



Chlorine isotope composition

S. J. Allin et al.

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# Chlorine isotope composition in chlorofluorocarbons CFC-11, CFC-12 and CFC-113 in firn, stratospheric and tropospheric air

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## Abstract

The stratospheric degradation of chlorofluorocarbons (CFCs) releases chlorine, which is a major contributor to the destruction of stratospheric ozone ( $O_3$ ). A recent study reported strong chlorine isotope fractionation during the breakdown of the most abundant CFC (CFC-12,  $CCl_2F_2$ ), similar to effects seen in nitrous oxide ( $N_2O$ ). Using air archives to obtain a long-term record of chlorine isotope ratios in CFCs could help to identify and quantify their sources and sinks. We analyse the three most abundant CFCs and show that CFC-11 ( $CCl_3F$ ) and CFC-113 ( $CClF_2CCl_2F$ ) exhibit significant stratospheric chlorine isotope fractionation, in common with CFC-12. The apparent isotope fractionation ( $\epsilon_{app}$ ) for mid- and high-latitude stratospheric samples are  $(-2.4 \pm 0.5)$  and  $(-2.3 \pm 0.4)$  ‰ for CFC-11,  $(-12.2 \pm 1.6)$  and  $(-6.8 \pm 0.8)$  ‰ for CFC-12 and  $(-3.5 \pm 1.5)$  and  $(-3.3 \pm 1.2)$  ‰ for CFC-113, respectively. Assuming a constant source isotope composition, we estimate the expected trends in the tropospheric isotope signature of these gases due to their stratospheric  $^{37}Cl$  enrichment and stratosphere–troposphere exchange. We compare these model results to the long-term  $\delta(^{37}Cl)$  trends of all three CFCs, measured on background tropospheric samples from the Cape Grim air archive (Tasmania, 1978–2010) and tropospheric firn air samples from Greenland (NEEM site) and Antarctica (Fletcher Promontory site). Model trends agree with tropospheric measurements within analytical uncertainties. From 1970 to the present-day, we find no evidence for variations in chlorine isotope ratios associated with changes in CFC manufacturing processes. Our study increases the suite of trace gases amenable to direct isotope ratio measurements in small air volumes, using a single-detector gas chromatography-mass spectrometry system.

## 1 Introduction

CFC-11 ( $CCl_3F$ ), CFC-12 ( $CCl_2F_2$ ) and CFC-113 ( $CClF_2CCl_2F$ ) are the most abundant chlorofluorocarbons (CFCs) in the atmosphere (Table 1). They are very stable and are

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not flammable or toxic, making them suitable as refrigerants, aerosol propellants and foam blowing agents. Measurements performed on air from porous firn (the top part of the polar ice sheet consisting of consolidated snow, progressively transforming into ice), show that they were absent from the atmosphere before the middle of the 20th century (e.g. Butler et al., 1999; Sturrock et al., 2002; Buizert et al., 2012). This suggests that they have no natural sources. CFC-11, CFC-12 and CFC-113 first appeared in the mid-1900s (Sturrock et al., 2002; Martinerie et al., 2009), which coincides with the start of their widespread anthropogenic production and consumption.

Atmospheric removal occurs in the stratosphere, where sink processes are dominated by photo-dissociation (Reaction R1) and reaction with O(<sup>1</sup>D) (Reaction R2), e.g.



Cl and ClO released in these reactions catalyse the destruction of O<sub>3</sub> (Molina and Rowland, 1974). Following the identification of a strong decrease in O<sub>3</sub> over Antarctica (Farman et al., 1985), swift international action worked to phase-out the production and consumption of CFCs under the Montreal Protocol and its subsequent amendments.

Measuring isotope ratios in atmospheric trace gases provides additional information regarding their sources and sinks, as well as the ways in which they contribute to biogeochemical cycles (Brenninkmeijer et al., 2003; Goldstein and Shaw, 2003). Long-term studies have investigated a range of species, for example, CH<sub>4</sub> (Etheridge et al., 1998), CO<sub>2</sub> (Francey et al., 1999; Rubino et al., 2013) and N<sub>2</sub>O (Röckmann et al., 2003; Kaiser et al., 2006).

Of particular relevance to this work is N<sub>2</sub>O, a much studied gas with similar sinks and a comparable lifetime to CFC-12. Various studies have used firn and ice core samples (Sowers et al., 2002; Röckmann et al., 2003; Bernard et al., 2006; Ishijima et al., 2007), as well as direct atmospheric samples (Yoshida and Toyoda, 2000; Kaiser et al., 2006) to investigate large scale N<sub>2</sub>O isotope and abundance changes. Source inputs depleted in <sup>15</sup>N and <sup>18</sup>O are thought to have caused the concurrent tropospheric



a relative  $^{37}\text{Cl}$ -enrichment of tropospheric CFC-12 with respect to the average source isotope ratio through stratosphere–troposphere air exchange, similar to observations made on other gases, such as  $\text{CH}_4$  (McCarthy et al., 2001),  $\text{H}_2$  (Batenburg et al., 2012) and  $\text{N}_2\text{O}$  (Röckmann et al., 2003).

The strong stratospheric chlorine isotope fractionation of CFC-12 (Laube et al., 2010a) and the substantial change inferred from  $\delta(^{13}\text{C})$  measurements (Zuiderweg et al., 2013), motivated us to conduct a more complete investigation of the chlorine isotope history of this species. This type of study has the potential to better constrain global sources and sinks (Röckmann et al., 2003). Here we use new measurements on stratospheric samples to quantify the apparent stratospheric  $^{37}\text{Cl}/^{35}\text{Cl}$  isotope fractionation ( $\varepsilon_{\text{app}}$ ) of chlorine not only in CFC-12, but also in CFC-11 and CFC-113. Alongside these values, stratosphere–troposphere exchange estimates are used to predict the tropospheric chlorine isotope history of these species. We then present new tropospheric measurements on samples from the Cape Grim air archive, as well as firn air from NEEM and Fletcher Promontory. These are compared to the model projections, leading to an evaluation of long-term chlorine isotope ratio changes in the three most abundant CFCs.

## 2 Methodology

Table 2 gives details of the samples used in this study. Mid- and high-latitude stratospheric air was collected on an M55 Geophysica research aircraft, as part of the European Union's FP7 project RECONCILE (von Hobe et al., 2013). Background tropospheric samples were sourced from Cape Grim (Tasmania) and firn air was sampled at the North Greenland Eemian Ice Drilling (NEEM) field campaign and Fletcher Promontory.

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## 2.1 Sample preparation and analysis

A gas chromatography-mass spectrometry (GC-MS) system, designed to make high precision measurements of pmol mol<sup>-1</sup> level compounds in small air samples, was used for this work. A pre-concentration system with an Agilent 6890 GC was coupled to a VG/Waters EBE tri-sector mass spectrometer. Details of previous work conducted using this system can be found in Laube et al. (2010b) and Sturges et al. (2012).

Magnesium perchlorate (Mg(ClO<sub>4</sub>)<sub>2</sub>) was used to dry the air introduced to the GC-MS system. Trace gases from 0.2 L (293 K, 1 bar) of air were pre-concentrated on Hayesep D (80/100 mesh), which was cooled to -78 °C by immersion in a dry ice/ethanol mixture. The sample was then trapped on a 49 m GS-GasPro column (internal diameter 0.32 mm) at -10 °C for 2 min. Heating to 200 °C at 10 °C min<sup>-1</sup> released the trapped compounds to the MS according to their retention strength.

The MS was operated in EI-SIR (Electron Impact-Selected Ion Recording) mode at a mass resolution of 1000. The fragment ions measured were C(<sup>35</sup>Cl)<sub>2</sub>F<sup>+</sup> (*m/z* 101), C<sup>35</sup>Cl<sup>37</sup>ClF<sup>+</sup> (*m/z* 103) and C(<sup>37</sup>Cl)<sub>2</sub>F<sup>+</sup> (*m/z* 105).

## 2.2 Data processing

In this study, isotope deltas ( $\delta$ ), expressed in per mill (‰), are used to denote the relative <sup>37</sup>Cl/<sup>35</sup>Cl ratio difference of CFCs in sample air with respect to a standard

$$\delta(^{37}\text{Cl}) = \frac{R_{\text{sample}}}{R_{\text{standard}}} - 1 \quad (1)$$

where *R* represents the <sup>37</sup>Cl/<sup>35</sup>Cl abundance ratio of a standard or sample. There is no internationally recognised isotope standard for these measurements. For this reason, all  $\delta$  values in this study are relative to a laboratory standard (AAL-071170). This is a background air sample collected in 2006 at Niwot Ridge, Colorado, by the National Oceanic and Atmospheric Administration (NOAA).

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This study considers isotopologue current ratios of the  $\text{CCl}_2\text{F}^+$  fragment, rather than directly measuring  $\text{Cl}^+$  ions. Equation (2) describes how the  $\text{C}({}^{35}\text{Cl})_2\text{F}^+$  ( $m/z$  101),  $\text{C}({}^{35}\text{Cl}{}^{37}\text{Cl})\text{F}^+$  ( $m/z$  103) and  $\text{C}({}^{37}\text{Cl})_2\text{F}^+$  ( $m/z$  105) fragments can be used to calculate the  ${}^{37}\text{Cl}/{}^{35}\text{Cl}$  isotope ratio from the  $\text{CCl}_2\text{F}$  measurements

$$R = \frac{A({}^{37}\text{Cl})}{A({}^{35}\text{Cl})} = \frac{\frac{A(\text{C}({}^{35}\text{Cl}{}^{37}\text{Cl})\text{F}^+)}{A(\text{C}({}^{35}\text{Cl})_2\text{F}^+)} + 2 \frac{A(\text{C}({}^{37}\text{Cl})_2\text{F}^+)}{A(\text{C}({}^{35}\text{Cl})_2\text{F}^+)}}{2 + \frac{A(\text{C}({}^{35}\text{Cl}{}^{37}\text{Cl})\text{F}^+)}{A(\text{C}({}^{35}\text{Cl})_2\text{F}^+)}} \quad (2)$$

where  $A$  represents the abundance of a given fragment. The  $\text{C}({}^{37}\text{Cl})_2\text{F}^+$  fragment was included in the analysis of 172 out of 448 samples. In all cases, delta values calculated using ratios derived from Eq. (2), agreed within  $1\sigma$  with delta values calculated directly from  $\text{C}({}^{35}\text{Cl}{}^{37}\text{Cl})\text{F}^+/\text{C}({}^{35}\text{Cl})_2\text{F}^+$  ratios (without the inclusion of the  $\text{C}({}^{37}\text{Cl})_2\text{F}^+$  fragment).

We use the linear regression slope of the  $\text{C}({}^{35}\text{Cl}{}^{37}\text{Cl})\text{F}^+$  against the  $\text{C}({}^{35}\text{Cl})_2\text{F}^+$  ion current over the duration of the peak to derive  $R$  and, based on this,  $\delta$  (Eq. 1). All sample measurements were bracketed by standards, to correct for instrumental drift.

No chromatographic interferences were found for these ions at the retention times corresponding to CFC-11 and CFC-12. However, CFC-113a has grown to a current atmospheric abundance of  $\sim 0.5 \text{ pmol mol}^{-1}$  since its emergence in the 1960s (Laube et al., 2014) and partially co-elutes with CFC-113. Interference from CFC-113a can be detected above the surrounding noise in one sample. This peak was manually excluded from the linear regression slope and a delta value calculated using Eq. (1). Within  $1\sigma$ , this value is the same as the original delta value calculated, without removing the CFC-113a interference.

Poor quality measurements were defined as a measurement with an  $R_{\text{sample}}$  SD (from repeat measurements) of more than three times the average SD of the  $R_{\text{standard}}$  values during an analysis period. Less than 1% of the measurements were rejected as poor quality. Measurements were also rejected if there was uncertainty regarding the integrity of the sample; small system leaks can be identified through the analysis



only occurs in the stratosphere

$$n_T \frac{dy_T}{dt} = P - F_{TS} + F_{ST} = P - F(y_T - y_S) \quad (4)$$

$$n_S \frac{dy_S}{dt} = F_{TS} - F_{ST} - L = F(y_T - y_S) - J n_S y_S. \quad (5)$$

The symbols have the following meaning:

5  $n$ : amount of air (in mol)

$y$ : mole fraction (in mol mol<sup>-1</sup>)

$F$ : trace gas flux (in mola<sup>-1</sup>)

$P$ : trace gas production (in mola<sup>-1</sup>)

$L$ : trace gas loss (in mola<sup>-1</sup>)

10  $J$ : loss rate coefficient (in a<sup>-1</sup>)

The global average lifetime is calculated as the ratio of total burden and total loss

$$\tau = \frac{n_S y_S + n_T y_T}{J n_S y_S} = \frac{n_S y_S + n_T y_T}{F(y_T - y_S) - n_S \frac{dy_S}{dt}}. \quad (6)$$

15 The parameters  $n_T$ ,  $n_S$  and  $F$  are taken from Holton (1990), Trenberth and Guillemot (1994) and Appenzeller et al. (1996). Using the tropospheric mole fraction history  $y_T(t)$  (Velders and Daniel, 2014) and an estimate of the global average lifetime  $\tau$ , Eqs. (4), (5) and (6) can be solved for  $y_S$ ,  $J$  and  $P$ . In practice, we adjust  $J$  so that the average  $\tau$  for the years 1999 to 2008 equals the lifetime estimates as per SPARC (2013) and use a fixed value of  $J$  for all model years.

To derive the corresponding isotope budget, we write Eqs. (4) and (5) for another isotopologue (primed quantities)

$$20 n_T \frac{dy'_T}{dt} = P' - F(y'_T - y'_S) \quad (7)$$

which can be written in  $\delta$  notation (relative to an arbitrary reference material) as

$$n_T \frac{d[y_T(1 + \delta_T)]}{dt} = P(1 + \delta_P) - F[y_T(1 + \delta_T) - y_S(1 + \delta_S)]. \quad (8)$$

Together with Eq. (4) this gives the tropospheric isotope budget

$$n_T y_T \frac{d\delta_T}{dt} = P(\delta_P - \delta_T) + F y_S (\delta_S - \delta_T). \quad (9)$$

5 The equivalent equation for the stratosphere is

$$n_S y_S \frac{d\delta_S}{dt} = F y_T (\delta_T - \delta_S) - n_S y_S J \varepsilon_J (1 + \delta_S) \quad (10)$$

where the photochemical isotope fractionation is

$$\varepsilon_J = \frac{J'}{J} - 1 \quad (11)$$

with  $J'$  being the loss rate coefficient for the isotope in the numerator of the isotope ratio (here,  $^{37}\text{Cl}$ ).

10  $\varepsilon_J$  (or  $J'$ ) are not known, so again following the analogue of  $\text{N}_2\text{O}$  (Röckmann et al., 2003), we use the observed apparent Rayleigh isotope fractionation in the stratosphere ( $\varepsilon_{\text{app}}$ ), to solve the isotope-specific part of the budget. Based on this apparent Rayleigh isotope fractionation, the relative isotope ratio difference between the stratosphere and  
 15 the troposphere  $\delta_{\text{ST}}$  can be calculated according to

$$\delta_{\text{ST}} = \frac{1 + \delta_S}{1 + \delta_T} - 1 = \frac{\delta_S - \delta_T}{1 + \delta_T} = \frac{-\delta_{\text{TS}}}{1 + \delta_{\text{TS}}} = \left(\frac{y_S}{y_T}\right)^{\varepsilon_{\text{app}}} - 1. \quad (12)$$

We adjust  $\varepsilon_J$  so that the average  $\varepsilon_{\text{app}}$  value for the years 1999 to 2008 equals our stratospheric observations and derive the projected atmospheric history of  $\delta_T$  and  $\delta_S$  from Eqs. (10) and (12), assuming a constant  $\delta_P$  value that results in  $\delta_T$  (2006) = 0, corresponding to the year our tropospheric air standard was filled.

## 2.4 Trace gas transport in firn

The migration of gases from the atmosphere through firn is largely controlled by diffusional and gravitational effects, such that: (1) less diffusive gases move through the firn more slowly than more diffusive gases, (2) heavier gases and isotopologues are enriched at depth relative to lighter gases and isotopologues. Consequently: (1) there is no unique “age-of-air” at a given depth in the firn, rather it is specific to an individual gas or isotopologue, (2) a gas can undergo significant isotope fractionation during its movement through the firn. A model of gas transport in firn must therefore include both diffusive transport and gravitational separation to reconstruct changes in atmospheric abundances and isotope ratios over time. These reconstructions have been accomplished here using the model developed at the Laboratoire de Glaciologie et Géophysique de l'Environnement (LGGE) and Grenoble Image Parole Signal Automatique (GISPA-lab). The physical basis of the model is described in Witrant et al. (2012). The model has been further refined to provide a new and more robust definition of the optimal solution, and a more rigorous treatment of isotopes based on a forward firn model written in terms of delta values (Witrant and Martinerie, 2013). The diffusive behaviour of gases is initialised for each firn column by using a set of gases with well-known atmospheric histories (Witrant et al., 2012).

More details on the firn modelling are given in the Supplement. The dates shown here for the firn data are based on median ages for each gas. The age spectrum associated with these median ages is also discussed in the Supplement. When plotting abundances and isotope ratios against median age, it is necessary to correct for gravitational and diffusive fractionation. In the case of abundances, gravitational settling is calculated as the difference between normal model runs and runs in which gravity is set to zero. For within-firn isotope fractionation, a correction is calculated using the reconstructed scenario of the gas (based on its major isotopologue) and assuming a constant atmospheric isotope delta. Changes in fractionation with depth therefore

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reflect firn fractionation processes alone; any discrepancies between projected and measured firn isotope deltas reflect changes in atmospheric isotope deltas over time.

Thermal diffusion was not incorporated in the model, despite an observed gradient of 2 to 3°C in the firn column at NEEM, due to recent local warming in Northwest Greenland (Carr et al., 2013). This is because although the thermal diffusion effect on the CFC mole fractions is on the order of 0.2% at NEEM, the effect on  $\delta(^{37}\text{Cl})$  is estimated to be less than 0.02‰ (Leuenberger and Lang, 2002).

### 3 Results and discussion

#### 3.1 Stratospheric data

After outlier correction (Sect. 2.2), a total of 31 and 28 (CFC-11), 33 and 54 (CFC-12) and 36 and 54 (CFC-113) measurements remain, for samples collected in the mid- and high-latitude stratosphere, respectively.

Negative apparent isotope fractionations ( $\epsilon_{\text{app}}$ ) have been calculated for all three species (Fig. 1). This indicates that sink processes prefer isotopically lighter molecules, causing the observed enrichment in  $^{37}\text{Cl}$ . Trend lines in Fig. 1 correspond to apparent fractionations ( $\epsilon_{\text{app}}$ ) of  $(-2.4 \pm 0.5)$  and  $(-2.3 \pm 0.4)$ ‰ (CFC-11),  $(-12.2 \pm 1.6)$  and  $(-6.8 \pm 0.8)$ ‰ (CFC-12) and  $(-3.5 \pm 1.5)$  and  $(-3.3 \pm 1.2)$ ‰ (CFC-113) for mid- and high-latitudes, respectively. As the sampling point moves from mid- to high-latitudes, the absolute CFC-12  $\epsilon_{\text{app}}$  value decreases. This effect is qualitatively the same for  $\text{N}_2\text{O}$  (Kaiser et al., 2006). CFC-11 and CFC-113 both show mid- and high-latitude  $\epsilon_{\text{app}}$  values that are indistinguishable within 1 SD. It could be that the latitude-dependence seen in CFC-12 and  $\text{N}_2\text{O}$  also exists in these compounds, but is being obscured by the size of analytical uncertainties, compared to the relatively small isotope changes being tracked.

Laube et al. (2010a) calculated the tropical CFC-12  $\epsilon_{\text{app}}$  to be  $(-12.1 \pm 1.7)$ ‰, which agrees with the mid-latitude value of  $(-12.2 \pm 1.5)$ ‰ presented here. Given the lower

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$\varepsilon_{\text{app}}$  value calculated from high-latitude samples, a difference between  $\varepsilon_{\text{app}}$  values at mid-latitudes and the tropics might be expected. The relative difference between mid-latitude/tropical and polar  $\varepsilon_{\text{app}}$  is much larger for CFC-12 than for  $\text{N}_2\text{O}$ . Due to the similar atmospheric lifetimes of CFC-12 and  $\text{N}_2\text{O}$ , this is unlikely to be due to dynamic effects as suggested for  $\text{N}_2\text{O}$  (Kaiser et al., 2006) and is at present unexplained. In the absence of globally representative  $\varepsilon_{\text{app}}$  values, we use the mid-latitude values to project the tropospheric isotope delta changes over time.

### 3.2 Emissions and transport modelling

Based on historical and predicted atmospheric mole fractions (Velders and Daniel, 2014), our 2-box approach estimates global emissions for CFC-11, CFC-12 and CFC-113 from 1950 to 2050. Figure 2 shows good agreement between these values and previously published estimates (Velders and Daniel, 2014). The proliferation of CFC production and consumption caused an initial sharp increase in emissions, followed by sustained high emissions during the 1970s and 1980s. The Montreal Protocol and its subsequent amendments ensured a significant reduction in CFC production, causing the concurrent decrease in emissions of all three species at the end of the 1980s.

Figure 3 shows how the tropospheric  $\delta(^{37}\text{Cl})$  of these three compounds would have changed with time, assuming no source variations. From their first release until the present-day, tropospheric CFC-11, CFC-12 and CFC-113  $\delta(^{37}\text{Cl})$  values are predicted to have increased by 1, 3 and 1 ‰, respectively. Maximum rates of increase are predicted to have occurred since 1990, which coincide with the sharp reduction in CFC emissions due to the introduction of legislation phasing out their production and consumption (Montzka et al., 2011). A drop in the anthropogenic input means that the existing atmospheric pool will become increasingly enriched, as demonstrated by the predicted acceleration of tropospheric  $\delta(^{37}\text{Cl})$  increase after 1990.

The shading in Fig. 3 represents the uncertainty envelope. Approximately equal contributions to the uncertainty arise from: (1) the differing reference interfaces for stratosphere–troposphere exchange used by Holton (1990) and Appenzeller

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et al. (1996), namely the 100 mbar and the 380 K isentrope, (2) the  $1\sigma$  uncertainty in  $\varepsilon_{\text{app}}$  values, (3) the range of “most likely” lifetime values from SPARC (2013).

### 3.3 Tropospheric data

Using the GC-MS instrument discussed in this work, Laube et al. (2010a) verified that their isotope ratio measurements were not biased by the response behaviour of the analytical system. They measured CFC-12 to a minimum of 77 pmol mol<sup>-1</sup>. This study includes two additional species and covers a wider range of sample mole fractions, meaning that a more extensive effort was necessary to remove the possibility of a bias in this work (details in Supplement).

From three independent data sets, a total of 44 (CFC-11), 74 (CFC-12) and 48 (CFC-113)  $\delta(^{37}\text{Cl})$  measurements are presented, covering the last 60, 50 and 45 years, respectively (Fig. 3). In general, there is good agreement between the Cape Grim and firn air measurements. This agreement is in line with expectations from well-mixed gases such as these. Before 1970, there are 2 (CFC-11), 1 (CFC-12) and 2 (CFC-113) measurements slightly higher than expected, which could represent source  $\delta(^{37}\text{Cl})$  changes. Emissions of these gases earlier in the record represent a larger proportion of their total atmospheric load. Our measurements are more sensitive to source  $\delta(^{37}\text{Cl})$  changes before 1970. However, a small correction was applied to 4 out of the 5 samples (details in Supplement) and these measurements were made on the lowest abundance samples leading to larger than mean analytical uncertainties. The limited data available before 1970 make it premature at this stage to report a source  $\delta(^{37}\text{Cl})$  change in these gases.

The predicted trends for all three species are small (Fig. 3), with the largest isotope changes expected to have occurred in the last two decades. From 1970 to the present-day, the modelled changes can be accounted for within analytical uncertainties (Fig. 3). This means that for these three CFCs, an isotopically constant source signature and the current understanding of transport and destruction processes are consistent with our observations in this period. Despite the noted parallels in atmospheric chemical

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behaviour between CFC-12 and  $\text{N}_2\text{O}$ , their atmospheric isotope delta histories differ because  $\text{N}_2\text{O}$  has been present in the atmosphere for at least 800 000 years longer than CFC-12 (Spahni et al., 2005; Schilt et al., 2010). When anthropogenic emissions began, atmospheric  $\text{N}_2\text{O}$  isotope deltas were the result of well-established release and removal processes, whereas a balanced system did not precede the industrial release of CFC-12. Also, input from  $^{15}\text{N}$ - and  $^{18}\text{O}$ -depleted sources have led to a decrease in  $\delta(^{15}\text{N}, \text{N}_2\text{O})$  and  $\delta(^{18}\text{O}, \text{N}_2\text{O})$  over the last century (Röckmann et al., 2003), while our results suggest that the  $\delta(^{37}\text{Cl}, \text{CFC-12})$  source has remained constant, at least since 1970.

Zuiderweg et al. (2013) cite changes in production processes as the most likely cause of the significant  $^{13}\text{C}$  enrichment of CFC-12. However, as they concede, there is no direct evidence that the reported methodology changes had any effect on the isotope signature of CFC-12. If variations in production processes did cause such a large change in the carbon isotope signature, it appears that the chlorine isotope signature either changed by a much smaller amount, or was unaffected.

## 4 Conclusions

Samples from the Cape Grim air archive (Tasmania, 1978–2010) and firn air samples from NEEM (Greenland) and Fletcher Promontory (Antarctica) have been used to infer an atmospheric history of chlorine isotopes in the three most abundant CFCs. These measurements cover the last 60 (CFC-11), 50 (CFC-12) and 45 (CFC-113) years. All three species are isotopically enriched in the stratosphere by destruction processes, leading to negative apparent isotope fractionations ( $\varepsilon_{\text{app}}$ ). This measured stratospheric isotope dependence and an assumed constant source  $\delta(^{37}\text{Cl})$  were used to reconstruct the long-term changes in their tropospheric isotope signatures. The predicted trends are small due to the long atmospheric lifetimes of the species and can largely be accounted for in our tropospheric measurements. Across all species, 5 pre-1970  $\delta(^{37}\text{Cl})$

values are higher than predicted. A source  $\delta(^{37}\text{Cl})$  change could explain this offset, but at present the available data are too limited in number and precision to confirm this.

Laube et al. (2010a) suggested that a more complete understanding of CFC-12 chlorine isotope changes could lead to better  $\text{O}_3$  recovery predictions and the identification of individual sources from their isotope signatures. Given the significant CFC-12 carbon isotope changes reported by Zuiderweg et al. (2013), it seemed likely that a characterisation of CFC chlorine isotopes would help to enable these goals. However, this study reveals only small long-term changes, with current analytical precisions not allowing for such an analysis. Measuring CFCs at their point of release (e.g. old refrigeration units) would help confirm whether there have been any source-driven chlorine isotope changes.

The instrument used in this work has the capacity to measure chlorine isotope ratios in other trace gases (e.g. CFC-115). Also, less abundant fragment ions can be measured (e.g. fragments containing  $^{13}\text{C}$ ), extending this pool of isotopic information to stable carbon isotopes. Finally, the high sensitivity of this instrument opens up the possibility of isotopically analysing the small air samples extracted from ice cores. Therefore, gases with both anthropogenic and natural sources could be investigated (e.g. carbonyl sulphide and methyl chloride).

**The Supplement related to this article is available online at doi:10.5194/acpd-14-31813-2014-supplement.**

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institutions in Belgium (FNRS-CFB and FWO), Canada (GSC), China (CAS), Denmark (FIST), France (IPEV, CNRS/INSU, CEA and ANR), Germany (AWI), Iceland (RannIs), Japan (NIPR), Korea (KOPRI), the Netherlands (NWO/ALW), Sweden (VR), Switzerland (SNF), UK (NERC) and the USA (US NSF, Office of Polar Programs). The Geophysica flights from Oberpfaffenhofen were funded by ESA under the PremieEx project and by Forschungszentrum Jülich. The Fletcher air pumping campaign was made possible with the support of BAS operations and logistics and funded by the Natural Environment Research Council. The Geophysica flights from Kiruna were funded by the EC as part of the FP7 project RECONCILE (Grant number: RECONCILE-226365-FP7-ENV-2008-1). Air sampling during these campaigns was funded by the Dutch Science foundation (NWO, grant number 865.07.001). In addition we are grateful for contributions from Michel Bolder and the Geophysica team (sample collection and campaign organisation).

**References**

- Appenzeller, C., Holton, J. R., and Rosenlof, K. H.: Seasonal variation of mass transport across the tropopause, *J. Geophys. Res.*, 101, 15071–15078, doi:10.1029/96JD00821, 1996.
- Batenburg, A. M., Schuck, T. J., Baker, A. K., Zahn, A., Brenninkmeijer, C. A. M., and Röckmann, T.: The stable isotopic composition of molecular hydrogen in the tropopause region probed by the CARIBIC aircraft, *Atmos. Chem. Phys.*, 12, 4633–4646, doi:10.5194/acp-12-4633-2012, 2012.
- Bernard, S., Röckmann, T., Kaiser, J., Barnola, J.-M., Fischer, H., Blunier, T., and Chappel-laz, J.: Constraints on N<sub>2</sub>O budget changes since pre-industrial time from new firn air and ice core isotope measurements, *Atmos. Chem. Phys.*, 6, 493–503, doi:10.5194/acp-6-493-2006, 2006.
- Brenninkmeijer, C. A. M., Janssen, C., Kaiser, J., Röckmann, T., Rhee, T. S., and Assonov, S. S.: Isotope effects in the chemistry of atmospheric trace gases, *Chem. Rev.*, 103, 5125–5162, doi:10.1021/cr020644k, 2003.
- Buizert, C., Martinerie, P., Petrenko, V. V., Severinghaus, J. P., Trudinger, C. M., Witrant, E., Rosen, J. L., Orsi, A. J., Rubino, M., Etheridge, D. M., Steele, L. P., Hogan, C., Laube, J. C., Sturges, W. T., Levchenko, V. A., Smith, A. M., Levin, I., Conway, T. J., Dlugokencky, E. J., Lang, P. M., Kawamura, K., Jenk, T. M., White, J. W. C., Sowers, T., Schwander, J., and

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Blunier, T.: Gas transport in firn: multiple-tracer characterisation and model intercomparison for NEEM, Northern Greenland, Atmos. Chem. Phys., 12, 4259–4277, doi:10.5194/acp-12-4259-2012, 2012.

Butler, J. H., Battle, M., Bender, M. L., Monzka, S. A., Clarke, A. D., Saltzman, E. S.,  
Sucher, C. M., Severinghaus, J. P., and Elkins, J. W.: A record of atmospheric halocarbons  
during the twentieth century from polar firn air, Nature, 399, 749–755, doi:10.1038/21586,  
1999.

Carr, J. R., Vieli, A., and Stokes, C.: Influence of sea ice decline, atmospheric warm-  
ing, and glacier width on marine-terminating outlet glacier behavior in northwest Green-  
land at seasonal to interannual timescales, J. Geophys. Res.-Earth, 118, 1210–1226,  
doi:10.1002/jgrf.20088, 2013.

Clerbaux, C. and Cunnold, D. M.: Long-lived compounds, in: Scientific Assessment of Ozone  
Depletion, Chapter 1, Global Ozone Research and Monitoring Project – Report No. 50, World  
Meteorological Organization, Geneva, Switzerland, 572 pp., 2007.

Daniel, J. S., Velders, G. J. M., Morgenstern, O., Toohey, D. W., Wallington, T. J., and Wueb-  
bles, D. J.: A focus on information and options for policymakers, in: Scientific Assessment of  
Ozone Depletion, World Meteorological Organization, Geneva, Chapter 5, 2011.

Etheridge, D. M., Steele, L. P., Francey, R. J., and Langenfelds, R. L.: Atmospheric methane  
between 1000 A.D., and present: evidence of anthropogenic emissions and climatic variabil-  
ity, J. Geophys. Res., 103, 15979–15993, doi:10.1029/98JD00923, 1998.

Farman, J. C., Gardiner, B. G., and Shanklin, J. D.: Large losses of total ozone in Antarctica  
reveal seasonal ClO<sub>x</sub>/NO<sub>x</sub> interaction, Nature, 315, 207–210, doi:10.1038/315207a0, 1985.

Francey, R. J., Allison, C. E., Etheridge, D. M., Trudinger, C. M., Enting, I. G., Leuenberger, M.,  
Langenfelds, R. L., Michel, E., and Steele, L. P.: A 1000-year high precision record of  $\delta^{13}\text{C}$   
in atmospheric CO<sub>2</sub>, Tellus B, 51, 170–193, doi:10.1034/j.1600-0889.1999.t01-1-00005.x,  
1999.

Goldstein, A. H. and Shaw, S. L.: Isotopes of volatile organic compounds: an emerging ap-  
proach for studying atmospheric budgets and chemistry, Chem. Rev., 103, 5025–5048,  
doi:10.1021/cr0206566, 2003.

Holton, J. R.: On the global exchange of mass between the stratosphere and troposphere, J.  
Atmos. Sci., 47, 392–395, doi:10.1175/1520-0469(1990)047<0392:OTGEOM>2.0.CO;2,  
1990.

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Ishijima, K., Sugawara, S., Kawamura, K., Hashida, G., Morimoto, S., Murayama, S., Aoki, S., and Nakazawa, T.: Temporal variations of the atmospheric nitrous oxide concentration and its  $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  for the latter half of the 20th century reconstructed from firn air analyses, *J. Geophys. Res.-Atmos.*, 112, D03305, doi:10.1029/2006JD007208, 2007.

5 Kaiser, J.: Stable Isotope Investigations of Atmospheric Nitrous Oxide, available at: <http://archimed.uni-mainz.de/pub/2003/0004/> (last access: December 2014), Ph. D. thesis, Johannes Gutenberg-Universität, Mainz, 2002.

Kaiser, J., Brenninkmeijer, C. A. M., and Röckmann, T.: Intramolecular  $^{15}\text{N}$  and  $^{18}\text{O}$  fractionation in the reaction of  $\text{N}_2\text{O}$  with  $\text{O}(^1\text{D})$  and its implications for the stratospheric  $\text{N}_2\text{O}$  isotope signature, *J. Geophys. Res.-Atmos.*, 107, 4214, doi:10.1029/2001JD001506, 2002.

10 Kaiser, J., Engel, A., Borchers, R., and Röckmann, T.: Probing stratospheric transport and chemistry with new balloon and aircraft observations of the meridional and vertical  $\text{N}_2\text{O}$  isotope distribution, *Atmos. Chem. Phys.*, 6, 3535–3556, doi:10.5194/acp-6-3535-2006, 2006.

15 Langenfelds, R. J., Fraser, P. J., Francey, R. J., Steele, L. P., Porter, L. W., and Allison, C. E.: The Cape Grim air archive: the first seventeen years, in: *Baseline 94–95*, edited by: Francey, R. J., Dick, A. L., and Derek, N., Bur. Meteorol., and CSIRO Atmos. Res., Melbourne, Victoria, Australia, 53–70, 1996.

Laube, J. C., Kaiser, J., Sturges, W. T., Bönisch, H., and Engel, A.: Chlorine isotope fractionation in the stratosphere, *Science*, 329, 1167, doi:10.1126/science.1191809, 2010a.

20 Laube, J. C., Martinerie, P., Witrant, E., Blunier, T., Schwander, J., Brenninkmeijer, C. A. M., Schuck, T. J., Bolder, M., Röckmann, T., van der Veen, C., Bönisch, H., Engel, A., Mills, G. P., Newland, M. J., Oram, D. E., Reeves, C. E., and Sturges, W. T.: Accelerating growth of HFC-227ea (1,1,1,2,3,3,3-heptafluoropropane) in the atmosphere, *Atmos. Chem. Phys.*, 10, 5903–5910, doi:10.5194/acp-10-5903-2010, 2010b.

25 Laube, J. C., Keil, A., Bönisch, H., Engel, A., Röckmann, T., Volk, C. M., and Sturges, W. T.: Observation-based assessment of stratospheric fractional release, lifetimes, and ozone depletion potentials of ten important source gases, *Atmos. Chem. Phys.*, 13, 2779–2791, doi:10.5194/acp-13-2779-2013, 2013.

30 Laube, J. C., Newland, M. J., Hogan, C., Brenninkmeijer, C. A. M., Fraser, P. J., Martinerie, P., Oram, D. E., Reeves, C. E., Röckmann, T., Schwander, J., Witrant, E., and Sturges, W. T.: Newly detected ozone-depleting substances in the atmosphere, *Nat. Geosci.*, 7, 266–269, doi:10.1038/ngeo2109, 2014.

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- Leuenberger, M. and Lang, C.: Thermal diffusion: an important aspect in studies of static air columns such as firn air, sand dunes, and soil air, in: *Isotope Aided Studies of Atmospheric Carbon Dioxide and Other Greenhouse Gases – Phase II*, IAEA, Vienna, 2002.
- Martinerie, P., Nourtier-Mazauric, E., Barnola, J.-M., Sturges, W. T., Worton, D. R., Atlas, E., Gohar, L. K., Shine, K. P., and Brasseur, G. P.: Long-lived halocarbon trends and budgets from atmospheric chemistry modelling constrained with measurements in polar firn, *Atmos. Chem. Phys.*, 9, 3911–3934, doi:10.5194/acp-9-3911-2009, 2009.
- McCarthy, M. C., Connell, P., and Boering, K. A.: Isotopic fractionation of methane in the stratosphere and its effect on free tropospheric isotopic compositions, *Geophys. Res. Lett.*, 28, 3657–3660, doi:10.1029/2001GL013159, 2001.
- Molina, M. J. and Rowland, F. S.: Stratospheric sink for chlorofluoromethanes: chlorine atom-catalysed destruction of ozone, *Nature*, 249, 810–812, doi:10.1038/249810a0, 1974.
- Montzka, S. A., Reimann, S., Engel, A., Krueger, K. O'Doherty, S., and Sturges, W. T.: Ozone-depleting substances (ODSs) and related chemicals, in: *Scientific Assessment of Ozone Depletion*, World Meteorological Organization, Geneva, Chapter 1, 2011.
- Nevison, C. D., Keeling, R. F., Weiss, R. F., Popp, B. N., Fraser, P. J., Porter, L. W., and Hess, P. G.: Southern Ocean ventilation inferred from seasonal cycles of atmospheric  $N_2O$  and  $O_2/N_2$  at Cape Grim, Tasmania, *Tellus B*, 57, 218–229, doi:10.1111/j.1600-0889.2005.00143.x, 2005.
- Nevison, C. D., Dlugokencky, E., Dutton, G., Elkins, J. W., Fraser, P., Hall, B., Krummel, P. B., Langenfelds, R. L., O'Doherty, S., Prinn, R. G., Steele, L. P., and Weiss, R. F.: Exploring causes of interannual variability in the seasonal cycles of tropospheric nitrous oxide, *Atmos. Chem. Phys.*, 11, 3713–3730, doi:10.5194/acp-11-3713-2011, 2011.
- Park, S., Croteau, P., Boering, K. A., Etheridge, D. M., Ferretti, D., Fraser, P. J., Kim, K. R., Krummel, P. B., Langenfelds, R. L., van Ommen, T. D., Steele, L. P., and Trudinger, C. M.: Trends and seasonal cycles in the isotopic composition of nitrous oxide since 1940, *Nat. Geosci.*, 5, 261–265, doi:10.1038/NGEO1421, 2012.
- Redeker, K. R., Davis, S., and Kalin, R. M.: Isotope values of atmospheric halocarbons and hydrocarbons from Irish urban, rural, and marine locations, *J. Geophys. Res.*, 112, D16307, doi:10.1029/2006JD007784, 2007.
- Röckmann, T. and Levin, I.: High-precision determination of the changing isotopic composition of atmospheric  $N_2O$  from 1990–2002, *J. Geophys. Res.-Atmos.*, 110, D21304, doi:10.1029/2005JD006066, 2005.

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Röckmann, T., Kaiser, J., and Brenninkmeijer, C. A. M.: The isotopic fingerprint of the pre-industrial and the anthropogenic N<sub>2</sub>O source, *Atmos. Chem. Phys.*, 3, 315–323, doi:10.5194/acp-3-315-2003, 2003.

Rubino, M., Etheridge, D. M., Trudinger, C. M., Allison, C. E., Battle, M. O., Langenfelds, R. L., Steele, L. P., Curran, M., Bender, M., White, J. W. C., Jenk, T. M., Blunier, T., and Francey, R. J.: A revised 1000 year atmospheric  $\delta^{13}\text{C-CO}_2$  record from Law Dome and South Pole, Antarctica, *J. Geophys. Res.-Atmos.*, 118, 8482–8499, doi:10.1002/jgrd.50668, 2013.

Schilt, A., Baumgartner, M., Blunier, T., Schwander, J., Spahni, R., Fischer, H., and Stocker, T. F.: Glacial–interglacial and millennial-scale variations in the atmospheric nitrous oxide concentration during the last 800,000 years, *Quaternary Sci. Rev.*, 29, 182–192, doi:10.1016/j.quascirev.2009.03.011, 2010.

Schwander, J., Barnola, J. M., Andrie, C., Leuenberger, M., Ludin, A., Raynaud, D., and Stauffer, B.: The age of the air in the firn and the ice at Summit, Greenland, *J. Geophys. Res.-Atmos.*, 98, 2831–2838, doi:10.1029/92JD02383, 1993.

Sowers, T., Rodebaugh, A., Yoshida, N., and Toyoda, S.: Extending records of the isotopic composition of the atmospheric N<sub>2</sub>O back to 1800 AD from air trapped in snow at the South Pole and the Greenland Ice Sheet Project II ice core, *Global Biogeochem. Cy.*, 16, 1129, doi:10.1029/2002GB001911, 2002.

Spahni, R., Chappellaz, J., Stocker, T. F., Louergue, L., Hausammann, G., Kawamura, K., Flückiger, J., Schwander, J., Raynaud, D., Masson-Delmotte, V., and Jouzel, J.: Atmospheric methane and nitrous oxide of the late pleistocene from Antarctic ice cores, *Science*, 310, 1317–1321, doi:10.1126/science.1120132, 2005.

SPARC: Report on the Lifetimes of Stratospheric Ozone-Depleting Substances, Their Replacements, and Related Species, edited by: Ko, M., Newman, P., Reimann, S., Strahan, S., SPARC Report No. 6, WCRP-15, Zurich, Switzerland, 2013.

Sturges, W. T., Oram, D. E., Laube, J. C., Reeves, C. E., Newland, M. J., Hogan, C., Martinie, P., Witrant, E., Brenninkmeijer, C. A. M., Schuck, T. J., and Fraser, P. J.: Emissions halted of the potent greenhouse gas SF<sub>5</sub>CF<sub>3</sub>, *Atmos. Chem. Phys.*, 12, 3653–3658, doi:10.5194/acp-12-3653-2012, 2012.

Sturrock, G. A., Etheridge, D. M., Trudinger, C. M., Fraser, P. J., and Smith, A. M.: Atmospheric histories of halocarbons from analysis of Antarctic firn air: major Montreal Protocol species, *J. Geophys. Res.-Atmos.*, 107, 4765, doi:10.1029/2002JD002548, 2002.

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Sugawara, S., Nakazawa, T., Shirakawa, Y., Kawamura, K., Aoki, S., Machida, T., and Honda, H.: Vertical profile of the carbon isotopic ratio of stratospheric methane over Japan, *Geophys. Res. Lett.*, 24, 2989–2992, doi:10.1029/97GL03044, 1997.

Trenberth, K. E. and Guillemot, C. J.: The total mass of the atmosphere, *J. Geophys. Res.-Atmos.*, 99, 23079–23088, doi:10.1029/94JD02043, 1994.

Velders, G. J. M. and Daniel, J. S.: Uncertainty analysis of projections of ozone-depleting substances: mixing ratios, EESC, ODPs, and GWPs, *Atmos. Chem. Phys.*, 14, 2757–2776, doi:10.5194/acp-14-2757-2014, 2014.

von Hobe, M., Bekki, S., Borrmann, S., Cairo, F., D'Amato, F., Di Donfrancesco, G., Dörnbrack, A., Ebersoldt, A., Ebert, M., Emde, C., Engel, I., Ern, M., Frey, W., Genco, S., Griessbach, S., Groöß, J.-U., Gulde, T., Günther, G., Hösen, E., Hoffmann, L., Homonai, V., Hoyle, C. R., Isaksen, I. S. A., Jackson, D. R., János, I. M., Jones, R. L., Kandler, K., Kalicinsky, C., Keil, A., Khaykin, S. M., Khosrawi, F., Kivi, R., Kuttippurath, J., Laube, J. C., Lefèvre, F., Lehmann, R., Ludmann, S., Luo, B. P., Marchand, M., Meyer, J., Mitev, V., Molleker, S., Müller, R., Oelhaf, H., Olschewski, F., Orsolini, Y., Peter, T., Pfeilsticker, K., Piesch, C., Pitts, M. C., Poole, L. R., Pope, F. D., Ravegnani, F., Rex, M., Riese, M., Röckmann, T., Rognerud, B., Roiger, A., Rolf, C., Santee, M. L., Scheibe, M., Schiller, C., Schlager, H., Siciliani de Cumis, M., Sitnikov, N., Søvde, O. A., Spang, R., Spelten, N., Stordal, F., Sumińska-Ebersoldt, O., Ulanovski, A., Ungermann, J., Viciani, S., Volk, C. M., vom Scheidt, M., von der Gathen, P., Walker, K., Wegner, T., Weigel, R., Weinbruch, S., Wetzel, G., Wienhold, F. G., Wohltmann, I., Woiwode, W., Young, I. A. K., Yushkov, V., Zobrist, B., and Stroh, F.: Reconciliation of essential process parameters for an enhanced predictability of Arctic stratospheric ozone loss and its climate interactions (RECONCILE): activities and results, *Atmos. Chem. Phys.*, 13, 9233–9268, doi:10.5194/acp-13-9233-2013, 2013.

Witrand, E. and Martinerie, P.: Input estimation from sparse measurements in LPV systems and isotopic ratios in polar firns, in: Proc. of the IFAC Joint Symposium on SSSC, TDS and FDA, The International Federation of Automatic Control, Grenoble, France, 659–664, doi:10.3182/20130204-3-FR-2033.00201, 2013.

Witrand, E., Martinerie, P., Hogan, C., Laube, J. C., Kawamura, K., Capron, E., Montzka, S. A., Dlugokencky, E. J., Etheridge, D., Blunier, T., and Sturges, W. T.: A new multi-gas constrained model of trace gas non-homogeneous transport in firn: evaluation and behaviour at eleven polar sites, *Atmos. Chem. Phys.*, 12, 11465–11483, doi:10.5194/acp-12-11465-2012, 2012.

Yoshida, N. and Toyoda, S.: Constraining the atmospheric N<sub>2</sub>O budget from intramolecular site preference in N<sub>2</sub>O isotopomers, *Nature*, 405, 330–334, doi:10.1038/35012558, 2000.

Zuiderweg, A., Kaiser, J., Laube, J. C., Röckmann, T., and Holzinger, R.: Stable carbon isotope fractionation in the UV photolysis of CFC-11 and CFC-12, *Atmos. Chem. Phys.*, 12, 4379–4385, doi:10.5194/acp-12-4379-2012, 2012.

Zuiderweg, A., Holzinger, R., Martinerie, P., Schneider, R., Kaiser, J., Witrant, E., Etheridge, D., Petrenko, V., Blunier, T., and Röckmann, T.: Extreme <sup>13</sup>C depletion of CCl<sub>2</sub>F<sub>2</sub> in firn air samples from NEEM, Greenland, *Atmos. Chem. Phys.*, 13, 599–609, doi:10.5194/acp-13-599-2013, 2013.

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**Table 1.** Key atmospheric characteristics of CFC-11, CFC-12 and CFC-113. (a) NOAA/AGAGE mean values taken from Montzka et al. (2011) and (b) data from SPARC (2013).

	Compound		
	CFC-11	CFC-12	CFC-113
(a) 2009 mole fraction pmol mol <sup>-1</sup>	243	535	76
(b) Atmospheric lifetime/years	52	102	93
(a) Ozone Depletion Potential (ODP)	1	0.82	0.85

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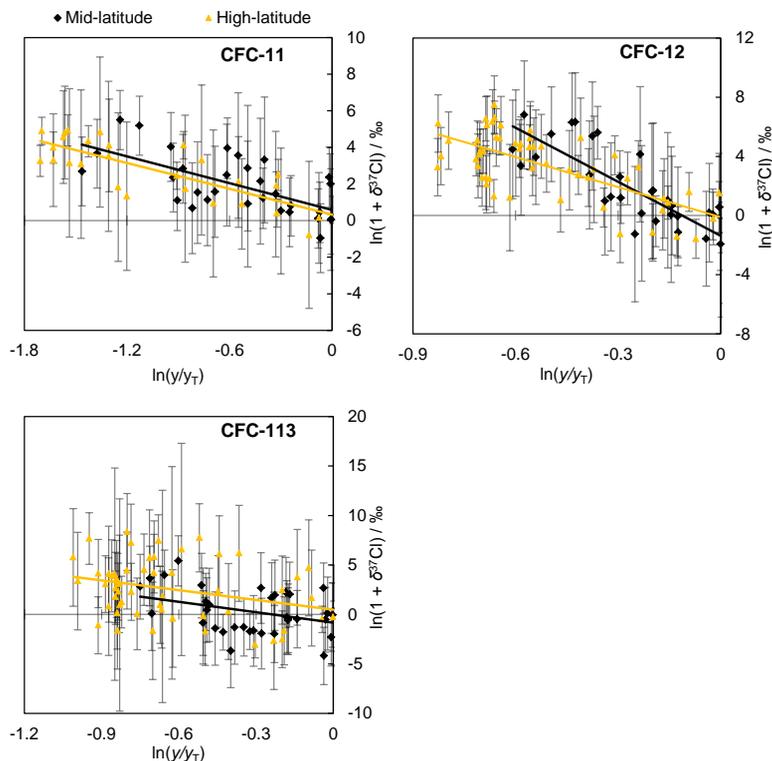
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**Table 2.** Details of the samples used in this study.

Sample type	Sampling location	Sampling dates	Approx. air age	Further details
Free tropospheric air	Cape Grim, Tasmania	1978–2010	1978–2010	Sampling procedure previously reported (e.g. Langenfelds et al., 1996)
Tropospheric firn air	Fletcher Promontory, Antarctica	Jan 2012	1970–2012	Drilling and firn air pumping operations conducted by the British Antarctic Survey
Tropospheric firn air	North Greenland	Jul 2009	1950–2009	Drilling and firn air pumping operations during the North Greenland Eemian Ice Drilling (NEEM) field campaign
Stratospheric air	Mid-latitude samples (48.1–53.4°)	2009	–	M55 Geophysica research aircraft, take off from Oberpfaffenhofen, Germany
Stratospheric air	High-latitude samples (66.3–76.2°)	2010	–	M55 Geophysica research aircraft, take off from Kiruna, Sweden

## Chlorine isotope composition

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**Figure 1.** Rayleigh fractionation plots of CFC-11, CFC-12 and CFC-113 chlorine isotope signatures, derived from mid-latitude (black diamonds) and high-latitude (orange triangles) stratospheric samples. Trend lines correspond to apparent fractionations ( $\epsilon_{\text{app}}$ ) of  $(-2.4 \pm 0.5)$  and  $(-2.3 \pm 0.4)\%$  (CFC-11),  $(-12.2 \pm 1.6)$  and  $(-6.8 \pm 0.8)\%$  (CFC-12) and  $(-3.5 \pm 1.5)$  and  $(-3.3 \pm 1.2)\%$  (CFC-113) for mid- and high-latitudes, respectively. The standard error of the gradient is quoted for  $\epsilon_{\text{app}}$  values. The average repeatability for individual samples was  $\pm 2.3\%$  (CFC-11),  $\pm 2.4\%$  (CFC-12) and  $\pm 3.9\%$  (CFC-113).  $1\sigma$  SD error bars are shown.

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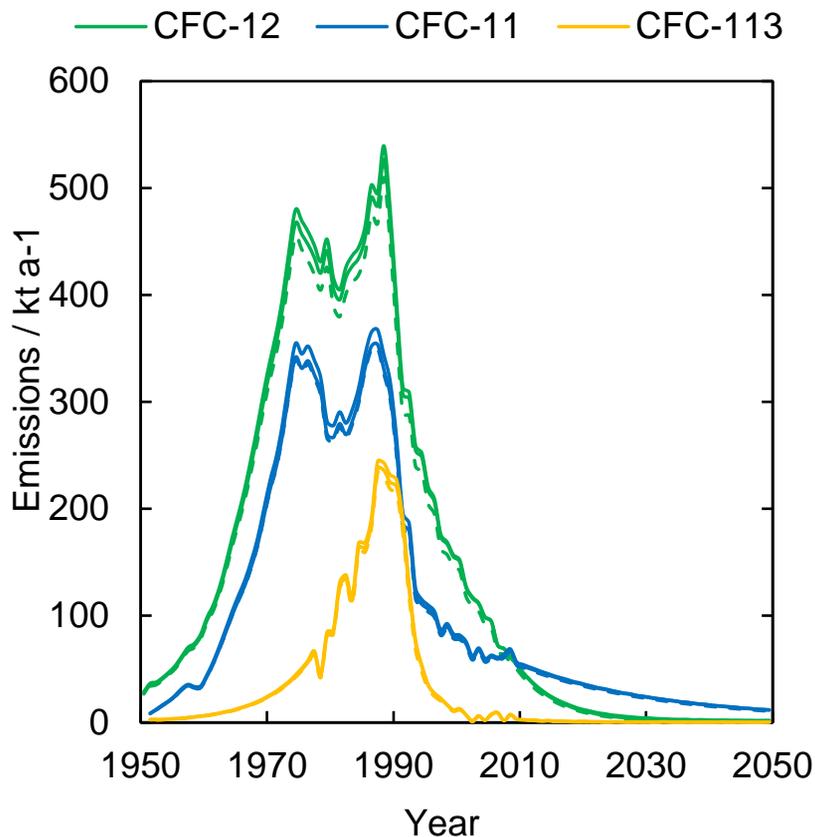
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**Figure 2.** Emissions estimates for CFC-11 (blue), CFC-12 (green) and CFC-113 (orange), based on global tropospheric mole fractions (SPARC, 2013). Each species is represented by three lines. The two emissions estimates from this study (solid lines) are produced using the stratosphere–troposphere exchange flux calculations of Holton (1990) and Appenzeller et al. (1996), with the latter giving slightly higher emissions. These are compared to values from Velders and Daniel (2014) (dashed lines).

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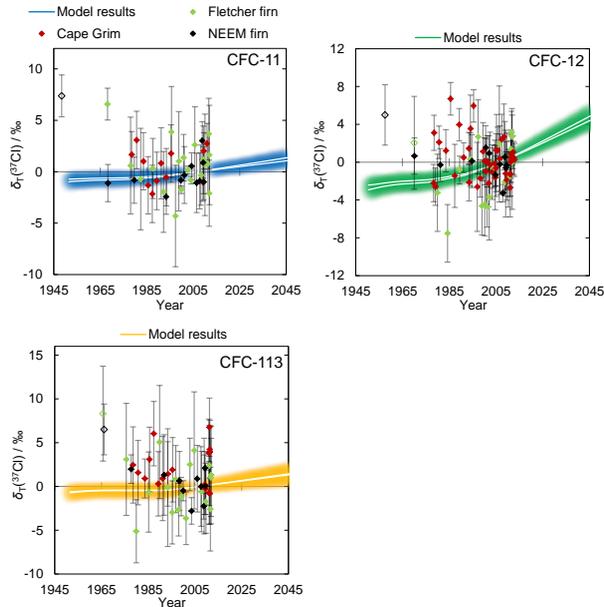
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**Figure 3.** Tropospheric  $\delta_{\text{T}}(^{37}\text{Cl})$  measurements, compared to model predictions. Black diamonds (NEEM firn air), green diamonds (Fletcher Promontory firn air) and red diamonds (Cape Grim air archive) represent the tropospheric chlorine isotope history of CFC-11, CFC-12 and CFC-113. The average repeatability for individual samples was  $\pm 2.7$  (CFC-11),  $\pm 2.7$  (CFC-12) and  $\pm 3.8$ ‰ (CFC-113).  $1\sigma$  SD error bars are shown. Open symbols indicate that the measurement was subject to a small correction, based on the dilution series analysis (see Supplement). White trend lines represent model estimates of the temporal evolution of  $\delta_{\text{T}}(^{37}\text{Cl})$  values. Two lines are displayed per species, one for each stratosphere–troposphere exchange flux estimate (Holton, 1990; Appenzeller et al., 1996). Blue (CFC-11), green (CFC-12) and orange (CFC-113) shading indicates the model uncertainty envelopes. These uncertainties are based on two independent sources of error: (1) the  $1\sigma$  uncertainty in  $\varepsilon_{\text{app}}$  values, (2) the range of “most likely” lifetime values from SPARC (2013). All delta values are relative to 2006 standard air and the model output has been adjusted so that it passes through 0‰ at 2006.