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**Sensitivity of polar
stratospheric ozone
loss**

S. R. Kawa et al.

Sensitivity of polar stratospheric ozone loss to uncertainties in chemical reaction kinetics

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Abstract

The impact and significance of uncertainties in model calculations of stratospheric ozone loss resulting from known uncertainty in chemical kinetics parameters is evaluated in trajectory chemistry simulations for the Antarctic and Arctic polar vortices. The uncertainty in modeled ozone loss is derived from Monte Carlo scenario simulations varying the kinetic (reaction and photolysis rate) parameters within their estimated uncertainty bounds. Simulations of a typical winter/spring Antarctic vortex scenario and Match scenarios in the Arctic produce large uncertainty in ozone loss rates and integrated seasonal loss. The simulations clearly indicate that the dominant source of model uncertainty in polar ozone loss is uncertainty in the Cl_2O_2 photolysis reaction, which arises from uncertainty in laboratory-measured molecular cross sections at atmospherically important wavelengths. This estimated uncertainty in $J_{\text{Cl}_2\text{O}_2}$ from laboratory measurements seriously hinders our ability to model polar ozone loss within useful quantitative error limits. Atmospheric observations, however, suggest that the Cl_2O_2 photolysis uncertainty may be less than that derived from the lab data. Comparisons to Match, South Pole ozonesonde, and Aura Microwave Limb Sounder (MLS) data all show that the nominal recommended rate simulations agree with data within uncertainties when the Cl_2O_2 photolysis error is reduced by a factor of two, in line with previous in situ ClO_x measurements. Comparisons to simulations using recent cross sections from Pope et al. (2007) are outside the constrained error bounds in each case. Other reactions producing significant sensitivity in polar ozone loss include $\text{BrO}+\text{ClO}$ and its branching ratios. These uncertainties challenge our confidence in modeling polar ozone depletion and projecting future changes in response to changing halogen emissions and climate. Further laboratory, theoretical, and possibly atmospheric studies are needed.

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1 Introduction

The annual loss of ozone (O_3) in the springtime polar lower stratosphere of both hemispheres is a key diagnostic for ozone assessment, recovery prediction, and chemistry interaction with climate change. To a large extent, our confidence in understanding and projecting changes in polar (and global) O_3 is based on our ability to simulate these loss processes in numerical models of chemistry and transport. The fidelity of the models is assessed in comparison with a wide range of observations. The models depend on laboratory-measured kinetic reaction rates and photolysis cross sections to simulate molecular interactions (Sander et al., 2006, hereafter referred to as JPL06). The rates of all of these reactions are subject to uncertainty, some of which is substantial. Given the complexity of the models, however, it is difficult to quantify uncertainty in many aspects of the system.

In this study we use trajectory box-model simulations for Antarctic and Arctic stratospheric O_3 to quantify the uncertainty in loss attributable to known reaction kinetic uncertainties. Following the method of earlier work, rates and uncertainties from the latest laboratory evaluation are applied in random combinations (Stolarski et al., 1978; Stolarski and Douglass, 1986; Fish and Burton, 1997; Considine et al., 1999). We compare the results with observations to evaluate which combinations are consistent with atmospheric data. This also allows us to identify the key reactions and rates contributing the largest potential errors as a guide for future work. Note that these simulations only test rate uncertainties, and their fidelity depends on the accuracy and completeness of the underlying chemical reaction set. Transport uncertainty is not included; however, the scenarios are chosen to minimize sensitivity to transport errors (see Antarctic trajectory discussion and references on Match trajectory uncertainties below).

The work is motivated by several recent observational and laboratory studies of processes involved in polar stratospheric O_3 loss that have prompted a reexamination of aspects of our understanding for this key indicator of global change (Pope et al., 2007;

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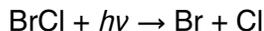
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von Hobe et al., 2007; Schofield et al., 2008). In particular, the rate of polar O₃ loss is predominantly controlled by catalytic halogen reactions:



5 and



followed by



10 Laboratory measurements of the Cl₂O₂ photolysis cross sections by Pope et al. (2007) imply a much slower rate of photolysis than indicated by previous measurements (Sander et al., 2006 and references within). This slower photolysis has a major impact in reducing the calculated O₃ loss rate in polar conditions. In addition there is significant uncertainty in the forward and reverse rates of the ClO/Cl₂O₂ equilibrium Reaction (R1) and their temperature dependence (von Hobe et al., 2007). Note that
15 throughout this paper Cl₂O₂ refers to the symmetric isomer of the ClO-dimer (ClOOC1). The addition of Br in the stratosphere from short-lived bromocarbons (Salawitch et al., 2005), not generally included in global chemistry-transport models (CTMs), increases the importance of Reaction (R3) and its branching ratios (a–c). Assessment of polar

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O₃ loss rates in Frieler et al. (2006) and WMO (2007) found that models generally required both additional stratospheric Br and a faster rate of Cl₂O₂ photolysis than current recommendations to match observations.

In this paper we attempt to assess the impact of these uncertainties in simulating polar O₃ loss against the backdrop of known uncertainties in kinetic rates using a quantitative model for the overall chemical error limits. Our overall objective is to evaluate the consistency of our theoretical understanding, model chemical mechanism, and kinetic rate parameters, including known kinetic uncertainties, with recent observations of Arctic and Antarctic winter/spring O₃ loss. Specifically, we 1) revisit the impact of kinetic uncertainties in models using JPL06 evaluations as well as new lab results (i.e., Pope et al., 2007), 2) assess the impact of constraints on photolysis uncertainty limits provided by atmospheric observations, and 3) identify the major uncertainty sources in simulating polar O₃ loss that result from uncertainties in kinetics as a potential guide to further lab measurements.

In the next section we outline the trajectory chemistry scenarios and Monte Carlo method used for calculating uncertainty bounds from the kinetic data. We also describe selection of data for comparison with the models. We then present the statistics of the calculations using JPL06 and the impact of constraining Cl₂O₂ photolysis error limits using atmospheric data. Following that we present comparisons with observations and implications for understanding processes and rates. We find that comparisons with ozonesonde and Aura Microwave Limb Sounder (MLS) data in the Antarctic and Match observations in the Arctic present a consistent picture of seasonal O₃ loss and chlorine partitioning vis-à-vis the kinetic rates and their uncertainties. The penultimate section summarizes key rate uncertainties in the polar O₃ loss reaction system with potential for future measurement work, and the final section provides summary remarks.

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2 Model scenarios and diagnostic observations

Results are presented below for the Antarctic and Arctic using slightly different calculation procedures and somewhat different observational data. The methods were developed independently but the findings are consistent between them, and both methods are described here. Our baseline for kinetic rates and uncertainties is JPL06 but in some cases reference is made to earlier studies using JPL02 (Sander et al., 2003), JPL97 (DeMore et al., 1997), and JPL94 (DeMore et al., 1994). Note that the recommendation for Cl_2O_2 photolysis cross sections and uncertainties has not changed since JPL97.

2.1 Antarctic model

A single, representative trajectory parcel was chosen for the Antarctic vortex from 29 July to 27 October. This parcel was selected from a run of 360 trajectory samples initialized at 80°S at 1° longitude increments on 27 September 2000 (a typical stratospheric winter). The trajectories were run backward 60 days to 29 July and then forward 30 days to 27 October. The trajectory winds were from the United Kingdom Meteorological Office operational analysis. The parcel was selected to be deep in the vortex at the time of maximum ozone loss, and have near-median potential temperature evolution and latitude-longitude excursions. In general, the vortex parcels follow statistically similar paths through this time period. This representative parcel's average latitude was 73°S varying between 89°S and 57°S at pressures from 40 to 67 hPa, and it diabatically descended over the course of the trajectory.

Chemical evolution along the trajectory was calculated using the standard Goddard stratospheric mechanism (Kawa et al., 1997), which is typical of current models. The initial chemical state of the trajectory was taken from a run of the global CTM with minor adjustments to the O_3 and reactive chlorine (Cly) abundances to more closely correspond to MLS measurements. The initialization on 29 July presents a fully activated partitioning of Cly. Polar stratospheric clouds (PSCs) form intermittently during the first

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half of the trajectory when the temperature is sufficiently low. An additional 5 pptv of Bry was included in the standard runs to represent the contribution from short-lived bromine-containing compounds reaching the stratosphere (WMO, 2007). Overhead O₃ and surface albedo for the photolysis calculations were derived from TOMS observations for 2000 interpolated to the trajectory (Pierson et al., 1999). The time series of O₃ along the trajectory for standard JPL06 chemistry can be seen in the central red curve in Fig. 1.

2.2 Kinetics uncertainties and Monte Carlo simulations

Kinetics uncertainties for the Antarctic calculations in this paper were taken from JPL06. Since all reaction rates must be positive, a lognormal distribution was assumed for the uncertainty in each rate coefficient, as described by Stolarski et al. (1978). That is, the nominal (median) value is multiplied and divided by a factor scaled to the JPL06 uncertainty estimates assuming a normal distribution of errors. The uncertainties in JPL06 are expressed as uncertainty in the measurement at a temperature of 298 K and an independent uncertainty in the temperature dependence. We convert these to uncertainties in the coefficients of the Arrhenius form of the reaction rate coefficient:

$$k(T) = A \exp(-E/R/T)$$

by the method described in JPL06. For uncertainties in termolecular reactions we have used the low-pressure limit form

$$k(T) = AT^{-n}$$

and the estimates for uncertainty in k₂₉₈ and temperature dependence as described in JPL06. The rate coefficients use the full formula given in JPL06, but the uncertainties were calculated using only the low-pressure limit. Uncertainties in photolysis coefficients were also taken from JPL06 and were applied uniformly at all wavelengths. The exception to this is the photolysis rate for Cl₂O₂, for which somewhat different assumptions were made as described later in this section.

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To evaluate the collective importance of the uncertainties in the 120 reactions and 37 photolysis (J) coefficients, we use a Monte Carlo technique previously described by Stolarski et al. (1978), Stolarski and Douglass (1986), and Considine et al. (1999). Briefly, a set of random rate coefficients and J coefficients, constrained to the recommended uncertainties, is produced for each simulation. This is repeated 1000 times to produce a distribution of constituent concentrations along the selected trajectory.

The Cl_2O_2 cross section presents a special case for this paper. The rate evaluation panels attempt to estimate statistical uncertainties on the photolysis rates based on expert knowledge and consensus regarding reported results (JPL06). For Cl_2O_2 cross sections, the lab measurements are very difficult, results vary widely, and the number of measurements is small (5 at stratospherically important wavelengths and temperatures), so the uncertainty estimate is correspondingly large (JPL06). We carried out 3 sets of Monte Carlo simulations (1000 members to each set). The first of these used the uncertainty stated in JPL06 (we used the uncertainty for the long wavelength portion of the spectrum, $>300\text{ nm}$). The second halved the uncertainty factor (3 to 1.5) to account for the constraints provided by atmospheric measurements as discussed below. The third used the recent Pope et al. (2007) cross sections as a baseline and the reduced uncertainty bounds.

As a further test, we evaluated the sensitivity of the O_3 loss along the selected trajectory to each of the rate coefficients by varying that coefficient individually by plus and minus one sigma.

2.3 Antarctic observations

The modeled evolution of O_3 is compared to ozonesondes at South Pole station (Hofmann et al., 1997) and McMurdo station (Nardi et al., 1999), and satellite data from Aura MLS (Santee et al., 2008). HCl from MLS is also compared to the model. For the sonde data at each site, we first interpolate the O_3 profile to 50 hPa and then average data on each day of the year from 1998 to 2007 excluding 2002 (because of the September major warming). For MLS, we interpolate potential temperature/equivalent

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latitude averaged daily data for 2005–2007 to the potential temperature/equivalent latitude of the sample trajectory selected above. The 3-year average and range of yearly-interpolated data are shown below. Similar results are obtained using zonal mean MLS at 76° S to 80° S between 46 and 68 hPa; the potential temperature/equivalent latitude averaging provides a convenient method to aggregate the data. Antarctic O₃ loss is fairly consistent from year to year in the late 1990s and 2000s except for 2002 and 2004, which had anomalously warm conditions (Hoppel et al., 2005; Santee et al., 2008). In each year, observed O₃ mixing ratios in the polar lower stratospheric vortex approach zero by mid-to-late September.

2.4 Arctic Match observations and model

For the Arctic, we concentrate on the polar winter of 1999–2000 because of the extensive in situ and remote measurement sets available during that year, including the SOLVE/THESEO and Match field campaigns (Newman et al., 2002; Rex et al., 2002). Polar O₃ loss is derived from regression analysis of sequential Match ozonesonde observations along air parcel trajectories (Rex et al., 1998; 2003). A photochemical box model is run along the identical trajectories used for the O₃ loss calculations. The chemical model uses a simplified mechanism that includes known reactions relevant to O₃ loss in the lower stratosphere, with Cly and Bry constrained by observations (Rex et al., 2003; Frieler et al., 2006). The Bry abundance is based on DOAS profile measurements over Sweden in 2000, reflecting a contribution of approximately 6 pptv from short-lived bromocarbons beyond that from a standard model (Frieler et al., 2006). In a method similar to that for the Antarctic, chemical model runs were done along the Match trajectories randomly varying the rates of the Match chemical mechanism according to the distribution of uncertainties given by JPL02.

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3 Monte Carlo simulation results

The full 1000-member ensemble of calculated O_3 time series for the Antarctic is shown in Fig. 1 based on rate uncertainties as tabulated in JPL06. A wide range in possible O_3 loss is found, from complete destruction before the end of August (day 238) to less than 50% loss by the end of October (day 304). This range is driven almost entirely by the stated uncertainty in Cl_2O_2 photolysis, which is a factor of three at wavelengths greater than 300 nm based on uncertainty in the measured molecular cross sections (JPL06). Almost all calculated lower stratospheric photolysis of Cl_2O_2 takes place at the longer wavelengths, where the laboratory measurements are most susceptible to possible contamination from photolysis of other chlorine species (Burkholder et al., 1990; Huder and DeMore, 1995; Pope et al., 2007). Based on JPL06 uncertainties (i.e., even without considering the results of Pope et al., 2007), modeled O_3 loss rates have an uncertainty of up to a factor of three. The same conclusion is reached for analysis of O_3 loss in the Arctic using Match trajectory-chemistry (Fig. 2). The upper and lower limits of the uncertainty interval (i.e., the inner 68% of the calculated ozone loss rates) differ by a factor of three. The primary message of this paper is that the stated uncertainty in the photolysis rate $J_{Cl_2O_2}$ from laboratory measurements precludes modeling of polar O_3 loss within any useful quantitative error limits, and, at this level of uncertainty, not much else matters. The loss rate uncertainties from $J_{Cl_2O_2}$ are larger than any produced by varying other photolysis rates, Bry , other tracers, reaction rates, trajectories, or transport within reasonable limits.

In order to move beyond the impact of nominal uncertainty in $J_{Cl_2O_2}$ to see what other rates and factors most strongly affect polar O_3 loss, we constrained the uncertainty in $J_{Cl_2O_2}$ based on analysis of in situ atmospheric measurements of ClO and Cl_2O_2 (Stimpfle et al., 2004). Comparing the measured ClO/Cl_2O_2 with model formulations for this ratio as a function of solar zenith angle, Stimpfle et al. found that values of $J_{Cl_2O_2}$ within about 50% of JPL02 are consistent with observations within the uncertainty of the measurements and the ClO/Cl_2O_2 equilibrium Reaction (R1). von

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Hobe et al. (2008) reached a very similar range of uncertainty for $J_{\text{Cl}_2\text{O}_2}$ based on an extensive analysis of observations and lab measurements, including uncertainties in the ClO/Cl₂O₂ equilibrium reactions. The analysis of in situ ClO measurements from Avallone and Toohey (2001) are also within this range. Therefore, we repeated our Monte Carlo uncertainty calculations using a halved uncertainty factor of 1.5 for $J_{\text{Cl}_2\text{O}_2}$ (which is the nominal uncertainty for photolysis at wavelengths less than 300 nm from JPL06), consistent with the results from atmospheric ClO and Cl₂O₂ measurements.

The results of the Antarctic trajectory scenario using $J_{\text{Cl}_2\text{O}_2} = J_{\text{Cl}_2\text{O}_2}(1 \pm \sigma)$, $\sigma = 0.5$, are shown in Fig. 3 (red line on the left side with darker grey shading). The uncertainty in the range of O₃ losses is still large. Using the date at which the O₃ mixing ratio first reaches a value less than 0.1 ppmv, the values range around a base value of day 260.9 (16 September) from day 251 (7 September) to day 276 (2 October) for the JPL06 case at the 95% confidence limits. Uncertainty in Cl₂O₂ photolysis is still the largest source of uncertainty in the O₃ loss (more below), but at this uncertainty level other error sources are discernable. For example, the scenarios using Pope et al. (2007) cross sections for the baseline are distinguishable from JPL06 at or near the 95% confidence level (Fig. 3a). The minimum O₃ in the base Pope et al. case reaches a minimum of 0.09 ppmv on day 294. The statistics of the distribution of O₃ on day 250 in the different scenarios is seen in Fig. 3b. There is little overlap between the JPL06 and Pope et al. photolysis distributions and both are slightly skewed toward low O₃ mixing ratios. In each, the base case, mean, and most probable value are within 0.1 ppmv of each other. The impact of uncertainty in other kinetic rates is discussed in Sect. 4 below.

In the course of developing these scenarios, we have tested a number of other sensitivities in polar O₃ loss that are worth comparing to the kinetics uncertainty range shown here. The O₃ loss rates are sensitive to the amounts of Cly and Bry as expected. Reducing Bry by 5 pptv (i.e., not adding Br from short-lived halocarbons) and 8 pptv from nominal (21 pptv) results in increasing the date of O₃ < 0.1 ppmv by 2.5 and 4.2 days, respectively, beyond the base case. Decreasing Cly by 15% (approximately the difference going from 2000 to 1992 in the lower stratospheric vortex) extends the

date of $O_3 < 0.1$ ppmv by 4 days. Note that Cly decreases by about 3% between 2000 and 2006 (Newman et al., 2006).

The O_3 loss is nearly insensitive to the rates of heterogeneous processes in the Antarctic scenario. Cly is nearly fully processed to reactive forms on 1 August and little recovery to reservoir species, HCl and $ClONO_2$, occurs in the low sun conditions through September. Varying the heterogeneous reaction rates by a factor of 2 changes the time of $O_3 < 0.1$ ppmv by less than an hour. Heterogeneous reactions on liquid ternary solution droplets and the occasional PSCs are more than fast enough to keep all Cly activated until the final shift to HCl that occurs when O_3 is depleted (see below). The loss rates are sensitive to temperature with the date of $O_3 < 0.1$ shifting by -1.7 to 2.3 days for a constant delta temperature of -5 and 5 K, respectively. Changes in the assumed surface albedo (0.1 to 0.6) have only a negligible effect (< 0.5 d) for the photolysis calculations in this scenario, while varying overhead O_3 (± 50 DU) changes the date by ± 1 day.

The shape of the O_3 loss curve and date of mixing ratio < 0.1 does depend strongly on the parcel trajectory through the Antarctic vortex. We have run thousands of trajectories from different initial conditions in the Antarctic over the time of our selected parcel to examine this sensitivity. A wide range of O_3 values is possible (although not as wide as in Fig. 1). The probability distribution function for day 250 (not shown) is relatively flat and the distribution of O_3 across the various trajectories is decidedly not Gaussian. As expected, our selected parcel is near the mean and most probable value of this distribution. In general, parcels at more equatorward latitudes and higher altitudes begin to lose O_3 sooner as the sun rises earlier, and complete O_3 loss occurs earlier provided they are completely activated and denitrified. The O_3 time series from our selected parcel corresponds closely to that of a zonal average of parcels at $75^\circ S$ ($\pm 2.5^\circ$). All parcels have some latitude excursions, which tend to hasten O_3 loss with respect to that at a fixed latitude. For example, we ran earlier calculations fixed at $80^\circ S$ and O_3 loss did not begin until about day 240, which did not seem realistic in comparison to observations that represent a potential mix of many actual trajectories. Finally,

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note that the conclusions discussed above regarding the uncertainty of O_3 loss resulting from uncertainty in Cl_2O_2 photolysis, the relative comparison between JPL06 and Pope et al., and the contribution of other kinetics uncertainties discussed below do not depend critically on the choice of trajectory. The large majority of Antarctic trajectories would show the same sensitivity.

3.1 Antarctic comparisons with data

Comparisons with averaged MLS O_3 and HCl from 2005–2007 are shown in Fig. 4a and b, respectively. The O_3 observations generally follow the JPL06 base case and fall within the 95% uncertainty limits given by the kinetics uncertainties on the JPL06 rate scenario (assuming an uncertainty factor of 1.5 for the Cl_2O_2 cross section). The data are outside the Pope et al. scenario bounds, at least until O_3 loss is near complete in October. The day-to-day fluctuations in MLS O_3 result from both equivalent latitude excursions of the trajectory and variations in the potential temperature/equivalent latitude averages of the data.

The MLS HCl comparison (Fig. 4b) is consistent with the O_3 time series. When O_3 mixing ratios fall below about 0.5 ppmv, HCl increases rapidly, driven by production from reaction of Cl and CH_4 (Douglass et al., 1995; Douglass and Kawa, 1999). The averaged MLS HCl rises more gradually than the unmixed trajectory scenario, but the mixing ratios remain within or near the model error bars given by the JPL06 rate uncertainties through October. Neither MLS dataset is consistent with the Pope et al. scenario, consistent with the results of Santee et al. (2008).

Comparison of the range of model O_3 with sonde data from South Pole (SP) and McMurdo for several years is shown in Fig. 5. In each year at each site, the measured O_3 decreases to near-zero mixing ratios. On most days at SP, the data fall within the 95% confidence limits on the JPL06 rate scenario and outside the Pope et al. scenario. Note that we might expect SP data to lag the trajectory scenario somewhat since, in general, the latitude of the trajectory is equatorward of SP and hence insolation is greater. SP data, however, generally show some impact of parcel trajectories from

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more equatorial latitudes. The sun does not rise at 50 hPa over SP until about day 246 (3 September), when the main O₃ decline begins, but some loss has already begun in August in most years. Interannual variation in the loss rate and minimum date is related to meteorological variability, but in all years from 1998 to 2007 (except 2002)

5 O₃ at the SP goes to near zero (<0.1 ppmv) within ±5 days of day 270 (see Hofmann et al., 1997, 2009 for updates through 2007). Average O₃ from McMurdo at 77.5°S starts out a bit higher than our trajectory initial condition, but generally falls within the JPL06 limits through September (day 274). McMurdo O₃ loss appears to lead SP by several days on average, but the range of the data is relatively large (about ±0.5 ppmv).
10 In October, McMurdo shows the effects of vortex distortion and mixing on local O₃ in some years.

3.2 Arctic comparisons with Match data

Comparison of the overall Match observed O₃ loss rate for the year 2000 with model simulations for a range of Cl₂O₂ cross section measurements is shown in Fig. 6. In this case error bars on the model calculation (red curve and bars in Fig. 6) represent the uncertainty in inferring the amount of activated Cly for the O₃ loss calculation, rather than the kinetics uncertainties. For Match, the best agreement with data occurs for the model based on Cl₂O₂ cross sections from Burkholder et al. (1990). The model based on the JPL06 recommendation is within the uncertainty of the data. Model and data errors bars nearly overlap using the Huder and DeMore (1995) cross sections, which are about a factor of 1.5 lower than JPL06, consistent with the constrained uncertainty discussion above. This analysis is consistent with 6 years of Match data, including the Antarctic in 2003 (Frieler et al., 2006). The model with Pope et al. (2007) cross sections is outside observational error limits.

25 In a complementary approach, the Match O₃ loss rates can be “inverted” using the chemistry model to infer the amount of activated Cly required to produce the observed loss rates (Rex et al., 2003). Consistent with the loss rate comparison in Fig. 6, the amount of inferred Cly is within uncertainties on the SOLVE observations (Stimpfle et

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al., 2004) using Burkholder or JPL06 cross sections (and enhanced Br) (Frieler et al., 2006). The amount of Cl_y required (3 to 10 ppbv) to produce observed loss using the Pope et al. cross sections is significantly more than the total chlorine in the stratosphere (~3.5 ppbv).

4 Sensitivity to key reactions and uncertainties

In addition to the Monte Carlo distribution of rate uncertainties run above, we also ran the Antarctic simulation varying rates one at a time at the $\pm 1\sigma$ uncertainty limits given in JPL06 (except $J_{\text{Cl}_2\text{O}_2}$ for which the uncertainty was set at 1.5 as discussed above). The reactions with the largest uncertainty impacts, in terms of date of initial O₃ < 0.1 ppmv, for the Antarctic scenario are given in Table 1. The uncertainty in Cl₂O₂ photolysis Reaction (R2) still has the largest impact on polar O₃ loss, but the uncertainty in ClO+BrO Reaction (R3) is close behind. Reaction (3b) has the largest effect among the possible product yields for Reaction (R3) because this channel directly recycles the Br and Cl in sunlight, while Reaction (R3c) sequesters Cl in OCIO at low zenith angles (producing the opposite sensitivity). The BrCl channel Reaction (R3a) is a minor pathway (~8%) (JPL06). Note that our modeling protocol treats uncertainty in each of these reaction pathways independently, which is probably not the case in the laboratory measurements, where the errors are likely correlated. Regardless of branching ratio, however, this exercise shows that uncertainty in this reaction causes a significant uncertainty in calculating polar O₃ loss.

The direct rate of O₃ loss with Br Reaction (R5) has some uncertainty impact, which will, of course, depend on the amount of Br_y in the vortex. The Cl₂O₂ formation and back Reaction (R1), which have substantial uncertainty from lab studies, do not have a large impact on polar O₃ loss because thermal dissociation is small compared to photolysis in sunlight, and photolysis Reaction (R2) is generally rate limiting compared to Cl₂O₂ formation Reaction (R1) in the O₃ loss cycle at high ClO mixing ratios (i.e., fully activated Cl_y). Reaction (R1) is, however, important to the ClO/Cl₂O₂ partitioning and

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interpretation of ambient Cl_y measurements (e.g., Stimpfle et al., 2004). Reactions that remove Cl and Br from the active catalytic cycle into HCl and HBr show minor impact. Sensitivity to all other reactions is negligible in this scenario.

The Arctic Match O₃ loss uncertainty is also dominated by the top two reaction sensitivities as in Table 1, with Reaction (R1) in a distant third place. These findings are similar to those of Fish and Burton (1997) in the Arctic using JPL94 kinetic rates, although Fish and Burton found that Reaction (R3b) contributed a slightly larger uncertainty than Reaction (R2). In lieu of a recommendation from JPL94, Fish and Burton estimated $\sigma(J_{\text{Cl}_2\text{O}_2})$ at a factor of 2, which they considered possibly overestimated in light of atmospheric ClO data. The nominal recommended cross sections in JPL06 are the same as those in JPL94, and, while the mean rates for Reaction (R3) have changed slightly, the uncertainty bounds have not. The different ranking of reaction uncertainties from Fish and Burton probably arises from the specifics of the trajectories used and the Cl_y and Br_y abundances.

5 Summary discussion

Quantitative evaluation of chemical uncertainty underlying stratospheric assessment models shows that known uncertainties in kinetic reaction rate parameters from laboratory measurements produce large uncertainty in polar O₃ loss calculated in simple, representative models. For both Antarctic and Arctic O₃ loss, the Monte Carlo uncertainty distribution is dominated by the recommended JPL06 uncertainty in photolysis cross sections for Cl₂O₂ at atmospherically relevant wavelengths. This uncertainty is large enough to include the recent cross section measurements of Pope et al. (2007), which imply greatly reduced polar O₃ loss rates in a standard model mechanism. New laboratory measurements are required to better interpret these and earlier results and diminish the error bounds. This finding is consistent with previous work, and several new lab studies are in progress (SPARC, 2009).

Evidence from atmospheric measurements, however, suggests that the Cl₂O₂ pho-

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5 photolysis rate can be constrained more tightly than the lab data indicate. When the photolysis rate uncertainty in the Monte Carlo simulations is constrained by atmospheric observations, the impact of varying Cl_2O_2 cross sections between JPL06 and Pope et al. is distinct at the 95% confidence level for spring/winter Antarctic O_3 loss. In this case, comparison to observations shows the ozonesonde, MLS, and Match data are consistent with JPL06 kinetic rates within model uncertainty. This lends confidence that the chemical mechanism and JPL06 rates in the models are representative and reasonably accurate. Note that if $J_{\text{Cl}_2\text{O}_2}$ is close to that produced from Pope et al., a large fraction of the observed O_3 loss cannot be explained by the standard chemistry mechanism. Efforts to simultaneously explain all the relevant theoretical, laboratory, and atmospheric data by identifying missing processes or using an alternate chemical mechanism have yet to yield a solution (SPARC, 2009).

15 Sensitivity tests have also shown that uncertainty in the rate and branching ratio for $\text{BrO} + \text{ClO}$ has a significant impact on uncertainty in polar O_3 loss calculations in both hemispheres. This finding is consistent with previous work based on earlier rate compilations at midlatitudes and in the Arctic, and leads us to a recommendation for further laboratory work on this reaction as well. The importance of this reaction is closely tied to the amount of bromine in the lower stratosphere, which also has significant uncertainty based on current measurements. Uncertainty in the key chemical reaction kinetics must be reduced in order to confidently assess the impact of changes in bromine and other trace gases on future projections of polar O_3 loss and recovery.

25 *Acknowledgements.* We would like to acknowledge support from the Atmospheric Chemistry Modeling and Analysis and the Modeling, Analysis, and Prediction Programs at NASA. Work at the Jet Propulsion Laboratory, California Institute of Technology, was done under contract with the National Aeronautics and Space Administration. Work at AWI was supported by the EC DG Research through the RECONCILE project. The McMurdo data used in this publication were obtained as part of the Network for the Detection of Atmospheric Composition Change (NDACC) and are publicly available (see <http://www.ndacc.org>).

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Table 1. First date of $O_3 < 0.1$ ppmv.

#	Reaction	Δt (days)	
		$+1\sigma$	-1σ
(R2)	$Cl_2O_2 + h\nu \rightarrow Cl + ClO_2$	-5.04	6.04
(R3b)	$BrO + ClO \rightarrow Br + ClOO$	-3.96	3.17
(R5)	$Br + O_3 \rightarrow BrO + O_2$	-0.83	1.00
(R3a)	$BrO + ClO \rightarrow BrCl + O_2$	-0.92	0.75
(R3c)	$BrO + ClO \rightarrow Br + OClO$	0.71	-0.17
(R1)	$ClO + ClO + M \rightarrow Cl_2O_2 + M$	-0.17	0.12
	$ClO + OH \rightarrow HCl + O_2$	0.79	0.00
	$Br + H_2CO \rightarrow HBr + HCO$	0.12	-0.08

base case=day 260.88

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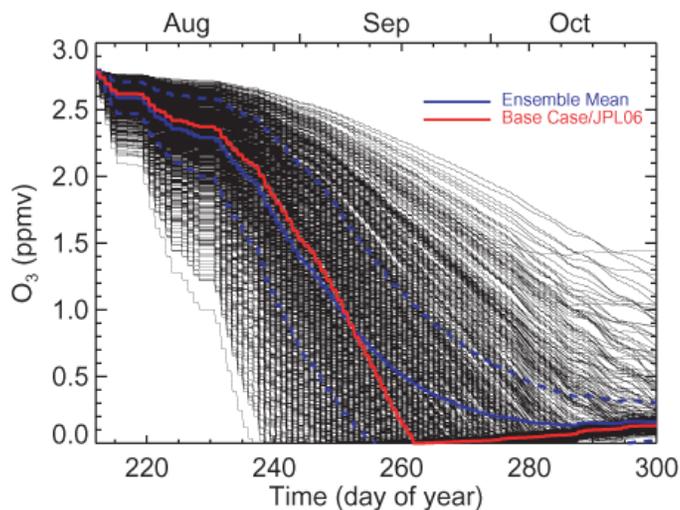


Fig. 1. Calculated O_3 as a function of time along the trajectory for simulations using each of the 1000 Monte Carlo reaction rate sets. The base case using nominal JPL06 rate recommendations is shown by the red curve. The solid and dashed blue lines are the mean and mean \pm standard deviation of O_3 from the 1000 cases at each time. This scenario uses the JPL06 recommended factor of 3 uncertainty on Cl_2O_2 photolysis as discussed in the text.

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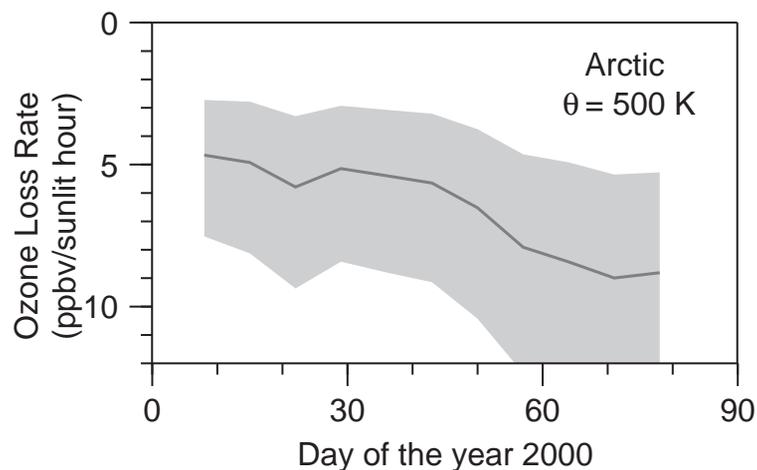


Fig. 2. Calculated O_3 loss rates as a function of time along the ensemble of Match trajectories at a potential temperature of 500 K varying the kinetic rates within JPL02 uncertainty limits and assuming complete activation of Cl_y . The median case for JPL02 $\pm 34\%$ of the distribution of reaction rate sets is shown (uncertainty interval of a lognormal distribution).

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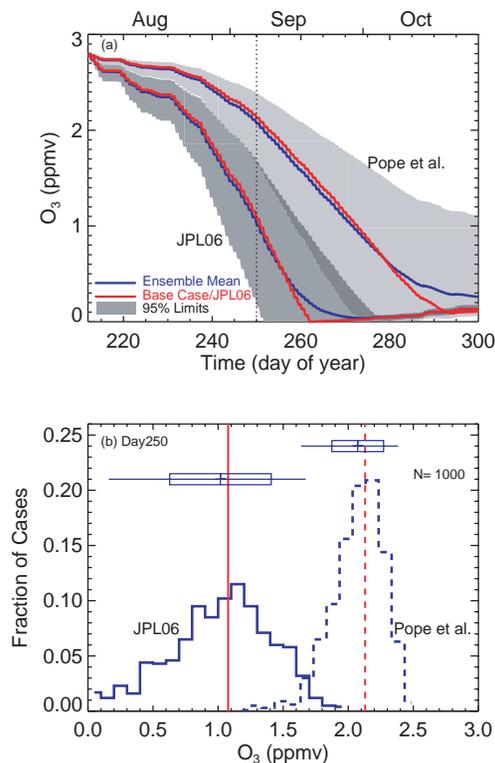


Fig. 3. O₃ loss calculations with uncertainty in $J_{\text{Cl}_2\text{O}_2}$ constrained to a factor of 1.5. **(a)** Gray shaded areas encompass the median 95% of O₃ mixing ratio values from the scenario distribution on each day for the JPL06 and Pope et al. cases. Red lines are the calculations using the nominal rates from JPL06 and JPL06 with Pope et al. cross sections substituted for $J_{\text{Cl}_2\text{O}_2}$. Solid blue lines in (a) are mean O₃. **(b)** Probability distributions of O₃ on day 250 from the Monte Carlo set of kinetic rates using JPL06 (solid) and JPL06+Pope et al. (dashed). Box and whisker forms show mean, ± 1 standard deviation, and 95% limits of the O₃ values.

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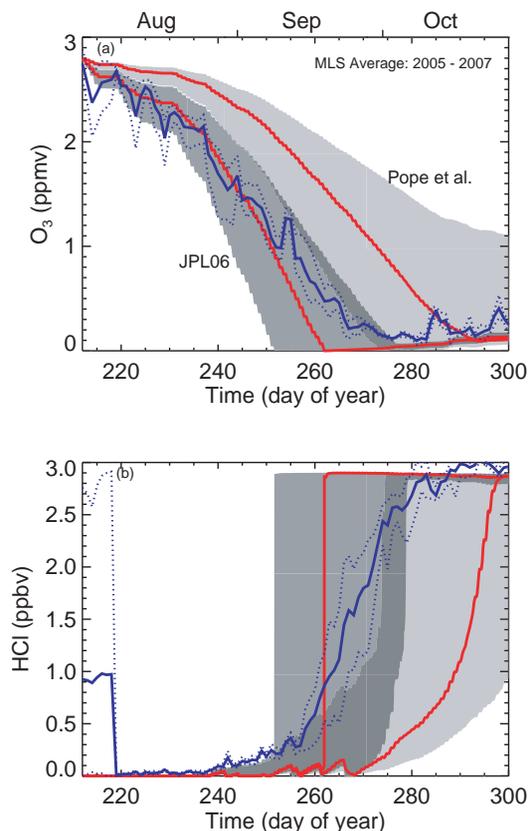


Fig. 4. Comparison of MLS O₃ **(a)** and HCl **(b)** data averaged for 2005–2007 (solid blue lines) with model simulation. Maxima and minima of 3 years of data for each day are shown by dotted lines. Equivalent latitude/potential temperature averaged MLS data are mapped to the equivalent latitude and potential temperature of the model trajectory for each day. Red lines and shading as in Fig. 3a.

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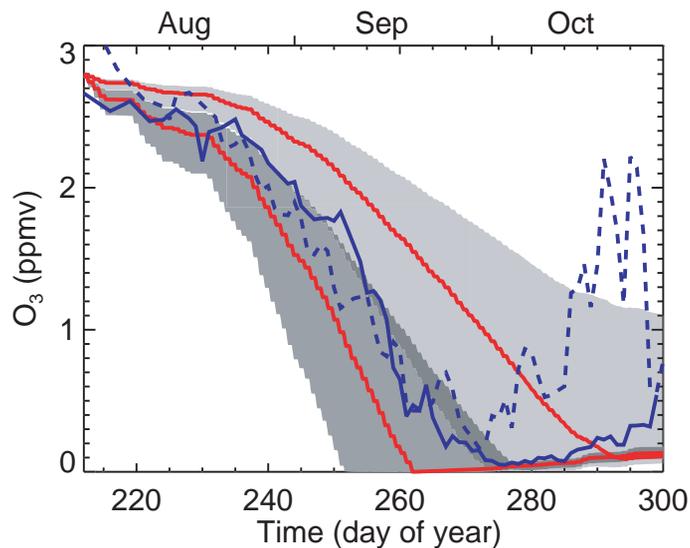


Fig. 5. South Pole (90° S, solid blue curve) and McMurdo (77.5° S, dashed blue curve) sonde data at 50 hPa averaged from 1998 to 2007 excluding 2002. Observations have been smoothed with a 3-point running average filter. Red lines and shading as in Fig. 3a.

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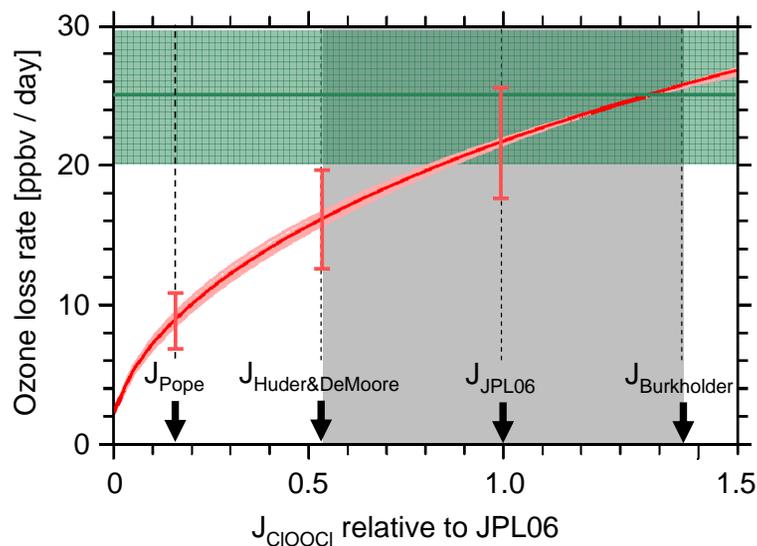


Fig. 6. Comparison of overall Match observed O_3 loss rate in early February 2000 (green line; green shading is 1σ uncertainty) with model calculations using four different evaluations for Cl_2O_2 photolysis cross sections. Red curve is fit to results of chemical calculations along Match trajectories. Model active chlorine abundances are constrained by measurements from SOLVE flight on 2 February 2000. Red error bars are uncertainties in calculated O_3 loss rates resulting from propagating the uncertainty in inferring the amount of activated Cl for through the model. Light red lines very close to the center red line reflect uncertainties from propagating uncertainties in the rate constant for the ClO self reaction and the ClOOCl equilibrium constant through the model. Gray shaded area denotes range of $J_{\text{Cl}_2\text{O}_2}$ values, relative to JPL06, that are consistent with O_3 observations.

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