

1 **Oxidative Potential of Ambient Water-Soluble PM_{2.5} in the**
2 **Southeastern United States: Contrasts in Sources and Health**
3 **Associations between Ascorbic Acid (AA) and Dithiothreitol**
4 **(DTT) Assays**

5
6 *Ting Fang¹, Vishal Verma², Josephine T. Bates³, Joseph Abrams⁴, Mitchel Klein⁴, Matthew J.*
7 *Strickland⁴, Stefanie E. Sarnat⁴, Howard H. Chang⁵, James A. Mulholland³, Paige E. Tolbert⁴,*
8 *Armistead G. Russell³, Rodney J. Weber*¹*

9
10 [1] School of Earth and Atmospheric Sciences, Georgia Institute of Technology, Atlanta, GA,
11 30332, USA

12 [2] Department of Civil and Environmental Engineering, University of Illinois Urbana-
13 Champaign, Champaign, IL, 61801, USA

14 [3] School of Civil and Environmental Engineering, Georgia Institute of Technology, Atlanta,
15 GA 30332, USA

16 [4] Department of Environmental Health, Rollins School of Public Health, Emory University,
17 Atlanta, GA 30322, USA

18 [5] Department of Biostatistics and Bioinformatics, Rollins School of Public Health, Emory
19 University, Atlanta, GA 30322, USA

20
21
22
23 *Corresponding author
24 Professor Rodney J. Weber, Ph.D.
25 Georgia Institute of Technology
26 School of Earth and Atmospheric Sciences
27 311 Ferst Drive Atlanta GA, 30332
28 office phone: 404.894.1750
29 Email: rodney.weber@eas.gatech.edu
30

31 **Abstract**

32 The ability of certain components of particulate matter to induce oxidative stress through catalytic
33 generation of reactive oxygen species (ROS) *in vivo* may be one mechanism accounting for
34 observed linkages between ambient aerosols and adverse health outcomes. A variety of assays
35 have been used to measure this so-called aerosol oxidative potential. We developed a semi-
36 automated system to quantify oxidative potential of filter aqueous extracts utilizing the
37 dithiothreitol (DTT) assay and have recently developed a similar semi-automated system using the
38 ascorbic acid (AA) assay. Approximately 500 PM_{2.5} filter samples collected in contrasting
39 locations in the Southeastern US were analyzed for a host of aerosol species, along with AA and
40 DTT activities. Here we present a detailed contrast in findings from these two assays. Water-
41 soluble AA activity was higher in summer/fall than in winter, with highest levels near highly
42 trafficked highways, whereas DTT activity was higher in winter compared to summer/fall and
43 more spatially homogeneous. AA activity was nearly exclusively correlated with water-soluble Cu
44 ($r = 0.70-0.94$ at most sites), whereas DTT activity was correlated with organic and metal species.
45 Source apportionment models, Positive Matrix Factorization (PMF) and a Chemical Mass Balance
46 Method with ensemble-averaged source impact profiles (CMB-E), suggest a strong contribution
47 from secondary processes (e.g., organic aerosol oxidation or metal mobilization by formation of
48 an aqueous particle with secondary acids) and traffic emissions to both AA and DTT activities in
49 urban Atlanta. Biomass burning was a large source for DTT activity, but insignificant for AA. AA
50 activity was not correlated with PM_{2.5} mass, while DTT activity co-varied strongly with mass ($r =$
51 $0.49-0.86$ across sites/seasons). Various linear models were developed to estimate AA and DTT
52 activities for the central Atlanta Jefferson Street site, based on the CMB-E sources. The models
53 were then used to estimate daily oxidative potential at this site over the 1998-2009 period. Time-
54 series epidemiological analyses were conducted to assess daily emergency department (ED) visits
55 data for the five-county Atlanta metropolitan area based on the estimated 10-year backcast
56 oxidative potential. Results suggest that estimated AA activity was not statistically associated with
57 any tested health outcomes, while DTT activity was associated with ED visits for both
58 asthma/wheeze and congestive heart failure. The findings point to the importance of both organic
59 components and transition metals from biomass burning and mobile sources to adverse health
60 outcomes in this region.

61

62 **Key Words: water-soluble DTT activity; water-soluble AA activity; oxidative potential;**
63 **source apportionment; health associations; cardiovascular; respiratory**

64 **1. Introduction**

65 Studies have linked exposure to fine particulate matter (PM_{2.5}) with increased respiratory
66 (Harkema et al., 2004; Aust, 2002; Schaumann et al., 2004) and cardiovascular (Pope et al.,
67 2004; Samet et al., 2000) diseases. PM_{2.5} consists of a wide range of chemical components of
68 potentially varying toxicity, implying that PM_{2.5} is not an ideal air quality metric for assessing
69 health impacts. For example, components such as ammonium, sulfate, nitrate, chloride, and some
70 chemical fraction of mineral dust, may be more benign than transition metals (Gasser et al.,
71 2009; Kodavanti et al., 2005; Akhtar et al., 2010), black carbon (or elemental carbon and
72 associated species) (Kleinman et al., 2007; Brunekreef et al., 1997), polycyclic aromatic
73 hydrocarbons (PAHs) (Lundstedt et al., 2007; Burchiel et al., 2005), and other specific organics
74 species (Nel et al., 2001). Although a small mass fraction of PM_{2.5}, these components could play
75 a disproportionately large role in the overall adverse health effects of PM_{2.5}. A comprehensive set
76 of mechanisms explaining the observed linkage between PM_{2.5} mass and adverse health effects
77 has not been established, but it has been hypothesized that one possible contributing
78 physiological route is a particle's ability to induce oxidative stress via catalytic generation of
79 reactive oxygen species (ROS) *in vivo*. A number of studies have associated particle oxidative
80 capacity with PM toxicity (Donaldson et al., 2005; Nel, 2005; Shi et al., 2003; Zielinski et al.,
81 1999), but without available large databases of ambient aerosol ROS, large population-based
82 epidemiologic studies of PM_{2.5} oxidative potential have not been possible.

83 A number of different assays have been developed to quantify the oxidative potential of PM
84 samples (Zomer et al., 2011; Mudway et al., 2011; Ayres et al., 2008; Jung et al., 2006; Cho et
85 al., 2005; Mudway et al., 2005; Venkatachari et al., 2005). Two commonly used approaches are
86 the dithiothreitol (DTT) assay (Cho et al., 2005) and the ascorbic acid (AA) assay (Ayres et al.,

87 2008; Mudway et al., 2005). It may be expected that these two different assays respond to
88 different aerosol components and are linked to different health endpoints. The AA assay has
89 been shown to be most sensitive to transition metals (Janssen et al., 2014; Strak et al., 2012;
90 DiStefano et al., 2009; Künzli et al., 2006) but quinone compounds may react with AA as well
91 (Roginsky et al., 1999). For the DTT assay, identified DTT-active PM components are organic
92 species, including water-soluble organic carbon (WSOC) (Verma et al., 2009; Cho et al., 2005),
93 or of increasing specificity, HUmic-Like Substances (HULIS) (Verma et al., 2012; Lin and Yu,
94 2011), and quinones (Chung et al., 2006; Kumagai et al., 2002) (a component of HULIS). Other
95 studies, however, have emphasized the role of transition metals, such as Cu and Mn (Vejerano et
96 al., 2015; Charrier and Anastasio, 2012). Some differences between studies may arise due to
97 differing source characteristics of the specific regions studied.

98 Both assays involve incubating the anti-oxidant (DTT or AA) with filter aqueous extracts of
99 PM_{2.5} at a controlled temperature (37 °C) and pH (7.4), and measuring the depletion of the
100 antioxidant over time, typically detected as a decrease in light absorption at a certain
101 wavelengths (412 and 265 nm for DTT and AA, respectively). The antioxidant loss rate is
102 interpreted as a measure of the ability of aerosol redox-active species to catalytically transfer
103 electrons from DTT or AA to oxygen (O₂). DTT can be considered a chemical surrogate to
104 cellular reductants, such as NADH or NADPH, which reduces O₂ to superoxide anion (O₂⁻) and
105 induces oxidative stress (Kumagai et al., 2002). Unlike DTT, AA is a physiological antioxidant
106 in lung lining fluid, which prevents the oxidation of lipids and proteins (Valko et al., 2005).
107 Asthmatic patients have markedly decreased concentration of AA in lung lining fluid compared
108 to healthy control subjects (Kelly et al., 1999). Therefore, the *in vitro* oxidation of these two

109 antioxidants by PM might represent the interaction of PM with biological antioxidants *in vivo*
110 leading to the induction of oxidative stress and ultimately adverse health effects.

111 Among the various available methods for measuring oxidative potential, these two assays are
112 relatively straightforward and reproducible, allowing high throughput routine measurements and
113 the generation of large data sets for exploring links between aerosol components and health
114 through epidemiology, or also as an initial screening step for identifying different redox
115 components for more detailed cell or animal studies (Ayres et al., 2008).

116 We recently developed a semi-automated system (Fang et al., 2015b) to measure DTT activity
117 and here describe its adaption to the AA assay. Utilizing our automated analytical system, we
118 measured the water-soluble oxidative potential of over 500 filter samples collected as part of the
119 Southeastern Center for Air Pollution & Epidemiology (SCAPE) study. Although in-soluble
120 components are important, since there is no current standard protocol for measuring the water-
121 insoluble oxidative potential, we focus solely on the water-soluble AA and DTT activities. We
122 evaluate and compare these two assays in order to identify specific aerosol components the AA
123 assay is responsive to. We perform a source apportionment analysis and assess these results
124 through observed AA activity seasonal and spatial variability. AA source profiles are used to
125 generate a model that estimates AA activities, which is then used to backcast AA levels over the
126 past 10 years for use in a time-series epidemiological analysis in the Atlanta metropolitan area.

127 Throughout, we compare the AA results to our previously published DTT findings (Bates et al.,
128 2015, Fang et al., 2015b, Verma et al., 2014) to provide a contrast between these two commonly
129 utilized assays to assess aerosol water-soluble oxidative potential and possible associations with
130 health endpoints.

131

132 2. Methods

133 2.1. Sampling

134 Sampling methods have been described in detail elsewhere (Fang et al., 2015b; Verma et al.,
135 2014). In brief, PM_{2.5} (quartz filters, Pallflex® Tissuquartz™, 8 × 10 inches) was sampled at
136 seven locations in the Southeastern US, with different source characteristics, using two sets of
137 high-volume samplers (Hi-Vol) (Thermo Anderson, flow rate normally 1.13 m³ min⁻¹).
138 Sampling in the metropolitan Atlanta area was carried out from June 2012 through March 2013
139 (noon - 11 a.m., 23 hours) and involved paired-sites with one Hi-Vol sampler fixed at an urban
140 background site (Jefferson Street, referred as JST) whilst the other sampler was deployed at three
141 other sites on a monthly basis, and at least twice during different seasons. These three sites were:
142 a rural site (Yorkville, YRK), a road-side site (RS, adjacent to the interstate highway I75/85),
143 and a near-road site (GT, 840 m from the RS site). Following sampling in Atlanta, the two
144 samplers were moved to Birmingham, AL (BHM, within a few kilometers of significant
145 transportation and industrial sources) and Centerville, AL (CTR, surrounded by forests and a
146 lightly traveled county road) for a month of sampling in June-July 2013, followed by one-month
147 August sampling at East St. Louis, IL, an urban residential/light commercial area about 3 km east
148 of the central business district of St. Louis, MO (Sauvain et al., 2008). Finally, a GT-RS pair was
149 conducted in September 2013. A table providing the sampling schedule and a map can be found
150 in the supporting material (Table S1 & Fig. S1). JST, YRK, BHM, and CTR are all part of the
151 Southeastern Aerosol Research and Characterization Study (SEARCH) network sites (Hansen et
152 al., 2003). Collected samples were immediately wrapped in prebaked aluminum foil and stored at

153 -18°C until analyzed. DTT, water-soluble organic carbon, and brown carbon analyses on the
154 filters were conducted within a year of sample collection, water-soluble elements were within a
155 year and half, and AA measurements were conducted within two years of sample collection.

156 **2.2. Oxidative potential measurement**

157 **Filter extraction:** One punch of the collected Hi-Vol filter (5.07 cm^2) was extracted in 30 mL of
158 deionized (DI) water ($> 18\text{ M}\Omega\text{ cm}^{-1}$) in a sterile polypropylene centrifuge tube (VWR
159 International LLC, Suwanee, GA, USA) by sonication using an Ultrasonic Cleanser (VWR
160 International LLC, West Chester, PA, USA) for half an hour. For those having activities close to
161 blanks, 15 mL was used instead. Extracts were then filtered using PTFE $0.45\text{ }\mu\text{m}$ syringe filters
162 (FisherbrandTM) to remove insoluble material larger than $0.45\text{ }\mu\text{m}$. Although OH may form
163 during sonication (Miljevic et al., 2014), it appears to have little effect on our ROS measurement
164 since we compared the water-soluble ROS activities from the same sample that had been
165 extracted by shaking for 3 hours vs sonication and found no significant differences (average ratio
166 and standard deviation is 1.08 ± 0.20 , $n = 7$).

167 **AA determination:** The method in this study was based on an ascorbate-only model (Mudway et
168 al., 2005; Ayres et al., 2008) that is a simplified and alternative high throughput approach to a
169 synthetic respiratory tract lining fluid model (RTLFL) containing ascorbate, urate, and reduced
170 glutathione (GSH) (Zielinski et al., 1999; Mudway et al., 2004). The method protocol is shown
171 in Figure 1 and system setup can be found in the Figure. S2 in the Supplement. The method
172 involves two steps.

173 The first step is an aerosol background measurement (Fig. 1). In order to control for the
174 contribution of absorbance of particles themselves at 265 nm wavelength, an AA-free control

175 was measured and subtracted from the sample absorbance readings. 2.4 mL aerosol extracts and
176 0.3 mL 0.5 mM Kbuffer were loaded into a reaction vial (sterile polypropylene centrifuge tube,
177 VWR International LLC, Suwanee, GA, USA) using a programmable syringe pump (A) with a 5
178 mL syringe (Kloehn, Inc., Las Vegas, NV, USA). Following mixing, 90 μ L of the mixture was
179 transferred to an intermediate vial using Pump B with a 250 μ L syringe, and diluted to 3 mL.
180 Pump A then withdrew the diluted mixture from the intermediate vial and pushed it through a
181 Liquid Wave-guide Capillary Cell (LWCC-M-100; World Precision Instruments, Inc., FL, USA)
182 with an optical path length of 100 mm. The waveguide was coupled to an online
183 spectrophotometer, which included a UV-VIS light source (Ocean Optics DT-Mini-2, Ocean
184 Optics, Inc., Dunedin, FL, USA), and a multi-wavelength light detector (USB4000 Miniature
185 Fiber Optic Spectrometer, Ocean Optics, Inc., Dunedin, FL, USA). Aerosol background
186 absorbance at 265 and 700 nm (baseline) were recorded at two-second intervals using data
187 acquisition software (SpectraSuite). For the samples collected in this study, backgrounds due to
188 the aerosol absorption at 265 nm were <10% of the sample absorbance readings. Prior to the
189 second step, the system performed a self-cleaning by flushing the intermediate vial and the two
190 syringes with DI water three times.

191 The second step is the AA measurement (Fig. 1). Following the aerosol background
192 measurement, Pump A discarded a fraction of the sample-Kbuffer mixture and left only 1.8 mL
193 in the reaction vial. 0.2 mL 2 mM AA solution was then loaded to the reaction vial using Pump
194 B. Both the reaction and intermediate vial were continuously shaken at 400 rpm in a
195 ThermoMixer (Eppendorf North America, Inc., Hauppauge, NY, USA), which also maintained
196 the incubation temperature at 37 °C. At five different specified times (7, 15, 24, 32, 40 minutes),
197 a small aliquot (100 μ L) was transferred to the intermediate vial, diluted to 3 mL, and pushed

198 through the LWCC, generating a total of five data points quantifying the remaining AA
199 concentration. The system then again performed a self-cleaning before analyzing the next
200 sample. A multi-position valve (14-port, VICI® Valco Instrument Co. Inc., USA) was used to
201 select samples for analysis. To ensure the suspension of PM in the extract, each sample was
202 mixed by pushing 5mL of air through the extract before loading to the reaction vial. The detailed
203 Kloehn control program code for all steps can be found in the Supporting Information.

204 Final AA activity is calculated as follows:

$$205 \quad \sigma AA = -\sigma Abs \times \frac{N_0}{Abs_0} \text{ (Eq. 1) ,}$$

$$206 \quad AA v = \frac{\sigma AA_s - \sigma AA_b}{\frac{V_a}{V_e} \times V_p} \text{ (Eq. 2) .}$$

207 Following the notation above, σAbs is the slope of absorbance versus time, where the absorbance
208 is the absorbance of each time interval subtracting the corresponding aerosol background
209 absorbance; Abs_0 is the initial absorbance calculated from the intercept of linear regression of
210 absorbance versus time; N_0 is the initial moles of AA added in the reaction vial (400 nmol);
211 σAA_s (σAA_b) is the rate of AA consumption for a sample (blank); V_e and V_a are the extraction
212 volume (30 or 15 mL) and sample volume added to the reaction (1.6 mL), respectively. V_p is the
213 ambient air volume (m^3) represented by the sample in the extraction volume. $AA v$ represents
214 volume normalized AA activity, in units of $nmol \text{ min}^{-1} m^{-3}$. Similarly, here $DTT v$ represents the
215 volume normalized DTT activity.

216 **2.3. Chemical analysis on PM filters**

217 **2.3.1. Water-soluble organic carbon and brown carbon**

218 An automated system (details in Fig. S3) was used to measure water-soluble organic carbon
219 (WSOC) and brown carbon (BrC) on the water-soluble extracts from the same Hi-Vol filters.
220 Filter extracts (~6mL, same extraction protocol outlined above), after loading onto a 5 mL
221 sample loop (Upchurch Scientific, Inc., Oak Harbor, WA), were first passed through a 1 m
222 LWCC (LWCC-2100; World Precision Instruments, Inc., FL, USA), where absorbance at 365
223 nm wavelength (BrC) was measured using an online spectrophotometer (Ocean Optics, Inc.,
224 Dunedin, FL, USA). The extracts then entered a TOC analyzer (Sievers Model 900, GE
225 Analytical Instruments, Boulder, CO, USA) for determining WSOC concentration.

226 **2.3.2. Water-soluble elements**

227 A similar automated system was developed to determine the water-soluble elements, including S
228 (Sulfur), Ca (Calcium), K (Potassium), Fe (Iron), Cu (Copper), Zn (Zinc), Ba (Barium), Pb
229 (Lead), As (Arsenic), Sr (Strontium), Se (Selenium), Br (Bromine), Mn (Manganese), and Ti
230 (Titanium). Details of the method are described in Fang et al. (2015a) and in the Supplement.

231 **2.3.3. PM_{2.5} mass**

232 PM_{2.5} mass concentration was measured by a Tapered Element Oscillating Microbalance
233 (TEOM) by Atmospheric Research Analysis (ARA, Inc.) at SEARCH sites (JST, YRK, BHM,
234 and CTR) and ESL. For the RS and GT sites, since PM_{2.5} mass were not available, the PM mass
235 concentrations were estimated from the sum of chemical components analyzed on the same Hi-
236 Vol filters (Verma et al., 2014) (Details in the Supplement).

237 **2.4. Source apportionment**

238 Source apportionment of AAv was performed using a Positive Matrix Factorization (PMF)
239 model (EPA PMF 5.0 software) (Paatero and Tapper, 1994) and a Chemical Mass Balance model
240 (version 8.2) with ensemble-averaged source impact profiles (CMB-E) (Balachandran et al.,
241 2012). PMF is a commonly used source apportionment approach that does not require source
242 profiles as CMB-E, whereas CMB-E has better performance and lower relative uncertainties as
243 compared to the PMF method (Balachandran et al., 2012). Source contributions to DTTv using
244 PMF and CMB-E are discussed in our other publications (Bates et al., 2015; Verma et al., 2014).
245 A PMF analysis on the water-soluble elements (S, K, Ca, Ti, Mn, Fe, Cu, Zn, As, Se, Br, Sr, Ba,
246 and Pb) and WSOC from JST, GT, and RS sites has been reported in Fang et al. (2015a). AAv
247 was simply added to the data sets to generate the AAv results shown here. The PMF results on
248 DTTv in our prior analyses were based on JST and GT sites.

249 In PMF, the uncertainties for each species were determined by multiplying the concentration by
250 overall uncertainties (%), which were obtained by propagating the uncertainties from filter
251 sampling (assumed to be 5%), extraction (assumed to be 5%), blanks (1 σ of multiple blanks),
252 calibration (1 σ of slope, for water-soluble elements), collocated measurements (for water-
253 soluble elements and AA, Fig. S4), and analytical uncertainties. The analytical uncertainties were
254 obtained by analyzing the same sample/standards multiple times; for example, a composite of
255 extracts from 11 samples for water-soluble elements (coefficient of variation, CV = 2-16 %);
256 9,10-phenanthrenequinone for AA (CV = 13%), and sucrose standard solutions for WSOC (CV
257 = 10%). Missing data were replaced by species medians with 400% uncertainty, and values
258 below LOD were assigned as half of LOD values with uncertainties of 5/6 the concentration

259 (Polissar et al., 1998). Uncertainty from collocated measurements was calculated as the relative
260 uncertainty of the slope ($1 \sigma/\text{slope}$), which was based on an orthogonal regression.

261 An ensemble-trained source apportionment approach (Balachandran et al., 2012) (CMB-E) was
262 also used to construct the source impacts on AAv based on PM_{2.5} species (sulfate, nitrate,
263 ammonium, OC, EC, and total metals) and AAv measured during SCAPE sampling periods
264 (2012-2013). The source profiles cover a range of sources, including light-duty gasoline vehicles
265 (LDGV), heavy-duty diesel vehicles (HDDV), ambient sulfate (AMSULF), ambient nitrate
266 (AMNITR), ammonium bisulfate (AMBSLF), not otherwise apportioned organic carbon
267 (OTHER_OC), dust, biomass burning (BURN), coal fired power plants, cement, and cooking.

268 An ensemble average was calculated for each source category using ten different runs developed
269 from four individual source apportionment methods. Since the filters for the source impact
270 profiles were collected from midnight to midnight, while the filters collected for AAv
271 measurement were collected from noon to 11 a.m. next day, the sources identified were linearly
272 interpolated using a fixed ratio. For example, two consecutive filters (filter 1 and 2) collected
273 from midnight to midnight, $12/24*\text{filter1} + 11/24*\text{filter2}$ would be used to produce the estimated
274 AAv to compared with actual measured data.

275 **2.5. Epidemiological assessment**

276 **2.5.1. Backcast-estimates of AA activities**

277 To undertake a time-series epidemiological analysis with sufficient power, retrospective data sets
278 of daily AAv levels from 1 August 1998 to 31 December 2009 at an Atlanta site representative
279 of the urban airshed air quality are needed. Previous epidemiological studies by the study team,
280 assessing Atlanta air quality and emergency department (ED) visits, have used data from the

281 SEARCH JST site, the anchor site for our AAv measurements. To generate daily estimates of
282 retrospective AAv at JST, first a linear model was used to estimate the contribution of various
283 sources to our observed AAv measured at JST. This was done through separate linear regressions
284 for AAv, with the ensemble-predicted sources as independent variables. In previous work
285 (Balachandran et al., 2012), a source times-series from August 1998 to December 2009 was
286 generated for JST using the same CMB-E model with the same independent variables measured
287 at JST. The AAv regression was then applied to this time series to construct a time series of
288 estimated AAv for the epidemiology study time period, during which direct measurements of
289 AAv were not available. In order to test the sensitivity of epidemiologic results to different
290 backcast models, two other models are generated for AAv (discussed in section 3.2.1). Identical
291 methods were applied to DTTv to obtain three different models for comparisons with those from
292 AAv.

293 **2.5.2. Epidemiological analyses**

294 Epidemiological time-series analysis (Strickland et al., 2010; Winquist et al., 2015) was
295 employed to assess associations of retrospective DTTv and AAv with health effects as reflected
296 in ED visits. Relationships between ED visits data from Atlanta area hospitals and typical
297 ambient air quality characteristics, as well as the impact of exposure misclassification and other
298 factors, have been extensively studied (Strickland et al., 2015; Darrow et al., 2014; Strickland et
299 al., 2014; Winquist et al., 2014; Goldman et al., 2012; Pachon et al., 2012; Strickland et al.,
300 2011; Strickland et al., 2010; Sarnat et al., 2010; Sarnat et al., 2008; Tolbert et al., 2007; Metzger
301 et al., 2004; Peel et al., 2005). For the present analysis, we apply these previously reported
302 epidemiologic modeling approach to the backcast-estimates of AAv and DTTv, in order to assess
303 associations of these newly developed air quality descriptors with selected outcomes in the ED

304 visits data collected from hospitals serving the five-county metropolitan Atlanta area during
305 1998-2009. The health outcomes investigated in the current analysis are daily visits for
306 respiratory diseases, including pneumonia (n=145,610 total visits for study period), chronic
307 obstructive pulmonary disease (n=49,251), and asthma/wheeze (n=263,665), and cardiovascular
308 diseases, including ischemic heart disease (n=73,477) and congestive heart failure (CHF)
309 (n=70,587). The air quality was modeled as a three-day moving average (“lag 0-2”, the moving
310 average of estimated pollutant level for that day, the previous day, and the day before). Poisson
311 generalized linear regression was performed; to control for temporal trends and meteorological
312 variables, models included cubic splines with monthly knots for time, linear, quadratic and cubic
313 terms for mean daily dew point (lag 0-2), maximum daily temperature (lag 0), and minimum
314 daily temperature (lag 1-2), indicators of hospital contribution time periods, season of year, day
315 of week and holiday, and interaction terms between season and maximum temperature, and
316 between season and day of week. These covariates were chosen based on prior studies
317 (Strickland et al., 2010; Winquist et al., 2015) which identified important confounders to the
318 relationship between daily ambient pollution levels and ED visits. Risk ratios (the relative risks
319 of ED visit associated with an increase of one interquartile range of the exposure metric) and
320 95% confidence intervals (CI) were used to describe the observed health associations. Risk ratios
321 with confidence intervals above 1 are indicative of statistically significant positive associations.
322 International classification of disease codes used to define the health outcomes can be found in
323 the Supplement.

324 3. Results and Discussion

325 3.1. AAv for measurements during 2012-2013 and comparisons to DTTv

326 3.1.1. Spatio-temporal distribution

327 Monthly average water-soluble AAv at various sampling sites are given in Fig. 2 (a). AAv was
328 heterogeneously distributed, indicated by the significant variability between sites. Highest AAv
329 were found at the roadside site (RS) and lowest at rural sites. For example, the ratio of average
330 AAv at RS to its paired Atlanta urban JST site, was 1.2 in fall and 1.6 in winter [(2.7 when
331 including the four high data points in Fig. 2 (a)] and RS to near-road GT was 1.7 in fall 2013.
332 AAv at the rural site was generally lower compared to the urban environments, the average
333 YRK/JST ratio was 0.7 in summer and 0.4 in winter, respectively. An exception is that BHM
334 (urban) and CTR (rural) had similar AAv ($BHM_{avg} = 0.75 \pm 0.45 \text{ nmol min}^{-1} \text{ m}^{-3}$ and $CTR_{avg} =$
335 $0.78 \pm 0.31 \text{ nmol min}^{-1} \text{ m}^{-3}$). Comparing AAv at different sites, BHM and the other urban site,
336 ESL (average AAv = $0.98 \pm 0.63 \text{ nmol min}^{-1} \text{ m}^{-3}$), had lower AAv relative to the Atlanta urban
337 sites (average of JST and GT in summer = $2.5 \pm 1.0 \text{ nmol min}^{-1} \text{ m}^{-3}$). The higher AAv near
338 traffic sources has also been found in other studies (Janssen et al., 2014; Strak et al., 2012;
339 Janssen et al., 2015). In contrast, Figure 2(b) shows that DTTv was largely spatially uniform,
340 differences between paired sites is much less than those for AAv.

341 The seasonal distribution can be examined from the Atlanta sites, JST, GT, RS, and YRK. The
342 AA activity was higher in summer/fall compared to winter; the ratio of summer or fall to winter
343 was 1.8, 2.9, 1.0, and 3.1 (average ratio is 2.2 ± 0.9) for JST, YRK, RS, and GT, respectively. In
344 contrast, JST DTTv had an opposite seasonal trend, with the highest level in December (winter),
345 while there was no significant seasonal variation observed at YRK, GT and the RS site. These
346 results indicate that there are differences in the sources for water-soluble AAv and DTTv, with

347 traffic emissions a more significant source for AAv. Correlation analysis with specific aerosol
348 components provides further insights.

349 **3.1.2. Correlations with chemical components**

350 To further identify the major sources for AAv and compare to DTTv, a correlation analysis was
351 performed between the assays and the following selected chemical components; BrC (an
352 indicator of incomplete combustion, i.e., biomass burning), WSOC and S (secondary processes),
353 Ca (mineral dust), and selected transition metals (Cu, Fe, Mn, and Zn) that have been related to
354 adverse health outcomes (Cheung et al., 2012; Kam et al., 2011; Shen and Anastasio, 2011;
355 Cheung et al., 2010; Akhtar et al., 2010; Landreman et al., 2008; Zhang et al., 2008; Kodavanti
356 et al., 2005). Correlation coefficients based on linear regressions between AAv or DTTv and
357 chemical species (Pearson's r) are shown graphically in Fig. 3. A detailed matrix showing the
358 correlations at individual sites is given in Table S2. To simplify Figure 3, JST and GT were
359 combined into one metric given their close proximity and high correlation ($r > 0.7$) for many PM
360 species, such as EC, WSOC, and water-soluble elements (Fang et al., 2015a; Verma et al., 2014).
361 As shown in Fig. 3, AAv was almost exclusively correlated with water-soluble Cu. The r value
362 ranged from 0.70-0.94 for most sites/seasons except RS in fall 2012, JST/GT in winter, ESL in
363 summer, and GT in fall 2013. High correlations between AAv and Cu are consistent with other
364 studies (Janssen et al., 2014; Künzli et al., 2006), although the correlation coefficients (r) in our
365 work (0.70-0.94) were higher (0.60-0.74 in other studies), possibly because we used water-
366 soluble Cu and the other studies used elemental (total) concentrations. Strak et al (2012) also
367 reports a higher r value between AAv and water-soluble Cu ($r = 0.82$) than that between AAv
368 and total Cu ($r = 0.76$) from the same sample set.

369 Compared to AAv, DTTv is more broadly correlated with aerosol species: high correlations were
370 observed with S ($r = 0.66-0.74$) and WSOC ($r = 0.71- 0.77$) in summer, which diminished in fall
371 ($r = 0.14-0.66$ for S and $r = 0.20-0.65$ for WSOC) and was weaker in winter ($r < 0.4$) for S.
372 Instead, higher correlations were found with BrC ($r = 0.78-0.88$) and WSOC ($r = 0.60-0.84$) in
373 winter. The decreasing correlation between DTTv and S going from summer to winter suggests
374 the important role of secondary processing in summer (Verma et al., 2009a; McWhinney et al.,
375 2013) and the increasing contribution of biomass burning to winter DTTv; AAv did not show
376 similar trends, i.e., AAv did not correlate with S ($r = -0.12-0.60$) and low r values were observed
377 with K in winter ($r = 0.07-0.19$, one exception was JST in December $r = 0.7$), suggesting
378 incomplete combustion (e.g., biomass burning) was not a significant source for AAv. Whereas
379 AAv was nearly exclusively correlated with Cu, DTTv was correlated with various metals,
380 including Ca, Mn, Fe, Cu, and Zn. In Fig. 2, counting the number of sampling sites at which r
381 values between AAv and various metals were larger than 0.65 (i.e. black solid bars), we
382 observed eight times for Cu and once for Mn ($r = 0.82$). Whereas for DTTv (see the striped bars
383 in Fig.3), we found twice ($r = 0.67$ and 0.77) for Ca, 3 times for Mn ($r = 0.65-0.75$), 6 times for
384 Fe ($r = 0.68-0.90$), once for Cu ($r = 0.68$), and 3 times for Zn ($r = 0.70-0.82$). There were,
385 however, no apparent seasonal patterns for the correlations between these water-soluble metals
386 and DTTv since they were related to mineral dust (68% of Ca, 45% of Mn, and 26% of Fe) and
387 vehicle brake/tire wear emissions (51% of Cu, 45% of Zn, 32% of Fe, and 17% of Mn).

388 The comparison of AAv and DTTv's correlation with $PM_{2.5}$ mass is noteworthy. DTTv was
389 fairly well correlated with $PM_{2.5}$ mass ($r = 0.49-0.86$, Fig. 3), whereas AAv did not correlate as
390 well ($r = -0.17$ to 0.59), as also found in Künzli et al. (2006) ($r = 0.34$). $PM_{2.5}$ mass has been
391 linked with adverse health endpoints in many epidemiological studies (Laden et al., 2000; Pope

392 et al., 2002; Pope et al., 2004; Metzger et al., 2004; Sarnat et al., 2008), thus the lack of
393 correlation of AAv with PM_{2.5} mass may be suggestive of a lack of linkage between AA-
394 measured oxidative potential and health endpoints (tested in section 3.2). In summary, AA and
395 DTT assays have differing associations with PM chemical species, further suggested by a lack of
396 strong correlation between the two assays at all sites ($r < 0.55$), similar to findings from other
397 studies ($r < 0.65$) (Janssen et al., 2014; Yang et al., 2014; Janssen et al., 2015). A source
398 apportionment analysis is performed to attempt to quantify contributions of various sources to
399 PM_{2.5} AAv.

400 **3.1.3. Source apportionment**

401 Various source contributions to water-soluble AAv and DTTv resolved from PMF and CMB-E
402 based on measured data during 2012-2013 are shown in Fig.4 (a, b) and (c, d), respectively. For
403 AAv, the detailed loadings of various species and time series of each factor suggested by PMF
404 can be found in Fig. S5.

405 **AA Sources:** Comparing Fig.4 (a) and (b) shows that CMB-E and PMF gave consistent and
406 complementary results for AAv sources. Consistent with the spatial distributions, CMB-E
407 indicated that vehicles comprise almost half of the total source contributions to PM_{2.5} AAv with
408 roughly equal contributions from light and heavy duty vehicles. PMF also found vehicle
409 emissions as a major source, but resolved the source as mechanical generation processes (44%
410 from brake/tire wear). Both methods also found sources of AAv associated with secondary
411 processes. CMB-E provides more details in that it separated out secondary organic and ambient
412 sulfate sources. This is consistent with our source apportionment analyses on water-soluble
413 metals (Fang et al., 2015a), which showed that Cu was mainly associated with secondary
414 formation and brake/tire wear, consistent with AAv being highly correlated with Cu. The 19%

415 other OC source in Figure 3b is related to un-apportioned OC, which includes secondary organic
416 aerosols from biogenic emissions, and possible additional contributions from other VOC sources.
417 At this point, the role it plays as a source of AAv is not clear. It should be noted that the residual
418 for PMF was -10% and that for CMB-E was 14%, which means the PMF model over-predicted
419 and CMB-E under-estimated AAv. The source apportionment analysis is consistent with the
420 spatial distribution, which indicated vehicle emissions as a main source for AAv activity.

421 **AAv compared to DTTv Sources:** Comparisons of the source apportionment results on water-
422 soluble AAv using PMF and CMB-E to a similar analyses for DTTv is insightful. PMF source
423 apportionment analyses [Fig. 4(a) and (c)] suggest a common contribution from traffic emissions
424 and secondary processes to both water-soluble AAv and DTTv, but the contributions were
425 stronger for AAv than DTTv. For example, 44% AAv was attributed to vehicles and 56% to
426 secondary processes, compared to 16% and 31% for DTTv, respectively. Higher fractional
427 contributions of these two sources for AAv is because unlike DTTv, biomass burning does not
428 contribute to AAv (1%), whereas it makes a large contribution to the overall study DTTv (35%).
429 CMB-E also found no contribution of biomass burning to AAv, but identified a fractional
430 contribution from biomass burning [36% BURN in Fig. 4(d)] to DTTv similar to PMF. CMB-E
431 points to ammonium sulfate (AMSULF) as a source for both AAv and DTTv. Neither AA nor
432 DTT assay responds to pure ammonium sulfate, meaning that ammonium sulfate is an indicator
433 of some source or process. It may be a marker for atmospheric processed or aged aerosols. For
434 example, both assays respond to water-soluble transition metals, and a significant fraction these
435 metals, when emitted, are not water-soluble [solubilities of Zn is ~50%, Cu and Mn 10-40%, Fe
436 < 10% (Birmili et al., 2006; Espinosa et al., 2002)]. Mobilization by acidic aerosols can increase
437 the soluble fraction, which requires a low aerosol pH and time, both can be linked to sulfate

438 aerosol. For example, at $\text{pH} < 2$, 1-2% of mineral dust Fe is mobilized within 3-5 days
439 (Meskhidze et al., 2003).

440 It is worth noting that both assays appear to be linked to emissions from traffic, but the actual
441 sources from traffic differ. AAv was almost exclusively associated with mechanically generated
442 aerosols (i.e. brake/tire wear), whereas for water-soluble DTTv, traffic emissions included both
443 metals and organic aerosol species, i.e., from mechanically generated (brake/tire wear) and
444 combustion (tail pipe emissions). Finally, correlations to specific aerosol species and source
445 apportionment analysis can be confounded by co-variability with other unmeasured components
446 or processes, as demonstrated by the associations with ammonium sulfate, or nonlinear responses
447 of these assays to specific components (Charrier et al., 2015). However, the major sources
448 identified for both AAv and DTTv, and the contrasts between their sources, is consistent with the
449 season trends and spatial distributions observed and discussed above for each assay, indicating
450 that the source apportionment analysis is robust.

451 **3.2. AAv association with health endpoints and contrasts to DTTv**

452 **3.2.1. Backcast-estimates of AAv using Source Impacts**

453 Although over roughly 1 year of AAv were generated for the central JST site in Atlanta, longer
454 data sets are generally needed for a time series epidemiological study. To generate these data, a
455 multiple linear regression was used to estimate AAv from the CMB-E identified sources. We
456 follow the same approach as that used for DTTv (Bates et al., 2015). Water-soluble AAv (nmol
457 $\text{min}^{-1} \text{m}^{-3}$) measured between 2012 and 2013 at JST were regressed against all CMB-E sources.
458 Insignificant sources (p of F-statistic of coefficient > 0.05 , Table S3) and the significant sources
459 with negative coefficients were removed. The latter occurred for BURN (biomass burning) and
460 AMNITR (ammonium nitrate), likely due to their opposite seasonal trends to the measured AAv.

461 These two sources also did not contribute to AA_v [see Fig. 4(a)]. The final regression for AA_v
462 is:

$$463 \quad AA_v^e = 0.079 + 0.19 LDGV + 0.23 HDDV + 0.063 AMSULF + 0.075 OTHER_OC \quad (\text{Eq. 3})$$

464 For direct comparison with DTT_v , we used the same criteria for including various sources in the
465 DTT_v regression model, with the result:

$$466 \quad DTT_v^e = 0.067 + 0.11 LDGV + 0.045 HDDV + 0.02 AMSULF + 0.069 BURN \quad (\text{Eq. 4})$$

467 (Note, the DTT_v^e regression is different from that in Bates et al., (2015) in that AMSULF (ammonium
468 sulfate) was included in this model)

469 AA_v^e and DTT_v^e are the estimated ROS activities of $PM_{2.5}$ ($\text{nmol min}^{-1} \text{m}^{-3}$), which are related to
470 the following sources ($\mu\text{g m}^{-3}$): light-duty gasoline vehicles (LDGV), heavy-duty diesel vehicles
471 (HDDV), ammonium sulfate (AMSULF), biomass burning (BURN), and other organic carbon
472 (OTHER_OC). The coefficients in the equations represent the intrinsic activities ($\text{nmol min}^{-1} \mu\text{g}^{-1}$)
473 of the sources, a measure of the strength of the source on a per $PM_{2.5}$ mass basis for water-
474 soluble AA_v or DTT_v . Interestingly, for both assays, the traffic sources (LDGV+HDDV) has the
475 highest ROS intrinsic activity, while secondary sources or biomass burning have relatively lower
476 ROS intrinsic activities. The high intrinsic activity in the traffic sources might be attributed to
477 metals that have much higher intrinsic ROS activities (Charrier and Anastasio, 2012; Verma et
478 al., 2015). The much higher coefficients of LDGV and HDDV in the AA_v^e regression than those
479 in DTT_v^e highlight the larger role of metals from these sources contributing to the overall AA_v^e .
480 Although biomass burning has a lower DTT intrinsic activity compared to the other sources in
481 the DTT_v^e model, it was the largest contributor to DTT_v^e due to the strength of this source over the
482 measurement period (e.g. large magnitude of BURN). The regression positive intercepts indicate

483 some unidentified source for AA_v and DTT_v . The ability of the models to predict AA_v^e and DTT_v^e
484 is given by the correlation between the model and measurements (AA_v^e vs AA_v and DTT_v^e vs
485 DTT_v). The r values are 0.60 and 0.68 for AA_v^e and DTT_v^e , respectively, indicating the models
486 can only account for about 40% of the observed variability. Regression coefficients, p-values,
487 and r values are summarized in Table S3.

488 To test the sensitivity of the epidemiological results to other predictive models, two other
489 regressions were used to predict AA_v and DTT_v : 1) all sources included, 2) only significant
490 sources with positive coefficients (above), but with AMSULF removed. The latter was done
491 because sulfate has substantially decreased over the last decade due to emission reductions in the
492 southeastern US (Hidy et al., 2014; Hand et al., 2012), which may have unknown effects on AA_v^e
493 and DTT_v^e . These models and various statistics are also summarized in Table S3. For both assays
494 the models with all sources included had highest correlations coefficients between model and
495 observed activities ($r \sim 0.7$, or model explains $\sim 50\%$ of the variability).

496 **3.2.2. Health associations from time-series epidemiological models**

497 Backcast AA_v^e and DTT_v^e were next generated for the study period corresponding to the health
498 (ED) data. The various regression models (including Eq.3 and 4) were used to generate daily
499 retrospective estimates of AA_v and DTT_v at the JST site for the period of 1998-2009, based on
500 existing source impacts generated in a previous study for the same site. AA_v^e and DTT_v^e were run
501 separately in epidemiological models of ED visits for selected outcomes (section 2.5.2). The risk
502 ratios for AA_v^e and DTT_v^e for asthma/wheeze and congestive heart failure are presented in Figure
503 5 (data given in Table S4). The other health outcomes (chronic obstructive pulmonary disease,

504 pneumonia, and ischemic heart disease) did not show significant associations with AA_V^e or DTT_V^e
505 (results given in Table S4).

506 For asthma/wheeze and congestive heart failure, although the risk ratios for an increase of an
507 interquartile range for AA_V^e were above 1 [1.005 and 1.003 for Asthma/wheeze and CHF,
508 respectively, Fig. 5(a)], the 95 % confidence intervals crossed 1 (0.994-1.015 and 0.986-1.020
509 for Asthma/wheeze and congestive heart failure, respectively), indicating a non-statistically
510 significant association between AA_V^e and the ED visits for these health outcomes. In contrast,
511 both of the health outcomes showed statistically significant associations with the DTT_V^e . The
512 same results were found for estimates based on the two other regressions [Figure 5(b) and (c)],
513 suggesting that the null relationship of AA_V^e and positive association of DTT_V^e with these health
514 outcomes are to some extent robust, despite the high uncertainties from the back-cast models. A
515 possible cause for the differences in AA_V^e and DTT_V^e health associations is, at least for this study
516 region, the more narrow selectivity of the AA assay to specific aerosol components (i.e., mostly
517 sensitive to Cu). The AA assay may not capture the overall oxidative potential of all the various
518 PM components as well as the DTT assay.

519 PM-induced oxidative stress in the exacerbation of asthma (Li et al., 2003) and the
520 cardiovascular system (Donaldson et al., 2001) has been proposed. Linkages seen here between
521 water-soluble ROS activity and morbidity due to asthma/wheeze and congestive heart failure are
522 consistent with these studies and the contrast with the AA_V results suggest the importance of
523 organic components and transition metals from biomass burning and vehicular emissions in the
524 Southeastern US, and support aerosol particle oxidative potential as a mechanism contributing to
525 these PM-induced adverse health effects. Although some studies have shown that ROS plays a
526 key role in COPD (O'Donnell et al., 2006), IHF (Lakshmi et al., 2009; Giordano, 2005), and

527 Pneumonia (Kuwano et al., 2003), we did not observed a significantly positive association of the
528 tested ROS assays with these health outcomes. Finally, although this work shows a contrast
529 between these two assays and association with health endpoints, Janssen et al. (2015) found
530 significant associations between both assays and nasal and airway inflammation based on a
531 different approach (panel study, n=31).

532 **4. Summary**

533 Approximately 500 PM_{2.5} high-volume filter samples collected in the Southeastern US were
534 analyzed for aerosol oxidative potential using the AA assay. The AA activities reported are from
535 the same filters for which water-soluble DTT activities had already been determined. We found
536 that water-soluble AA activity on a per air volume basis (AAv) was highest near roadways and
537 lowest at rural sites. AAv was higher in summer/fall than winter. These results are in contrast to
538 DTTv, which was more spatially uniform and had an opposite seasonal trend at the urban Atlanta
539 site (higher in winter than summer/fall). AAv was most consistently correlated with water-
540 soluble metals (especially water-soluble Cu), whereas DTTv was correlated with organic species
541 and water-soluble metals (Fe, Cu, Zn, Mn, and Ca), and also PM_{2.5} mass. A source
542 apportionment analysis indicated that traffic emissions and secondary processes were strong
543 contributors to both AAv and DTTv in urban Atlanta. For AAv only road dust was responsible,
544 in contrast to both combustion emissions and road dust contributing to the DTTv from this
545 source. Biomass burning did not contribute to AAv, but was a substantial source for DTTv,
546 consistent with AAv being mainly associated with transition metals. These source apportionment
547 results are also consistent with observed seasonal trends and spatial distributions, for both assays.
548 Time-series large population epidemiological analyses using backcast-estimates of AAv and
549 DTTv from a number of linear models based on 10-year historical source impacts suggest that

550 AAv was not linked with any emergency department (ED) visits for all tested health outcomes at
551 95% confidence intervals. DTTv was associated with ED visits for both asthma/wheeze and
552 congestive heart failure, for all the linear models tested. Neither AAv nor DTTv was associated
553 with chronic obstructive pulmonary disease (COPD), Ischemic heart failure (IHD) or pneumonia
554 at a statistically significant level. Based on the wide-ranging comparisons between these assays,
555 we conclude that, for the region investigated, the DTT assay was a more comprehensive multi-
556 pollutant indicator of PM_{2.5} oxidative potential than the AA assay. This can be useful when
557 deciding on what assay to utilize to address the goal of a specific study. DTT is potentially a
558 more valuable parameter to include in PM health-related studies because of its broader
559 sensitivity to aerosol components associated with oxidative potential. Finally, the ability to
560 readily measure both PM_{2.5} AA and DTT with automated systems enables large scale studies
561 involving direct measurements of PM oxidative potential. These types of future studies are
562 needed to test if our health findings based on backcast-estimated AA and DTT levels are robust
563 and applicable to other regions.

564

565 **Acknowledgement**

566 This publication was made possible by US EPA grant R834799. The contents are solely the
567 responsibility of the grantee and do not necessarily represent the official views of the US EPA.
568 Further, US EPA does not endorse the purchase of any commercial products or services
569 mentioned in the publication. The authors thank Laura King for assistance in collecting samples,
570 R. Erik Weber and Janessa Riana Rowland for assistance in lab work and the SEARCH
571 personnel for their many contributions supporting the field deployments. T. Fang acknowledges
572 the support from the Oversea Study Program of Guangzhou Elite Project.

573 **References**

- 574 Akhtar, U. S., McWhinney, R. D., Rastogi, N., Abbatt, J. P. D., Evans, G. J., and Scott, J. A.: Cytotoxic
575 and proinflammatory effects of ambient and source-related particulate matter (PM) in relation to the
576 production of reactive oxygen species (ROS) and cytokine adsorption by particles, *Inhalation Toxicology*,
577 22, 37-47, 10.3109/08958378.2010.518377, 2010.
- 578 Aust, A., Ball, J., Hu, A., Lighty, J., Smith, K., Straccia, A., Veranth, J., and Young, W.: Particle
579 characteristics responsible for effects on human lung epithelial cells, *Res Rep Health Eff Inst*, 1 - 65,
580 2002.
- 581 Ayres, J. G., Borm, P., Cassee, F. R., Castranova, V., Donaldson, K., Ghio, A., Harrison, R. M., Hider,
582 R., Kelly, F., and Kooter, I. M.: Evaluating the toxicity of airborne particulate matter and nanoparticles by
583 measuring oxidative stress potential-a workshop report and consensus statement, *Inhalation toxicology*,
584 20, 75-99, 2008.
- 585 Balachandran, S., Pachon, J. E., Hu, Y., Lee, D., Mulholland, J. A., and Russell, A. G.: Ensemble-trained
586 source apportionment of fine particulate matter and method uncertainty analysis, *Atmospheric*
587 *Environment*, 61, 387-394, <http://dx.doi.org/10.1016/j.atmosenv.2012.07.031>, 2012.
- 588 Bates, J.T., Weber, R. J., Abrams, J., Verma, V., Fang, T., Klein, M., Strickland, M. J., Sarnat, S. E.,
589 Chang, H. H., Mulholland, J. A., Tolbert, P. E., Russell, A. G.: Reactive Oxygen Species Generation
590 Linked to Sources of Atmospheric Particulate Matter and Cardiorespiratory Effects, *Environmental*
591 *Science & Technology*, in review.
- 592 Birmili, W., Allen, A. G., Bary, F., and Harrison, R. M.: Trace Metal Concentrations and Water Solubility
593 in Size-Fractionated Atmospheric Particles and Influence of Road Traffic, *Environ. Sci. Technol.*, 40,
594 1144-1153, doi:10.1021/es0486925, 2006.
- 595 Brunekreef, B., Janssen, N. A., de Hartog, J., Harssema, H., Knape, M., and van Vliet, P.: Air pollution
596 from truck traffic and lung function in children living near motorways, *Epidemiology*, 8, 298-303,
597 10.1097/00001648-199705000-00012, 1997.
- 598 Burchiel, S. W., Lauer, F. T., Dunaway, S. L., Zawadzki, J., McDonald, J. D., and Reed, M. D.:
599 Hardwood smoke alters murine splenic T cell responses to mitogens following a 6-month whole body
600 inhalation exposure, *Toxicology and Applied Pharmacology*, 202, 229-236,
601 <http://dx.doi.org/10.1016/j.taap.2004.06.024>, 2005.

602 Charrier, J. G., and Anastasio, C.: On dithiothreitol (DTT) as a measure of oxidative potential for ambient
603 particles: evidence for the importance of soluble transition metals, *Atmos. Chem. Phys.*, 12, 9321-9333,
604 10.5194/acp-12-9321-2012, 2012.

605 Cheung, K., Shafer, M. M., Schauer, J. J., and Sioutas, C.: Diurnal trends in oxidative potential of coarse
606 particulate matter in the Los Angeles basin and their relation to sources and chemical composition,
607 *Environmental Science & Technology*, 46, 3779-3787, 10.1021/es204211v, 2012.

608 Cheung, K. L., Ntziachristos, L., Tzamkiozis, T., Schauer, J. J., Samaras, Z., Moore, K. F., and Sioutas,
609 C.: Emissions of particulate trace elements, metals and organic species from gasoline, diesel, and
610 biodiesel passenger vehicles and their relation to oxidative potential, *Aerosol Science and Technology*,
611 44, 500-513, 10.1080/02786821003758294, 2010.

612 Cho, A. K., Sioutas, C., Miguel, A. H., Kumagai, Y., Schmitz, D. A., Singh, M., Eiguren-Fernandez, A.,
613 and Froines, J. R.: Redox activity of airborne particulate matter at different sites in the Los Angeles
614 Basin, *Environmental research*, 99, 40-47, 2005.

615 Chung, M. Y., Lazaro, R. A., Lim, D., Jackson, J., Lyon, J., Rendulic, D., and Hasson, A. S.: Aerosol-
616 borne quinones and reactive oxygen species generation by particulate matter extracts, *Environmental
617 Science & Technology*, 40, 4880-4886, 10.1021/es0515957, 2006.

618 Darrow, L. A., Klein, M., Flanders, W. D., Mulholland, J. A., Tolbert, P. E., and Strickland, M. J.: Air
619 Pollution and Acute Respiratory Infections Among Children 0–4 Years of Age: An 18-Year Time-Series
620 Study, *American Journal of Epidemiology*, 180, 968-977, 10.1093/aje/kwu234, 2014.

621 DiStefano, E., Eiguren-Fernandez, A., Delfino, R. J., Sioutas, C., Froines, J. R., and Cho, A. K.:
622 Determination of metal-based hydroxyl radical generating capacity of ambient and diesel exhaust
623 particles, *Inhal Toxicol*, 21, 731-738, 10.1080/08958370802491433, 2009.

624 Donaldson, K., Stone, V., Seaton, A., and MacNee, W.: Ambient particle inhalation and the
625 cardiovascular system: potential mechanisms, *Environmental Health Perspectives*, 109, 523-527, 2001.

626 Donaldson, K., Tran, L., Jimenez, L., Duffin, R., Newby, D., Mills, N., MacNee, W., and Stone, V.:
627 Combustion-derived nanoparticles: a review of their toxicology following inhalation exposure, *Part Fibre
628 Toxicol*, 2, 10, 10.1186/1743-8977-2-10, 2005.

629 Espinosa, A. J. F., Rodríguez, M. T., de la Rosa, F. J. B., and Sánchez, J. C. J.: A chemical speciation of
630 trace metals for fine urban particles, *Atmos. Environ.*, 36, 773–780, doi:10.1016/S1352-2310(01)00534-
631 9, 2002.

632 Fang, T., Guo, H., Verma, V., Peltier, R. E., and Weber, R. J.: PM_{2.5} water-soluble elements in the
633 southeastern United States: automated analytical method development, spatiotemporal distributions,
634 source apportionment, and implications for health studies, *Atmos. Chem. Phys. Discuss.*, 15, 17189-
635 17227, 10.5194/acpd-15-17189-2015, 2015a.

636 Fang, T., Verma, V., Guo, H., King, L. E., Edgerton, E. S., and Weber, R. J.: A semi-automated system
637 for quantifying the oxidative potential of ambient particles in aqueous extracts using the dithiothreitol
638 (DTT) assay: results from the Southeastern Center for Air Pollution and Epidemiology (SCAPE), *Atmos.
639 Meas. Tech.*, 8, 471-482, 10.5194/amt-8-471-2015, 2015b.

640 Gasser, M., Riediker, M., Mueller, L., Perrenoud, A., Blank, F., Gehr, P., and Rothen-Rutishauser, B.:
641 Toxic effects of brake wear particles on epithelial lung cells in vitro, *Particle and Fibre Toxicology*, 6, 30,
642 10.1186/1743-8977-6-30, 2009.

643 Giordano, F. J.: Oxygen, oxidative stress, hypoxia, and heart failure, *Journal of Clinical Investigation*,
644 115, 500-508, 10.1172/JCI200524408, 2005.

645 Goldman, G. T., Mulholland, J. A., Russell, A. G., Gass, K., Strickland, M. J., and Tolbert, P. E.:
646 Characterization of ambient air pollution measurement error in a time-series health study using a
647 geostatistical simulation approach, *Atmospheric Environment*, 57, 101-108,
648 10.1016/j.atmosenv.2012.04.045, 2012.

649 Hand, J. L., Schichtel, B. A., Malm, W. C., and Pitchford, M. L.: Particulate sulfate ion concentration and
650 SO₂ emission trends in the United States from the early 1990s through 2010, *Atmos. Chem. Phys.*, 12,
651 10353-10365, 10.5194/acp-12-10353-2012, 2012.

652 Hansen, D. A., Edgerton, E. S., Hartsell, B. E., Jansen, J. J., Kandasamy, N., Hidy, G. M., and Blanchard,
653 C. L.: The Southeastern aerosol research and characterization study: Part 1—Overview, *Journal of the Air
654 & Waste Management Association*, 53, 1460-1471, 10.1080/10473289.2003.10466318, 2003.

655 Harkema, J. R., Keeler, G., Wagner, J., Morishita, M., Timm, E., Hotchkiss, J., Marsik, F., Dvonch, T.,
656 Kaminski, N., and Barr, E.: Effects of concentrated ambient particles on normal and hypersecretory
657 airways in rats, *Res Rep Health Eff Inst*, 120, 1-68, 2004.

658 Harrison, R. M., and Yin, J.: Particulate matter in the atmosphere: which particle properties are important
659 for its effects on health?, *Science of The Total Environment*, 249, 85-101,
660 [http://dx.doi.org/10.1016/S0048-9697\(99\)00513-6](http://dx.doi.org/10.1016/S0048-9697(99)00513-6), 2000.

661 Hidy, G. M., Blanchard, C. L., Baumann, K., Edgerton, E., Tanenbaum, S., Shaw, S., Knipping, E.,
662 Tombach, I., Jansen, J., and Walters, J.: Chemical climatology of the southeastern United States, 1999-
663 2013, *Atmos. Chem. Phys.*, 14, 11893-11914, 10.5194/acp-14-11893-2014, 2014.

664 Janssen, N. A. H., Yang, A., Strak, M., Steenhof, M., Hellack, B., Gerlofs-Nijland, M. E., Kuhlbusch, T.,
665 Kelly, F., Harrison, R., Brunekreef, B., Hoek, G., and Cassee, F.: Oxidative potential of particulate matter
666 collected at sites with different source characteristics, *Science of The Total Environment*, 472, 572-581,
667 <http://dx.doi.org/10.1016/j.scitotenv.2013.11.099>, 2014.

668 Janssen, N. A. H., Strak, M., Yang, A., Hellack, B., Kelly, F. J., Kuhlbusch, T. A. J., Harrison, R. M.,
669 Brunekreef, B., Cassee, F. R., Steenhof, M., and Hoek, G.: Associations between three specific a-cellular
670 measures of the oxidative potential of particulate matter and markers of acute airway and nasal
671 inflammation in healthy volunteers, *Occupational and environmental medicine*, 72, 49-56, 2015.

672 Jung, H., Guo, B., Anastasio, C., and Kennedy, I. M.: Quantitative measurements of the generation of
673 hydroxyl radicals by soot particles in a surrogate lung fluid, *Atmos. Environ.*, 40, 1043–1052, 2006.

674 Kam, W., Ning, Z., Shafer, M. M., Schauer, J. J., and Sioutas, C.: Chemical characterization and redox
675 potential of coarse and fine particulate matter (PM) in underground and ground-level rail systems of the
676 Los Angeles metro, *Environmental Science & Technology*, 45, 6769-6776, 10.1021/es201195e, 2011.

677 Kelly, F. J., Mudway, I., Blomberg, A., Frew, A., and Sandström, T.: Altered lung antioxidant status in
678 patients with mild asthma, *The Lancet*, 354, 482-483, [http://dx.doi.org/10.1016/S0140-6736\(99\)01812-7](http://dx.doi.org/10.1016/S0140-6736(99)01812-7),
679 1999.

680 Kleinman, M. T., Sioutas, C., Froines, J. R., Fanning, E., Hamade, A., Mendez, L., Meacher, D., and
681 Oldham, M.: Inhalation of concentrated ambient particulate matter near a heavily trafficked road
682 stimulates antigen-induced airway responses in mice, *Inhalation toxicology*, 19, 117-126,
683 10.1080/08958370701495345, 2007.

684 Kodavanti, U. P., Schladweiler, M. C., Ledbetter, A. D., McGee, J. K., Walsh, L., Gilmour, P. S.,
685 Highfill, J. W., Davies, D., Pinkerton, K. E., Richards, J. H., Crissman, K., Andrews, D., and Costa, D.
686 L.: Consistent pulmonary and systemic responses from inhalation of fine concentrated ambient particles:
687 roles of rat strains used and physicochemical properties, *Environmental Health Perspectives*, 113, 1561-
688 1568, 10.1289/ehp.7868, 2005.

689 Kumagai, Y., Koide, S., Taguchi, K., Endo, A., Nakai, Y., Yoshikawa, T., and Shimojo, N.: Oxidation of
690 proximal protein sulfhydryls by phenanthraquinone, a component of diesel exhaust particles, *Chemical
691 Research in Toxicology*, 15, 483-489, 10.1021/tx0100993, 2002.

692 Künzli, N., Mudway, I. S., Götschi, T., Shi, T., Kelly, F. J., Cook, S., Burney, P., Forsberg, B.,
693 Gauderman, J. W., Hazenkamp, M. E., Heinrich, J., Jarvis, D., Norbäck, D., Payo-Losa, F., Poli, A.,
694 Sunyer, J., and Borm, P. J. A.: Comparison of Oxidative Properties, Light Absorbance, and Total and
695 Elemental Mass Concentration of Ambient PM(2.5) Collected at 20 European Sites, *Environmental
696 Health Perspectives*, 114, 684-690, 10.1289/ehp.8584, 2006.

697 Kuwano, K., Nakashima, N., Inoshima, I., Hagimoto, N., Fujita, M., Yoshimi, M., Maeyama, T.,
698 Hamada, N., Watanabe, K., and Hara, N.: Oxidative stress in lung epithelial cells from patients with
699 idiopathic interstitial pneumonias, *European Respiratory Journal*, 21, 232-240,
700 10.1183/09031936.03.00063203, 2003.

701 Laden, F., Neas, L. M., Dockery, D. W., and Schwartz, J.: Association of fine particulate matter from
702 different sources with daily mortality in six U.S. cities, *Environmental Health Perspectives*, 108, 941-947,
703 10.2307/3435052, 2000.

704 Lakshmi, S. V., Padmaja, G., Kuppasamy, P., and Kutala, V. K.: Oxidative stress in cardiovascular
705 disease, *Indian journal of biochemistry & biophysics*, 46, 421-440, 2009.

706 Landreman, A. P., Shafer, M. M., Hemming, J. C., Hannigan, M. P., and Schauer, J. J.: A macrophage-
707 based method for the assessment of the Reactive Oxygen Species (ROS) activity of atmospheric
708 particulate matter (PM) and application to routine (daily-24 h) aerosol monitoring studies, *Aerosol
709 Science and Technology*, 42, 946-957, 10.1080/02786820802363819, 2008.

710 Li, N., Hao, M., Phalen, R. F., Hinds, W. C., and Nel, A. E.: Particulate air pollutants and asthma: A
711 paradigm for the role of oxidative stress in PM-induced adverse health effects, *Clinical Immunology*, 109,
712 250-265, <http://dx.doi.org/10.1016/j.clim.2003.08.006>, 2003.

713 Lin, P., and Yu, J. Z.: Generation of reactive oxygen species mediated by Humic-like Substances in
714 atmospheric aerosols, *Environmental Science & Technology*, 45, 10362-10368, 10.1021/es2028229,
715 2011.

716 Lundstedt, S., White, P. A., Lemieux, C. L., Lynes, K. D., Lambert, I. B., Öberg, L., Haglund, P., and
717 Tysklind, M.: Sources, fate, and toxic hazards of oxygenated polycyclic aromatic hydrocarbons (PAHs) at
718 PAH- contaminated sites, *AMBIO: A Journal of the Human Environment*, 36, 475-485, 10.1579/0044-
719 7447(2007)36[475:sfatho]2.0.co;2, 2007.

720 McWhinney, R. D., Zhou, S., and Abbatt, J. P. D.: Naphthalene SOA: redox activity and naphthoquinone
721 gas-particle partitioning, *Atmos. Chem. Phys.*, 13, 9731-9744, 10.5194/acp-13-9731-2013, 2013.

722 Meskhidze, N., Chameides, W. L., Nenes, A., and Chen, G.: Iron mobilization in mineral dust: Can
723 anthropogenic SO₂ emissions affect ocean productivity?, *Geophysical Research Letters*, 30 (21),
724 doi:10.1029/2003GL018035, 2003.

725 Metzger, K. B., Tolbert, P. E., Klein, M., Peel, J. L., Flanders, W. D., Todd, K., Mulholland, J. A., Ryan,
726 P. B., and Frumkin, H.: Ambient air pollution and cardiovascular emergency department visits,
727 *Epidemiology*, 15, 46-56, 10.1097/01.EDE.0000101748.28283.97, 2004.

728 Miljevic, B., F. Hedayat, S. Stevanovic, K. E. Fairfull-Smith, S. E. Bottle, and Z. D. Ristovski (2014), To
729 Sonicate or Not to Sonicate PM Filters: Reactive Oxygen Species Generation Upon Ultrasonic Irradiation,
730 *Aerosol Science and Technology*, 48(12), DOI:10.1080/02786826.2014.981330, 1276-1284.

731 Mudway, I., Duggan, S., Venkataraman, C., Habib, G., Kelly, F., and Grigg, J.: Combustion of dried
732 animal dung as biofuel results in the generation of highly redox active fine particulates, *Particle and Fibre
733 Toxicology*, 2, 6, 10.1186/1743-8977-2-6, 2005.

734 Mudway, I. S., Stenfors, N., Duggan, S. T., Roxborough, H., Zielinski, H., Marklund, S. L., Blomberg,
735 A., Frew, A. J., Sandström, T., and Kelly, F. J.: An in vitro and in vivo investigation of the effects of
736 diesel exhaust on human airway lining fluid antioxidants, *Archives of Biochemistry and Biophysics*, 423,
737 200-212, <http://dx.doi.org/10.1016/j.abb.2003.12.018>, 2004.

738 Nel, A.: Air pollution-related illness: effects of particles, *Science*, 308, 804-806,
739 10.1126/science.1108752, 2005.

740 Nel, A. E., Diaz-Sanchez, D., and Li, N.: The role of particulate pollutants in pulmonary inflammation
741 and asthma: evidence for the involvement of organic chemicals and oxidative stress, *Current opinion in
742 pulmonary medicine*, 7, 20-26, 10.1097/00063198-200101000-00004, 2001.

743 O'Donnell, R., Breen, D., Wilson, S., and Djukanovic, R.: Inflammatory cells in the airways in COPD,
744 *Thorax*, 61, 448-454, 10.1136/thx.2004.024463, 2006.

745 Paatero, P., and Tapper, U.: Positive matrix factorization: A non-negative factor model with optimal
746 utilization of error estimates of data values, *Environmetrics*, 5, 111-126, 10.1002/env.3170050203, 1994.

747 Pachon, J. E., Balachandran, S., Hu, Y., Mulholland, J. A., Darrow, L. A., Sarnat, J. A., Tolbert, P. E.,
748 and Russell, A. G.: Development of Outcome-based, Multipollutant Mobile Source Indicators, *Journal of
749 the Air & Waste Management Association*, 62, 431-442, 10.1080/10473289.2012.656218, 2012.

750 Peel, J. L., Tolbert, P. E., Klein, M., Metzger, K. B., Flanders, W. D., Todd, K., Mulholland, J. A., Ryan,
751 P. B., and Frumkin, H.: Ambient air pollution and respiratory emergency department visits,
752 *Epidemiology*, 16, 164-174, 10.1097/01.ede.0000152905.42113.db, 2005.

753 Polissar, A. V., Hopke, P. K., Paatero, P., Malm, W. C., and Sisler, J. F.: Atmospheric aerosol over
754 Alaska: 2. Elemental composition and sources, *Journal of Geophysical Research: Atmospheres*, 103,
755 19045-19057, 10.1029/98JD01212, 1998.

756 Pope, C. A., Burnett, R. T., Thun, M. J., Calle, E. E., Krewski, D., Ito, K., and Thurston, G. D.: Lung
757 cancer, cardiopulmonary mortality, and long-term exposure to fine particulate air pollution, *Journal of the*
758 *American Medical Association*, 287, 1132-1141, 10.1001/jama.287.9.1132, 2002.

759 Pope, C. A., Burnett, R. T., Thurston, G. D., Thun, M. J., Calle, E. E., Krewski, D., and Godleski, J. J.:
760 Cardiovascular mortality and long-term exposure to particulate air pollution: epidemiological evidence of
761 general pathophysiological pathways of disease, *Circulation*, 109, 71-77,
762 10.1161/01.CIR.0000108927.80044.7F, 2004.

763 Roginsky, V. A., Barsukova, T. K., and Stegmann, H. B.: Kinetics of redox interaction between
764 substituted quinones and ascorbate under aerobic conditions, *Chem Biol Interact*, 121, 177-197, 1999.

765 Samet, J. M., Dominici, F., Curriero, F. C., Coursac, I., and Zeger, S. L.: Fine particulate air pollution and
766 mortality in 20 U.S. cities, 1987–1994, *The New England Journal of Medicine*, 343, 1742-1749,
767 doi:10.1056/NEJM200012143432401, 2000.

768 Sarnat, J. A., Marmur, A., Klein, M., Kim, E., Russell, A. G., Sarnat, S. E., Mulholland, J. A., Hopke, P.
769 K., and Tolbert, P. E.: Fine particle sources and cardiorespiratory morbidity: an application of chemical
770 mass balance and factor analytical source-apportionment methods, *Environ Health Perspect*, 116, 459-
771 466, 10.1289/ehp.10873, 2008.

772 Sarnat, S. E., Klein, M., Sarnat, J. A., Flanders, W. D., Waller, L. A., Mulholland, J. A., Russell, A. G.,
773 and Tolbert, P. E.: An examination of exposure measurement error from air pollutant spatial variability in
774 time-series studies, *Journal of exposure science & environmental epidemiology*, 20, 135-146,
775 10.1038/jes.2009.10, 2010.

776 Sauvain, J.-J., Deslarzes, S., and Riediker, M.: Nanoparticle reactivity toward dithiothreitol,
777 *Nanotoxicology*, 2, 121-129, 10.1080/17435390802245716, 2008.

778 Schaumann, F., Borm, P. J. A., Herbrich, A., Knoch, J., Pitz, M., Schins, R. P. F., Luettig, B., Hohlfeld, J.
779 M., Heinrich, J., and Krug, N.: Metal-rich ambient particles (particulate matter_{2.5}) cause airway
780 inflammation in healthy subjects, *American Journal of Respiratory and Critical Care Medicine*, 170, 898-
781 903, 10.1164/rccm.200403-423OC, 2004.

782 Shen, H., and Anastasio, C.: Formation of hydroxyl radical from San Joaquin Valley particles extracted in
783 a cell-free surrogate lung fluid, *Atmospheric chemistry and physics* 11, 9671-9682, 10.5194/acp-11-9671-
784 2011, 2011.

785 Shi, T., Schins, R. P., Knaapen, A. M., Kuhlbusch, T., Pitz, M., Heinrich, J., and Borm, P. J.: Hydroxyl
786 radical generation by electron paramagnetic resonance as a new method to monitor ambient particulate
787 matter composition, *Journal of environmental monitoring : JEM*, 5, 550-556, 10.1039/B303928P, 2003.

788 Strak, M., Janssen, N. A., Godri, K. J., Gosens, I., Mudway, I. S., Cassee, F. R., Lebret, E., Kelly, F. J.,
789 Harrison, R. M., Brunekreef, B., Steenhof, M., and Hoek, G.: Respiratory health effects of airborne
790 particulate matter: the role of particle size, composition, and oxidative potential-the RAPTES project,
791 *Environ Health Perspect*, 120, 1183-1189, 10.1289/ehp.1104389, 2012.

792 Strickland, M. J., Darrow, L. A., Klein, M., Flanders, W. D., Sarnat, J. A., Waller, L. A., Sarnat, S. E.,
793 Mulholland, J. A., and Tolbert, P. E.: Short-term Associations between Ambient Air Pollutants and
794 Pediatric Asthma Emergency Department Visits, *American Journal of Respiratory and Critical Care*
795 *Medicine*, 182, 307-316, 10.1164/rccm.200908-1201OC, 2010.

796 Strickland, M. J., Darrow, L. A., Mulholland, J. A., Klein, M., Flanders, W. D., Winquist, A., and
797 Tolbert, P. E.: Implications of different approaches for characterizing ambient air pollutant concentrations
798 within the urban airshed for time-series studies and health benefits analyses, *Environmental Health*, 10,
799 36, 10.1186/1476-069X-10-36, 2011.

800 Strickland, M. J., Klein, M., Flanders, W. D., Chang, H. H., Mulholland, J. A., Tolbert, P. E., and
801 Darrow, L. A.: Modification of the Effect of Ambient Air Pollution on Pediatric Asthma Emergency
802 Visits: Susceptible Subpopulations, *Epidemiology*, 25(6), 843-850, 10.1097/EDE.000000000000170,
803 2014.

804 Strickland, M. J., Gass, K. M., Goldman, G. T., and Mulholland, J. A.: Effects of ambient air pollution
805 measurement error on health effect estimates in time-series studies: a simulation-based analysis, *J Expos
806 Sci Environ Epidemiol*, 25, 160-166, 10.1038/jes.2013.16, 2015.

807 Tolbert, P. E., Klein, M., Peel, J. L., Sarnat, S. E., and Sarnat, J. A.: Multipollutant modeling issues in a
808 study of ambient air quality and emergency department visits in Atlanta, *Journal of Exposure Science and
809 Environmental Epidemiology*, 17, S29-S35, 10.1038/sj.jes.7500625, 2007.

810 Valko, M., Morris, H., and Cronin, M. T.: Metals, toxicity and oxidative stress, *Current medicinal
811 chemistry*, 12, 1161-1208, 10.2174/0929867053764635 2005.

812 Vejerano, E. P., Ma, Y., Holder, A. L., Pruden, A., Elankumaran, S., and Marr, L. C.: Toxicity of
813 particulate matter from incineration of nanowaste, *Environmental Science: Nano*, 2, 143-154,
814 10.1039/C4EN00182F, 2015.

815 Venkatachari, P., Hopke, P. K., Grover, B. D., and Eatough, D. J.: Measurement of particle-bound
816 reactive oxygen species in rubidoux aerosols, *J. Atmos. Chem.*, 50, 49–58, 2005.

817 Verma, V., Ning, Z., Cho, A. K., Schauer, J. J., Shafer, M. M., and Sioutas, C.: Redox activity of urban
818 quasi-ultrafine particles from primary and secondary sources, *Atmospheric Environment*, 43, 6360-6368,
819 10.1016/j.atmosenv.2009.09.019, 2009.

820 Verma, V., Rico-Martinez, R., Kotra, N., King, L. E., Liu, J., Snell, T. W., and Weber, R. J.: Contribution
821 of water-soluble and insoluble components and their hydrophobic/hydrophilic subfractions to the reactive
822 oxygen species-generating potential of fine ambient aerosols, *Environmental Science & Technology*, 46,
823 11384-11392, 10.1021/es302484r, 2012.

824 Verma, V., Fang, T., Guo, H., King, L. E., Bates, J. T., Peltier, R. E., Edgerton, E. S., Russell, A. G., and
825 Weber, R. J.: Reactive oxygen species associated with water-soluble PM_{2.5} in the southeastern United
826 States: spatiotemporal trends and source apportionment, *Atmos. Chem. Phys.*, 14, 12915-12930,
827 10.5194/acp-14-12915-2014, 2014.

828 Verma, V., Fang, T., Xu, L., Peltier, R. E., Russell, A. G., Ng, N. L., and Weber, R. J.: Organic aerosols
829 associated with the generation of Reactive Oxygen Species (ROS) by water-soluble PM_{2.5},
830 *Environmental Science & Technology*, 49, 4646-4656, 10.1021/es505577w, 2015.

831 Winquist, A., Kirrane, E., Klein, M., Strickland, M., Darrow, L. A., Sarnat, S. E., Gass, K., Mulholland,
832 J., Russell, A., and Tolbert, P.: Joint Effects of Ambient Air Pollutants on Pediatric Asthma Emergency
833 Department Visits in Atlanta, 1998–2004, *Epidemiology*, 25(5), 666-673, doi:
834 10.1097/EDE.000000000000146, 2014.

835 Winquist, A., Schauer, J. J., Turner, J. R., Klein, M., and Sarnat, S. E.: Impact of ambient fine particulate
836 matter carbon measurement methods on observed associations with acute cardiorespiratory morbidity, *J
837 Expos Sci Environ Epidemiol*, 25, 215-221, 10.1038/jes.2014.55, 2015.

838 Yang, A., Jedynska, A., Hellack, B., Kooter, I., Hoek, G., Brunekreef, B., Kuhlbusch, T. A. J., Cassee, F.
839 R., and Janssen, N. A. H.: Measurement of the oxidative potential of PM_{2.5} and its constituents: The
840 effect of extraction solvent and filter type, *Atmospheric Environment*, 83, 35-42,
841 <http://dx.doi.org/10.1016/j.atmosenv.2013.10.049>, 2014.

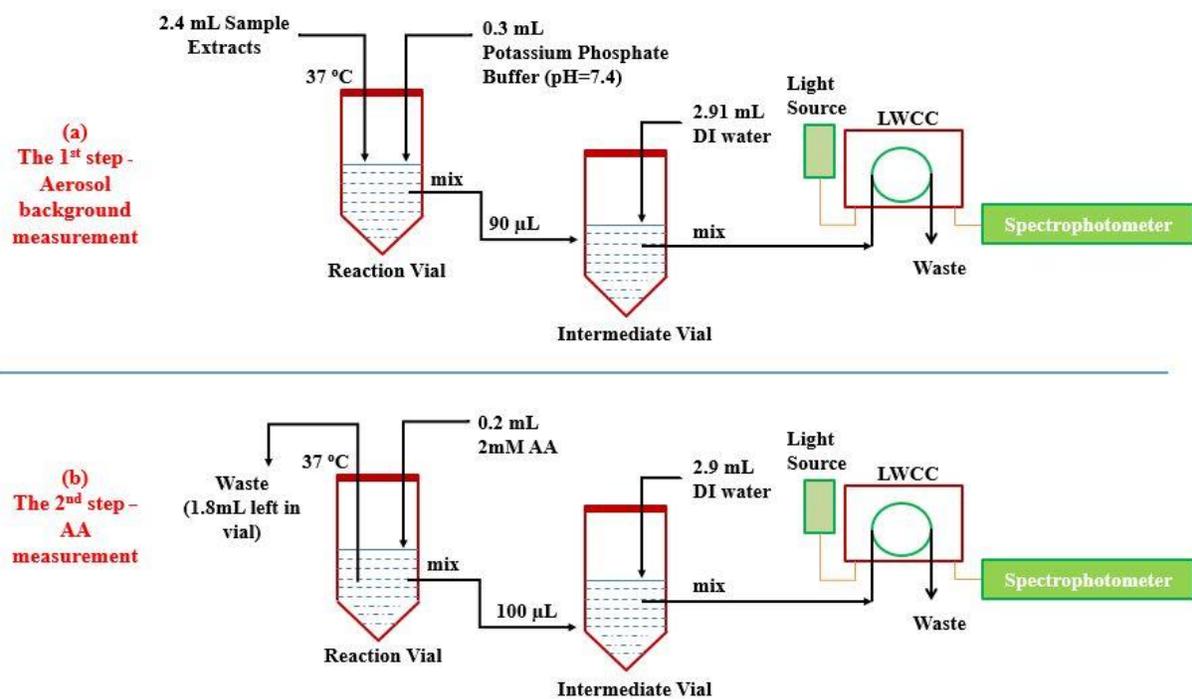
842 Zhang, Y., Schauer, J. J., Shafer, M. M., Hannigan, M. P., and Dutton, S. J.: Source apportionment of in
843 vitro Reactive Oxygen Species bioassay activity from atmospheric particulate matter, *Environmental
844 Science & Technology*, 42, 7502-7509, 10.1021/es800126y, 2008.

845 Zielinski, H., Mudway, I. S., Berube, K. A., Murphy, S., Richards, R., and Kelly, F. J.: Modeling the
846 interactions of particulates with epithelial lining fluid antioxidants, *The American journal of physiology*,
847 277

848 Zomer, B., Collé, L., Jedynska, A., Pasterkamp, G., Kooter, I., and Bloemen, H.: Chemiluminescent
849 reductive acridinium triggering (CRAT) – mechanism and applications, *Anal. Bioanal. Chem.*, 401,
850 2945–2954, 2011., L719-726, 1999.

851

852



853

854 **Figure 1.** Protocol schematics for conducting Ascorbic Acid assay

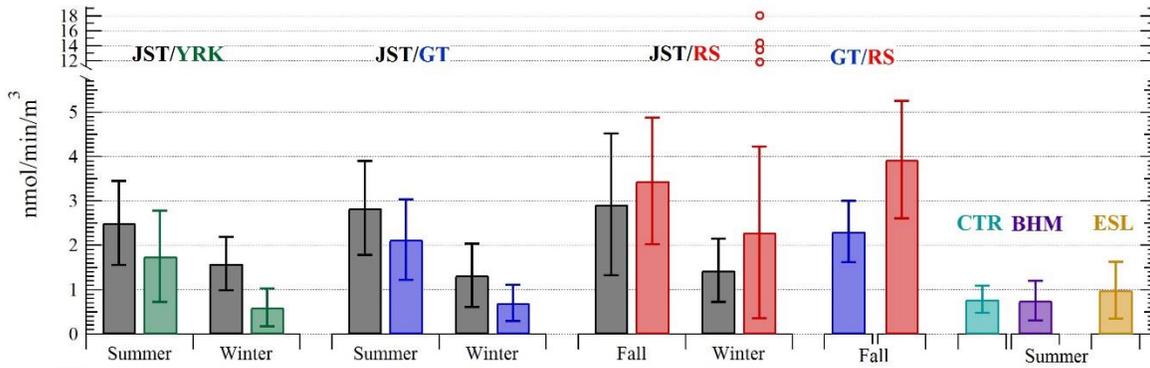
855

856

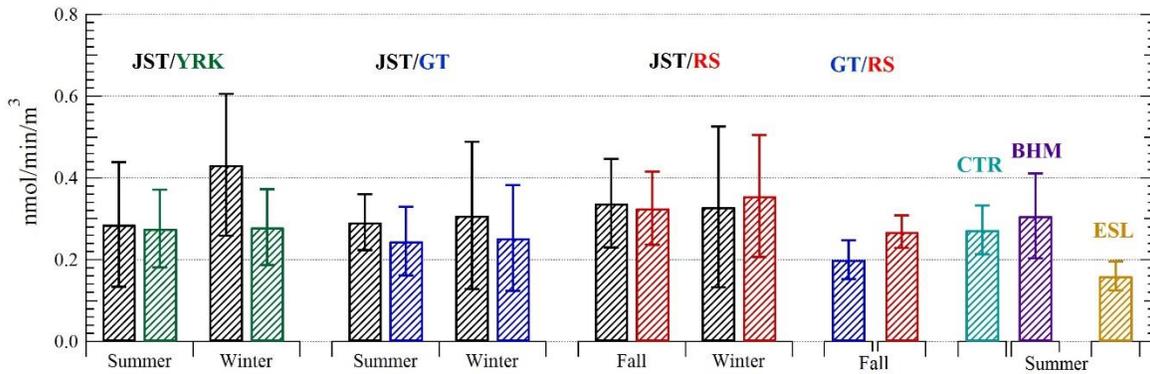
857

858

859 (a) AAv (nmol/min/m³)



860 (b) DTTv (nmol/min/m³)



859

860

861 **Figure 2.** Monthly average (\pm SD) of PM_{2.5} oxidative potential based on the (a) AA and (b) DTT
862 assays from the water-soluble extracts from filters collected at three urban (JST, BHM, and
863 ESL), two rural (YRK and CTR), a near-road (GT), and a road-side (RS) site in the Southeastern
864 United States.

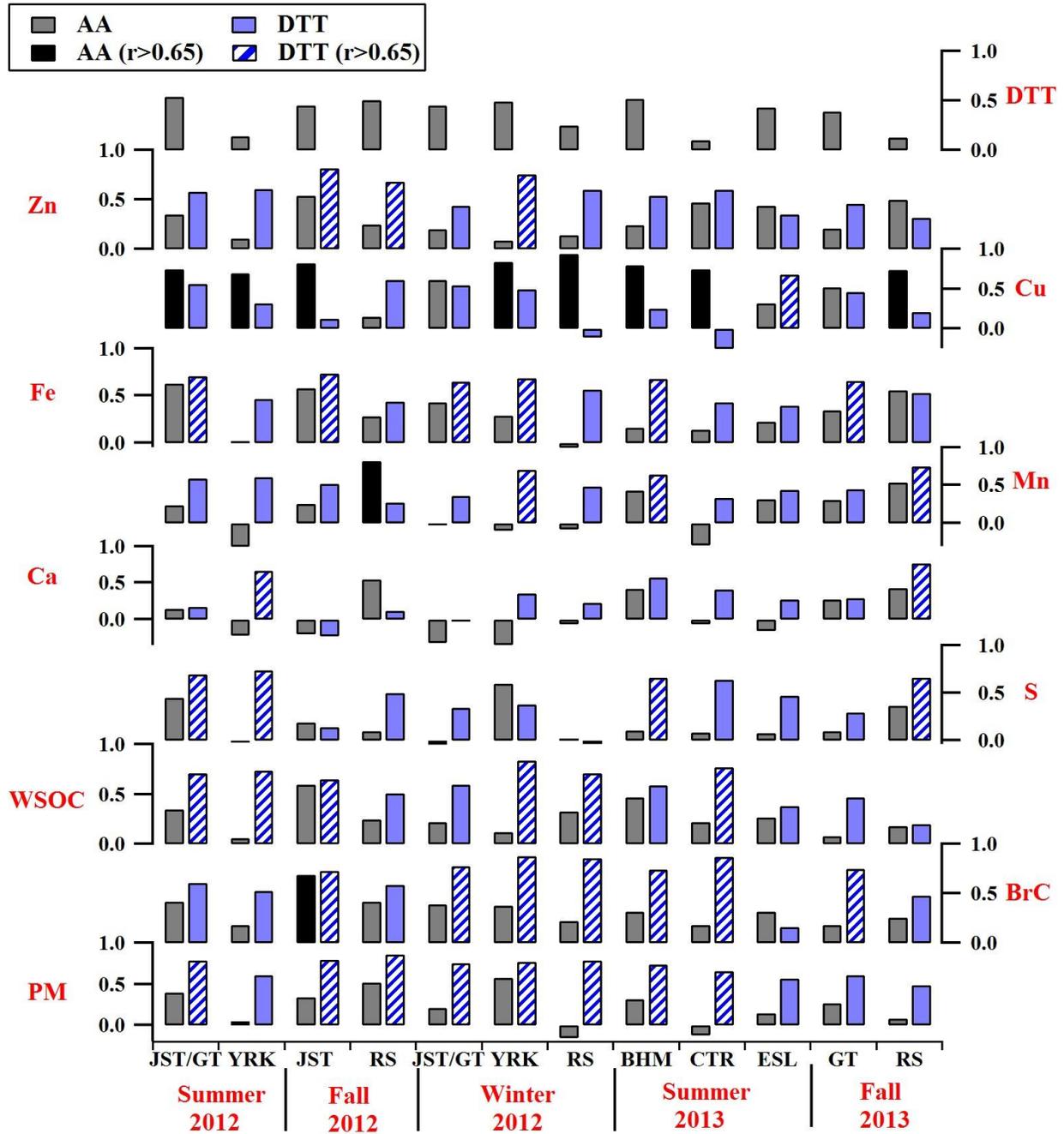
865

866

867

868

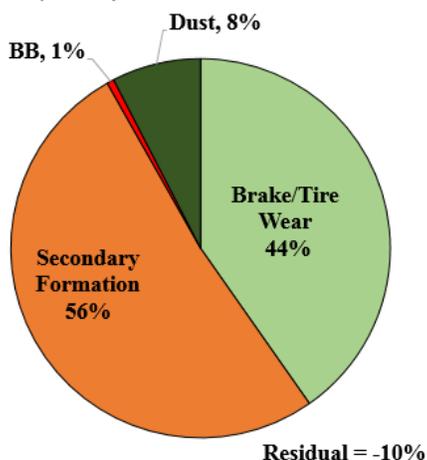
869



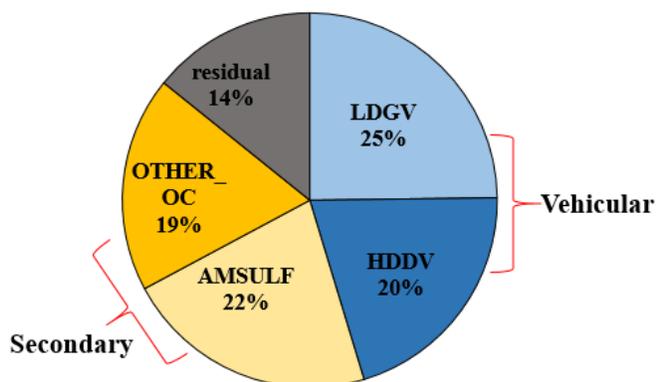
871

872 **Figure 3.** Correlation coefficient (Pearson's r) of fine particle water-soluble AA or DTT
 873 activities with PM_{2.5} mass and selected chemical species at various sites in the Southeastern US.
 874 A more detailed correlation table is provided in Table S2.

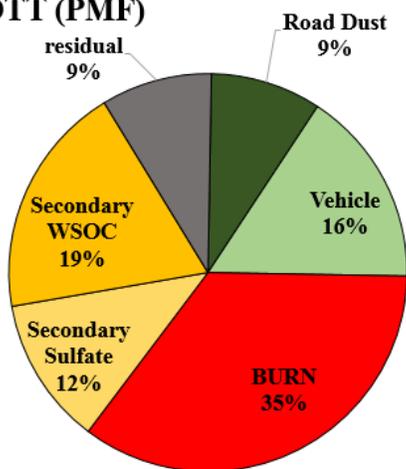
(a) AA (PMF)



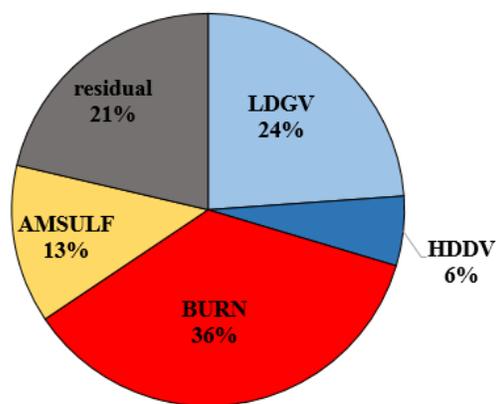
(b) AA (CMB-E)



(c) DTT (PMF)

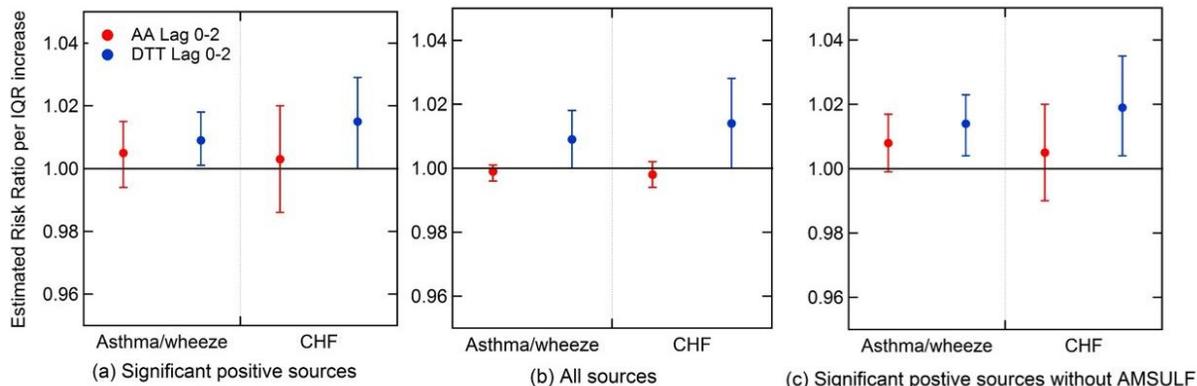


(d) DTT (CMB-E)



875

876 **Figure 4.** Contribution of various factors resolved by PMF (a, c), and ensemble (b, d), to the
877 water-soluble AA (a, b) and DTT (c, d) activities measured during 2012-2013. BURN – biomass
878 burning; AMSULF – ammonium sulfate; HDDV – heavy-duty diesel vehicles; LDGV – light-
879 duty gasoline vehicles; OTHER_OC – other organic carbon which secondary organic aerosols
880 from biogenic emissions, and possible additional contributions from other VOC sources .



881

882 **Figure 5.** Associations between backcast-estimated AA and DTT activities based on estimated
883 sources for the previous 10 years (1998-2009) and emergency department (ED) visits for
884 asthma/wheeze and congestive heart failure (CHF) in the greater metropolitan Atlanta, GA,
885 region. The estimated AA and DTT were based on linear regression models that includes (a) only
886 statistically significant (p of F-statistic of coefficient < 0.05) sources with positive coefficients; (b)
887 all sources; and (c) significant positive sources without AMSULF (ammonium sulfate). The
888 models were generated from a multiple regression of the measured AA activities or DTT, on a
889 per volume air bases, with all sources from CMB-E as independent variables. Risk ratios and
890 associated 95% confidence intervals are presented for an increase of one interquartile range
891 (IQR) increment of the exposure metric. A risk ratio with 95% confidence intervals (CI) for
892 interquartile range above 1 indicates a statistically significant positive association. Risk ratio data
893 and related statistics can be found in Table S4.