Interactive comment on “Identification of particulate organosulfates in three megacities at the middle and lower reaches of the Yangtze River” by X. K. Wang et al.

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RE: A point-to-point response to referee #4’s comments

“Identification of Particulate Organosulfates in Three Megacities at the Middle and Lower Reaches of the Yangtze River” (acp-2015-393) by X. K. Wang, S. Rossignol, Y. Ma, L. Yao, M. Y. Wang, J. M. Chen, C. George, and L. Wang

We are grateful to referee #4 for his/her valuable technical comments, encouraging us to further develop our experimental section. A point-to-point response to this reviewer’s comments is given below.

General comments:

1. “I feel that the manuscript is not suitable for publication in its current form for a number of reasons. My first concern is lack of chromatograms and mass spectra in the manuscript or supporting information. I understand that showing a large number of extracted ion chromatograms is not practical. However, the authors need to show at least chromatograms from major peaks in the main manuscript or supporting information. The authors can discuss about the differences in organosulfates at different locations directly using chromatograms. That is far more effective than descriptive texts.”

Reply: We do share with the referee his/her concern about the practical aspects of adding too many chromatograms. However, as recommended we have added extracted ion chromatograms of major m/z in the main text and in the supplement and now start our discussion around those. In addition, we have also added retention times in the supporting information (new Table S3). Nevertheless, a figure such as Figure 3 or Figure 6 appears more appropriate to discuss the complete dataset.

2. “Second, the authors discuss about seasonal and diurnal variations from extremely small sets of data. While the authors can discuss about the differences in these sampling days, they should not be discussed as seasonal or diurnal patterns.”

Reply: Our point of view was not to infer any general conclusion about seasonal or diurnal trends of OSs in China but more to describe our observations and suggest some explanations. Nevertheless, we agree that seasonal/diurnal comparisons from average values in Table 1 are not so relevant and should come along with moderate conclusions. For the Shanghai samples and corresponding manuscript section, even if we do agree that no definitive conclusion can be made about seasonal/diurnal trends with such a small dataset, we think that describing our results this way is informative and could help future more complete field campaign to identify specific points to clarify, as the presence of nitrooxy-OSs during daytime or the importance of the precursors’ seasonality in different locations. The manuscript was nevertheless thoroughly edited to limit the
conclusions about seasonal trends i.e., preferentially highlighting the nature/variety of
the detected OSs rather than their seasonal/diurnal trends. For example, new section
3.5 is now entitled with "Comparison of OSs in the SH samples" (Page 17, Line 460).
3. "Third, I find the manuscript extremely difficult to follow. It is extremely densely
written. The authors should seek a language editing service to make sure that the
manuscript can be read smoothly."

Reply: The manuscript have been lightened and clarified as much as possible and
reviewed by a language editing service. Naturally all technical comments will be imple-
mented in the revised version.

Technical comments:
4. Page 21419 line 13 onwards: The authors should also address Kahnt et al (2015)
here. Kahnt, A., Behrouzi, S., Vermeylen, R., Shalamzari, M. S., Vercauteren, J.,
Roekens, E., Glaeyts, M., and Maenhaut, W.: One-year study of nitroorganic com-
pounds and their relation to wood burning in PM10 aerosol from a rural site in Belgium,

Reply: The reference was addressed. We now state in our revised manuscript (Page
4, Line 76-79) that "For example, the α-pinene-related nitrooxy-OSs were detected at
high concentrations during the winter period, followed by autumn, spring, and summer
at a rural background site in Hamme, Belgium (Kahnt et al., 2013)".
5. Page 21420 line 10: Do the authors mean 'OS' measurements instead of identifica-
tion? I am not aware of a study dealing with OS identification from Shanghai aerosols.
Reply: Ma et al. (2014) have identified 17 OSs in SH with the corresponding authentic
standard compounds.
6. Page 21421 line 4: Do the authors mean 24 h samples? Daily samples mean a
sampling was performed every day. I can only find two 24 h samples in the Table 1.

Reply: In the present manuscript, “daily sample” has been replaced by the sample IDs.
7. Page 21422 line 11: ‘Th’ and ‘m/z’ are used simultaneously. Both are a unit of
massto-charge ratio, and it is redundant here. I recommend using ‘m/z’ as recom-
mented by IUPAC.

Reply: Only "m/z" is used in the revised manuscript (Page 7, Line 179-180).
8. Page 21423 line 1 and throughout the manuscript: A term ‘identification’ should
be reserved for a compound that is positively identified from the comparison to an
authentic standard compound. ‘Pseudo-molecular ion’ should be ‘quasi-molecular ion’.

Reply: We revised our title into “Molecular Characterization of Atmospheric Particulate
Organosulfates in Three Megacities at the Middle and Lower Reaches of the Yangze
River”. Also, we now state in our revised manuscript (Page 4, Line 85-86) that "Re-
cently, ultra-high-resolution mass spectrometry has been applied for tentative deter-
mination of the molecular formulas of OSs in atmospheric aerosols samples” to avoid
any misinterpretation of the term “identification”. “Pseudo-molecular ion” has been re-
placed with ‘quasi-molecular ion’ throughout the manuscript.
9. Page 21423 line 16: The authors claim the detection of about 200 organosul-
fates from the UHPLC/(−)ESI-MS analysis but I am a little skeptical if these compounds
eulated from the column as peaks. The authors must show more extracted ion chro-
matograms in the main manuscript and supporting information. In supporting informa-
tion, the authors show only four extracted ion chromatograms for an m/z value range
between m/z 351.1849 and m/z 351.1853 without showing their mass spectra. From
the description of the data processing, the authors did not seem to consider an iso-
tope distribution of a detected compound, and relied solely on the ratio of H/C, O/C,
N/C, S/C, and DBE. The authors should show corresponding mass spectra to assure
readers that the compounds shown here are indeed organosulfates.

Reply: We have added some more ion chromatograms in the manuscript and in the

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supplement (new Figure S1). Showing additional ion chromatograms and mass spectra would be just overwhelming and unpractical. However, retention times, potential precursors, and corresponding references for the identified OSs have been included in the supplement (new Tables S3). Molecular formula assignment was performed including the following elements: C, H, N, S and O. Isotopes are a classical way to confirm molecular formula assignments when the resolution of the instrument is not sufficient. Here, in most cases, the isotopic peak is not detected due to the low intensity of the related quasi-molecular ion. Nevertheless, we are confident in our assignments as, within an error of 2 ppm, only one molecular formula is generally chemically relevant. Please also refer to our reply to the very last comment of referee #1 on this point.

10. Pages 21422-21423: In connection to the previous concern, how did the authors calibrate m/z values? Have the authors used a lock mass function or were they calibrated externally? This should be clearly stated in the manuscript.

Reply: We apologize for not having provided this information before and have amended the manuscript accordingly. We now state in our revised manuscript (Page 7, Line 180-182) that "The Q-Exactive mass spectrometer was externally mass calibrated daily using a 2 mM sodium acetate solution that provides a series of negative adduct ions in the range of 50–750 m/z".

11. Figure S1 in supporting information: Figure S1 should show the intensity of the peak.

Reply: Figure S1 is shown to illustrate the separation of various OSs. Due to the different PM loadings in the samples, the intensity of the peaks in Figure S1 varies significantly but does not deliver additional useful information. Therefore, in our opinion, relative abundance is retained for the y-axis.

12. Page 21425 line 10: How do the authors know that they are all nitrooxyorganosulfates? They are other structures that can contain nitrogen such as heterocyclic compounds, nitrophenolic compounds, etc.

Reply: MSn experiments were not performed in our study, and hence we cannot exclude other possibilities such as heterocyclic compounds, nitrophenolic compounds, etc. We now state (Page 8, Line 205-209) that "Compounds that satisfy these criteria and present a number of oxygen atoms greater than or equal to 4s+3n (4s+3n≤o) were tentatively regarded as OSs or nitrooxy-OSs. However, other S- and N-containing compounds, such as sulfonates or compounds bearing nitro groups, may also be involved (e.g., Riva et al., 2015b; El Haddad et al., 2013)".

13. Figure 2. The authors should caution the readers that these are a number of isomers separated by the authors’ method. For example, the C10H17O7N1S1 compounds show only three isomers for a certain method but it can be separated into six isomers when the method is further optimized.

Reply: We now state in our revised manuscript (Page 8, Line 212-213) that "Note that the obtained number of isomers may vary significantly when the separation method is further optimized".

14. Page 21426 line 27-29 and throughout manuscript: Is it a nitrate group or nitrooxy group? How do the authors know without MSn experiments?

Reply: Please refer to our reply to the 12th comments from the referee #4.

15. Page 21427: The authors cannot discuss seasonal differences from such a limited number of samples.

Reply: The manuscript was thoroughly edited to limit the conclusions about seasonal trends, preferentially highlighting the nature/variety of the detected OSs rather than their trends. For example, new section 3.5 is now entitled with "Comparison of OSs in the SH samples" (Page 17, Line 460). In addition, air quality and meteorological data on the sampling days have been compared with the seasonal averages in the supplement (new Table S1), indicating that those parameters on the sampling days are quite close to the seasonal average.
16. Page 21429 onwards about KMD and VK diagrams: I find this section very difficult to read and get information out of it. I recommend the authors summarizing most important information here instead of describing every single detail about the diagrams.

Reply: The manuscript was lightened and clarified as much as possible and reviewed by a language editing service.

17. Page 21432: These isoprene originating organosulfates tend to elute very early in the chromatogram, and their MS intensities can be potentially influenced by co-eluting compounds in this region (ion suppression). How have the authors corrected for this? If not, how do the authors know that they are not affected by the ion suppression?

Reply: No correction has been made. We do agree that co-eluting compounds may lead to ion suppression. However, the samples were not analyzed here by ESI direct infusion. LC separation allowed to minimize artifacts due to the matrix, diluting the low amount of sample injected (5 \( \mu \)L) in the LC solvent flow (300 \( \mu \)L min\(^{-1}\)) and resolving most the matrix components (these latter being previously solvent extracted and filtered). In addition, By extracting early-eluting inorganics from our mass spectra, we found that their retention times were much earlier than those of isoprene OSs. Hence, the ion suppression would be minimum under our experimental conditions.

18. Page 21433 onwards: Seasonal and diurnal variations cannot be discussed Él when the number of sampling day is so limited even they are similar to average seasonal conditions.

Reply: Please refer to our reply to the 2nd comment from referee #4.


Interactive comment on Atmos. Chem. Phys. Discuss., 15, 21415, 2015.