

# Response to comments

Manuscript Number: acp-2017-23

Title: *In situ chemical measurement of individual cloud residue particles at a mountain site, South China. Qin hao Lin et al.*

Received and published: 20 March 2017

## Referee #2: J. Schneider

*In their manuscript "In situ chemical measurement of individual cloud residue particles at a mountain site, South China", Qin hao Lin and co-workers report on the analysis of single particles from cloud residues using a single particle aerosol mass spectrometer. They observed a high fraction of EC-containing particle in the residuals and detected amines with a high variability. Nitrate was found to be increased in residuals compared to ambient particles, while sulfate showed a dependency on the chemical composition of the residues. The topic of the paper is well suited for ACP, and the data itself are interesting, because single particle measurements of cloud residuals are still sparse. However, the manuscript suffers from many unclear statements and some severe uncertainties regarding the analysis of interstitial particles. I have many points where more information is needed or where I disagree. None of these points alone would be a "major" comment, but the multitude of my remarks and questions suggest to require a major revision and to reconsider the manuscript after my comments listed below have been addressed.*

*We would like to thank Prof. J. Schneider for his useful comments and recommendations to improve the manuscript. We agree with the comments, and careful revision has been made accordingly, please refer to the following responses for details.*

Comments and remarks:

*Title: I suggest to change the title to "In situ chemical composition measurement of individual cloud residue particles at a mountain site, South China"*

*We have changed accordingly. Please refer to Lines 1-2 of the revised manuscript.*

*Page 4, lines 64 – 68:*

*More references are needed here to discuss the anthropogenic influence on cloud particles, not just two papers on single particle analysis.*

37

38 As also suggested by Reviewer 1, We have added references (Stier et al., 2005; Sorooshian  
39 et al., 2007b; Lohmann et al., 2007; Rosenfeld et al., 2008; Roth et al., 2016; Seinfeld. et  
40 al., 2016; Li et al., 2017) to discuss the anthropogenic influence on cloud particles.  
41 Anthropogenic particles can increase number concentration of small cloud droplets, in turn,  
42 affect reflectivity and life time of clouds (Rosenfeld et al., 2008; Stier et al., 2005;  
43 Lohmann et al., 2007). In-situ cloud chemical measurements show varied chemical  
44 composition of cloud droplets at various regions (Sorooshian et al., 2007a; Roth et al., 2016;  
45 Li et al., 2017). Although a large number of aerosol/cloud studies over the past 20 years,  
46 the uncertainty for evaluating radiative forcing due to aerosol-cloud interactions has not  
47 been reduced (Seinfeld. et al., 2016). Please refer to Lines 55-62 of the revised manuscript.

48

49

50 *Page 5, line 79: Replace "Nf of sulfate" by "NF of sulfate-containing particles"*

51

52 We have changed it accordingly. Please refer to Line 74 of the revised manuscript.

53

54 *Page 5, line 80: Replace "other study" by "other studies"*

55

56 We have changed it accordingly. Please refer to Line 76 of the revised manuscript.

57

58 *Page 7, line 122: Was the humidity measured in the evaporation chamber? How do you*  
59 *make sure that all water evaporates?*

60

61 Relative humidity (RH) was around 30% in the evaporation chamber, thus it can be  
62 assumed that the majority of water was evaporated. Please refer to Lines 131-134 of the  
63 revised manuscript.

64

65 *Page 8, lines 147-149: Did you do the size calibration on the mountain top station? What*  
66 *was the ambient pressure during the measurements and during the calibration? Did you*  
67 *check the inlet flow or the pressure inside the aerodynamic lens?*

68

69 We did the size calibration on the mountain top station. Polystyrene latex spheres  
70 (Nanosphere Size Standards, Duke Scientific Corp., Palo Alto) of 0.2-2.0  $\mu\text{m}$  in diameter  
71 were used to calibrate the sizes of the detected particles on the mountain top station. The  
72 ambient pressure was 830 hPa (826-842 hPa) during the measurements and during the

73 calibration. The pressure inside the aerodynamic lens maintains about 3 hPa during the  
74 measurements and during the calibration. Please refer to Lines 152-155 of the revised  
75 manuscript.

76

77 *Page 8, line 161: The SMPS does not measure the cloud droplet concentration but the*  
78 *cloud residue concentration. Cloud droplets would have to be measured outside in the*  
79 *cloud (by FSSP or similar instrumentation).*

80

81 We have corrected the mistake. “cloud droplet concentration” was replaced with “cloud  
82 residual concentration”. Please refer to Lines 165-168 of the revised manuscript.

83

84 *Page 9, line 171: As I will outline in more detail below, I doubt the existence of interstitial*  
85 *particles in this size range.*

86

87 The period of collecting interstitial particles on 22-23 Jan encountered initial mixing of  
88 northerly cloud-free air (dry and cold airstreams) and southwesterly cloudy air (moist  
89 airflows). The dry northern air mass might lower supersaturation, only larger particles  
90 could be activated. This might result in non-activated particles observed to be above 200  
91 nm here (Mertes et al., 2005; Kleinman et al., 2012; Hammer et al., 2014). To make it more  
92 accurate, we prefer to name “non-activated particles”, rather than “interstitial particles”.  
93 We have clarified them. Please refer to Lines 452-461 of the revised manuscript.

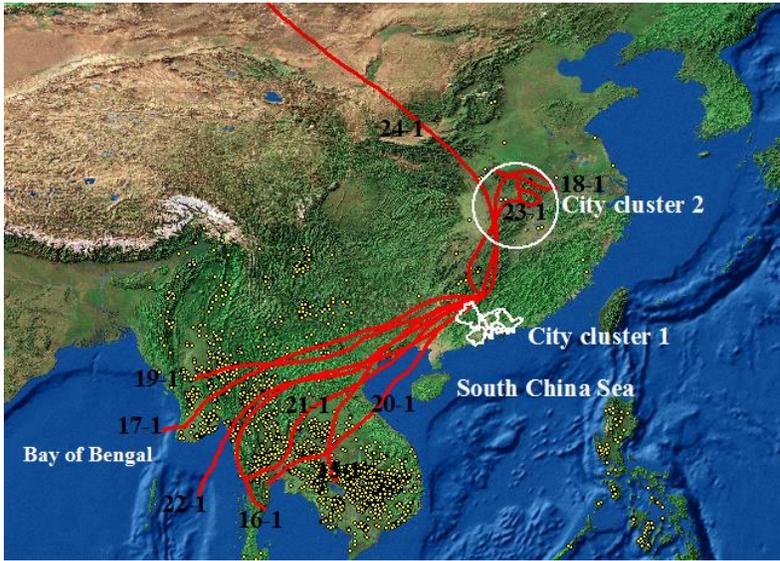
94

95 *Page 9 lines 184-185 and Figure 2: More info on the trajectories is needed: How did the*  
96 *vertical evolution look like? How well is the mountain represented in the model? Is 1800*  
97 *m the best altitude that represents the mountain site? Please add also the most important*  
98 *megacities to the map to help estimating the influence of anthropogenic emissions.*

99

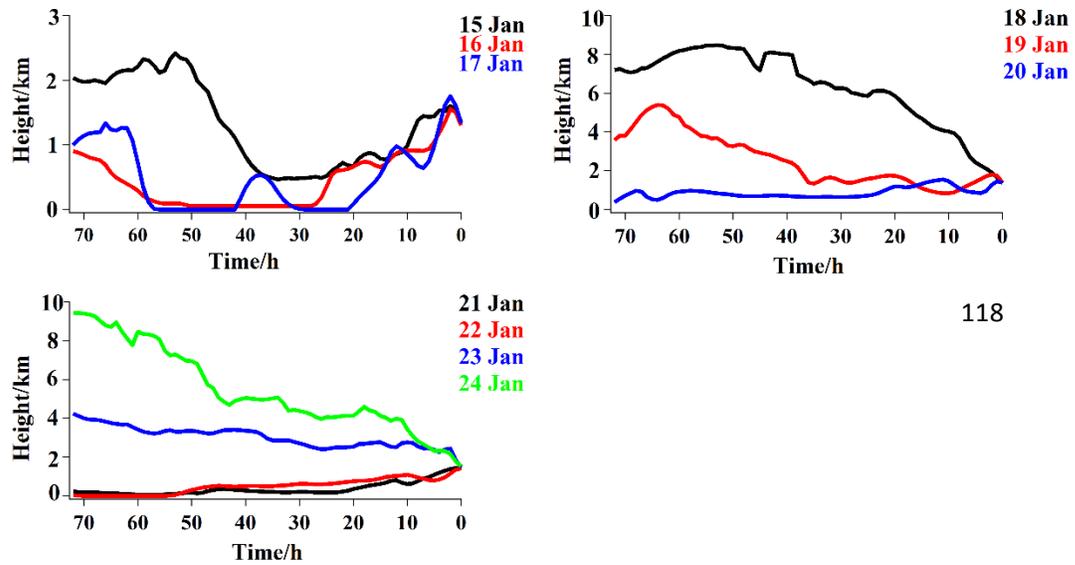
100 We have added the vertical evolution of the trajectories. The beginning of southwesterly  
101 air masses traversed at lower heights relative to northerly air masses. Please refer to the  
102 Figure 2 (b). Heights of the HYSPLIT model in the study region (a spatial resolution of  
103  $0.5^\circ \times 0.5^\circ$ ) was averaged 500 m a.s.l, which was lower than height of the observed site  
104 (1,690 m a.s.l). Therefore, a height of 1,800 m a.s.l. (approximately 100 m above the  
105 observed site) was used as an endpoint in the model. Continental air masses crossed  
106 industrial areas where located in the Yangtze River Mid-Reaches city cluster (Figure 2a).  
107 The site was possibly affected by industrial emissions under the influence of continental  
108 air masses. Please refer to Lines 189-196 of the revised manuscript.

109 (a)



110

111 (b)



118

124 Figure 2: (a) HYSPLIT back trajectories (72 h) for air masses at 1,800 m during the whole  
125 sampling period. The white borders and circle refers to the Pearl River Delta (city cluster  
126 1) and Yangtze River Mid-Reaches city cluster (city cluster 2), respectively. The yellow  
127 dots represent fire dots during the study periods. The fire dots are available at  
128 <https://earthdata.nasa.gov/>; (b) Heights (above model ground) of the air masses as a  
129 function of time.

130

131

132

133 Page 10, line 207: I suggest moving Figure S3 to the main paper.

134

135 Figure S3 has been moved to the main paper, please refer to Figure 4 in the revised  
136 manuscript.

137

138 Page 11, lines 221-222: But Roth et al. found a clear enhancement of amines in residues  
139 compared to the background aerosol (9% to about 2%).

140

141 We have added a comparison Nf of amine-containing particles between cloud residues and  
142 background aerosol reported by Roth et al., (2016). Please refer to Lines 265-267 of the  
143 revised manuscript.

144

145 Page 12, lines 235-237: I agree that dust is found more frequently in the coarse particle  
146 size range, but then I would expect to see an increase of the dust fraction in the residues  
147 with increasing diameter. This is not seen in Figure S3. Is the identification of dust reliable?  
148 What about the Fe-containing particles? They might be dust as well.

149

150 We agree with the comment. As a matter of fact, Nf of dust cloud residues generally  
151 increased with increasing diameter. Please refer to Lines 276-278 of the revised manuscript.

152

153 Approximately 16% of the Fe cloud residues contained Ca peak (m/z 40). Relatively weak  
154 Na and K peaks in the Fe particles possibly contributes to anthropogenic sources (Zhang  
155 et al., 2014), especially northern air mass across iron/steel industrial activities in Yangtze  
156 River Mid-Reaches city clusters (Figure 2). These might suggest that Fe cloud residues  
157 was likely to have come from mixed sources. Please refer to Lines 290-294 of the revised  
158 manuscript.

159

160

161

162

163

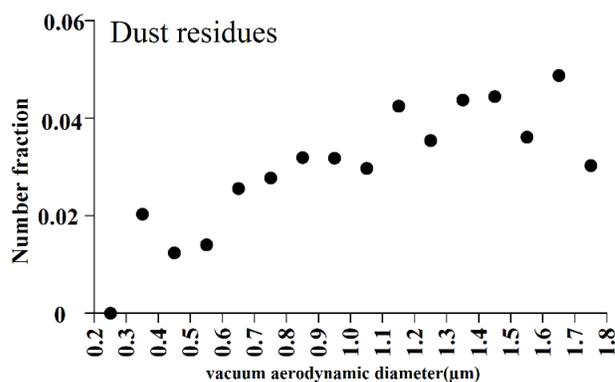
164

165

166

167

168



169 *Page 12, lines 238-243: What could be the source of these Pb- and Fe-containing particles?*  
170 *See also comment above. Can the Fe-containing particles belong to the dust-type?*

171

172 As mentioned above, the Fe cloud residues contain Ca peak (m/z 40, 16% by number) and  
173 relatively weak Na and K peaks, which possibly contributes to anthropogenic sources  
174 (Zhang et al., 2014), especially northern air mass across iron/steel industrial activities in  
175 Yangtze River Mid-Reaches city clusters (Figure 2). These might suggest that this particle  
176 type likely originated from mixed sources. Please refer to Lines 290-294 of the revised  
177 manuscript.

178

179 The Pb particles showed its typical ions at m/z 208Pb<sup>+</sup> and internally mixed with K and Cl.  
180 Previous studies found that K and Cl internally mixed with Pb particles have a possible  
181 origination of waste incineration (Zhang et al., 2009) or iron and steel facility (Tsai et al.,  
182 2007). Please refer to Lines 315-318 of the revised manuscript.

183

184 *Page 12, lines 244-252: If these particles were from sea salt, they should contain chloride*  
185 *ions. That is hard to see in Figure 3. Are these Na-rich particles correlated with air masses*  
186 *coming from the ocean?*

187

188 Na-rich particles result from varied sources of industrial emissions or sea salt particles and  
189 dry lake beds (Moffet et al. 2008). The Nf of the Na-rich cloud residues did not increase  
190 from continental (Northerly) air mass to maritime (southwesterly) air mass on 21 Jan (3.3%  
191 versus 2.4% by number). However, sea salt ion peak areas (m/z, 81/83Na<sub>2</sub><sup>35</sup>Cl/Na<sub>2</sub><sup>37</sup>Cl)  
192 were enhanced for Na-rich particles origination from maritime air mass relative to  
193 continental air mass (3.8 ±12.4 times). Continental air masses crossed industrial areas  
194 where located in the Yangtze River Mid-Reaches city cluster (Figure 2). Industrial  
195 emissions was a possible contributor to Na-rich particles under the influence of continental  
196 air masses (Wang et al. 2016). This might suggests that the Na-rich particles were  
197 contributed by both the industrial emissions and sea salt. Therefore, under the influence of  
198 maritime air mass, the signals for sea salt contribution became stronger. Please refer to  
199 Lines 299-310 of the revised manuscript.

200

201 *Page 13, line 259 and Figure 5: How do you distinguish between sulfuric acid and sulfate?*  
202 *Besides, spelling (sulfate, sulphuric acid) should be consistent ("f" or "ph").*

203

204 Sulfate ion peak at m/z -97 HSO<sub>4</sub><sup>-</sup> and sulfuric acid cluster ion peak at m/z -195 [H(HSO<sub>4</sub>)<sub>2</sub>]

205 were given in previous single particle studies (Pratt et al., 2009; Rehbein et al., 2011).  
206 "Sulphuric acid" has been replaced by "Sulfuric acid". Please refer to the caption of Figure  
207 5.

208

209 *Page 13, lines 265-267: What other forms of nitrate do you suggest to be present on the*  
210 *Na-rich and dust residues? What about uptake of nitric acid from the gas-phase by the*  
211 *cloud droplets? How certain is the identification of ammonium? Which peak was used?*

212

213 The Na-rich and Dust types were mainly composed of alkaline ion peaks ( $m/z$ ,  $23\text{Na}^+$ ,  $39\text{K}^+$   
214 and  $40\text{Ca}^+$ ) in the position mass spectra (Figure 3). This suggests that rather than  $\text{NH}_4\text{NO}_3$ ,  
215 nitrate might exist in the form of  $\text{Ca}(\text{NO}_3)_2$ ,  $\text{NaNO}_3$  or  $\text{KNO}_3$  in the dust and Na-rich cloud  
216 residues. Please refer to Lines 340-344 of the revised manuscript.

217

218 We agree with the comment. We have discussed the contribution of uptake of gas-phase  
219  $\text{HNO}_3$  to enhanced nitrate in the cloud residues and cited Schneider et al. (2017). Please  
220 refer to Lines 334-335 of the revised manuscript.

221

222 Generally, a  $\text{NH}_4^+$  ion signal ( $m/z$  18) was used for identification of ammonium in the  
223 analysis of single particle mass spectrometry (Pratt et al., 2009). Please refer to Line 338  
224 of the revised manuscript.

225

226 *Page 13, line 268-272: The stability of ammonium nitrate depends also on the humidity. In*  
227 *the book by Seinfeld and Pandis (2nd edition, Wiley and Sons, 2006, Chapter 10.4.3) it is*  
228 *shown that at 30% RH ammonium nitrate does not exist above 30 C. I would assume that*  
229 *the dry carrier gas in the evaporation section is below 30% RH. Thus, it may well be that*  
230  *$\text{NH}_4\text{NO}_3$  evaporates in your system.*

231

232 We agree with the comment. We have clarified the artificial effect on ammonium nitrate in  
233 the cloud residues. Please refer to Lines 344-346 of the revised manuscript.

234

235 *Page 13, lines 275-276: The sentences "The presence of abundant sulfate in aged EC cloud*  
236 *residues was considered to be a good CCN species before activation:" needs rephrasing.*  
237 *It is not clear to me what you want to say. Do you mean "aged EC particles mixed with*  
238 *sulfate are good CCN"?*

239

240 The sentence has been changed to "aged EC particles mixed with sulfate are good CCN".

241 Please refer to Lines 350 of the revised manuscript.

242

243 *Page 13, line 279: Ammonium will most likely play a key role in the form of ammonium*  
244 *sulfate or ammonium nitrate. In organic particles, amines may play that role*  
245 *(methylamines). Again: how do you identify ammonium and how do you distinguish*  
246 *between amines and ammonium?*

247

248 We agree with the comment. Ammonium will most likely play a key role in the form of  
249 ammonium sulfate or ammonium nitrate in the OC and aged EC cloud residues (Zhang et  
250 al., 2017). Please refer to Lines 353-354 of the revised manuscript.

251

252 A  $\text{NH}_4^+$  ion signal (m/z 18) was used for identification of ammonium. The existences of  
253 m/z 59  $\text{N}(\text{CH}_3)_3^+$  (trimethylamine, TMA) and related amine ion signals m/z  
254  $58\text{C}_2\text{H}_5\text{NHCH}_2^+$  (diethylamine, DEA) and m/z  $86\text{C}_5\text{H}_{12}\text{N}^+$  (triethylamine, TEA) were used  
255 for identification of amines (Angelino et al., 2001). Please refer to Lines 254-257 and 338  
256 of the revised manuscript.

257

258 *Page 14, line 281 (and Figure 5): Why does oxalate nor correlate with OC?*

259

260 Classification of the OC particles mainly based on intense organic carbon ion signals (e.g.,  
261 m/z  $27\text{C}_2\text{H}_3^+$ ,  $37\text{C}_3\text{H}^+$ ,  $43\text{C}_2\text{H}_3\text{O}^+$  and  $51\text{C}_4\text{H}_3^+$ ). However, majority of oxalate-containing  
262 particles internally mixed with the K-rich type. Therefore, oxalate was classified to the K-  
263 rich type, probably contributed from biomass burning. Noted that K-rich could contain a  
264 large abundant of organics (Pratt et al. 2011), however, the signals of organics were covered  
265 by the potassium due to its high sensitive to the laser. Please refer to Lines 371-378 of the  
266 revised manuscript.

267

268 *Page 14, lines 284-285: What do you mean by "enrichment of TMA in amine cloud*  
269 *residuals"? You observe that in 93% of those cloud residuals that are assigned to the*  
270 *"amine" type contain TMA. That is not surprising, more surprising is that it's not 100%.*  
271 *But that's inside the measurement uncertainties, to my opinion.*

272

273 We have changed "enrichment of TMA in amine cloud residuals" to "presence of TMA in  
274 amine cloud residuals". Please refer to Lines 360-361 of the revised manuscript.

275

276 Amine family signals m/z  $58\text{C}_2\text{H}_5\text{NHCH}_2^+$  and m/z  $86\text{C}_5\text{H}_{12}\text{N}^+$  were also selected to

277 identify amines (Angelino et al., 2001), leading to only 93% of the "amine" residues  
278 containing TMA. Particles that exist a peak signal  $m/z$  58C<sub>2</sub>H<sub>5</sub>NHCH<sub>2</sub><sup>+</sup>, were found to  
279 account for 99% of the Amine residues. Please refer to Lines 254-257 of the revised  
280 manuscript.

281

282 *Page 14, lines 294-295: "This may result in 33% by number to the Amine residues*  
283 *containing oxalate." Please rephrase, not clear what you want to say.*

284

285 We have rephrased this sentence to "it may facilitate the entrainment of oxalate (33% by  
286 number) in the Amine residues. ". Please refer to Lines 370-371 of the revised manuscript.

287

288 *Page 14, line 298: What does "unscaled" mean? These are absolute particle numbers.*

289

290 Considering that the SPAMS mainly detected in size range 0.2-2.0  $\mu\text{m}$  and has size-  
291 dependent transmission efficiency. Detected particles were not corrected by a SMPS.  
292 Therefore, detected particles cannot represent real atmospheric particle level. "unscaled"  
293 has been changed to "detected particle counts". Please refer to Line 381 of the revised  
294 manuscript.

295

296 *Page 14, lines 302-303: You say that the air masses change from northerly on 18 Jan to*  
297 *southwesterly on 19 Jan, but the particles remain similar from 17 Jan (around noon) to 20*  
298 *Jan (noon). On the other hand, the change in particle types is very abrupt from cloud*  
299 *residuals to ambient on Jan 17.*

300

301 Southwesterly wind flow on 19-20 Jan was too weak ( $\sim 2.75 \text{ m s}^{-1}$ ) to dilute particles  
302 originated from northerly air masses (Figure 1). Additionally, high RH (90%) contour line  
303 at height 1,500 m (a.s.l.) gradually moved to north China from 19 to 20 Jan (Figure S5).  
304 These might lead to similar residual particle types observed from 19 Jan to 20 Jan, although  
305 the site encountered southwesterly cloudy air on 19-20 Jan (Figure 2). Please refer to Lines  
306 400-404 of the revised manuscript.

307

308 Ambient RH showed an abrupt decrease from nearly 100% at 10:00 to 85% at 11:00 on 17  
309 Jan (Figure 1). The entrained particles originated from northern air mass might have  
310 insufficient supersaturation to activate as cloud droplets. It leads to a very abrupt change  
311 in Nf of particle types from cloud residues to ambient particles on Jan 17. Please refer to  
312 Lines 382-390 of the revised manuscript.

313

314

315 *Page 15, lines 322-325: Do verify the possible transport of biomass burning particles to*  
316 *the site, the vertical history of the trajectories is required.*

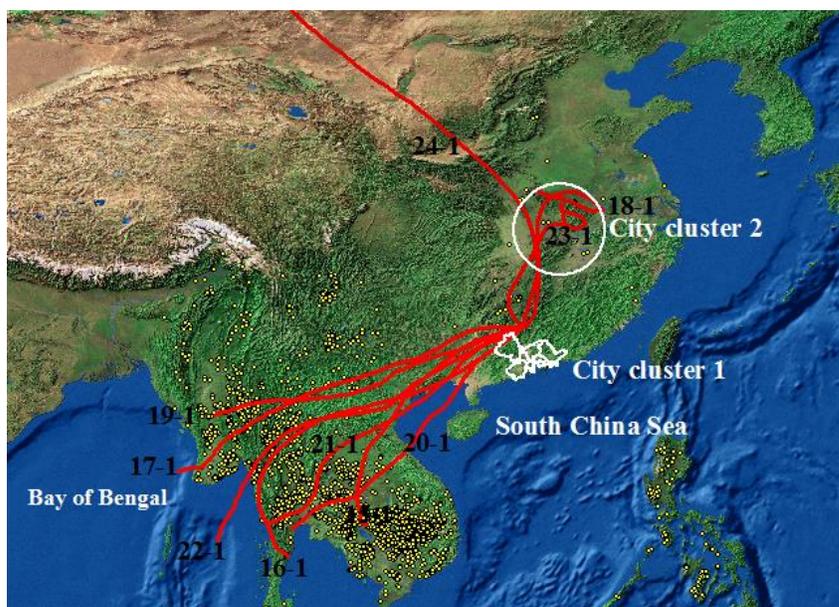
317

318 We have added the vertical evolution of the trajectories. The beginning of trajectories  
319 traversed at low heights (about 0-2 km above model ground) of Southeast Asia, where  
320 abundant fire dots occurred. Please refer to the Figure 2 (b).

321

322

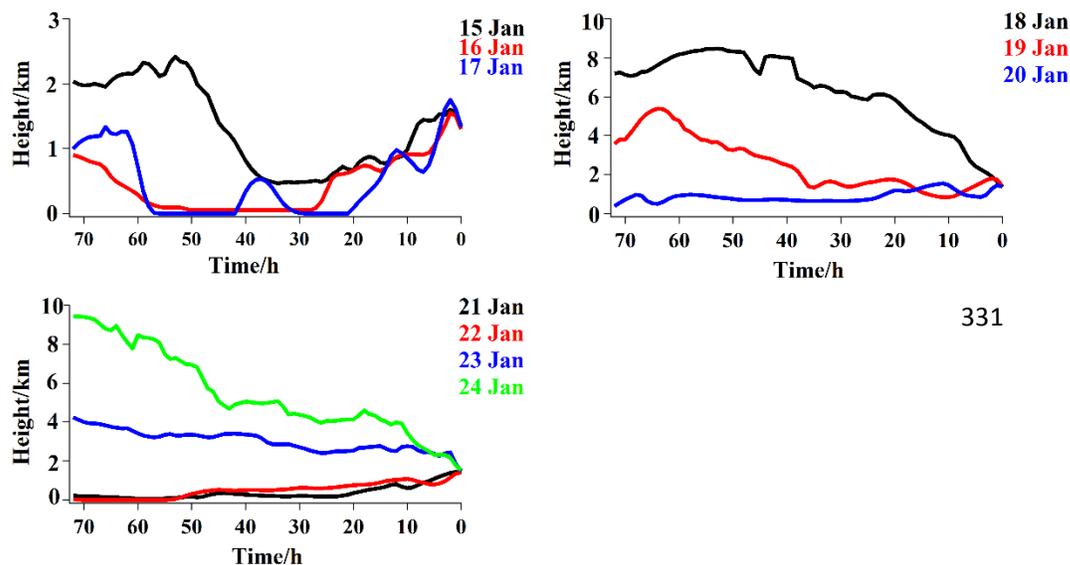
(a)



323

324

(b)



331

337 Figure 2: (a) HYSPLIT back trajectories (72 h) for air masses at 1,800 m during the whole  
338 sampling period. The white borders and circle refers to the Pearl River Delta (city cluster  
339 1) and Yangtze River Mid-Reaches city clusters (city cluster 2), respectively. The yellow  
340 dots represent fire dots during the study periods. The fire dots are available at  
341 <https://earthdata.nasa.gov/>; (b) Heights (above model ground) of the air masses as a  
342 function of time.

343

344 *Page 16, lines 337-229: "Note that after the activation of amine particles, the partitioning*  
345 *of the gas amine on cloud droplets may further contribute to the enhanced Amine cloud*  
346 *residues". That is true, but holds also for other species, as nitrate (HNO<sub>3</sub>) or water-soluble*  
347 *OC.*

348

349 We have strengthened the important contribution of uptake of gaseous HNO<sub>3</sub> or water-  
350 soluble OC to cloud droplets. Please refer to Lines 334-335, 364-366 and 490-494 of the  
351 revised manuscript.

352

353 *Section 3.5: Here I have a major concern: You report interstitial particles containing*  
354 *sulfate and nitrate in the size range between 200 and 1300 nm. It is very hard to believe*  
355 *(not to say impossible) that such large particles are not activated in a cloud.*

356

357 *Later (page 17, lines 366-368) you write " However, few studies have focused on this issue,*  
358 *in part because interstitial particles show a smaller size than that detected by single-*  
359 *particle mass spectrometry (Roth et al., 2016)." Since the SPAMS has a very similar lower*  
360 *detection size range as the ALABAMA used by my group in Roth et al., 2016), you can not*  
361 *expect that you detect non-activated interstitial particles which should be in the size range*  
362 *below 200 nm.*

363 *My suspicion is: The clouds became thinner, and entrainment of cloud-free air has mixed*  
364 *"normal" aerosol particles into the cloud. But such particles cannot be referred to as*  
365 *"interstitial". As long as you don't have cloud microphysics (number and size of cloud*  
366 *droplets) or at least liquid water content (Particle Volume Monitor) available, I would*  
367 *suggest to remove this chapter on interstitial particles.*

368

369 The period of collecting interstitial particles on 22-23 Jan encountered initial mixing of  
370 northerly cloud-free air (dry and cold airstreams) and southwesterly cloudy air (moist  
371 airflows). The dry northern air mass might lower supersaturation, only larger particles  
372 could be activated. This might result in above 200 nm non-activated particles observed

373 here (Mertes et al., 2005; Kleinman et al., 2012; Hammer et al., 2014). To make it more  
374 accurate, we prefer to name “non-activated particles”, rather than “interstitial particles”.  
375 We have clarified them. Please refer to Lines 453-461 of the revised manuscript.

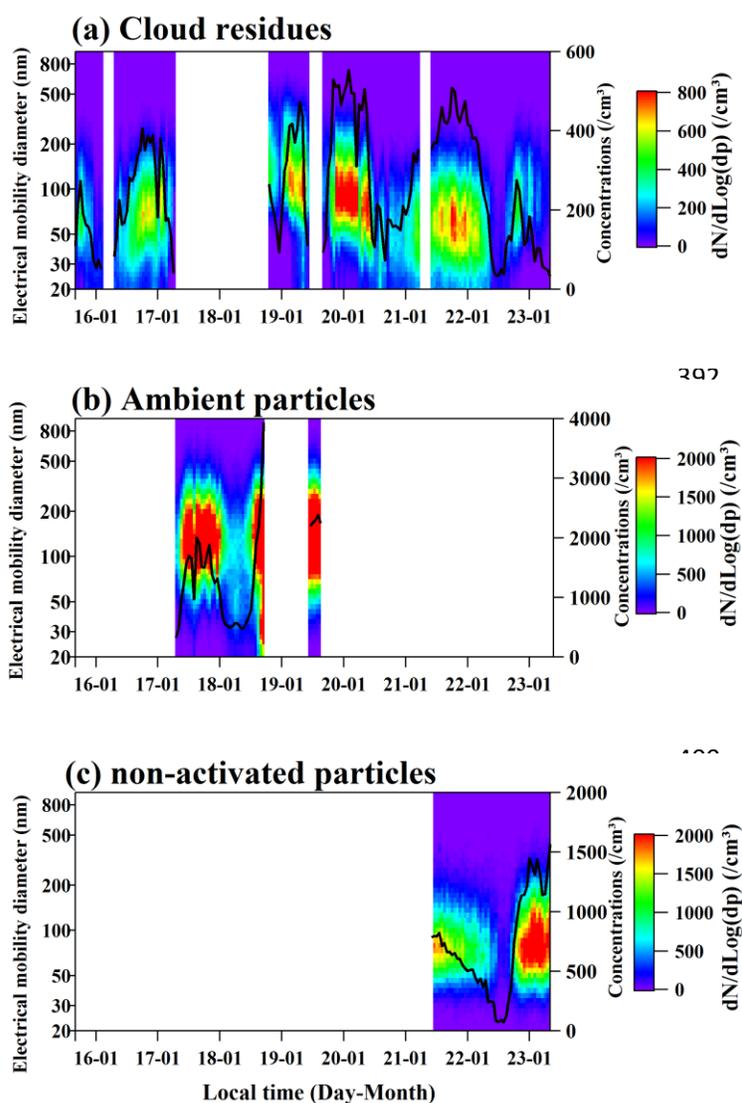
376

377 *Page 17, line 358 / Table 1: I would prefer a graph with bars or pie charts. I also strongly*  
378 *recommend showing the SMPS size distributions from residues, ambient and interstitial.*  
379 *That might help to identify the issues with the large interstitial particles.*

380

381 Table 1 was replaced by pie charts. Please refer to Figure 7 in the revised manuscript. The  
382 SMPS size distributions from residues, ambient and interstitial particle were provided in  
383 Figure S2.

384



408 Figure S2: Size distribution (electrical mobility diameter 20-900 nm) of cloud residues (a),

409 ambient (b) and non-activated (c) particles were measured a scanning mobility particle  
410 sizer (SMPS). Black lines represent particles concentrations integrated by the SMPS. The  
411 data of cloud residual concentrations was corrected by enrichment factor of 5.25.

412

413 *Page 18, lines 374-375 / Figure 7 & 8: How are the difference mass spectra of Figure 7*  
414 *and 8 calculated? Is it ambient - residues and interstitial – residues? Or vice versa? How*  
415 *were the spectra normalized? Please explain.*

416

417 We have provide the information in the captions of Figures 8 and 9 of the revised  
418 manuscript. Please refer to Lines 484-485 of the revised manuscript.

419

420 Figures 8: Mass spectral subtraction plot of the average mass spectrum corresponding to  
421 cloud residues minus ambient particles. Positive area peaks correspond to higher  
422 abundance in cloud residues, whereas negative area peaks show higher intensity in ambient  
423 particles.

424

425 Figure 9: Mass spectral subtraction plot of the average mass spectrum corresponding to  
426 cloud residues particles minus non-activated particles. Positive area peaks correspond to  
427 higher abundance in cloud residues, whereas negative area peaks show higher intensity in  
428 non-activated particles.

429

430

431 *Page 18, lines 376-382: Why not? I drew the same conclusion as Hayden et al. (2008) in*  
432 *my 2017 paper (Schneider et al., 2017, please note the update from ACPD 2016 to ACP*  
433 *2017). HNO<sub>3</sub> uptake may not be the source of the particles but explains the high amount*  
434 *of nitrate found on many particles, also on the Na-rich and dust particles discussed above.*

435

436 We agree with the comment. We have strengthened the contribution of uptake of gaseous  
437 HNO<sub>3</sub> to the enhanced nitrate in the cloud residues. We have update the citation of  
438 Schneider et al. (2017) from ACPD 2016 to ACP 2017. Please refer to Lines 334-335 and  
439 490-494 of the revised manuscript.

440

441 *Page 18, lines 384-386: I agree with that, but wouldn't that support the idea of uptake of*  
442 *HNO<sub>3</sub> from the gas phase? If the nitrate content does not play the major role in the*  
443 *activation, but more nitrate is found in the residues, that's an argument for HNO<sub>3</sub> uptake.*

444

445 We agree with the comment. We have strengthened the contribution of uptake of gaseous  
446 HNO<sub>3</sub> to the enhanced nitrate in the cloud residues. Please refer to Lines 334-335 and 490-  
447 494 of the revised manuscript.

448

449 *Page 18, lines 387-388: Can intensity simply be compared like this? What about size effects*  
450 *and matrix effects? But again, an explanation how Figures 7 and 8 were calculated might*  
451 *help here.*

452

453 We agree with the comment. Size and matrix might affect the mass spectra of single  
454 particles. Such comparison has been performed in previous single particle studies (Moffet et  
455 al., 2008; Pratt et al., 2011). In addition to comparison of certain compound's intensity, its  
456 size distribution and number fractions of cloud residues was compared with ambient or  
457 non-activated particles, to discuss size effect. Please refer to Lines 487-492 and 504-505  
458 of the revised manuscript.

459

460 Figure 8 and 9 show differences in average mass spectra for cloud residues versus ambient  
461 particles, as well as cloud residues versus non-activated particles, respectively. Intensity  
462 refers to peak area. Please refer to Lines 484-485 of the revised manuscript.

463

464 Figures 8: Mass spectral subtraction plot of the average mass spectrum corresponding to  
465 cloud residues minus ambient particles. Positive area peaks correspond to higher  
466 abundance in cloud residues, whereas negative area peaks show higher intensity in ambient  
467 particles.

468

469 Figure 9: Mass spectral subtraction plot of the average mass spectrum corresponding to  
470 cloud residues particles minus non-activated particles. Positive area peaks correspond to  
471 higher abundance in cloud residues, whereas negative area peaks show higher intensity in  
472 non-activated particles.

473

474

475 *Page 18, lines 391-392: "Compared with interstitial particles, sulfate enhanced in the Fe*  
476 *cloud residues." I think an "is" is missing here.*

477

478 We have changed "Compared with interstitial particles, sulfate enhanced in the Fe cloud  
479 residues." to "Compared with non-activated particles, sulfate was found to enhance in the  
480 Fe cloud residues.". Please refer to Lines 503-504 of the revised manuscript.

481

482 Page 19, lines 398-399: Better: "The in-cloud process has been reported to be an important  
483 pathway: : :"

484

485 We have changed accordingly. Please refer to Lines 510-511 of the revised manuscript.

486

487 Page 20, lines 421-422: The Jungfraujoch is a station located mostly in the free  
488 troposphere and in a remote region, so the biomass burning contribution can be expected  
489 to be lower than at other sites.

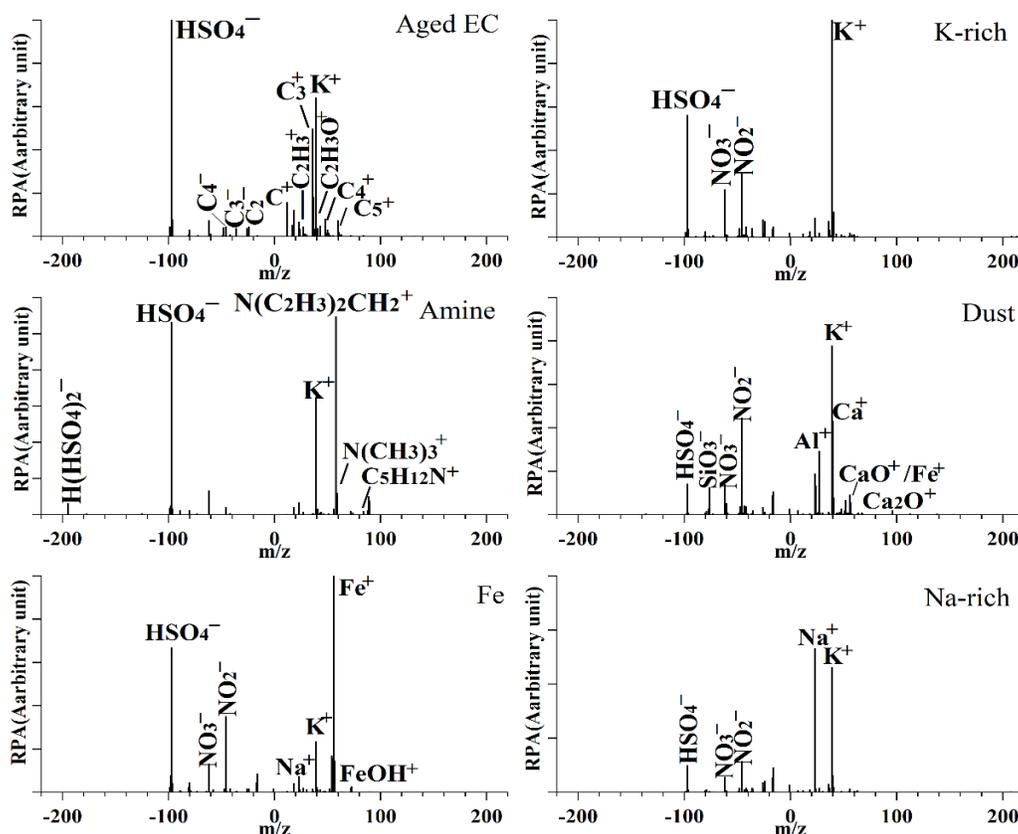
490 We agree with the comment. We have discussed less number fraction of biomass burning  
491 in the observed cloud residues at the Jungfraujoch station, where located mostly in the free  
492 troposphere and in a remote region. Please refer to Lines 251-253 of the revised manuscript.

493

494 Figures

495 Figure 3: Please improve resolution. Labels can't be read upon zooming in.

496 We have improved Figure 3 resolution. Please refer to the modified Figure 3. Average mass  
497 spectrum of Pb, OC and Other types have been moved to the supplemental information  
498 (Figure S4)



499

500 Figure 3: Averaged positive and negative mass spectra for the main 6 particle types (Aged  
501 EC, K-rich, Amine, Dust, Fe, Na-rich) of the sampled particles during the whole sampling  
502 period. RPA in the vertical axis refers to relative peak area.  $m/z$  in the horizontal axis  
503 represents mass-to-charge ratio.

504

505

506

507

508

509

510

511

512

513

514

515

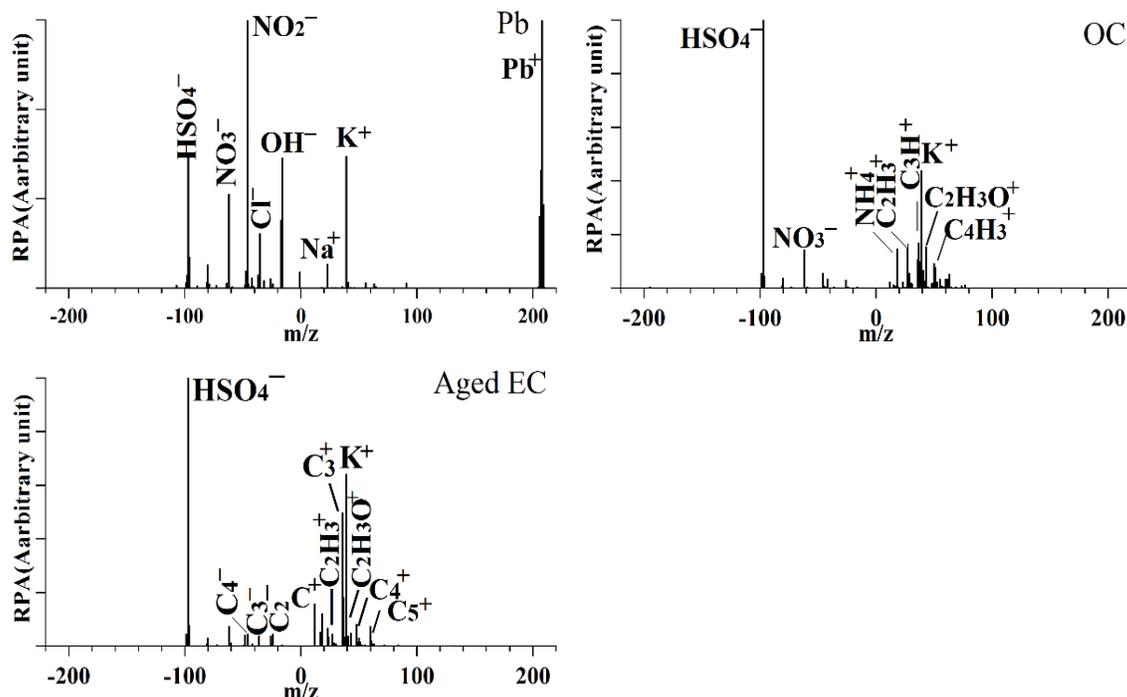
516

517

518

519

520



521 Figure S4: Averaged positive and negative mass spectra for Pb, OC and Other types of the  
522 sampled particles during the whole sampling period. RPA in the vertical axis refers to  
523 relative peak area.  $m/z$  in the horizontal axis represents mass-to-charge ratio.

524

525 *Figure 6: The ambient particle time series (b) are broader than the corresponding gaps in*  
526 *(a). Please make the Figure broader. You can move the legend with the particle types to*  
527 *above or below the graphs, plus the legend is only needed once.*

528

529 Ambient and cloud residues were collected at the same hour, which lead to ambient particle  
530 time series (b) broader than the corresponding gaps in (a). Figure 6 has been changed  
531 accordingly. Please refer to the modified Figure 6.

532

533

534

535

536

537

538

539

540

541

542

543

544

545

546

547

548

549

550

551

552

553

554

555

556

557

558

559

560

561

562

563

564

565

566

567

568

569

570

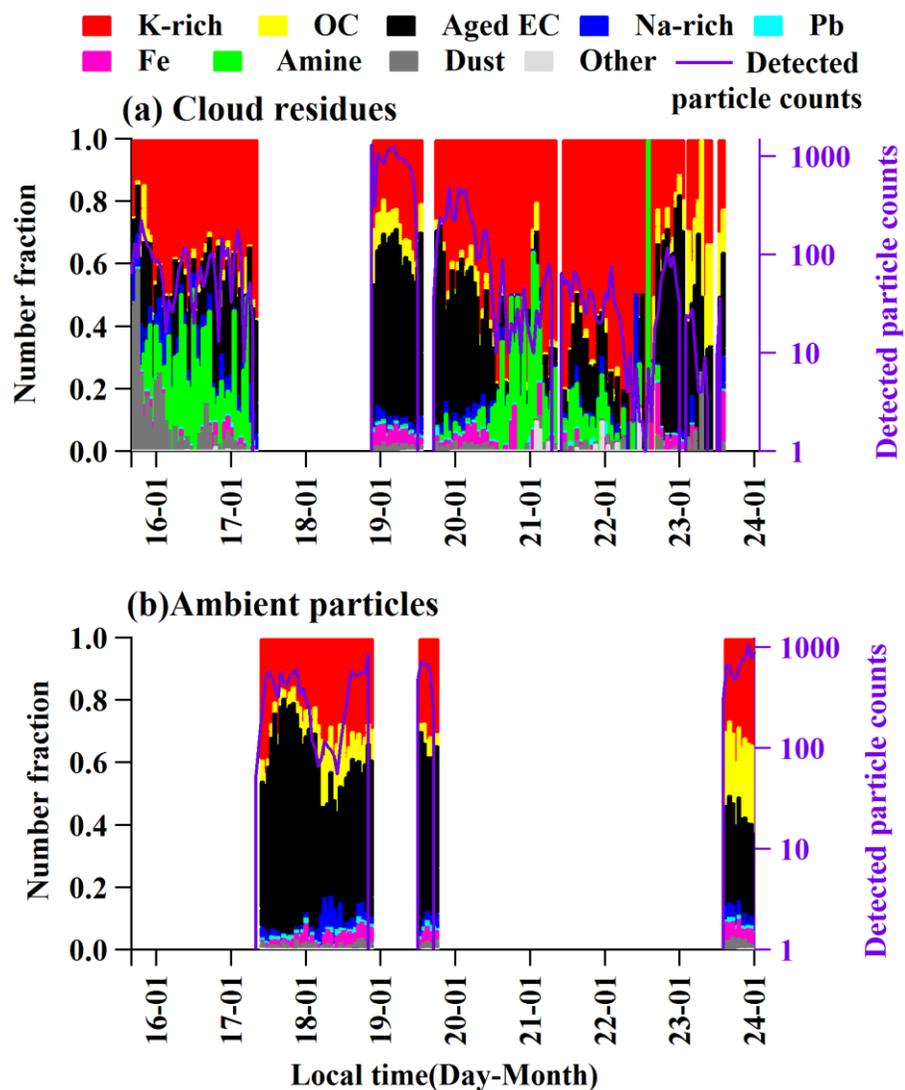


Figure 6: The hourly average variations in the cloud residual and ambient particles during the whole sampling period.

References:

Hammer, E., Gysel, M., Roberts, G.C., Elias, T., Hofer, J., Hoyle, C.R., Bukowiecki, N., Dupont, J.C., Burnet, F., Baltensperger, U. and Weingartner, E.: Size-dependent particle activation properties in fog during the ParisFog 2012/13 field campaign, *Atmos. Chem. Phys.*, 14, 10517-10533, 2014.

Kleinman, L.I., Daum, P.H., Lee, Y.-N., Lewis, E.R., Sedlacek III, A., Senum, G., Springston, S., Wang, J., Hubbe, J. and Jayne, J.: Aerosol concentration and size distribution measured below, in, and above cloud from the DOE G-1 during VOCALS-Rex, *Atmos. Chem. Phys.*, 12, 207-223, 2012.

Li, T., Wang Y., Zhou J., Wang T., Ding A., Nie W., Xue L., Wang X., and Wang W.: Evolution of trace

571 elements in the planetary boundary layer in southern China: Effects of dust storms and aerosol-cloud  
572 interactions, *J. Geophys. Res. Atmos.*, 122, doi:10.1002/2016JD025541, 2017.

573 Lohmann, U., Stier, P., Hoose, C., Ferrachat, S., Kloster, S., Roeckner, E., Zhang, J.: Cloud microphysics  
574 and aerosol indirect effects in the global climate model ECHAM5-HAM, *Atmos. Chem. Phys.*, 7,  
575 3425-3446, 2007.

576 Mertes, S., Lehmann, K., Nowak, A., Massling, A. and Wiedensohler, A.: Link between aerosol  
577 hygroscopic growth and dropletactivation observed for hill-capped clouds at connected flow  
578 conditions during FEBUKO, *Atmos. Environ.*, 39, 4247-4256, 2005.

579 Moffet, R.C., Foy, B. d., Molina, L. a., Molina, M., and Prather, K.: Measurement of ambient aerosols  
580 in northern Mexico City by single particle mass spectrometry, *Atmos. Chem. Phys.*, 8, 4499-4516,  
581 doi:10.5194/acp-8-4499-2008, 2008.

582 Pratt, K. A., Heymsfield, A. J., Twohy, C. H., Murphy, S. M., DeMott, P. J., Hudson, J. G., Subramanian,  
583 R., Wang, Z., Seinfeld, J. H., and Prather, K. A.: In Situ Chemical Characterization of Aged Biomass-  
584 Burning Aerosols Impacting Cold Wave Clouds, *J. Atmos. Sci.*, 67, 2451-2468,  
585 doi:10.1175/2010jas3330.1, 2010a.

586 Pratt, K.A., Murphy, S., Subramanian, R., DeMott, P., Kok, G., Campos, T., Rogers, D., Prenni, A.,  
587 Heymsfield, A., Seinfeld, J.: Flight-based chemical characterization of biomass burning aerosols  
588 within two prescribed burn smoke plumes, *Atmos. Chem. Phys.*, 11, 12549-12565, doi:10.5194/acp-  
589 11-12549-2011, 2011.

590 Rosenfeld, D., Lohmann, U., Raga, G.B., O'Dowd, C.D., Kulmala, M., Fuzzi, S., Reissell, A., Andreae,  
591 M.O.: Flood or drought: how do aerosols affect precipitation? *Science*, 321, 1309-1313, doi:  
592 10.1126/science.1160606, 2008.

593 Roth, A., Schneider, J., Klimach, T., Mertes, S., van Pinxteren, D., Herrmann, H., and Borrmann, S.:  
594 Aerosol properties, source identification, and cloud processing in orographic clouds measured by  
595 single particle mass spectrometry on a central European mountain site during HCCT-2010, *Atmos.*  
596 *Chem. Phys.*, 16, 505-524, doi:10.5194/acp-16-505-2016, 2016.

597 Schneider, J., Mertes, S., van Pinxteren, D., Herrmann, H., and Borrmann, S.: Uptake of nitric acid,  
598 ammonia, and organics in orographic clouds: Mass spectrometric analyses of droplet residual and  
599 interstitial aerosol particles, *Atmos. Chem. Phys.*, 17, 1571-1593, doi:10.5194/acp-17-1571-2017,  
600 2017.

601 Seinfeld, J.H., Bretherton, C., Carslaw, K.S., Coe, H., DeMott, P.J., Dunlea, E.J., Feingold, G., Ghan,  
602 S., Guenther, A.B., Kahn, R.: Improving our fundamental understanding of the role of aerosol-cloud  
603 interactions in the climate system. *Proc. Natl. Acad. Sci. USA*, 113, 5781-5790, doi:  
604 10.1073/pnas.1514043113, 2016.

605 Sorooshian, A., Ng, N. L., Chan, A. W. H., Feingold, G., Flagan, R. C., and Seinfeld, J. H.: Particulate  
606 organic acids and overall water-soluble aerosol composition measurements from the 2006 Gulf of

607 Mexico Atmospheric Composition and Climate Study (GoMACCS), *J. Geophys. Res. Atmos.*,  
608 112(D13), doi:10.1029/2007JD008537, 2007a.

609 Stier, P., Feichter, J., Kinne, S., Kloster, S., Vignati, E., Wilson, J., Ganzeveld, L., Tegen, I., Werner, M.,  
610 Balkanski, Y.: The aerosol-climate model ECHAM5-HAM. *Atmos. Chem. Phys.*, 5, 1125-1156. 2005.

611 Tang, M., Cziczo, D. J., and Grassian, V. H.: Interactions of Water with Mineral Dust Aerosol: Water  
612 Adsorption, Hygroscopicity, Cloud Condensation, and Ice Nucleation, *Chem. Rev.*,  
613 doi:10.1021/acs.chemrev.5b00529, 2016.

614 Tsai, J.H., Lin, K.H., Chen, C.Y., Ding, J.Y., Choa, C.G. and Chiang, H.L.: Chemical constituents in  
615 particulate emissions from an integrated iron and steel facility, *J. Hazard. Mater.*, 147, 111-119.  
616 2007.

617 Wang, H., An, J., Shen, L., Zhu, B., Xia, L., Duan, Q. and Zou, J.: Mixing state of ambient aerosols in  
618 Nanjing city by single particle mass spectrometry, *Atmos. Environ.*, 132, 123-132, 2016.

619 Zhang, G., Bi, X., Lou, S., Li, L., Wang, H., Wang, X., Zhou, Z., Sheng, G., Fu, J. and Chen, C.: Source  
620 and mixing state of iron-containing particles in Shanghai by individual particle analysis,  
621 *Chemosphere*, 95, 9-16, 2014.

622 Zhang, G., Lin, Q., Peng, L., Bi, X., Lei, M., Chen, D., Brechtel, F.J., Chen, X., Yan, W., Wang, X., Peng,  
623 P., Sheng, G., Zhou, Z.: Single particle mixing state and cloud scavenging of black carbon at a  
624 high-altitude mountain site in south China, *J. Geophys. Res. Atmos.* Submitted, 2017.

625 Zhang, Y.P., Wang, X.F., Chen, H., Yang, X., Chen, J.M. and Allen, J.O.: Source apportionment of lead-  
626 containing aerosol particles in Shanghai using single particle mass spectrometry, *Chemosphere*, 74,  
627 501-507, 2009.

