



Seasonal characteristics, formation mechanisms and geographical origins of PM_{2.5} in two megacities in Sichuan Basin, China

Huanbo Wang^{1,2}, Mi Tian¹, Yang Chen¹, Guangming Shi¹, Yuan Liu¹, Fumo Yang^{1,2,3,4*}, Leiming Zhang⁵,
Liqun Deng⁶, Jiayan Yu⁷, Chao Peng¹, and Xuyao Cao¹

¹Research Center for Atmospheric Environment, Chongqing Institute of Green and Intelligent Technology, Chinese Academy of Sciences, Chongqing, 400714, China

²Chongqing University, Chongqing, 400044, China

³Center for Excellence in Regional Atmospheric Environment, Institute of Urban Environment, Chinese Academy of Sciences, Xiamen, 361021, China

⁴Yangtze Normal University, Chongqing, 408100, China

⁵Environment and Climate Change Canada, Toronto, Canada

⁶Sichuan Academy of Environmental Sciences, Chengdu, 610041, China

⁷Chongqing Environmental Monitoring Center, Chongqing 401147, China

Correspondence to: Fumo Yang (fmyang@cigit.ac.cn)



1 **Abstract.** To investigate the characteristics of PM_{2.5} and its major chemical components, formation
2 mechanisms, and geographical origins in the two biggest cities, Chengdu (CD) and Chongqing (CQ) in
3 Sichuan Basin, the most densely populated basin in China, daily PM_{2.5} samples were collected
4 simultaneously at one urban site in each city from October 2014 to July 2015. Annual mean
5 concentrations of PM_{2.5} were 67.0 ± 43.4 and $70.9 \pm 41.4 \mu\text{g m}^{-3}$ at CD and CQ, respectively. Secondary
6 inorganic aerosols (SNA) and organic matter (OM) accounted for 41.1% and 26.1%, respectively, of
7 PM_{2.5} mass at CD, and 37.4% and 29.6% at CQ. Seasonal variations of PM_{2.5} and its major chemical
8 components were significant, usually with the highest values in winter and the lowest in summer. SNA
9 and OM were 1.7-3.4 times higher on polluted days than on clean days at both sites, whereas their
10 percentage contributions to PM_{2.5} varied differently among the components and between the two sites.
11 Gas-phase oxidation probably played an important role on the formation of secondary aerosols when
12 PM_{2.5} mass varied in the range of 75-150 $\mu\text{g m}^{-3}$, while heterogeneous transformation was likely the
13 major mechanism on the heavy polluted days. Geographical regions causing high PM_{2.5} were identified
14 to mainly distribute within the basin at both sites based on potential source contribution function (PSCF)
15 analysis.



16 1 Introduction

17 Fine particles (PM_{2.5}, particulate matter with an aerodynamic diameter smaller than 2.5 μm) have
18 adverse effects on human health (Anderson et al., 2012; Lepeule et al., 2012; Taus et al., 2008),
19 deteriorate air quality (Zhang et al., 2008; Paraskevopoulou et al., 2015), reduce atmospheric visibility
20 (Fu et al., 2016; Cao et al., 2012; Baumer et al., 2008), impact climate (Ramanathan and Feng, 2009;
21 Hitznerberger et al., 1999; Mahowald, 2011), and affect ecosystem (Larssen et al., 2006). In the past two
22 decades, China has experienced serious PM_{2.5} pollution due to the rapidly increasing energy
23 consumption through economic development, industrialization and urbanization (Tie and Cao, 2009).
24 The National Ambient Air Quality Standards (NAAQS) for PM_{2.5} was promulgated by the Chinese
25 government in 2012, and strict strategies have been implemented nationwide, e.g. controlling SO₂
26 emissions by installing desulphurization system in coal-fired power plants and conversion of fuel to
27 natural gas (Lu et al., 2011), mitigating NO_x emissions through traffic restrictions, and reducing
28 biomass burning through straw shredding. Despite these efforts, there are still many cities that have not
29 met the current NAAQS. According to the ‘2013-2015 Reports on the State of Environment of China’,
30 annual mean concentration of PM_{2.5} in all the 74 major cities over China was 72, 64, and 50 μg m⁻³ in
31 2013, 2014 and 2015, respectively, and only 4.1%, 12.2% and 22.5% of the monitored cities met the
32 NAAQS (35 μg m⁻³).

33 Previous studies showed that Beijing-Tianjin-Hebei area (BTH), Yangtze River Delta (YRD), Pearl
34 River Delta (PRD), and Sichuan Basin were the four main regions in China with severe aerosol
35 pollution. While many studies have been conducted in BTH, PRD and YRD regions to understand the
36 general characteristics of PM_{2.5} and its chemical components, formation mechanism, and sources (Ji et
37 al., 2016; Li et al., 2015; Quan et al., 2015; Tan et al., 2016; Yang et al., 2015; Zhang et al., 2013; Zhao
38 et al., 2015; Zhao et al., 2013a; Cheng et al., 2015; Zheng et al., 2015a; Yang et al., 2011a), only a few
39 studies have focused on Sichuan Basin (Tao et al., 2014; Tian et al., 2013; Yang et al., 2011b).
40 Covering an area of 260,000 km² with a population of around 100 million in southwest China, the
41 Sichuan Basin is a subtropical expanse of low hills and plains and the most populated basin in China. It
42 is completely encircled by high mountains, and characterized by persistently high relative humidity, and
43 extremely low wind speeds all the year-round (Mengting et al., 2016; Chen and Xie, 2013). It is
44 supposed that the characteristics of PM_{2.5} in Sichuan Basin should be very different from those in
45 eastern coastal China (i.e. PRD and YRD) and North China Plain (i.e. BTH) due to its special



46 topography and meteorological conditions, besides emission sources. More specifically, the terrain of
47 the two megacities in the basin are distinct from each other, i.e., Chongqing is a mountainous
48 municipality lying on the eastern margin of the basin while Chengdu is located in a completely flat west.
49 Therefore, there is a great interest in comparing the chemical components of $PM_{2.5}$ and characterizing
50 pollution episodes between Chengdu and Chongqing.

51 The present study aims to fill this gap by measuring chemically-resolved $PM_{2.5}$ in Chengdu and
52 Chongqing in four seasons during 2014-2015. The main objectives are to: (1) characterize the seasonal
53 and site differences of $PM_{2.5}$ and its major chemical components at the two urban sites in Sichuan Basin;
54 (2) compare the $PM_{2.5}$ chemical components under different $PM_{2.5}$ levels and identify the major
55 chemical components that are responsible for $PM_{2.5}$ pollution episodes; (3) explore the possible
56 formation mechanism of the secondary organic and inorganic aerosols; and (4) reveal the geographical
57 regions causing high $PM_{2.5}$ levels through potential source contribution function (PSCF) analysis. The
58 comparison of the differences of $PM_{2.5}$ in this paper provides insight regarding the extent to which fine
59 particulate pollution can vary in terms of in chemical composition, formation mechanisms, and
60 geographical origins between the two megacities in a great basin. This information has implications for
61 better understanding the reasons for heavy haze pollution in this unique basin.

62 **2 Methodology**

63 **2.1 Sampling sites**

64 $PM_{2.5}$ samples were collected at two urban sites, one in Chengdu and another in Chongqing, the two
65 largest cities in Sichuan Basin, southwest China (Wang et al., 2017). The two sampling sites are located
66 260 km apart (Fig. 1). The sampling site in Chengdu (CD) is located on the roof of a sixth floor building
67 in the Sichuan Academy of Environmental Science (104°4' E, 30°37' N) with no large surrounding
68 industries but heavy traffic. The closest main road (Renmin South road of Chengdu) is about 20 m east
69 of the sampling site. The sampling site in Chongqing (CQ) is located on the rooftop of Chongqing
70 Monitoring Center (106°30' E, 29°37' N). The highway G50 is 250 m away from this sampling site. The
71 two selected sampling sites are considered to represent typical urban environment in their respective
72 cities without the influence of local point sources from industry activities.



73 2.2 Sample collection

74 Daily (23-h) integrated PM_{2.5} samples were collected in four months, each in a different season: autumn
75 (23 October to 18 November, 2014), winter (6 January to 2 February, 2015), spring (2 to 29 April, 2015),
76 and summer (2 to 30 July, 2015). At both sites, PM_{2.5} samples were collected in parallel on Teflon filters
77 (Whatman Corp., 47 mm) and quartz filters (Whatman Corp., 47 mm) and used for different chemical
78 analyses. At the CD site, PM_{2.5} sampling was carried out using an versatile air pollutant sampler (Wang
79 et al., 2017). One channel was used to load PM_{2.5} sample on Teflon filter for mass and trace elements
80 analysis and the other one was equipped with quartz filter for water-soluble inorganic ions and
81 carbonaceous components analysis. The sampler was running at 15 L min⁻¹ for each channel. At the CQ
82 site, a low-volume aerosol sampler (BGI Corp., from Omni, USA) operating at a flow rate of 5 L min⁻¹
83 was used to collect PM_{2.5} samples on Teflon filter, and another sampler (Thermo Scientific Corp.
84 Partisol 2000i, USA) with a flow rate of 16.7 L min⁻¹ was used to collect PM_{2.5} samples on quartz filter.

85 Before sampling, all the quartz filters were preheated at 450°C for 4 h to remove the organic
86 compounds. All sampled filters were stored in clean Petri slides in the dark and at -18°C until analysis
87 to prevent the evaporation of volatile compounds. Before and after sample collection, all the Teflon
88 filters were weighted at least three times using an microbalance (Sartorius, ME 5-F, Germany) after
89 their stabilization for 48 h under controlled conditions (temperature: 20~23°C, relative humidity:
90 45~50%). Differences among replicate weights were mostly less than 15 µg for each sample.

91 2.3 Chemical analysis

92 For the analysis of water-soluble inorganic ions, a quarter of each quartz filter was first extracted using
93 ultrapure water in an ultrasonic bath for 30 min, and then filtered through a 0.45 µm pore syringe filter.
94 Anions (SO₄²⁻, NO₃⁻ and Cl⁻) and cations (Na⁺, NH₄⁺, K⁺, Mg²⁺ and Ca²⁺) were determined using ion
95 chromatograph (Dionex Corp., Dionex 600, USA). Anions were separated using AS11-HC column with
96 30 mM KOH as an eluent at a flow rate of 1.0 ml min⁻¹. Cations were determined using CS12A column
97 with 20 mM MSA (methanesulfonic acid) at a flow rate of 1.0 ml min⁻¹. Individual standard solutions of
98 all investigated anions and cations (1000 mg L⁻¹, o2si, USA) were diluted to construct the calibration
99 curves. The correlation coefficients of the linear regression of the standard curves were all above 0.999.
100 Field blanks were prepared and analyzed together with the samples and then subtracted from the
101 samples. The concentrations of the water-soluble inorganic ions in the field blanks were in the range of



102 0.008-0.13 $\mu\text{g m}^{-3}$. The relative standard deviation of each ion was better than 8% for the reproducibility
103 test.

104 Organic carbon (OC) and elemental carbon (EC) were measured by thermal-optical reflectance
105 (TOR) method using a DRI OC/EC analyzer (Atmoslytic Inc., USA). The methodology for OC/EC
106 analysis was based on TOR method as described in Chow et al. (2007). For calibration and quality
107 control, measurement with filter blank, standard sucrose solution and replicate analysis were performed.
108 Blank corrections were performed by subtracting the blank values from the sampled ones. The
109 concentration of EC in field blanks was zero while OC were below 0.7 $\mu\text{g C cm}^{-2}$. The repeatability was
110 better than 15%.

111 The elements including Al, Si, Ca, Fe, and Ti were analyzed on Teflon filter using X-ray
112 fluorescence analyzer (Epsilon 5ED-XRF, PAN'alytical Corp., Netherlands), the QA/QC procedures of
113 the XRF analysis have been described in Cao et al. (2012).

114 The gaseous species were continuously measured by a set of online gas analyzers, including
115 EC9850 SO_2 analyzer, 9841 $\text{NO}/\text{NO}_2/\text{NO}_x$ analyzer, 9830 CO analyzer, and 9810 O_3 analyzer (Ecotech
116 Corp., Australia) at the CD site, and Thermo 42i $\text{NO}/\text{NO}_2/\text{NO}_x$ analyzer, 43i SO_2 analyzer, 48i CO
117 analyzer, and 49i O_3 analyzer (Thermo Scientific Corp., USA) at the CQ site. The mass concentration of
118 $\text{PM}_{2.5}$ was automatically measured by online particulate monitor instruments (BAM1020, Met one
119 Corp., USA, at CD and 5030 SHARP, Thermo Scientific Corp, USA, at CQ). Hourly meteorological
120 parameters, including ambient temperature (T), relative humidity (RH), wind speed (WS) and direction,
121 barometric pressure (P), and solar radiation (SR), were obtained from an automatic weather station at
122 each site.

123 2.4 Data analysis

124 The EC-tracer method has been widely used to estimate SOC (Turpin and Huntzicker, 1995; Castro et
125 al., 1999), which can be expressed as

$$126 \text{POC} = (\text{OC}/\text{EC})_{\text{prim}} \times \text{EC} \quad (1)$$

$$127 \text{SOC} = \text{OC} - \text{POC} \quad (2)$$

128 Where POC, SOC and OC represent estimated primary OC, secondary OC and measured total OC,
129 respectively. $(\text{OC}/\text{EC})_{\text{min}}$ was simplified as the $(\text{OC}/\text{EC})_{\text{prim}}$ to estimate SOC in this study. $(\text{OC}/\text{EC})_{\text{min}}$
130 was 2.4, 2.6, 1.6 and 2.2 in autumn, winter, spring and summer at CD, respectively, and 1.9, 2.8, 1.1
131 and 1.5 at CQ. The estimated SOC was only an approximation with uncertainties, e.g., from influence



132 of biomass burning (Ding et al., 2012).

133 The coefficient of divergence (COD) has been used to evaluate the spatial similarity of chemical
134 compositions at different sites (Wongphatarakul et al., 1998; Qu et al., 2015), which is defined as

$$135 \quad COD_{jk} = \sqrt{\frac{1}{p} \sum_{i=1}^p \left(\frac{x_{ij} - x_{ik}}{x_{ij} + x_{ik}} \right)^2} \quad (3)$$

136 Where x_{ij} and x_{ik} represent the average concentration for a chemical component i at site j and k ,
137 respectively, p is the number of chemical components. Generally, a COD value lower than 0.2 indicates
138 a relatively similarity of spatial distribution.

139 2.5 Geographical origins of PM_{2.5}

140 72-h air mass back trajectories were generated based on the Hybrid Single Particle Lagrangian
141 Integrated Trajectory (HYSPLIT) model using 40 km gridded meteorological data (GDAS 40 km) for
142 PM_{2.5} measurements from October 2014 to July 2015 at both sites. Four trajectories at 04:00, 10:00,
143 16:00, and 22:00 UTC every day with the starting height of 300 m above ground level were calculated
144 (Squizzato and Masiol, 2015). The trajectories coupled with daily PM_{2.5} concentrations were used for
145 PSCF analysis, with the threshold criterion in PSCF analysis being set at the upper 50% of PM_{2.5}. The
146 trajectory covered area was in the range of 20–45° N and 90–120° E and divided into 0.5°×0.5° grid cells.

147 3 Results and discussion

148 3.1 Overview of PM_{2.5} mass concentrations and major components

149 Table 1 presents seasonal and annual mean concentrations of PM_{2.5} and its major chemical components
150 at CD and CQ during the sampling periods. Daily PM_{2.5} ranged from 11.6 to 224.7 $\mu\text{g m}^{-3}$ with annual
151 average being $67.0 \pm 43.4 \mu\text{g m}^{-3}$ at CD site and $70.9 \pm 41.4 \mu\text{g m}^{-3}$ at CQ site. The annual average
152 values were about two times of the NAAQS annual limit. Secondary inorganic aerosol (SNA, the sum
153 of SO_4^{2-} , NO_3^- and NH_4^+) and carbonaceous species together represented more than 70% of PM_{2.5} at
154 both sites (Fig. 2). The annual mean concentrations of the total SNA were 27.6 and 26.5 $\mu\text{g m}^{-3}$ at CD
155 and CQ, respectively, contributing 41.1% and 37.4% of PM_{2.5}. Among these, SO_4^{2-} , NO_3^- and NH_4^+
156 were 11.2, 9.1, and 7.2 $\mu\text{g m}^{-3}$, respectively, or 16.8%, 13.6% and 10.8% of PM_{2.5} mass at CD, and 12.2,
157 7.7 and 6.6 $\mu\text{g m}^{-3}$ or 17.2%, 10.9% and 9.2% of PM_{2.5} mass at CQ. Organic matters (OM), estimated
158 from OC by using a conversion factor of 1.6 to account for other elements present in organic



159 compounds (Turpin and Lim, 2001), were the most abundant species in $PM_{2.5}$, accounting for 26.1%
160 and 29.6% of $PM_{2.5}$ mass at CD and CQ, respectively. In contrast, EC only comprised of around 6% at
161 both sites. The annual mean OC and EC at CQ were 20% and 25% higher than those at CD. Fine soil
162 (FS) can be estimated by summing the oxides of the elements mainly associated with soil, i.e., Al_2O_3 ,
163 SiO_2 , CaO , FeO , Fe_2O_3 , and TiO_2 (Huang et al., 2014). The annual mean concentration of FS at CQ was
164 $6.7 \mu g m^{-3}$ (or 9.5% of $PM_{2.5}$), which was about two times of that at CD ($3.8 \mu g m^{-3}$ or 5.7% of $PM_{2.5}$).
165 The minor components such as K^+ and Cl^- constituted less than 5% of $PM_{2.5}$. The unaccounted portion
166 of $PM_{2.5}$ reached 18.3% at CD and 15.3% at CQ, which was likely related to the uncertainties in the
167 multiplication factor used for estimation of OM and FS, other unidentified species, and measurement
168 uncertainties.

169 3.2 Seasonal variations

170 Figure 3 shows the seasonal variations of mass concentrations of $PM_{2.5}$ and its major chemical
171 components at CD and CQ sites. Seasonal-average $PM_{2.5}$ was the highest in winter at both sites, which
172 was 1.8-2.5 times of those in the other seasons. Besides the high emissions of SO_2 and NO_x in winter,
173 stagnant air condition with frequent calm winds and low boundary layer height was another major cause
174 of the highest $PM_{2.5}$ in this season (Chen and Xie, 2013). Seasonal differences among the other three
175 seasons were generally small, e.g., less than 40%. All $PM_{2.5}$ components but FS followed the seasonal
176 pattern of $PM_{2.5}$ with the highest concentrations in winter, but with subtle differences. The majority of
177 the components showed a summer minimum, but not SO_4^{2-} . The minimum of OC at CD and FS at CQ
178 appeared in spring and autumn, respectively. Higher SO_4^{2-} concentrations in summer were likely due to
179 the enhanced photochemical reactions associated with higher temperature and stronger solar radiation in
180 summer. It is also noted that the seasonal variations in NO_3^- were much larger than those in SO_4^{2-} and
181 NH_4^+ , which can be explained by the enhanced formation of NO_3^- under high relative humidity in winter,
182 and volatility of NH_4NO_3 in summer (Pathak et al., 2009; Quan et al., 2015; Squizzato et al., 2013). In
183 addition, thermodynamically driven behavior of NH_4NO_3 was another factor for the lower NO_3^-
184 concentrations in summer (Wang et al., 2016; Kuprov et al., 2014). High levels of Cl^- and K^+ in winter
185 and of FS in spring should be caused by biomass burning or spring dust storms (Tao et al. 2013, 2014).

186 Both OC and EC showed the highest concentrations in winter at CD and CQ, whereas seasonal
187 differences of those carbonaceous components were less distinct in other seasons, e.g. the variations of
188 OC and EC among other three seasons were less than 30%. Seasonal average SOC was unexpectedly



189 the highest in winter at both sites, different from the anticipated high value in summer in consideration
190 of strong photochemical reaction. Condensation of semi-volatile organic aerosols in winter seemed to
191 play a larger role than photochemical reaction in summer, knowing that low temperature favors
192 condensation process (Sahu et al., 2011; Cesari et al., 2016). Although high O₃ and strong solar
193 radiation condition in summer was conducive to strong photochemical reactions, high temperature
194 favors gas-particle partitioning in the gaseous phase and thus limited the increase of SOC (Strader et al.,
195 1999).

196 The seasonal relative contributions of major chemical component to PM_{2.5} are shown in Fig. 2. The
197 seasonal average contributions of SNA to PM_{2.5} only varied within a small range from 39.5% to 43.2%
198 at CD, whereas in a relatively larger range from 31.0% in summer to 37.1-41.5% in other seasons at CQ.
199 The smaller contribution in summer at CQ was mainly due to the lower NO₃⁻ concentrations. At both
200 CD and CQ, NO₃⁻ and NH₄⁺ showed the highest contributions in winter and the lowest one in summer,
201 whereas an opposite trend was found for SO₄²⁻. The contributions of carbonaceous components (the sum
202 of OC and EC) generally followed the seasonal patterns of SNA, accounting for 26.7-38.8% of PM_{2.5}
203 mass. Among these, OM showed the lowest fractions in PM_{2.5} in spring (21.1%) at CD and highest
204 value in winter (33.6%) at CQ, while the percentages of OM in other seasons were similar at both sites,
205 around 27%. The seasonal variations of EC fractions were not obvious, with a slightly higher value in
206 spring. The highest contributions from FS was more than 10%, appeared in spring at both sites.

207 3.3 Site differences

208 A comparison between the two sites in terms of seasonal-average concentrations of major chemical
209 components are shown in Fig. 4. Despite the 260 km distance between the two sampling sites, a
210 moderate similarity was observed in autumn, winter and spring on the basis of low COD values
211 (0.15-0.18), indicating the regional-scale PM_{2.5} pollution pattern in Sichuan Basin and the similarity in
212 major emission sources for both sites. However, it is worth to note that several components differed by
213 up to a factor of 2.5 in their season-average concentrations, e.g. OC and EC in winter and spring, and
214 Cl⁻ and FS in all the four seasons. These discrepancies were partly caused by the different atmospheric
215 chemical processes and local sources between the two sites on certain days. In summer, the differences
216 for several major chemical components (FS, OC, SO₄²⁻, NO₃⁻, EC) between the two sites were larger
217 than in other seasons, causing the high COD value (0.33) in this season.

218 Good correlations between the two sites were found for daily SNA, OC, EC and K⁺ concentrations



219 in autumn, winter and spring (Table 2). However, for NO_3^- , a significant correlation was identified only
220 in autumn, likely due to the strong impact of local vehicle emissions and the subsequent atmospheric
221 processes forming NO_3^- . Similarly, a moderate correlation was observed just in winter for both Cl^- and
222 FS. In summer, weak or no correlations were identified between the two sites for almost all chemical
223 components. Therefore, there existed a fairly uniform distribution of most major components in $\text{PM}_{2.5}$
224 throughout the basin in all the seasons except summer.

225 **3.4 Chemical characteristics of $\text{PM}_{2.5}$ on clean and polluted days**

226 **3.4.1 Key chemical species causing polluted days**

227 To explore the major chemical pollutants responsible for polluted days, daily $\text{PM}_{2.5}$ data were
228 categorized into two groups, i.e. clean days and polluted days based on $\text{PM}_{2.5}$ concentrations below and
229 above the NAAQS guideline value of $75 \mu\text{g m}^{-3}$, respectively. 34 and 31 polluted days were counted at
230 CD and CQ site, respectively, accounting for 30.4% and 28.6% of the entire sampling days. The number
231 of polluted days at CD was 8, 21, 4 and 1 in autumn, winter, spring and summer, accounting for 29.6%,
232 75%, 14.3% and 3.4% of the total sampling days in each season, respectively, and at CQ they were 4, 19,
233 6 and 2, accounting for 14.8%, 67.9%, 21.4% and 6.9%. Note that these numbers were only from
234 one-month (around 28 days) sampling in each season. Considering the similar meteorological
235 conditions and pollution levels during autumn and spring, these two seasons were combined together
236 and was referred to as the warm season, while winter was referred to as the cold season. Because of the
237 very small number of polluted days in summer at both sites, the $\text{PM}_{2.5}$ data in this season were not
238 discussed.

239 $\text{PM}_{2.5}$ concentrations increased dramatically on polluted days compared to clean days (Fig. 5). For
240 example, $\text{PM}_{2.5}$ concentrations were more than doubled on polluted days during the entire sampling
241 periods at both sites. The two dominant groups of components in $\text{PM}_{2.5}$, SNA and OC, were 2.5-2.8
242 times higher on polluted days in both cold and warm seasons at CD. However, larger variations were
243 found at CQ with SNA increased by 2.7 and 1.7 times, and OM by 3.4 and 2.1 times in the cold and
244 warm season, respectively. Thus, while the enhancement of SNA and OC on polluted days were similar
245 at CD, OC increased much more at CQ, indicating some different contributing factors to the high $\text{PM}_{2.5}$
246 pollution at the two sites. Pollutants accumulation under stagnant meteorological conditions might be a
247 main factor at CD based on the similar magnitudes of the enhancements of $\text{PM}_{2.5}$ and its dominant



248 components, while additional processes should have increased OC more than other components at CQ.

249 In the cold season, the percentage contributions of SNA to $PM_{2.5}$ were similar on clean and polluted
250 days, 38-41% at both sites (Fig. S1). However, the percentage contribution of OM to $PM_{2.5}$ decreased
251 from 30.1% on clean days to 27.5% on polluted days at CD, and increased from 26.9% to 34.9% at CQ.
252 A different pattern was seen in the warm season, with no significant variations of OM fractions in $PM_{2.5}$
253 between the clean and polluted days at either site, but an increased SNA contribution by around 7% at
254 CD and decreased contribution by 14% at CQ. The percentage contributions of SNA and OM to $PM_{2.5}$
255 discussed above were different from those found in eastern coastal China and North China Plain, where
256 considerable increases were found for SNA and decreases for OM on polluted days than clean days
257 (Tan et al., 2009; Wang et al., 2015; Quan et al., 2014; Zhang et al., 2015a; Zhang et al., 2016; Cheng et
258 al., 2015). This emphasized again the unique characteristics of $PM_{2.5}$ pollution in Sichuan Basin due to
259 its particular topography and meteorological conditions.

260 Concentrations of the individual SNA species (SO_4^{2-} , NO_3^- and NH_4^+) increased by a factor of
261 1.2-3.3 on polluted days compared to clean days in all the cases (Fig. 5). But the percentage
262 contributions differed among the species as NO_3^- increased and SO_4^{2-} decreased on polluted days. The
263 concentration of FS increased slightly at CD (less than $3 \mu g m^{-3}$) but significantly (from 5.4 on clean
264 days to $14.7 \mu g m^{-3}$ on polluted days) at CQ in warm season. The percentage contribution of FS to $PM_{2.5}$
265 reached 15.3% on polluted days at CQ in warm season.

266 3.4.2 Transformation mechanisms of secondary aerosols

267 Meteorological conditions, atmospheric chemical processes and long-range transport are all responsible
268 for $PM_{2.5}$ accumulation on polluted days (Zheng et al., 2015b). CO and EC are directly emitted from
269 combustion processes and are not very reactive, thus their concentrations in the air are strongly
270 controlled by meteorological parameters within a relatively short period. Both CO and EC
271 concentrations increased on polluted days (Fig. 5 and Fig. S2), suggesting the important role the
272 meteorological condition played on $PM_{2.5}$ accumulation. As expected, very weak winds (less than $0.7 m$
273 s^{-1}) were observed on polluted days, which hindered the pollutants horizontal transport. CO can be
274 considered as a reference pollutant species whose temporal variations were mainly from the impact of
275 meteorological conditions. The impact of other factors on other pollutants can thus be explored by
276 scaling the concentrations of other pollutants to that of CO. For example, $PM_{2.5}$ was enhanced by a
277 factor of 2.0-2.7 on polluted days than clean days in the two seasons and at the two sites, but the



278 CO-scaled $PM_{2.5}$ only showed an enhancement of a factor of 1.5-1.8 (Fig. 6), and the latter values were
279 likely from the enhanced secondary aerosol formation.

280 As shown in Fig. 6, the CO-scaled SNA was 60-90% higher on polluted days with individual
281 species 40-120% higher (except in the warm season at CQ), even though their gaseous precursor (SO_2
282 and NO_2 , no data for NH_3) were only less than 30% higher. This suggests stronger chemical
283 transformation from gaseous precursors to particle formation on polluted days. Sulfur oxidation ratio
284 ($SOR = n-SO_4^{2-}/(n-SO_4^{2-}+n-SO_2)$) and nitrogen oxidation ratio ($NOR = n-NO_3^-/(n-NO_3^-+n-NO_2)$) were
285 defined to evaluate the degree of secondary transformation (n refers to as the molar concentration). In
286 the cold season, NOR increased from 0.09 on clean days to 0.16 on polluted days at CD and from 0.07
287 to 0.14 at CQ. SOR increased only slightly, from 0.31 to 0.35 at CD and 0.28 to 0.35 at CQ. In the
288 warm season, NOR and SOR exhibited a similar pattern as those in the cold season with the exception
289 of NOR at CQ, which might be related to the high temperature in the warm season. The CO-scaled SOC
290 increased by a factor of 2.6 and 1.5 on polluted days in the cold and warm season at CQ, but no
291 significant change or decrease was found at CD. Moreover, increased SOC/OC only occurred at CQ in
292 the cold season (Fig. 6). The different pattern in SOC (or SOC/OC) than SNA (or SOR and NOR)
293 suggests that secondary organic aerosols (SOA) production is of less important than SNA production in
294 most occasions except in the cold season at CQ.

295 The formation mechanisms of secondary aerosols were further explored based on the diurnal
296 variations of their gaseous precursors and the relevant meteorological parameters. Considering the
297 different characteristics of $PM_{2.5}$ pollution, the polluted days were further divided into moderate
298 polluted days ($75 < PM_{2.5} \leq 150 \mu g m^{-3}$) and heavy polluted days ($PM_{2.5} > 150 \mu g m^{-3}$). Figure 7 and Fig.
299 S2 describe the diurnal variations of T, RH, WS, SR, NO, CO, SO_2 , NO_2 and O_3 concentrations in the
300 cold season at CD and CQ. Most gaseous species except O_3 showed relatively higher concentrations on
301 polluted days, but exhibited different diurnal patterns under different pollution levels. For example, NO
302 and CO both reached their peak values at around 10:00 am, consistent with the morning traffic rush
303 hours (Fig. S2). For SO_2 and NO_2 , similar diurnal variations were seen on clean and moderate polluted
304 days, but a different one on heavy polluted days, with the latter having long-lasting (from 12:00 to
305 18:00 pm) high concentrations likely caused by transport as supported by stronger afternoon winds on
306 these days (Fig. 7). O_3 showed a typical diurnal variation with a peak value at around 16:00 pm.
307 Daytime (11:00-20:00 LT) O_3 was much higher on moderate polluted days, but slightly lower on heavy



308 polluted days compared to those on clean days. Aerosols were generally considered as a constrain factor
309 to O₃ production due to their absorbing or scattering the UV radiation, which would reduce solar
310 radiation and consequently decrease the photochemical activity and O₃ levels (Zhang et al., 2015b;
311 Zheng et al., 2015b; Tian et al., 2016). The O₃ data on heavy polluted days seemed to support this
312 hypothesis, but not those on moderate polluted days. The diurnal variations of O₃ tracked the pattern of
313 solar radiation (Fig. S2), which might be associated with the special climate characteristics in Chengdu
314 with extremely low annual sunshine totals and mostly overcast days, especially in winter. As a result,
315 the impact of weather conditions on solar radiation might overwhelm that of aerosol pollution at CD
316 and caused unusual diurnal variations of O₃ and solar radiation. In the warm season, no heavy PM_{2.5}
317 pollution occurred at CD and CQ, and the diurnal patterns of gaseous precursors and meteorological
318 parameters were similar as those in the cold season (Fig. 8 and Fig. S3).

319 Besides the well-known photochemical processes forming SO₄²⁻ and NO₃⁻ (Stockwell and Calvert,
320 1983; Blitz et al., 2003; Calvert and Stockwell, 1983), heterogeneous reactions might also be important
321 formation mechanisms for these SNA species (Quan et al., 2015; Zheng et al., 2015a; Zhao et al.,
322 2013b). Similarly, SOA is mainly formed through photochemical oxidation of primary VOCs followed
323 by condensation of SVOC onto particles as well as through aqueous-phase reactions (Ervens et al.,
324 2011). While photochemical reactions are mostly influenced by temperature and solar radiation,
325 heterogeneous reactions are associated with high RH. Meteorological conditions and O₃ levels
326 associated with different pollution levels suggest that gas-phase oxidation reaction was likely the major
327 formation mechanism of secondary aerosols on moderate polluted days (high O₃ levels and SR, low RH)
328 while heterogeneous reactions likely played a more important role on heavy polluted days (low O₃
329 levels and slightly higher RH). This hypothesis needs further verification using high-resolution data.

330 3.4.3 Impact of NH₃ amount and RH on NO₃⁻ concentrations

331 In Sect. 3.4.2, it was found that the CO-scaled NO₃⁻ increased dramatically in the cold season but
332 decreased in the warm season on polluted days at CQ. To explain the different season patterns, major
333 factors affecting NO₃⁻ are explored, including NH₃ levels and RH. The neutralization ratio (NR) is
334 defined as

$$335 \quad NR = \frac{[NH_4^+]}{2[SO_4^{2-}] + [NO_3^-]} \quad (4)$$

336 Where the concentrations of NH₄⁺, SO₄²⁻ and NO₃⁻ were expressed as molar concentrations. NR in cold



337 season ranged from 1.0 to 1.2 at the two sites, indicating fully neutralization of SO_4^{2-} and NO_3^- by NH_3 .
338 In this case, NH_3 was not a limiting factor for NO_3^- . NR in the warm season was 1.05 at CD, but less
339 than 1.0 (0.94-0.95) at CQ. NH_3 seemed to be a limiting factor for NO_3^- formation at CQ on both clean
340 and polluted days in the warm season. Although NH_4^+ increased on polluted days (by 30% based on
341 CO-scaled concentrations), SO_4^{2-} increased even more (by 60% based on scaled concentration),
342 resulting in incomplete neutralization of NO_3^- and little variations of its concentrations.

343 Another consideration is the thermodynamic equilibrium between particulate NH_4NO_3 and gaseous
344 HNO_3 , which depends on temperature and RH. Kuprov et al. (2014) found that the equilibrium would
345 be shifted toward the particulate phase when RH was above the deliquescence relative humidity (DRH)
346 of NH_4NO_3 , and the dissociation constant decreased to about an order of magnitude when RH was
347 above 75%. As shown in Fig. S4, on polluted days, RH was lower than DRH most of the time in the
348 warm season at CQ estimated according to Mozurkewich (1993), which explains the lower NO_3^-
349 concentrations. In contrast, RH was above the DRH most of the time at CD. Thus, despite similar
350 concentrations of SO_4^{2-} and NH_4^+ between CD and CQ in summer, NO_3^- was 50% lower at CQ than that
351 at CD due to the lower ambient RH at CQ.

352 3.5 Geographical origins causing high $\text{PM}_{2.5}$ pollution

353 PSCF analysis was applied to investigate the potential source regions contributing to high $\text{PM}_{2.5}$
354 pollution (Fig. 9). Basically, all the major source areas for high $\text{PM}_{2.5}$ were distributed within the Basin.
355 Long-range transports normally occurred in North Plain and eastern coastal regions were not observed
356 at CD and CQ site (Zhao et al., 2015; Zhang et al., 2013). At CD, the major source areas in the cold
357 season included the areas of the northeastern, southeastern and southern Chengdu and in some areas of
358 Chongqing, and in warm season included areas scattered between Chengdu and Chongqing (e.g.,
359 Neijiang) besides the area south of Chengdu. At CQ in the cold season, the northeast area of Chongqing
360 was identified as strong sources, where a number of industries were located, such as Changshou
361 chemical industrial ozone. In the warm season, the areas of the southern Chongqing were also found to
362 be potential sources. Overall, $\text{PM}_{2.5}$ pollution at CQ was characterized by significant local contribution
363 from major sources located in or nearby Chongqing. In contrast, regional transport in Sichuan Basin
364 from southeast, south and southwest of Chengdu had a major impact on $\text{PM}_{2.5}$ pollution at CD.



365 4 Conclusions

366 Seasonal patterns, formation mechanisms, and sources origins of PM_{2.5} at the two megacities in Sichuan
367 Basin were explored using chemically resolved daily PM_{2.5} samples collected in four consecutive
368 seasons. On about 30% of the days, daily PM_{2.5} exceeded NAAQS, with SNA and OC concentrations
369 1.7-2.8 and 2.1-3.4 times, respectively, higher than those on clean days. The percentage contributions of
370 SNA and OM to PM_{2.5} mass only differed slightly between polluted and clean days at CD, while it
371 increased significantly for OM at CQ in cold season. This phenomenon was different from those found
372 in the other regions of China, implying the unique roles played by local topography, meteorology, and
373 emissions sources in this region. Most chemical components of PM_{2.5} exhibited the highest
374 concentrations in the winter (with exception of FS) and the lowest in the summer (with exception of
375 SO₄²⁻). The site differences in PM_{2.5} and SNA between CD and CQ were not significant in all the
376 seasons except in the summer, while higher OC and FS concentrations were observed at CQ than at CD
377 in most seasons. The different diurnal patterns of gaseous precursors and meteorological parameters
378 under different PM_{2.5} level implied the more important roles of heterogeneous transformation on the
379 heavy polluted days. The ammonia-limited and high temperature conditions also played a critical role in
380 low NO₃⁻ levels in the warm season at CQ. Major sources of PM_{2.5} were concentrated in Chengdu,
381 Chongqing, and the areas in between, with the regional sources affecting more at CD site and the local
382 emissions at CQ site. Future studies should aim to combine high-resolution PM_{2.5} data and at suburban
383 and rural locations to gain a thorough understanding of PM_{2.5} sources and formation mechanisms.

384

385 *Competing interests.* The authors declare that they have no conflict of interest.

386 *Acknowledgements.* This work was supported by the National Natural Science Foundation of China (No.
387 41405027, 41375123, and 41403089), the "Strategic Priority Research Program" of the Chinese
388 Academy of Sciences (No. KJZD-EW-TZ-G06), the West Action Plan of the Chinese Academy of
389 Science (No. KZCX2-XB3-14), and Chongqing Science and Technology Commission (No.
390 cstc2014yykfC20003, cstckjxjljrc13). We are grateful to Yumeng Zhu and Jun Wang for sample
391 collection.

392

393 **Reference**

- 394 Anderson, G. B., Krall, J. R., Peng, R. D., and Bell, M. L.: Is the Relation Between Ozone and Mortality Confounded by
395 Chemical Components of Particulate Matter? Analysis of 7 Components in 57 US Communities, *Am J Epidemiol*, 176,
396 726-732, 10.1093/aje/kws188, 2012.
- 397 Baumer, D., Vogel, B., Versick, S., Rinke, R., Mohler, O., and Schnaiter, M.: Relationship of visibility, aerosol optical
398 thickness and aerosol size distribution in an ageing air mass over South-West Germany, *Atmos Environ*, 42, 989-998,
399 10.1016/j.atmosenv.2007.10.017, 2008.
- 400 Blitz, M. A., Hughes, K. J., and Pilling, M. J.: Determination of the high-pressure limiting rate coefficient and the enthalpy
401 of reaction for OH+SO₂, *J Phys Chem A*, 107, 1971-1978, 10.1021/jp026524y, 2003.
- 402 Calvert, J. G., and Stockwell, W. R.: Acid Generation in the Troposphere by Gas-Phase Chemistry, *Environ Sci Technol*, 17,
403 A428-A443, DOI 10.1021/es00115a002, 1983.
- 404 Cao, J. J., Wang, Q. Y., Chow, J. C., Watson, J. G., Tie, X. X., Shen, Z. X., Wang, P., and An, Z. S.: Impacts of aerosol
405 compositions on visibility impairment in Xi'an, China, *Atmos Environ*, 59, 559-566, 10.1016/j.atmosenv.2012.05.036,
406 2012.
- 407 Castro, L. M., Pio, C. A., Harrison, R. M., and Smith, D. J. T.: Carbonaceous aerosol in urban and rural European
408 atmospheres: estimation of secondary organic carbon concentrations, *Atmos Environ*, 33, 2771-2781, Doi
409 10.1016/S1352-2310(98)00331-8, 1999.
- 410 Cesari, D., Donato, A., Conte, M., Merico, E., Giangreco, A., Giangreco, F., and Contini, D.: An inter-comparison of PM_{2.5}
411 at urban and urban background sites: Chemical characterization and source apportionment, *Atmos Res*, 174, 106-119,
412 10.1016/j.atmosres.2016.02.004, 2016.
- 413 Chen, Y., and Xie, S. D.: Long-term trends and characteristics of visibility in two megacities in southwest China: Chengdu
414 and Chongqing, *J Air Waste Manage*, 63, 1058-1069, 10.1080/10962247.2013.791348, 2013.
- 415 Cheng, Y., He, K. B., Du, Z. Y., Zheng, M., Duan, F. K., and Ma, Y. L.: Humidity plays an important role in the PM_{2.5}
416 pollution in Beijing, *Environ Pollut*, 197, 68-75, 10.1016/j.envpol.2014.11.028, 2015.
- 417 Chow, J. C., Watson, J. G., Chen, L. W. A., Chang, M. C. O., Robinson, N. F., Trimble, D., and Kohl, S.: The IMPROVE-A
418 temperature protocol for thermal/optical carbon analysis: maintaining consistency with a long-term database, *J Air Waste*
419 *Manage*, 57, 1014-1023, 10.3155/1047-3289.57.9.1014, 2007.
- 420 Ding, X., Wang, X. M., Gao, B., Fu, X. X., He, Q. F., Zhao, X. Y., Yu, J. Z., and Zheng, M.: Tracer-based estimation of
421 secondary organic carbon in the Pearl River Delta, south China, *J Geophys Res-Atmos*, 117, Artn
422 D0531310.1029/2011jd016596, 2012.
- 423 Ervens, B., Turpin, B. J., and Weber, R. J.: Secondary organic aerosol formation in cloud droplets and aqueous particles
424 (aqSOA): a review of laboratory, field and model studies, *Atmos Chem Phys*, 11, 11069-11102,
425 10.5194/acp-11-11069-2011, 2011.
- 426 Fu, X. X., Wang, X. M., Hu, Q. H., Li, G. H., Ding, X., Zhang, Y. L., He, Q. F., Liu, T. Y., Zhang, Z., Yu, Q. Q., Shen, R. Q.,
427 and Bi, X. H.: Changes in visibility with PM_{2.5} composition and relative humidity at a background site in the Pearl River
428 Delta region, *J Environ Sci-China*, 40, 10-19, 10.1016/j.jes.2015.12.001, 2016.
- 429 Hitzengerger, R., Berner, A., Giebl, H., Kromp, R., Larson, S. M., Rouc, A., Koch, A., Marischka, S., and Puxbaum, H.:
430 Contribution of carbonaceous material to cloud condensation nuclei concentrations in European background (Mt.
431 Sonnblick) and urban (Vienna) aerosols, *Atmos Environ*, 33, 2647-2659, Doi 10.1016/S1352-2310(98)00391-4, 1999.
- 432 Huang, X. H. H., Bian, Q. J., Ng, W. M., Louie, P. K. K., and Yu, J. Z.: Characterization of PM_{2.5} Major Components and
433 Source Investigation in Suburban Hong Kong: A One Year Monitoring Study, *Aerosol Air Qual Res*, 14, 237-250,
434 10.4209/aaqr.2013.01.0020, 2014.
- 435 Ji, D. S., Zhang, J. K., He, J., Wang, X. J., Pang, B., Liu, Z. R., Wang, L. L., and Wang, Y. S.: Characteristics of atmospheric
436 organic and elemental carbon aerosols in urban Beijing, China, *Atmos Environ*, 125, 293-306,
437 10.1016/j.atmosenv.2015.11.020, 2016.



- 438 Kuprov, R., Eatough, D. J., Cruickshank, T., Olson, N., Cropper, P. M., and Hansen, J. C.: Composition and secondary
439 formation of fine particulate matter in the Salt Lake Valley: Winter 2009, *J Air Waste Manage*, 64, 957-969,
440 10.1080/10962247.2014.903878, 2014.
- 441 Larssen, T., Lydersen, E., Tang, D. G., He, Y., Gao, J. X., Liu, H. Y., Duan, L., Seip, H. M., Vogt, R. D., Mulder, J., Shao, M.,
442 Wang, Y. H., Shang, H., Zhang, X. S., Solberg, S., Aas, W., Okland, T., Eilertsen, O., Angell, V., Liu, Q. R., Zhao, D. W.,
443 Xiang, R. J., Xiao, J. S., and Luo, J. H.: Acid rain in China, *Environ Sci Technol*, 40, 418-425, Doi 10.1021/Es0626133,
444 2006.
- 445 Lepeule, J., Laden, F., Dockery, D., and Schwartz, J.: Chronic Exposure to Fine Particles and Mortality: An Extended
446 Follow-up of the Harvard Six Cities Study from 1974 to 2009, *Environ Health Persp*, 120, 965-970, 10.1289/ehp.1104660,
447 2012.
- 448 Li, B., Zhang, J., Zhao, Y., Yuan, S. Y., Zhao, Q. Y., Shen, G. F., and Wu, H. S.: Seasonal variation of urban carbonaceous
449 aerosols in a typical city Nanjing in Yangtze River Delta, China, *Atmos Environ*, 106, 223-231,
450 10.1016/j.atmosenv.2015.01.064, 2015.
- 451 Lu, Z., Zhang, Q., and Streets, D. G.: Sulfur dioxide and primary carbonaceous aerosol emissions in China and India,
452 1996-2010, *Atmos Chem Phys*, 11, 9839-9864, 10.5194/acp-11-9839-2011, 2011.
- 453 Mahowald, N.: Aerosol Indirect Effect on Biogeochemical Cycles and Climate, *Science*, 334, 794-796,
454 10.1126/science.1207374, 2011.
- 455 Mengting, G., Xuhu, C., and Yu, S.: Characteristics of Low Wind-Speed Meteorology in China, *Acta Scientiarum*
456 *Naturalium Universitatis Pekinensis*, 52, 219-226, 2016.
- 457 Mozurkewich, M.: The Dissociation-Constant of Ammonium-Nitrate and Its Dependence on Temperature,
458 Relative-Humidity and Particle-Size, *Atmos Environ a-Gen*, 27, 261-270, Doi 10.1016/0960-1686(93)90356-4, 1993.
- 459 Paraskevopoulou, D., Liakakou, E., Gerasopoulos, E., and Mihalopoulos, N.: Sources of atmospheric aerosol from long-term
460 measurements (5 years) of chemical composition in Athens, Greece, *Sci Total Environ*, 527, 165-178,
461 10.1016/j.scitotenv.2015.04.022, 2015.
- 462 Pathak, R. K., Wu, W. S., and Wang, T.: Summertime PM_{2.5} ionic species in four major cities of China: nitrate formation in
463 an ammonia-deficient atmosphere, *Atmos Chem Phys*, 9, 1711-1722, 2009.
- 464 Qu, W. J., Wang, J., Zhang, X. Y., Wang, D., and Sheng, L. F.: Influence of relative humidity on aerosol composition:
465 Impacts on light extinction and visibility impairment at two sites in coastal area of China, *Atmos Res*, 153, 500-511,
466 10.1016/j.atmosres.2014.10.009, 2015.
- 467 Quan, J. N., Tie, X. X., Zhang, Q., Liu, Q., Li, X., Gao, Y., and Zhao, D. L.: Characteristics of heavy aerosol pollution during
468 the 2012-2013 winter in Beijing, China, *Atmos Environ*, 88, 83-89, 10.1016/j.atmosenv.2014.01.058, 2014.
- 469 Quan, J. N., Liu, Q., Li, X., Gao, Y., Jia, X. C., Sheng, J. J., and Liu, Y. G.: Effect of heterogeneous aqueous reactions on the
470 secondary formation of inorganic aerosols during haze events, *Atmos Environ*, 122, 306-312,
471 10.1016/j.atmosenv.2015.09.068, 2015.
- 472 Ramanathan, V., and Feng, Y.: Air pollution, greenhouse gases and climate change: Global and regional perspectives, *Atmos*
473 *Environ*, 43, 37-50, 10.1016/j.atmosenv.2008.09.063, 2009.
- 474 Sahu, L. K., Kondo, Y., Miyazaki, Y., Pongkiatkul, P., and Oanh, N. T. K.: Seasonal and diurnal variations of black carbon
475 and organic carbon aerosols in Bangkok, *J Geophys Res-Atmos*, 116, Artn D15302 10.1029/2010jd015563, 2011.
- 476 Squizzato, S., Masiol, M., Brunelli, A., Pistollato, S., Tarabotti, E., Rampazzo, G., and Pavoni, B.: Factors determining the
477 formation of secondary inorganic aerosol: a case study in the Po Valley (Italy), *Atmos Chem Phys*, 13, 1927-1939,
478 10.5194/acp-13-1927-2013, 2013.
- 479 Squizzato, S., and Masiol, M.: Application of meteorology-based methods to determine local and external contributions to
480 particulate matter pollution: A case study in Venice (Italy), *Atmos Environ*, 119, 69-81, 10.1016/j.atmosenv.2015.08.026,
481 2015.
- 482 Stockwell, W. R., and Calvert, J. G.: The Mechanism of the HO-SO₂ Reaction, *Atmos Environ*, 17, 2231-2235, Doi
483 10.1016/0004-6981(83)90220-2, 1983.



- 484 Strader, R., Lurmann, F., and Pandis, S. N.: Evaluation of secondary organic aerosol formation in winter, *Atmos Environ*, 33,
485 4849-4863, Doi 10.1016/S1352-2310(99)00310-6, 1999.
- 486 Tan, J. H., Duan, J. C., He, K. B., Ma, Y. L., Duan, F. K., Chen, Y., and Fu, J. M.: Chemical characteristics of PM_{2.5} during a
487 typical haze episode in Guangzhou, *J Environ Sci-China*, 21, 774-781, 10.1016/S1001-0742(08)62340-2, 2009.
- 488 Tan, J. H., Duan, J. C., Ma, Y. L., He, K. B., Cheng, Y., Deng, S. X., Huang, Y. L., and Si-Tu, S. P.: Long-term trends of
489 chemical characteristics and sources of fine particle in Foshan City, Pearl River Delta: 2008-2014, *Sci Total Environ*, 565,
490 519-528, 10.1016/j.scitotenv.2016.05.059, 2016.
- 491 Tao, J., Zhang, L. M., Engling, G., Zhang, R. J., Yang, Y. H., Cao, J. J., Zhu, C. S., Wang, Q. Y., and Luo, L.: Chemical
492 composition of PM_{2.5} in an urban environment in Chengdu, China: Importance of springtime dust storms and biomass
493 burning, *Atmos Res*, 122, 270-283, 10.1016/j.atmosres.2012.11.004, 2013.
- 494 Tao, J., Gao, J., Zhang, L., Zhang, R., Che, H., Zhang, Z., Lin, Z., Jing, J., Cao, J., and Hsu, S. C.: PM_{2.5} pollution in a
495 megacity of southwest China: source apportionment and implication, *Atmos Chem Phys*, 14, 8679-8699,
496 10.5194/acp-14-8679-2014, 2014.
- 497 Taus, N., Tarulescu, S., Idomir, M., and Taus, R.: Respiratory exposure to air pollutants, *J Environ Prot Ecol*, 9, 15-25, 2008.
- 498 Tian, M., Wang, H. B., Chen, Y., Yang, F. M., Zhang, X. H., Zou, Q., Zhang, R. Q., Ma, Y. L., and He, K. B.: Characteristics
499 of aerosol pollution during heavy haze events in Suzhou, China, *Atmos Chem Phys*, 16, 7357-7371,
500 10.5194/acp-16-7357-2016, 2016.
- 501 Tian, Y. Z., Wu, J. H., Shi, G. L., Wu, J. Y., Zhang, Y. F., Zhou, L. D., Zhang, P., and Feng, Y. C.: Long-term variation of the
502 levels, compositions and sources of size-resolved particulate matter in a megacity in China, *Sci Total Environ*, 463,
503 462-468, 10.1016/j.scitotenv.2013.06.055, 2013.
- 504 Tie, X. X., and Cao, J. J.: Aerosol pollution in China: Present and future impact on environment, *Particuology*, 7, 426-431,
505 10.1016/j.partic.2009.09.003, 2009.
- 506 Turpin, B. J., and Huntzicker, J. J.: Identification of Secondary Organic Aerosol Episodes and Quantitation of Primary and
507 Secondary Organic Aerosol Concentrations during Scaqs, *Atmos Environ*, 29, 3527-3544, Doi
508 10.1016/1352-2310(94)00276-Q, 1995.
- 509 Turpin, B. J., and Lim, H. J.: Species contributions to PM_{2.5} mass concentrations: Revisiting common assumptions for
510 estimating organic mass, *Aerosol Sci Tech*, 35, 602-610, Doi 10.1080/02786820152051454, 2001.
- 511 Wang, D. F., Zhou, B., Fu, Q. Y., Zhao, Q. B., Zhang, Q., Chen, J. M., Yang, X., Duan, Y. S., and Li, J.: Intense secondary
512 aerosol formation due to strong atmospheric photochemical reactions in summer: observations at a rural site in eastern
513 Yangtze River Delta of China, *Sci Total Environ*, 571, 1454-1466, 10.1016/j.scitotenv.2016.06.212, 2016.
- 514 Wang, H. B., Shi, G. M., Tian, M., Zhang, L. M., Chen, Y., Yang, F. M., and Cao, X. Y.: Aerosol optical properties and
515 chemical composition apportionment in Sichuan Basin, China, *Sci Total Environ*, 577, 245-257,
516 10.1016/j.scitotenv.2016.10.173, 2017.
- 517 Wang, H. L., Qiao, L. P., Lou, S. R., Zhou, M., Chen, J. M., Wang, Q., Tao, S. K., Chen, C. H., Huang, H. Y., Li, L., and
518 Huang, C.: PM_{2.5} pollution episode and its contributors from 2011 to 2013 in urban Shanghai, China, *Atmos Environ*, 123,
519 298-305, 10.1016/j.atmosenv.2015.08.018, 2015.
- 520 Wongphatarakul, V., Friedlander, S. K., and Pinto, J. P.: A comparative study of PM_{2.5} ambient aerosol chemical databases,
521 *Environ Sci Technol*, 32, 3926-3934, Doi 10.1021/Es9800582, 1998.
- 522 Yang, F., Huang, L., Duan, F., Zhang, W., He, K., Ma, Y., Brook, J. R., Tan, J., Zhao, Q., and Cheng, Y.: Carbonaceous
523 species in PM_{2.5} at a pair of rural/urban sites in Beijing, 2005-2008, *Atmos Chem Phys*, 11, 7893-7903,
524 10.5194/acp-11-7893-2011, 2011a.
- 525 Yang, F., Tan, J., Zhao, Q., Du, Z., He, K., Ma, Y., Duan, F., Chen, G., and Zhao, Q.: Characteristics of PM_{2.5} speciation in
526 representative megacities and across China, *Atmos Chem Phys*, 11, 5207-5219, 10.5194/acp-11-5207-2011, 2011b.
- 527 Yang, Y. R., Liu, X. G., Qu, Y., An, J. L., Jiang, R., Zhang, Y. H., Sun, Y. L., Wu, Z. J., Zhang, F., Xu, W. Q., and Ma, Q. X.:
528 Characteristics and formation mechanism of continuous hazes in China: a case study during the autumn of 2014 in the
529 North China Plain, *Atmos Chem Phys*, 15, 8165-8178, 10.5194/acp-15-8165-2015, 2015.



- 530 Zhang, Q., Quan, J. N., Tie, X. X., Li, X., Liu, Q., Gao, Y., and Zhao, D. L.: Effects of meteorology and secondary particle
531 formation on visibility during heavy haze events in Beijing, China, *Sci Total Environ*, 502, 578-584,
532 10.1016/j.scitotenv.2014.09.079, 2015a.
- 533 Zhang, Q., Shen, Z. X., Cao, J. J., Zhang, R. J., Zhang, L. M., Huang, R. J., Zheng, C. J., Wang, L. Q., Liu, S. X., Xu, H. M.,
534 Zheng, C. L., and Liu, P. P.: Variations in PM_{2.5}, TSP, BC, and trace gases (NO₂, SO₂, and O₃) between haze and non-haze
535 episodes in winter over Xi'an, China, *Atmos Environ*, 112, 64-71, 10.1016/j.atmosenv.2015.04.033, 2015b.
- 536 Zhang, R., Jing, J., Tao, J., Hsu, S. C., Wang, G., Cao, J., Lee, C. S. L., Zhu, L., Chen, Z., Zhao, Y., and Shen, Z.: Chemical
537 characterization and source apportionment of PM_{2.5} in Beijing: seasonal perspective, *Atmos Chem Phys*, 13, 7053-7074,
538 10.5194/acp-13-7053-2013, 2013.
- 539 Zhang, Y., Huang, W., Cai, T. Q., Fang, D. Q., Wang, Y. Q., Song, J., Hu, M., and Zhang, Y. X.: Concentrations and chemical
540 compositions of fine particles (PM_{2.5}) during haze and non-haze days in Beijing, *Atmos Res*, 174, 62-69,
541 10.1016/j.atmosres.2016.02.003, 2016.
- 542 Zhang, Y. H., Hu, M., Zhong, L. J., Wiedensohler, A., Liu, S. C., Andreae, M. O., Wang, W., and Fan, S. J.: Regional
543 Integrated Experiments on Air Quality over Pearl River Delta 2004 (PRIDE-PRD2004): Overview, *Atmos Environ*, 42,
544 6157-6173, 10.1016/j.atmosenv.2008.03.025, 2008.
- 545 Zhao, M. F., Huang, Z. S., Qiao, T., Zhang, Y. K., Xiu, G. L., and Yu, J. Z.: Chemical characterization, the transport
546 pathways and potential sources of PM_{2.5} in Shanghai: Seasonal variations, *Atmos Res*, 158, 66-78,
547 10.1016/j.atmosres.2015.02.003, 2015.
- 548 Zhao, P. S., Dong, F., He, D., Zhao, X. J., Zhang, X. L., Zhang, W. Z., Yao, Q., and Liu, H. Y.: Characteristics of
549 concentrations and chemical compositions for PM_{2.5} in the region of Beijing, Tianjin, and Hebei, China, *Atmos Chem*
550 *Phys*, 13, 4631-4644, 10.5194/acp-13-4631-2013, 2013a.
- 551 Zhao, X. J., Zhao, P. S., Xu, J., Meng, W., Pu, W. W., Dong, F., He, D., and Shi, Q. F.: Analysis of a winter regional haze
552 event and its formation mechanism in the North China Plain, *Atmos Chem Phys*, 13, 5685-5696,
553 10.5194/acp-13-5685-2013, 2013b.
- 554 Zheng, B., Zhang, Q., Zhang, Y., He, K. B., Wang, K., Zheng, G. J., Duan, F. K., Ma, Y. L., and Kimoto, T.: Heterogeneous
555 chemistry: a mechanism missing in current models to explain secondary inorganic aerosol formation during the January
556 2013 haze episode in North China, *Atmos Chem Phys*, 15, 2031-2049, 10.5194/acp-15-2031-2015, 2015a.
- 557 Zheng, G. J., Duan, F. K., Su, H., Ma, Y. L., Cheng, Y., Zheng, B., Zhang, Q., Huang, T., Kimoto, T., Chang, D., Poschl, U.,
558 Cheng, Y. F., and He, K. B.: Exploring the severe winter haze in Beijing: the impact of synoptic weather, regional
559 transport and heterogeneous reactions, *Atmos Chem Phys*, 15, 2969-2983, 10.5194/acp-15-2969-2015, 2015b.



Table 1 Annual and seasonal mean concentrations of PM_{2.5} and major chemical components ($\mu\text{g m}^{-3}$) at CD and CQ during 2014-2015.

	CD					CQ				
	Autumn	Winter	Spring	Summer	Annual	Autumn	Winter	Spring	Summer	Annual
PM _{2.5}	62.1±38.4	113.5±47.8	48.0±25.2	45.1±15.2	67.0±43.4	56.3±23.6	115.1±53.9	58.3±24.6	54.2±16.2	70.9±41.4
SO ₄ ²⁻	10.5±6.5	16.4±7.1	8.3±5.9	9.7±4.7	11.2±6.8	9.9±4.7	17.5±7.4	10.4±6.5	11.1±5.7	12.2±6.8
NO ₃ ⁻	9.3±7.4	17.5±8.8	5.9±3.6	3.9±2.2	9.1±8.0	7.8±3.8	15.8±9.5	5.9±5.0	1.6±1.3	7.7±7.6
NH ₄ ⁺	6.9±4.8	12.7±5.4	5.1±3.2	4.2±1.9	7.2±5.2	5.7±2.8	11.3±5.2	5.2±3.0	4.0±2.1	6.6±4.4
Cl ⁻	1.9±1.2	3.4±1.9	0.6±0.4	0.2±0.2	1.5±1.7	0.8±0.4	1.6±1.2	0.5±0.5	0.04±0.03	0.7±0.9
K ⁺	0.6±0.4	1.2±0.6	0.6±0.5	0.5±0.2	0.7±0.5	0.5±0.2	1.2±0.7	0.5±0.2	0.3±0.1	0.6±0.5
OC	10.4±6.1	19.7±8.4	6.3±3.7	7.4±1.5	10.9±7.6	9.7±4.7	24.2±13.6	10.0±5.1	8.5±3.4	13.1±10.0
EC	3.0±2.1	6.3±3.0	2.7±2.3	2.5±0.7	3.6±2.7	3.8±1.7	5.9±3.2	4.7±3.0	3.7±1.5	4.5±2.6
FS	3.2±1.6	4.5±2.0	4.8±3.0	2.7±1.5	3.8±2.2	5.0±2.8	6.3±3.3	9.1±7.6	6.5±4.0	6.7±5.0



Table 2 Pearson's correlation analysis for daily concentrations of major components in PM_{2.5} between CD and CQ, significant correlation ($p < 0.01$) are bold faced.

	SO ₄ ²⁻	NO ₃ ⁻	NH ₄ ⁺	Cl ⁻	K ⁺	OC	EC	FS
Autumn	0.78	0.72	0.75	0.10	0.76	0.87	0.79	0.08
Winter	0.63	0.49	0.60	0.51	0.77	0.83	0.80	0.64
Spring	0.76	0.39	0.59	0.09	0.68	0.78	0.74	0.29
Summer	0.49	-0.13	0.40	0.37	0.36	0.43	0.23	-0.02
Annual	0.72	0.71	0.76	0.67	0.79	0.80	0.72	0.31

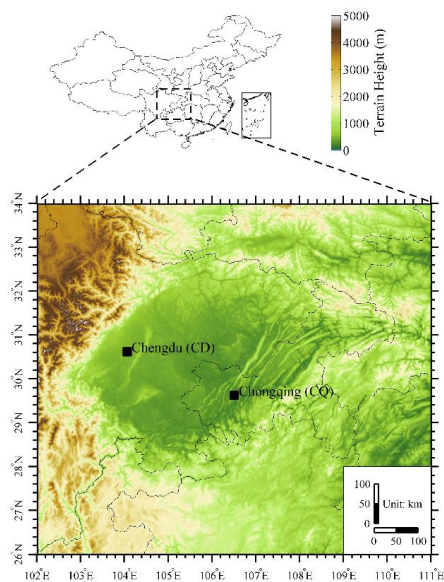


Figure 1. Locations of the sampling sites at Chengdu (CD) and Chongqing (CQ).

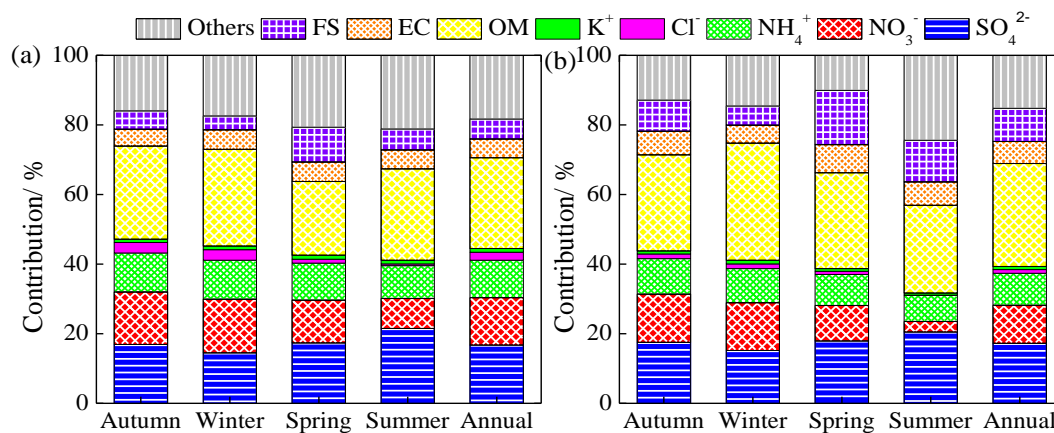


Figure 2. Seasonal and annual contributions of individual chemical components to PM_{2.5} at CD (a) and CQ (b).

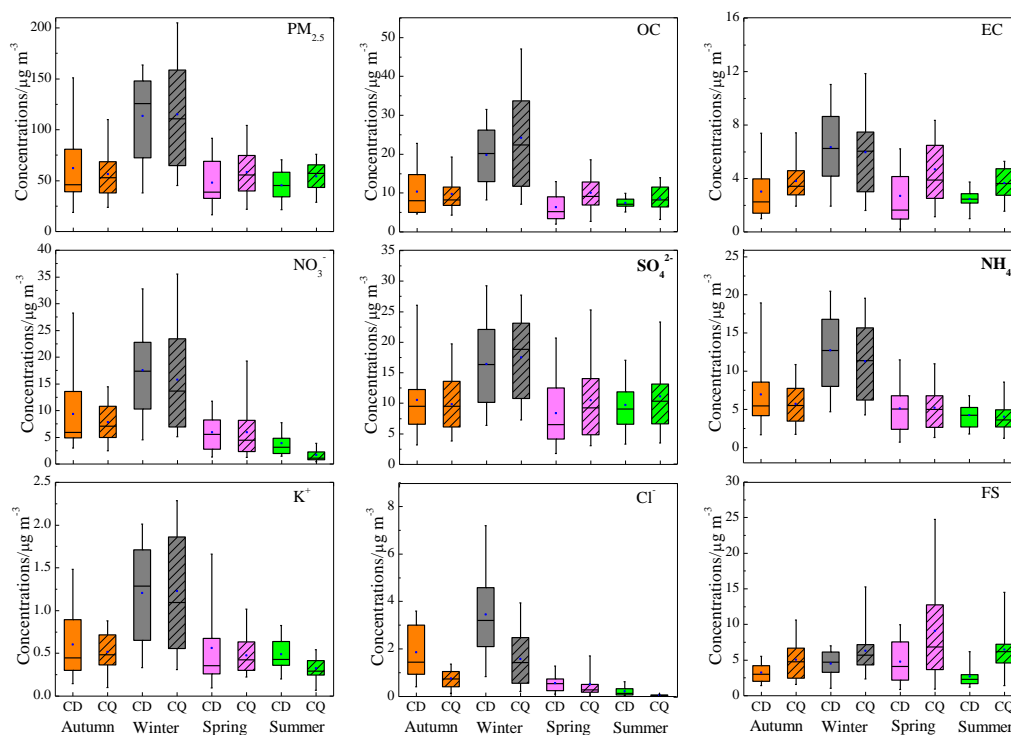


Figure 3. Seasonal variations of major chemical compositions in $PM_{2.5}$. The 5th, 25th, median, 75th, and 95th percentiles are shown in the figure.

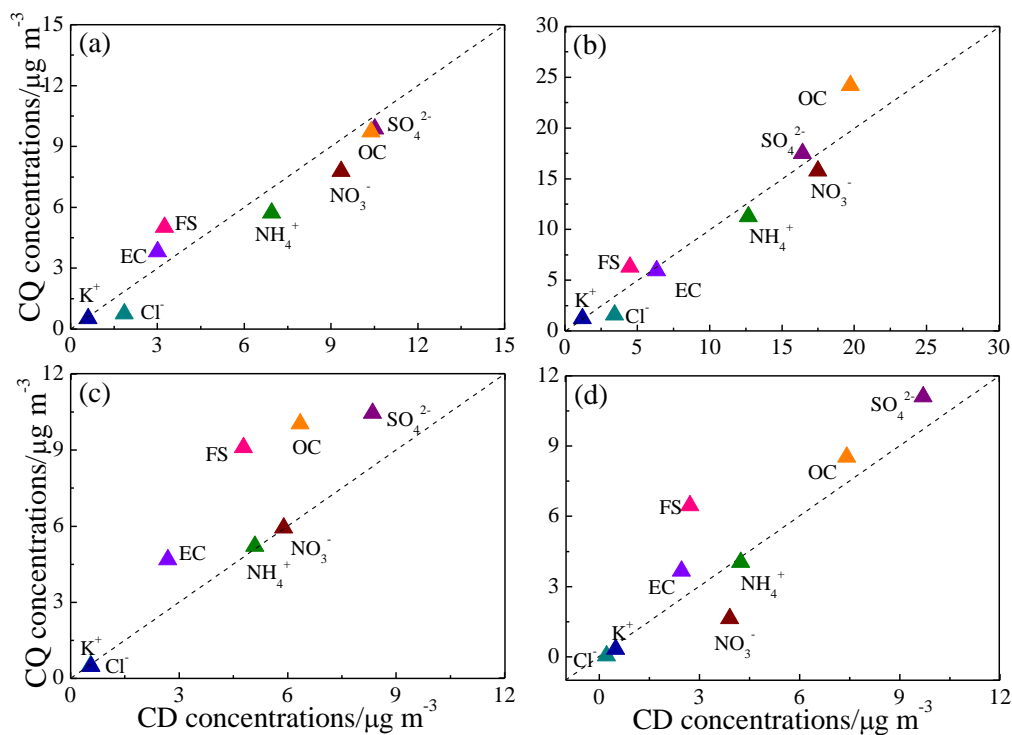


Figure 4. Seasonal mean concentrations of major components in autumn (a), winter (b), spring (c), and summer (d) at CD and CQ sites.

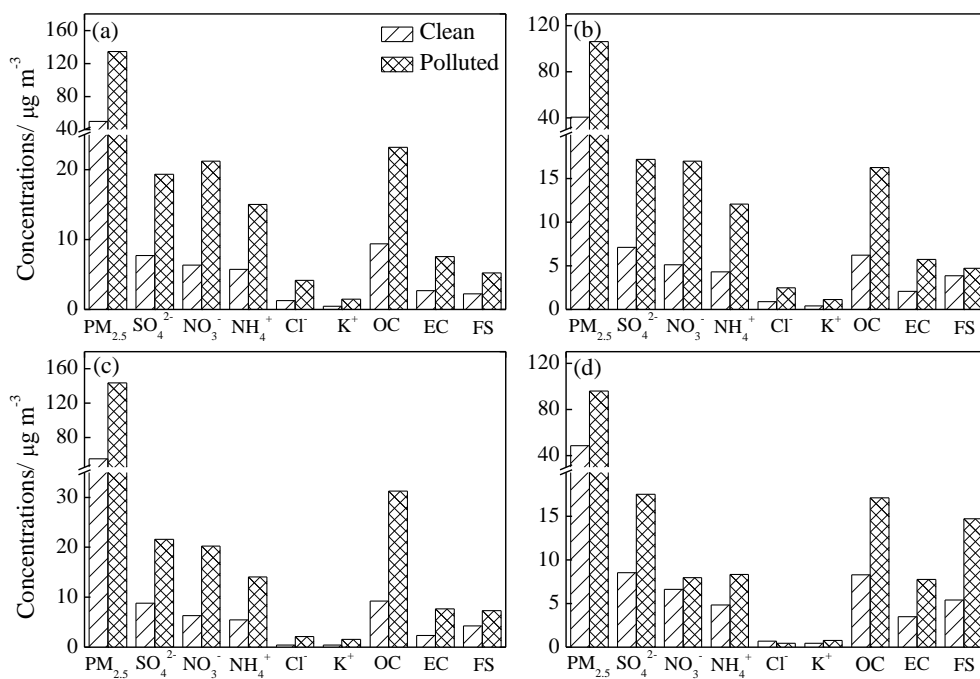


Figure 5. $PM_{2.5}$ and major chemical components on clean and polluted days in the cold (left column) and warm (right column) seasons at CD (upper row) and CQ (lower row).

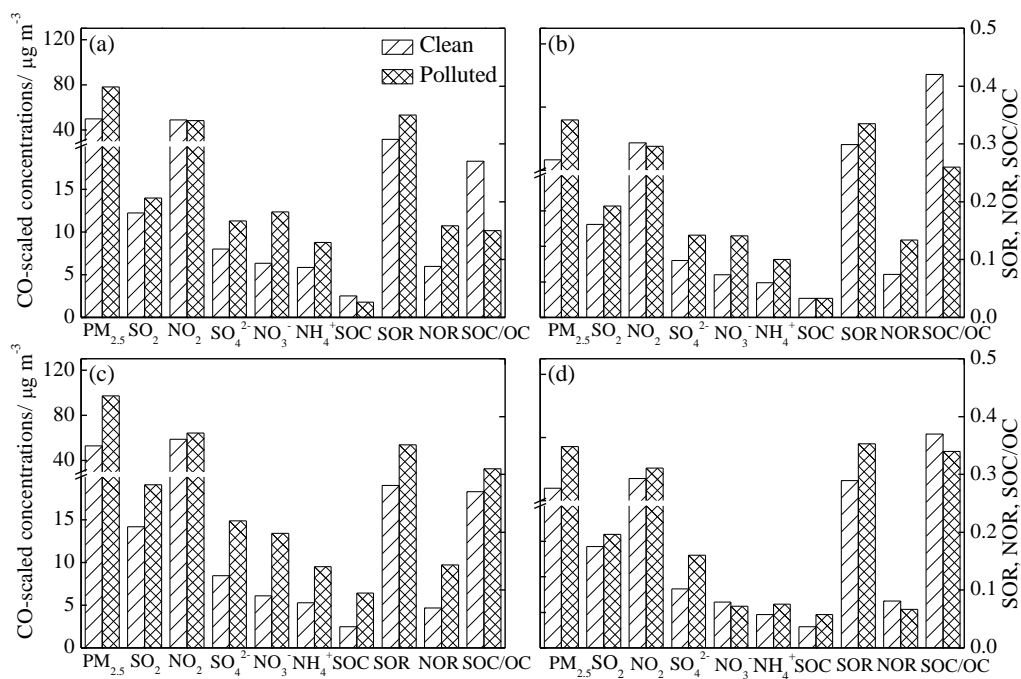


Figure 6. CO-scaled concentrations of various pollutants and the values of SOR, NOR, and SOC/OC in the cold (left column) and warm (right column) seasons at CD (upper row) and CQ (lower row).

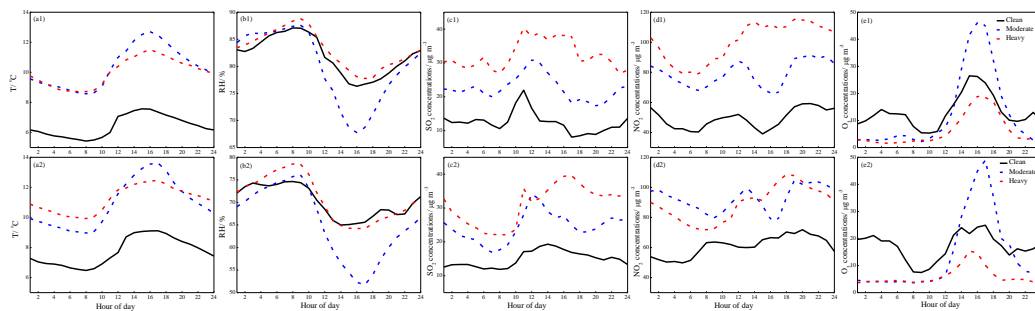


Figure 7. Diurnal variations of T, RH, SO₂, NO₂, and O₃ under different pollution-level conditions in the cold season at CD (upper row) and CQ (lower row). Clean days: PM_{2.5} ≤ 75 µg m⁻³; moderate polluted days: 75 < PM_{2.5} ≤ 150 µg m⁻³; heavy polluted days: PM_{2.5} > 150 µg m⁻³.

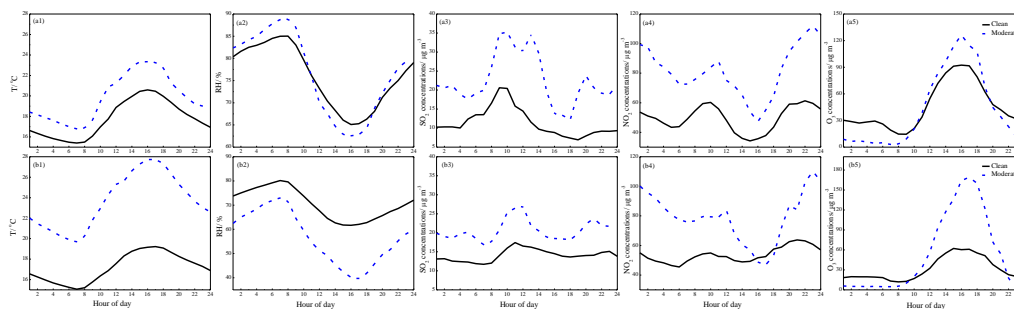


Figure 8. Same as in Figure 7 except in the warm season and only for two pollution-levels.

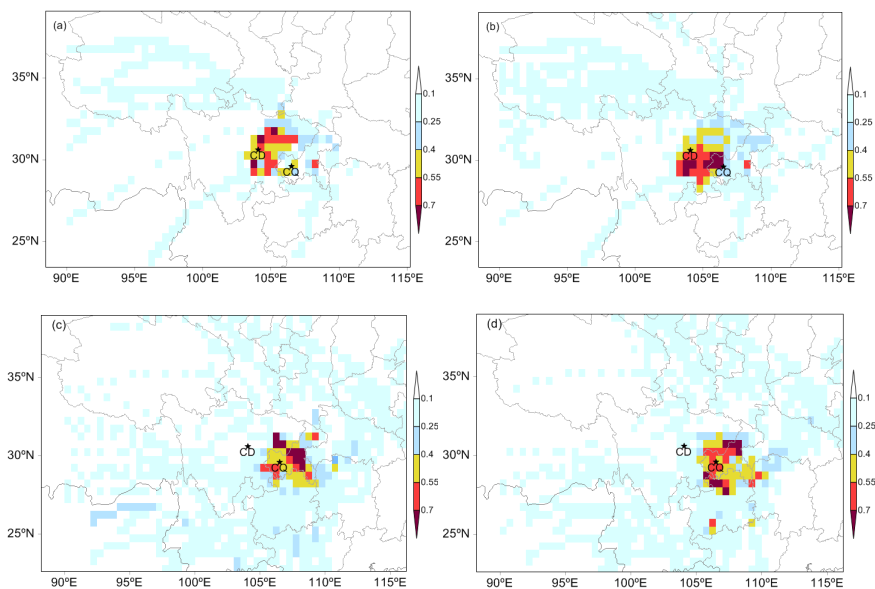


Figure 9. PSCF maps of PM_{2.5} in the cold (left column) and warm (right column) seasons at CD (upper row) and CQ (lower row).