2.5}$ in PM$_{10}$, referred to as PM$_{2.5}$/PM$_{10}$, during the summer and autumn episodes and non-episodes. PM$_{10}$ and PM$_{2.5}$ concentrations increased significantly ($p < 0.05$) during the episodes in both summer and autumn. The PM$_{2.5}$/PM$_{10}$ value also increased remarkably on episode days compared to that on non-episode days, except for episode 2 (45.9% ± 2.5%), which suggests that more secondary species and/or primary fine particles (e.g., primary OC and EC generated from combustion) were
generated or released during the episodes. In contrast, the lower PM$_{2.5}$/PM$_{10}$ value during episode 2 might imply a strong source of coarse particles. Indeed, this inference was confirmed by the source apportionment analysis in section 3.3.3.

Figure 2. Daily concentrations of PM$_{10}$ and PM$_{2.5}$ in May, June, October, and November 2014. Episode 1, May 16 to 22; episode 2, May 25 to 30; episode 3, June 5 to 15; episode 4, October 15 to 20; episode 5, October 24 to 28; Episode 6, November 14 to 23.

Table 2. Mean PM$_{10}$, PM$_{2.5}$, and PM$_{2.5}$/PM$_{10}$ with 95% CI during PM$_{2.5}$ episodes and non-episodes in Wuhan. Non-episode 1 and Non-episode 2 represent the non-episode periods in summer and autumn, respectively.

<table>
<thead>
<tr>
<th></th>
<th>PM$_{10}$ (μg/m$^3$)</th>
<th>PM$_{2.5}$ (μg/m$^3$)</th>
<th>PM$<em>{2.5}$/PM$</em>{10}$ (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Episode 1</td>
<td>154.3 ± 10.1</td>
<td>123.0 ± 9.1</td>
<td>72.8 ± 2.6</td>
</tr>
<tr>
<td>Episode 2</td>
<td>230.1 ± 19.1</td>
<td>98.9 ± 5.7</td>
<td>45.9 ± 2.5</td>
</tr>
<tr>
<td>Episode 3</td>
<td>191.4 ± 9.8</td>
<td>126.7 ± 7.0</td>
<td>66.9 ± 1.8</td>
</tr>
<tr>
<td><strong>Non-episode 1</strong></td>
<td><strong>98.5 ± 3.9</strong></td>
<td><strong>56.6 ± 1.7</strong></td>
<td><strong>58.9 ± 1.5</strong></td>
</tr>
<tr>
<td>Episode 4</td>
<td>221.8 ± 8.9</td>
<td>148.6 ± 5.2</td>
<td>67.9 ± 2.0</td>
</tr>
<tr>
<td>Episode 5</td>
<td>154.2 ± 10.4</td>
<td>108.2 ± 6.8</td>
<td>69.3 ± 3.1</td>
</tr>
<tr>
<td>Episode 6</td>
<td>157.3 ± 9.0</td>
<td>120.0 ± 7.6</td>
<td>71.2 ± 2.1</td>
</tr>
<tr>
<td><strong>Non-episode 2</strong></td>
<td><strong>88.7 ± 3.4</strong></td>
<td><strong>64.2 ± 2.2</strong></td>
<td><strong>65.3 ± 1.3</strong></td>
</tr>
</tbody>
</table>
3.2 Chemical composition of PM$_{2.5}$

Figure 3 shows the daily variations of PM$_{2.5}$ and its composition. As the instrument for the analysis of WSIs was initially deployed in September 2014, data are not available for May and June. The carbonaceous aerosol (18.5 ± 1.2 μg/m$^3$) and elements (6.0 ± 0.3 μg/m$^3$) accounted for 19.1% ± 0.6% and 6.2% ± 0.2% of PM$_{2.5}$ in summer, respectively. In autumn, WSIs were the most abundant component in PM$_{2.5}$ (64.4 ± 2.5 μg/m$^3$; 68.6% ± 1.9%), followed by carbonaceous aerosol (24.3 ± 1.0 μg/m$^3$; 25.5% ± 0.8%) and elements (4.5 ± 0.2 μg/m$^3$; 4.6% ± 0.1%). The secondary inorganic ions SO$_4^{2-}$ (18.8 ± 0.6 μg/m$^3$), NO$_3^-$ (18.7 ± 0.8 μg/m$^3$), and NH$_4^+$ (12.0 ± 0.4 μg/m$^3$) dominated the WSIs, with the average contribution of 34.0% ± 0.6%, 30.1% ± 0.5%, and 20.4% ± 0.1%, respectively.

Figure 3. Daily variations of PM$_{2.5}$ and its components. Pie charts represent the composition of elements and water-soluble ions, respectively. Pink shaded areas represent episodes.

The charge balance between the anions and cations was usually used to predict the existing forms of SIAs and the acidity of PM$_{2.5}$. Figure 4 shows the relative abundance of molar charges of SIAs, which were located fairly close to the one-to-one line on both episode and non-episode days. This finding suggests that NH$_4$NO$_3$ and (NH$_4$)$_2$SO$_4$ were coexisting forms of the SIAs in PM$_{2.5}$ in Wuhan. When extending NH$_4^+$ to total cations (NH$_4^+$, Ca$^{2+}$, Mg$^{2+}$, Na$^+$, and K$^+$) and NO$_3^-$ and SO$_4^{2-}$ to total anions (NO$_3^-$, SO$_4^{2-}$, and Cl$^-$), the molar charges of the cations and anions were balanced (slope, 0.98; $R^2 = 0.98$), as shown in Figure S6 in the Supplement, indicating that PM$_{2.5}$
was neutralized during autumn in Wuhan.

Figure 4. Relative abundance of molar charges of PM$_{2.5}$ during autumn in Wuhan.

For the carbonaceous aerosol, OC ($14.8 \pm 0.5$ μg/m$^3$) and EC ($3.6 \pm 0.1$ μg/m$^3$) accounted for 79.9% ± 0.3% and 20.2% ± 0.3% of the total carbon, respectively. In general, SOC was expected to exist when the OC/EC ratio was greater than 2 (Duan et al., 2005; Chow et al., 1996), and the proportion of SOC increased with the increase in OC/EC ratio. The average OC/EC ratio was 4.8 ± 0.1 in Wuhan, which suggests that SOC (i.e., carbon fraction of SOA) was an important component in PM$_{2.5}$. Indeed, as the constituents of OC, SOC and POC can be distinguished with the EC-tracer method, following Equations 7 and 8 (Cabada et al., 2004):

POC = (OC/EC)$_{\text{prim}}$ × EC + OC$_{\text{non-comb}}$  \hspace{1cm} (Eq.7)
SOC = OC – POC  \hspace{1cm} (Eq.8)

where (OC/EC)$_{\text{prim}}$ was the ratio of primary OC to EC, obtained from the pairs of OC and EC with the OC/EC ratios among the 10% lowest; and OC$_{\text{non-comb}}$ was the primary OC that was not related to combustion activities. These values were determined by the slope and intercept of the linear regression between primary OC and EC, respectively (Figure 5). Because the abundance of SOC depended largely upon the oxidative capacity of the atmosphere, the oxidative radical (HO$_2$) was simulated with the PBM-MCM model and compared with the pattern of SOC. More details about the simulation are provided in section 3.4. Figure 6 shows the hourly concentrations of SOC and POC and the average diurnal patterns of SOC, POC, and HO$_2$. In general, the POC levels ($8.6 \pm 0.2$ μg/m$^3$) were slightly higher than those of SOC ($6.4 \pm 0.3$ μg/m$^3$; $p < 0.05$). The
difference was greatest in November, when the concentration was $9.5 \pm 0.4$ and $4.7 \pm 0.3 \, \mu g/m^3$ for POC and SOC, respectively. Because the production of SOC was closely related to the atmospheric oxidative capacity, the lowest fraction of SOC in November might be attributable to the weakest oxidative capacity; for example, the $O_3$ level was lowest in November ($14.3 \pm 1.0$ ppbv). Two peaks were found for the simulated diurnal pattern of $HO_2$, which might be caused by strong solar radiation at noon and in the early afternoon and by reactions among alkenes and $O_3$ and $NO_3$ at night (Emmerson et al., 2005; Kanaya et al., 1999). The diurnal patterns of POC and SOC revealed that POC levels were relatively stable throughout the day. The increase in the POC level in the early morning (06:00 to 08:00) and late afternoon and early evening (16:00 to 20:00) was likely related to increases in vehicular emissions during rush hours, and the decrease from 08:00 to 15:00 might be caused by the extension of the boundary layer. In contrast, the SOC level showed two peaks at around 12:00 and 19:00, which was consistent with the diurnal variation of the simulated $HO_2$, suggesting that the formation of SOC was closely related to the oxidative radicals in the atmosphere. (A detailed relationship is discussed in section 3.4.3.)

![Figure 5](image-url)

**Figure 5.** Regression between OC and EC with the 10% lowest OC/EC ratios during (a) summer and (b) autumn in Wuhan.
Figure 6. Hourly concentrations of SOC and POC. Insert graph presents average diurnal variations of SOC, POC, and HO$_2$.

Among the elements, potassium (K; 2060.7 ± 82.3 ng/m$^3$), iron (Fe; 996.5 ± 34.3 ng/m$^3$), and calcium (Ca; 774.1 ± 39.4 ng/m$^3$) were the most abundant species, accounting for 47.0% ± 2.2%, 21.4% ± 0.3%, and 15.6% ± 0.3% of the total analyzed elements, respectively. Correlation analysis indicated that Fe had good correlation with Ca ($R^2 = 0.66$; Figure S7 in the Supplement), whereas weak correlations of K with Fe ($R^2 = 0.14$) and Ca ($R^2 = 0.09$) were found, suggesting that Fe and Ca shared common sources that were different from the sources of K. Because Fe and Ca are typical crustal elements, fugitive dust (e.g., dust from traffic, construction and demolition works, yards, and bare soil) was their most likely source. In contrast, apart from emissions from mineral sources, K is also emitted from biomass burning. As such, K was believed to be mainly emitted from biomass burning in this study, which is further supported by the moderate correlations of K with OC ($R^2 = 0.52$) and EC ($R^2 = 0.48$) because biomass burning also emits OC and EC (Saarikoski et al., 2007; Echalar et al., 1995).

### 3.3 Causes of PM$_{2.5}$ episodes

#### 3.3.1 Meteorological conditions

The processes of particle formation, dispersion, and deposition are closely related to meteorological conditions. To interpret the possible causes of the PM$_{2.5}$ episodes, Figure 8 shows the patterns of wind direction and speed, temperature, relative humidity, and atmospheric pressure in Wuhan during the monitoring period. In general, southeast winds prevailed at the sampling site with a wind speed of approximately 1.0 m/s. The low wind speed indicates the
dominance of local air masses. However, due to the high stability and long lifetime of PM$_{2.5}$, the regional and superregional impact could not be eliminated. In comparison with those in summer, the wind speed (summer, $1.1 \pm 0.04$ m/s; autumn, $0.8 \pm 0.03$ m/s) and temperature (summer, $25.6 \pm 0.2$ m/s; autumn, $17.5 \pm 0.3$ m/s) were significantly ($p < 0.05$) lower in autumn, whereas the atmospheric pressure (summer, $1006.9 \pm 0.2$ hPa; autumn, $1020.9 \pm 0.2$ hPa) was much higher. During the episodes, the wind speed was generally lower than during non-episodes, with the exception of episode 5. This might be one cause for the episodes, but it does not fully explain the great enhancements of PM$_{2.5}$, because the wind speeds were very low and the differences between the episodes and non-episodes were minor. The atmospheric pressure was not very high during episodes 1 through 5, suggesting that the synoptic system was not responsible for the occurrence of these PM$_{2.5}$ episodes. However, the atmospheric pressure was remarkably higher ($p < 0.05$) in episode 6 ($1024 \pm 1$ hPa) than in non-episode 2 ($1021 \pm 0.3$ hPa), which might have suppressed the diffusion of PM$_{2.5}$ and the gaseous precursors. In addition, the temperature was lower (episode 6, $14.0^\circ$C $\pm 0.4^\circ$C; non-episode 2, $17.2^\circ$C $\pm 0.3^\circ$C; $p < 0.05$), which favors the gas-to-particle partitioning of semivolatile and non–thermal stabilized species. As a consequence, these effects might have elevated the PM$_{2.5}$ concentrations in episode 6, which is discussed further in section 3.4.2.
Table 3 summarizes the mass concentrations and percentages of the main components in PM$_{2.5}$. The mass concentrations of PM$_{2.5}$ components significantly increased from non-episode days to episode days ($p < 0.05$). In contrast, the percentages of the chemical components in PM$_{2.5}$ varied by species. In summer, the fractions of EC and K in PM$_{2.5}$ experienced significant increases from non-episode 1 (EC, 4.8% ± 0.2%; K, 2.0% ± 0.1%) to episode 1 (EC, 5.7% ± 0.5%; K, 4.4% ± 0.3%) and episode 3 (EC, 5.3% ± 0.2%; K, 3.0% ± 0.2%). Because EC is the tracer of incomplete combustion (Chow et al., 1996) and K is the indicator of biomass burning (Saarakoski et al., 2007; Echalar et al., 1995), the higher percentages of EC and K in episodes 1 and 3 imply the outstanding contribution of biomass burning. In contrast, the fraction of OC in PM$_{2.5}$ remained stable on both episode and non-episode days ($p > 0.05$), possibly because the high temperatures in summer hindered the gas-to-particle partitioning of semivolatile organics (Takekawa et al., 2003). Furthermore, the percentages of Ca (2.9% ± 0.4%) and Fe (2.7% ± 0.3%) significantly increased during episode 2 ($p < 0.05$) compared to those in non-episode 1 (Ca, 1.1% ± 0.1%; Fe, 1.5% ± 0.1%), which shows that fugitive dust made a considerable contribution to PM$_{2.5}$ in episode 2. In addition, biomass burning might also have contributed to PM$_{2.5}$, in view of the increase in the percentage of K (non-episode 1, 2.0% ± 0.1%; episode 2, 3.2% ± 0.2%).

In autumn, the percentage of K significantly ($p < 0.05$) increased during episode 4 (3.1% ± 0.1% vs. 2.1% ± 0.1% in non-episode 2), as did that of OC (27.3% ± 0.7% vs. 20.9% ± 0.8% in non-episode 2), suggesting the dominant role of biomass burning in episode 4. Furthermore, the fractions of OC in episode 5 (23.8% ± 1.5%) and NO$_3^-$ in episode 6 (26.1% ± 1.0%) were obviously higher than those in non-episode 2 (OC, 20.9% ± 0.8%; NO$_3^-$, 19.8% ± 0.9%). Due to the complexity of the sources of OC and NO$_3^-$, the causes of episodes 5 and 6 are further explored in the following sections.

In summary, episodes 1, 3, and 4 were greatly affected by biomass burning. This finding was further confirmed by the significant increases in the gaseous tracers of biomass burning such as ethyne (C$_2$H$_2$) and methyl chloride (CH$_3$Cl) (Guo et al., 2011b; Simoneit et al., 2002) during these episodes ($p < 0.05$; see Figure S8 in the Supplement).
Table 3. Concentrations (μg/m³) and percentages (in parentheses) of the main components of PM$_{2.5}$ during non-episodes and episodes. Bold font demonstrates significant increase in percentage of PM$_{2.5}$ components during episodes compared to non-episodes.

<table>
<thead>
<tr>
<th></th>
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<td>Episode 5</td>
<td>Episode 6</td>
<td>Non-episode 2</td>
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<td>OC</td>
<td>18.5±1.3</td>
<td>16.3±3.3</td>
<td>19.8±2.5</td>
<td>7.9±0.3</td>
<td>35.1±1.7</td>
<td>24.9±1.9</td>
<td>22.7±1.5</td>
</tr>
<tr>
<td></td>
<td>(15.1%±0.8%)</td>
<td>(14.0%±1.9%)</td>
<td>(14.5%±1.0%)</td>
<td>(15.4%±0.7%)</td>
<td>(27.3%±0.7%)</td>
<td>(23.8%±1.5%)</td>
<td>(21.4%±0.8%)</td>
</tr>
<tr>
<td>EC</td>
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<td>2.8±0.1</td>
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<td>4.2±0.4</td>
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</tr>
<tr>
<td></td>
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<td>(4.1%±0.5%)</td>
<td>(5.3%±0.2%)</td>
<td>(4.8%±0.2%)</td>
<td>(4.2%±0.3%)</td>
<td>(4.1%±0.4%)</td>
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<td>NA</td>
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<td>25.9±2.4</td>
<td>21.3±2.0</td>
</tr>
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<tr>
<td>NO$_3^-$</td>
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<td>NA</td>
<td>NA</td>
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<td>24.7±2.9</td>
<td><strong>30.8±3.6</strong></td>
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<tr>
<td>NH$_4^+$</td>
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<td>NA</td>
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<td>16.8±1.8</td>
</tr>
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<td></td>
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<td></td>
<td></td>
</tr>
<tr>
<td>K</td>
<td><strong>5.3±0.6</strong></td>
<td><strong>3.4±0.4</strong></td>
<td><strong>3.8±0.5</strong></td>
<td>1.1±0.1</td>
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</tr>
<tr>
<td></td>
<td>(4.4%±0.3%)</td>
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<td>(3.0%±0.2%)</td>
<td>(2.0%±0.1%)</td>
<td>(3.1%±0.1%)</td>
<td>(2.2%±0.1%)</td>
<td>(2.2%±0.1%)</td>
</tr>
<tr>
<td>Ca</td>
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<td>1.1±0.1</td>
<td>0.8±0.2</td>
<td>0.8±0.1</td>
</tr>
<tr>
<td></td>
<td>(1.1%±0.1%)</td>
<td>(2.9%±0.4%)</td>
<td>(0.8%±0.1%)</td>
<td>(1.1%±0.1%)</td>
<td>(0.9%±0.1%)</td>
<td>(0.8%±0.2%)</td>
<td>(0.8%±0.1%)</td>
</tr>
<tr>
<td>Fe</td>
<td>1.4±0.1</td>
<td><strong>2.8±0.3</strong></td>
<td>1.1±0.1</td>
<td>0.8±0.05</td>
<td>1.5±0.1</td>
<td>1.2±0.2</td>
<td>1.2±0.1</td>
</tr>
<tr>
<td></td>
<td>(1.3%±0.1%)</td>
<td>(2.7%±0.3%)</td>
<td>(1.0%±0.1%)</td>
<td>(1.5%±0.1%)</td>
<td>(1.2%±0.2%)</td>
<td>(1.2%±0.2%)</td>
<td>(1.1%±0.1%)</td>
</tr>
</tbody>
</table>
3.3.3 Source apportionment

To clarify the sources of PM$_{2.5}$ and quantify their contributions, the hourly data on PM$_{2.5}$ components were applied to PMF for source apportionment. Five and six sources were resolved for summer and autumn, respectively, as shown in Figure 10 and Figure 11. The source of SIA was missing in summer, due to the lack of WSI data. For other sources, the profile of each corresponding source was similar in summer and autumn. Factor 1 had high loadings of crustal elements (i.e., Ba, Ca, Mn, and Fe), indicating the source of fugitive dust. Factor 2 was likely associated with oil refinery and usage, in view of the high percentages of V and Ni, which often originate from the combustion of heavy oil (Barwise et al., 1990; Nriagu and Pacyna, 1988). Factor 3 was distinguished by the high loadings of OC, EC, and K, indicating the biomass burning source (Zhang et al., 2013; Friedli et al., 2003). Factor 4 seemed to represent vehicle emissions, due to the dominance of Pb, As, Cu, and Se. Cu is the tracer of road traffic because it is widely used as the antioxidant in brake pads (de Fatima Andrade et al., 2012). Although lead-containing gasoline has been forbidden in China since 2000, high levels of Pb are often reported in traffic sources (Yang et al., 2013; Song et al., 2012), which might be due to the use of Pb-containing materials in car components, such as lead wheel weights, solder in electronics, and lead-acid batteries (Song et al., 2012). OC and Hg dominated in factor 5, and Cl$^-$ also showed high loading in this factor in the autumn profile. It is well documented that Hg and Cl$^-$ are largely emitted from coal combustion (Wang et al., 2010; Ye et al., 2003). Hence, this factor was assigned as coal combustion. Finally, a source of SIA with high loadings of NO$_3^-$, SO$_4^{2-}$, and NH$_4^+$ was resolved in autumn.

Figures S9 and S10 in the Supplement show the day-to-day variations of the source contributions in summer and autumn, respectively. Overall, biomass burning was the largest contributor (45.0% ± 0.03%) to PM$_{2.5}$ in summer. However, the contribution of biomass burning in autumn (23.7% ± 0.5%) was lower than that of SIA (38.6% ± 0.7%). Bearing in mind the uncertainties caused by the lack of WSIs, the greater contribution of biomass burning in summer might be associated with the lower WSIs. For example, NO$_3^-$ was expected to be much lower in summer due to its thermal decomposition at high temperatures. Table 4 summarizes the source contributions during episodes and non-episodes. Noticeably, the contributions of biomass burning were significantly higher ($p < 0.05$) in episode 1 (59.2 ± 6.3 μg/m$^3$; 46.6% ± 3.0%), episode 3 (64.9 ± 3.3 μg/m$^3$; 50.8% ± 1.2%), and episode 4 (48.7 ± 2.9 μg/m$^3$; 44.8% ± 2.6%) than those in the corresponding
non-episodes, confirming that biomass burning was the main cause of these PM$_{2.5}$ episodes. In addition, vehicle emissions made a greater contribution to episode 4 (14.9 ± 1.2 μg/m$^3$; 13.7% ± 1.1%) than to non-episode 2. In contrast, the contribution of fugitive dust (6.5 ± 1.3 μg/m$^3$; 5.6% ± 1.0%) in episode 2 was remarkably ($p < 0.05$) higher than in non-episode 1 (1.1 ± 0.1 μg/m$^3$; 1.8% ± 0.2%). This finding was consistent with the inference that episode 2 was attributable to fugitive dust according to the low PM$_{2.5}$/PM$_{10}$ ratio (45.9% ± 2.5%) and high levels of crustal elements (Ca, 2.9% ± 0.4%; Fe, 2.7% ± 0.3%) in this episode. In addition, vehicle emissions and SIA both experienced significant increases in episodes 5 and 6 ($p < 0.05$). In fact, the increase of OC in episode 5 (see Table 3) was mainly caused by vehicle emissions and coal combustion (Table S3 in the Supplement). For episode 6, in addition to the increase in OC, SIA was also an important contributor, particularly for NO$_3^-$, which increased from 9.9 ± 1.2 μg/m$^3$ in non-episode 2 to 21.4 ± 3.3 μg/m$^3$ in episode 6 (Table S3).

Figure 10. Profiles of PM$_{2.5}$ sources in summer. Error bars represent 95% CI estimated by bootstrap method in PMF.
Figure 11. Profiles of PM$_{2.5}$ sources in autumn. Error bars represent 95% CI estimated by bootstrap method in PMF.
Table 4. Mass concentration (μg/m³) and percentage contribution (in parentheses) of sources to sum of resolved species in PM$_{2.5}$. Bold font represents a significant increase in percentage contribution in episodes compared to non-episodes.

<table>
<thead>
<tr>
<th>Source Type</th>
<th>Episode 1</th>
<th>Episode 2</th>
<th>Episode 3</th>
<th>Non-episode 1</th>
<th>Episode 4</th>
<th>Episode 5</th>
<th>Episode 6</th>
<th>Non-episode 2</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>10.0±1.3</td>
<td>29.5±6.1</td>
<td>6.4±0.5</td>
<td>4.8±0.6</td>
<td>3.3±0.3</td>
<td>2.3±0.5</td>
<td>2.6±0.3</td>
<td>0.9±0.1</td>
</tr>
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<td>Fugitive dust</td>
<td>(8.7%±1.2%)</td>
<td>(27.5%±5.5%)</td>
<td>(5.3%±0.4%)</td>
<td>(8.7%±1.0%)</td>
<td>(3.0%±0.2%)</td>
<td>(2.7%±0.5%)</td>
<td>(3.0%±0.3%)</td>
<td>(1.6%±0.2%)</td>
</tr>
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<td>Oil refinery and usage</td>
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<td>Biomass burning</td>
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<td>48.7±2.9</td>
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<td>Vehicle emissions</td>
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<td>Coal Combustion</td>
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<td>20.9±4.7</td>
<td>41.6±3.3</td>
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<td>SIA</td>
<td>—</td>
<td>—</td>
<td>—</td>
<td>—</td>
<td>23.3±3.3</td>
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<td></td>
<td>(9.6%±1.2)%</td>
<td>(11.9%±3.2%)</td>
<td>(7.9%±0.9%)</td>
<td>(16.3%±1.0%)</td>
<td>(11.9%±1.3%)</td>
<td>(17.5%±2.7%)</td>
<td>(16.3%±1.7%)</td>
<td>(18.1%±1.8%)</td>
</tr>
</tbody>
</table>

Note: (parentheses) denote percentage contribution.
3.3.4 Open fires and air mass trajectories

To further confirm the biomass burning activities during the PM$_{2.5}$ episodes, the wildfire distribution (downloaded from NASA Firms Web Fire Mapper, and accessible at https://firms.modaps.eosdis.nasa.gov/firemap/) and 72-h backward air mass trajectories (simulated by Hysplit v4.9 model) are plotted in Figure 12. Because the concentrations, compositions and source contributions of PM$_{2.5}$ were averaged over the entire period of each episode, the wildfire distribution and backward trajectories were also averaged for the entire period of each episode. Consistent with the great contributions to PM$_{2.5}$ of biomass burning, the air masses arriving in Wuhan had passed over the areas where intensive open fires were detected in episodes 1, 3, and 4. In episode 2, wildfires were widespread in northeast China. However, the air mass trajectories were mainly from the south and northwest and evaded the burning areas, which might explain why biomass burning was not a predominant factor in episode 2. The sparse wildfires in episodes 5 and 6 coincided with the source apportionment result that biomass burning did not significantly elevate the concentration of PM$_{2.5}$.

Figure 12. Wildfire distribution and 72-h backward air mass trajectories. Red squares demonstrate potential areas where biomass burning aggravated particulate pollution in Wuhan.
3.4 Formation mechanisms

3.4.1 Model validation

In this study, the PBM-MCM model was used to help investigate the formation mechanisms of NO$_3^-$ and SOC. Before application, the model was validated via O$_3$ simulation. Figure S11 compares the daily averages and diurnal variations of O$_3$ between the simulation and the observations. It was found that the model simulated O$_3$ variation well in both daily and diurnal patterns. However, it generally overestimated the O$_3$ levels in November. The meteorological parameters indicated that the frequency of foggy days was extremely high (36.7%) in November, possibly resulting in a weakening of solar radiation and consequently of photochemical reactivity.

To quantitatively evaluate the performance of the model, the index of agreement (IOA) was calculated using Equation 9.

\[
IOA = 1 - \frac{\sum_{i=1}^{n}(O_i - S_i)^2}{\sum_{i=1}^{n}(O_i - \bar{O})^2 + \sum_{i=1}^{n}(S_i - \bar{S})^2}
\]

(Eq.9)

where $\bar{O}$ is the average of $n$ samples and $O_i$ and $S_i$ represent the observed and simulated values, respectively. Within the interval of [0, 1], higher IOA values indicated better agreement between the simulation and observation.

By calculation, the IOA reached 0.86, indicating excellent performance of the model in the O$_3$ simulation. Because O$_3$ production is closely associated with oxidative radicals, intermediates, and products, the robust O$_3$ simulation gave us full confidence to accept the simulated N$_2$O$_5$, HO$_2$, and SVOCs.

3.4.2 NO$_3^-$ formation

The composition analysis indicated that the proportion of NO$_3^-$ increased remarkably during episode 6. To interpret this phenomenon, the formation mechanisms of NO$_3^-$ were investigated. Figure 14 shows the hourly variations of the calculated and observed NO$_3^-$ and the contribution of R3 (i.e., N$_2$O$_5$ + H$_2$O $\rightarrow$ 2HNO$_3$), among which NO$_3^-$ cal 1, NO$_3^-$ cal 2, and NO$_3^-$ obs referred to homogeneous formation (R1 and R2), total formation (R1, R2, and R3), and field measurement of NO$_3^-$, respectively. Although the particle-bound NO$_3^-$ was influenced by many factors (i.e., formation, deposition, and dispersion), the calculations generally well reproduced the measured NO$_3^-$ in episode 6, with a high correlation coefficient ($R^2 = 0.63$) and an IOA of 0.78. However, on November 23, 2014, the observed NO$_3^-$ decreased rapidly from 09:00, which was not captured by the calculations. This discrepancy was likely caused by the weather conditions on...
that day, because (1) the average wind speed increased from 1.7 m/s before 09:00 to 2.7 m/s after
09:00 and even reached 4.0 m/s at 14:00; and (2) moderate rain began at 12:00 and continued
until 23:00, with total precipitation of 24 mm. Indeed, this was the beginning of a 7-day rainy
period, which ended episode 6 with a sharp decrease of the PM$_{2.5}$ concentration (approximately
175 μg/m$^3$; see Figure 2).

Because the values of NO$_3$$_{\text{cal } 1}$ were very close to those of NO$_3$$_{\text{cal } 2}$, the variation of NO$_3$ in
episode 6 could be well explained by the homogeneous formation (R1 and R2), whereas the
heterogeneous reaction of N$_2$O$_5$ on aerosol surfaces (R3) made only a minor contribution to the
total NO$_3$ (i.e., nearly nil from 0:00 to 17:00 and 3.7% ± 0.6% from 18:00 to 23:00). Because
the homogeneous formation of NO$_3$ was closely related to the concentrations of HNO$_3$ (g) and
NH$_3$ (g) and the temperature (see R1 and R2), Table 4 compares the temperature, HNO$_3$ (g), NH$_3$
(g), NO, NO$_2$, O$_3$, and the simulated OH and HO$_2$ (a measure of oxidative capacity [Cheng et al.,
2010]) between episode 6 and non-episode 2. It was found that the levels of HNO$_3$ (g) (0.65 ±
0.01 ppbv) and NH$_3$ (g) (13.48 ± 0.72 ppbv) in episode 6 were significantly higher than those
during the non-episode 2 (0.47 ± 0.03 and 9.54 ± 0.37 ppbv for HNO$_3$ and NH$_3$, respectively),
which might substantially favor the formation of NH$_4$NO$_3$. Because HNO$_3$ (g) is generally
formed by oxidation of NO$_x$, the production of HNO$_3$ (g) should be closely related to the
oxidative capacity of the air and the level of NO$_x$. In episode 6, the levels of O$_3$ (17.09 ± 2.04
ppbv), OH ([3.8 ± 1.3] × 10$^5$ molecules/cm$^3$), and HO$_2$ ([1.1 ± 0.3] × 10$^7$ molecules/cm$^3$) were
noticeably lower than those in non-episode 2 (O$_3$, 24.57 ± 1.64 ppbv; OH, [7.2 ± 0.9] × 10$^5$
molecules/cm$^3$; HO$_2$, [2.0 ± 0.2] × 10$^7$ molecules/cm$^3$), indicating a weaker oxidative capacity.
However, the levels of NO (43.55 ± 11.65 ppbv) and NO$_2$ (44.93 ± 2.29 ppbv) were much higher
than those in non-episode 2 (14.70 ± 2.40 and 29.46 ± 0.95 ppbv for NO and NO$_2$, respectively),
possibly leading to the enhancement of HNO$_3$ (g) in episode 6. Furthermore, the particle-bound
NO$_3$ was of low thermal stability (Querol et al., 2004), and the temperature lowered ~2.3 C° in
episode 6, which suppressed the decomposition and volatilization of NH$_4$NO$_3$. Therefore, the
high levels of NO$_x$ and NH$_3$ and the low temperature were both responsible for the increase in
NO$_3$ in episode 6.
Table 4. Comparison of temperature, HNO₃ (g), NH₃ (g), NO, NO₂, O₃, and simulated OH and HO₂ between episode 6 and non-episode 2.

<table>
<thead>
<tr>
<th></th>
<th>Episode 6</th>
<th>Non-episode 2</th>
</tr>
</thead>
<tbody>
<tr>
<td>Temperature (°C)</td>
<td>14.9 ± 0.5</td>
<td>17.2 ± 0.3</td>
</tr>
<tr>
<td>HNO₃ (ppbv)</td>
<td>0.65 ± 0.01</td>
<td>0.47 ± 0.03</td>
</tr>
<tr>
<td>NH₃ (ppbv)</td>
<td>13.48 ± 0.72</td>
<td>9.54 ± 0.37</td>
</tr>
<tr>
<td>NO (ppbv)</td>
<td>43.55 ± 11.65</td>
<td>14.70 ± 2.40</td>
</tr>
<tr>
<td>NO₂ (ppbv)</td>
<td>44.93 ± 2.29</td>
<td>29.46 ± 0.95</td>
</tr>
<tr>
<td>O₃ (ppbv)</td>
<td>17.09 ± 2.04</td>
<td>24.57 ± 1.64</td>
</tr>
<tr>
<td>OH (molecules/cm³)</td>
<td>(3.8 ± 1.3) × 10⁵</td>
<td>(7.2 ± 0.9) × 10⁵</td>
</tr>
<tr>
<td>HO₂ (molecules/cm³)</td>
<td>(1.1 ± 0.3) × 10⁷</td>
<td>(2.0 ± 0.2) × 10⁷</td>
</tr>
</tbody>
</table>

3.4.3 SOC formation

In addition to the high levels of NO₃⁻ in episode 6, the proportions of OC also increased during the autumn episodes. Because SOC is an important fraction in OC that often grows as an air mass ages, it could help to explain the increase of OC in the autumn episodes by exploring the possible formation mechanisms of SOC. It is well known that SOC formation is closely related
to SVOCs, which are formed from the reactions with oxidative radicals (i.e., RO₂, NO₃, and HO₂; Kanakidou et al., 2005; Forstner et al., 1997). Hence, the relationship between SOC and SVOCs was investigated. The SVOCs were simulated with the PBM-MCM model and SOC was calculated with the EC-tracer method mentioned in section 3.2.1. The speciation of SVOCs and their precursors can be found in the Table S4 in the Supplementary Material. Briefly, the precursors of SVOCs include isoprene, aromatics, and C₇-C₁₂ n-alkanes.

Figure 15 presents the daily and diurnal variations in SOC and SVOCs. It was found that SOC correlated well with SVOCs in both daily (R² = 0.52) and diurnal (R² = 0.63) patterns in autumn, indicating that the simulated SVOCs were responsible for the production of SOC. The oxidation products of aromatics and isoprene were the main constituents of the SVOCs, with average contributions of 42.5% ± 2.8% and 39.4% ± 2.0%, respectively. Among the aromatics, xylenes made the greatest contribution (15.0% ± 0.7%) to the SVOCs, followed by trimethylbenzenes (11.5% ± 0.7%), ethylbenzene (8.8% ± 0.5%), toluene (5.1% ± 0.7%), and benzene (2.2% ± 0.2%). Compared to those in non-episode 2 (i.e., 40.7% ± 3.4% and 41.1% ± 2.4% contributed by aromatics and isoprene, respectively), the contribution of aromatics to SVOCs increased to 46.3% ± 4.1% during the episodes, whereas the proportion of isoprene oxidation products decreased to 36.1% ± 3.7%, suggesting that the increase in aromatics was the main cause of the autumn episodes. To quantify the contribution of biomass burning to SOC, the observed VOCs were apportioned to different sources, including biomass burning with CH₃CN as the tracer. The source profiles are provided in Figure S12 in the Supplementary Material. According to the SVOCs simulated on the basis of VOCs emitted from biomass burning, the SVOCs were elevated by 15.4% ± 1.3% due to biomass burning during the episodes.

In contrast, the correlations were much worse in summer (R² = 0.01 and 0.31 for daily and diurnal variations, respectively). The high frequency (50.8%) of rainy days in summer was a factor for the poor correlation; for example, the level of SOC was low during the late period of June when the precipitation lasted for about 10 days, and the model overestimated the SVOCs without considering the influence of precipitation. The correlations between SOC and SVOCs (R² = 0.14 and 0.19 for the daily and diurnal variations, respectively) were still poor even after the rainy days were excluded. Hence, the poor correlation should also be related to other factors, such as incomplete consideration of the contribution of biogenic VOCs. Although isoprene was included as a precursor of the SVOCs, other biogenic species (i.e., α-pinene, β-pinene, and
monoterpenes) that were proven to be important precursors of SOC (Kanakidou et al., 2005) were not monitored in this study. Moreover, the level of biogenic VOCs was much higher in summer than in autumn. Taking isoprene as an example, the mixing ratio of isoprene was 66.7 ± 4.9 pptv in summer and only 37.2 ± 2.6 pptv in autumn. The higher missing level of biogenic VOCs in summer led to a higher deficit of SVOCs, perhaps causing the poorer correlation between SOC and SVOCs. Nevertheless, this notion needs requires validation with more comprehensive data on biogenic VOCs.

Figure 14. Correlations between calculated SOC and simulated SVOCs in (a) day-to-day variation and (b) diurnal pattern.

4. Conclusions
In summer and autumn 2014, the concentrations of PM$_{2.5}$ and its components were continuously monitored in Wuhan; six PM$_{2.5}$ episodes were captured. The analysis of PM$_{2.5}$ concentrations and compositions found that Wuhan suffered from relatively high levels of PM$_{2.5}$, even in the warm seasons. Secondary inorganic ions were the most predominant species in PM$_{2.5}$ in the form of NH$_4$NO$_3$ and (NH$_4$)$_2$SO$_4$. Comparable levels of SO$_4^{2-}$ and NO$_3^-$ indicate that stationary and mobile sources had equivalent importance in Wuhan. With the EC-tracer method, it was found that the POC level was slightly higher than that of SOC, and both increased significantly during the episodes. K was the most abundant element in PM$_{2.5}$, implying biomass burning in and
around Wuhan during the sampling campaign. Indeed, the source apportionment revealed that biomass burning was the main cause of increases in PM$_{2.5}$ in episodes 1, 3, and 4. Fugitive dust was the leading factor in episode 2. However, episodes 5 and 6 were mainly attributable to vehicle emissions and SIAs. Study of the formation mechanism of NO$_3^-$ and SOC found that NO$_3^-$ was mainly generated from the homogeneous reactions in episode 6, and the high levels of NO$_x$ and NH$_3$ and the low temperature caused the increase in NO$_3^-$. Furthermore, the daily and diurnal variations of SOC correlated well with those of SVOCs in autumn. Aromatics and isoprene were the main precursors of SOC, and the contribution of aromatics increased during the episodes. However, the correlation between SOC and SVOCs was much worse in summer, possibly as a result of the incompleteness of the biogenic VOC input in the simulation of SVOCs.

This study advances our understanding of the chemical characteristics of PM$_{2.5}$ in warm seasons in Wuhan and for the first time quantifies the contribution of biomass burning to PM$_{2.5}$. The investigation of SOC formation will also inspire the application of the explicit chemical mechanisms on the study of SOA.

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