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apportionment of
methane and nitrous
oxide in California's
San Joaquin Valley

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Source apportionment of methane and nitrous oxide in California's San Joaquin Valley at CalNex 2010 via positive matrix factorization

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processes including fossil fuel combustion have been estimated to account for 15 % of total global anthropogenic N₂O emissions (Denman et al., 2007).

In 2006, the state of California adopted Assembly Bill 32 (AB32) into a law known as the Global Warming Solutions Act, which committed the state to cap and reduce anthropogenic GHG emissions to 1990 levels by 2020. A statewide GHG emission inventory (CARB, 2013) maintained by the Air Resources Board of California (CARB) is used to report, verify and regulate emissions from GHG sources. In 2011, CH₄ accounted for 32.5 million metric tonnes (MMT) CO₂-eq representing 6.2 % of the statewide GHG emissions, while N₂O emissions totaled 6 MMT CO₂-eq representing about 3 % of the GHG emissions inventory (Fig. 1). CARB's accurate knowledge of GHG sources and statewide emissions is key to the success of any climate change mitigation strategy under AB32. CARB's GHG inventory is a "bottom-up" summation of emissions derived from emission factors and activity data. The bottom-up approach is reasonably accurate for estimation and verification of emissions from mobile and point sources (vehicle tailpipes, power plant stacks etc.) where the input variables are well-understood and well-quantified. The main anthropogenic sources of CH₄ in the CARB inventory include ruminant livestock and manure management, landfills, wastewater treatment, fugitive and process losses from oil and gas production and transmission, and rice cultivation while the major N₂O sources are agricultural soil management, livestock manure management and vehicle fuel combustion (CARB, 2013). The emission factors for many of these sources have large uncertainties as they are biological and their production and release mechanisms are inadequately understood thus making these sources unsuitable for direct measurements (e.g. emissions of N₂O from farmlands). Many of these sources (e.g. CH₄ from landfills) are susceptible to spatial heterogeneity and seasonal variability. Unfortunately, a more detailed understanding of source characteristics is made difficult because CH₄ and N₂O are often emitted from a mix of point and area sources within the same source facility (e.g. dairies in the agricultural sector) making bottom-up estimation uncertain. There is a lack of direct measurement data or "top-down" measurement-based approaches to independently validate seasonal trends and

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inventory estimates of CH₄ and N₂O in California's Central Valley, which has a mix of several agricultural sources and oil and gas operations, both of which are known major sources of GHGs.

In the recent past, regional emission estimates derived from measurements from a tall tower at Walnut Grove in Central California coupled with inverse dispersion techniques (Fischer et al., 2009) reported underestimation of CH₄ and N₂O emissions especially in the Central Valley. Comparison of regional surface footprints determined from WRF-STILT algorithm between October–December 2007 indicate posterior CH₄ emissions to be higher than California-specific inventory estimates by 37 ± 21 % (Zhao et al., 2009). Predicted livestock CH₄ emissions are 63 ± 22 % higher than a priori estimates. A study over a longer period (December 2007–November 2008) at the same tower (Jeong et al., 2012a) generated posterior CH₄ estimates that were 55–84 % larger than California-specific prior emissions for a region within 150 km from the tower. For N₂O, inverse estimates for the same sub-regions (using either EDGAR32 and EDGAR42 a priori maps) were about twice as much as a priori EDGAR inventories (Jeong et al., 2012b). Recent studies have incorporated WRF-STILT inverse analysis on airborne observations across California (Santoni et al., 2012). The authors conclude that CARB CH₄ budget is underestimated by a factor of 1.64 with aircraft-derived emissions from cattle and manure management, landfills, rice, and natural gas infrastructure being around 75, 22, 460, and 430 % more than CARB's current estimates for these categories, respectively. Statistical source footprints of CH₄ emissions generated using FLEXPART-WRF modeling and CalNex-Bakersfield CH₄ concentration data are consistent with locations of dairies in the region (Gentner et al., 2014a). The authors conclude that the majority of CH₄ emissions in the region originate from dairy operations. Scaled-up CH₄ rice cultivation estimates derived from aircraft CH₄/CO₂ flux ratio observations over rice paddies in the Sacramento valley during the growing season when emissions are at their strongest (Peischl et al., 2012) are around three times larger than inventory estimates. CH₄ budgets derived for the Los Angeles (LA) basin from aircraft observations (Peischl et al., 2013) and studies involving comparison

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(Ryerson et al., 2013). The SJV represents the southern half of California's Central Valley. It is 60 to 100 km wide, surrounded on three sides by mountains, with the Coastal Ranges to the west, the Sierra Nevada Mountains to the east, and the Tehachapi Mountains to the southeast.

The measurement site was located to the southeast of the Bakersfield urban core in Kern County (Fig. 2). The east–west Highway 58 is located about 0.8 km to the north; the north–south Highway 99 about 7 km to the west. The city's main waste water treatment plant (WWTP) and its settling ponds are located to the east and south of the site (< 2.5 km), respectively. Numerous dairy and livestock operations are located to the south-southwest of the site at 10 km distance or farther. The metropolitan region has three major oil refineries located within 10 km from the site (two to the northwest; one to the southeast). A majority of Kern County's high-production active oil fields (> 10 000 barrels (bbl) per day) (CDC, 2013) are located to the west/northwest and are distant (~ 40–100 km). Kern River oilfield (~ 60 000 bbl day⁻¹), one of the largest in the country, and Kern Front (~ 11 000 bbl day⁻¹) are located about 10–15 km to the north. There are several other oil fields dotted within the urban core (5–20 km) which are less productive (< 2000 bbl day⁻¹) or not active (< 100 bbl day⁻¹). The whole region is covered with agricultural farmlands with almonds, grapes, citrus, carrots and pistachios amongst the top commodities by value and acreage (KernAg, 2010).

The meteorology and transport of air masses in the southern SJV is complex and has been addressed previously (Bao et al., 2007; Beaver and Palazoglu, 2009). The wind rose plots (Fig. 3) shown here present a simplified distribution of microscale wind speed and direction at the site, the latter often being non-linear over larger spatial scales. The plots depict broad differences in meteorology during daytime and nighttime. A mesoscale representation of the site meteorology during this study period was evaluated through back-trajectory footprints generated from each hourly sample using FLEXPART Lagrangian transport model with WRF meteorological modeling (Gentner et al., 2014a). The 6 and 12 h back trajectory footprints are generated on a 4 km × 4 km resolution with simulations originating from top of the 18 m tall tower. The site experi-

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mally insulated temperature controlled 7 foot wide cargo wagon trailer developed by the GHG instrument manufacturers (LGR Inc.). CO was coincidentally measured using another instrument (Teledyne API, USA, Model # M300EU2) with a precision of 0.5 % of reading and output as 1 min averages. The mixing ratios from the two collocated CO instruments correlated well ($r \sim 0.99$) and provided a good stability check for the LGR instrumentation. Scaled Teledyne CO data was used to gap-fill the LGR CO data. The coincident gas-phase VOC measurements were made using a gas chromatograph (GC) with a quadrupole mass selective detector and a flame ionization detector (Gentner et al., 2012).

Hourly calibration checks of the three GHGs and CO were performed using near-ambient level scuba tank standards through the entire campaign. During data processing, final concentrations were generated from the raw data values using scaling factors obtained from comparison of measured and target concentrations during calibration checks. Diurnal plots of measured species are generated from 1 min averages. PMF analyses in the following sections are based on 30 min averages to match the time resolution of VOC measurements. The metrological data measured at the top of the tower included relative humidity (RH), temperature (T), and wind speed (WS) and direction (WD).

3.2 Positive Matrix Factorization (PMF)

Source apportionment techniques like PMF have been used in the past to apportion ambient concentration datasets into mutually co-varying groups of species. PMF is especially suitable for studies where a priori knowledge of number of sources impacting the measurements, chemical nature of source profiles and relative contribution of each source to the concentration time series of a measured compound are unknown or cannot be assumed. PMF has been applied to ambient particulate matter studies (Kim et al., 2004; Lee et al., 1999); in determining sources of atmospheric organic aerosols (OA) (Ulbrich et al., 2009; Slowik et al., 2010; Williams et al., 2010); and in gas phase measurements of VOCs in major metropolitan cities (Bon et al., 2011; Brown et al.,

2007). PMF is a receptor-only unmixing model which breaks down a measured data set containing time series of a number of compounds into a mass balance of an arbitrary number of constant source factor profiles (FP) with varying concentrations over the time of the data set (time series *or* TS) (Ulbrich et al., 2009).

5 In real world ambient scenarios, sources of emissions are often not known or well-understood. PMF technique requires no a priori information about the number or composition of factor profiles or time trends of those profiles. The constraint of non-negativity in PMF ensures that all values in the derived factor profiles and their contributions are constrained to be positive leading to physically meaningful solutions. PMF attributes a measure of experimental uncertainty (or weight) to each input measurement. 10 Data point weights allow the level of influence to be related to the level of confidence the analyst has in the measured data (Hopke, 2000). In this way, problematic data such as outliers, below-detection-limit (BDL), or altogether missing data can still be substituted into the model with appropriated weight adjustment (Comero et al., 2009) allowing for a larger input data set, and hence a more robust analysis. PMF results are quantitative; it is possible to obtain chemical composition of sources determined by the model (Comero et al., 2009). PMF is not data-sensitive and can be applied to data sets that are not homogenous and/or require normalization without introducing artifacts.

3.3 Mathematical framework of PMF

20 The PMF model is described in greater detail elsewhere (Paatero and Tapper, 1994; Paatero, 1997; Comero et al., 2009; Ulbrich et al., 2009) and we will briefly mention some concepts relevant to the understanding of the analysis carried out in this study. The PMF input parameters involve a $m \times n$ data matrix \mathbf{X} with i rows containing mixing ratios at sampling time t_i and j columns containing time series of each tracer j . 25 A corresponding uncertainty matrix \mathbf{S} reports measurement precision (uncertainty) of

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of CH₄ vs. ppt level of propene) and improve the visual attributes of PMF output plots to follow. Data points denoting zero enhancement (lower limit) were replaced by a very small positive number (i.e. exp(-5)) to avoid “zeros” in the data matrix **X**.

$$x_{ij} = (\text{Mixing ratio}_{ij} - \text{Background}_j) / (\text{Maximum mixing ratio}_j - \text{Background}_j) \quad (4)$$

5 For the VOCs, guidelines set forth by (Williams et al., 2010) were adopted to calculate the uncertainty estimates. An analytical uncertainty (AU) of 10 % was used; a limit of detection (LOD) of 1 ppt and a limit of quantification (LOQ) of 2 ppt (Gentner et al., 2012) was used to calculate the total uncertainty for each x_{ij} :

$$s_{ij} \equiv 2 \times \text{LOD}, \quad \text{if } x_{ij} \leq \text{LOD}, \quad (5a)$$

$$10 \quad s_{ij} \equiv \text{LOQ}, \quad \text{if } \text{LOD} < x_{ij} \leq \text{LOQ}, \quad (5b)$$

$$s_{ij} \equiv \left((\text{AU} \times x_{ij})^2 + (\text{LOD})^2 \right)^{0.5}, \quad \text{if } x_{ij} > \text{LOQ} \quad (5c)$$

Using this approach, the detection limit dictates the errors for low enhancements (near LOD) while the errors for larger enhancements of VOCs are tied more to the magnitude of the data value (x_{ij}) itself.

15 The GHG and CO measurements have high precision and significantly lower detection limits than ambient levels. The relatively low values of GHGs in the uncertainty matrix, compared to VOCs, is substituted with those calculated using a custom approach. The GHG and CO uncertainties are assumed to be proportional to the square root of the data value and an arbitrary scaling factor determined through trial and error
20 in order to produce lower values of QQ_{exp}^{-1} :

$$s_{ij} \equiv A \times (x_{ij})^{0.5}, \quad \text{where } A = 1 \text{ (for CH}_4\text{)}, 0.25 \text{ (for CO}_2\text{)}, 0.5 \text{ (for CO)}, 0.1 \text{ (for N}_2\text{O)} \quad (6)$$

This method attributes larger percentage uncertainties to smaller enhancements and hence lesser weight in the final solution and vice versa. This approach leads to an

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uncertainty matrix that attributes relatively similar percentage errors to both GHGs and VOCs, which should lead to a better fitting of the data through PMF.

Missing values are replaced by geometric mean of the tracer time series and their accompanying uncertainties are set at four times this geometric mean (Polissar et al., 1998) to decrease their weight in the solution. Based on the a priori treatment of the entire input data (scaling) and the corresponding outputs of the PMF analysis, a weighting-approach (for measurements from different instruments) as used in (Slowik et al., 2010) is not found to be necessary.

3.5 PMF source analysis

We use the customized software tool (PMF Evaluation Tool v2.04, PET) developed by Ulbrich et al. (2009) in Igor Pro (Wavemetrics Inc., Portland, Oregon) to run PMF, evaluate the outputs and generate statistics. The PET calls the PMF2 algorithm (described in detail in Ulbrich et al., 2009) to solve the bilinear model for a given set of matrices \mathbf{X} and \mathbf{S} for different numbers of factors p and for different values of FPEAK or SEED (defined and described later). The tool also stores the results for each of these combinations in a user friendly interface that allows simultaneous display of the factor profiles (FP) and time series (TS) of a chosen solution along with residual plots for individual tracers. A detailed explanation of PMF analysis performed in this study is provided in the Supplement. The supplement describes the PMF methodology of how the final number of user-defined factors was chosen (Sect. S1), the outcomes of linear transformations (rotations) of various PMF solutions (Sect. S2) and how uncertainties in the chosen solution were derived (Sect. S3). The standard deviations in the mass fractions of individual tracers in each factor profile and time series of each factor mass is evaluated using a bootstrapping analysis (Norris et al., 2008; Ulbrich et al., 2009) and described in Sect. S3. These error estimates are combined and propagated to derive PMF-based uncertainties for each factor's contribution to source-apportioned diurnal enhancements for a specific compound (Sect. 4).

4 Results and discussion

In Bakersfield, there are a multitude of pollutant sources, ranging from local to regional, from biogenic to anthropogenic, and from primary to secondary. We recognize that PMF analysis is not capable of precise separation of all sources. In PMF analysis, the analyst chooses the number of factor profiles to include in the solution and assigns a source category interpretation for each identified factor. The PMF factors are not unique sources but really statistical combinations of coincident sources. The chemical profile of each factor may contain some contributions from multiple sources that are co-located, or have a similar diurnal pattern. Such limitations have been observed previously by Williams et al. (2010) while applying PMF in an urban-industrial setting like Riverside, California. The user must infer the dominant source contributions to these individual factors. Our factor profile (FP) nomenclature is based on the closest explanation of the nature and distribution of emission sources in the region. The source factor names should be treated with caution bearing in mind the physical constraints of the solution and not used to over-explain our interpretation of the region's CH₄ and N₂O inventories.

A seven factor solution has been chosen to optimally explain the variability of the included trace gases. The factors have been named based on our interpretation of the emission "source" categories they represent, with corresponding colors which remain consistent in the discussion across the rest of the paper: evaporative and fugitive (black), dairy and livestock (orange), motor vehicles (red), agricultural + soil management (purple), daytime biogenics + secondary organics (light blue), non-vehicular urban (green) and nighttime anthropogenic + terpene biogenics (navy blue). Figure 4 presents the Factor Profile (FP) plots of each factor. The sum of the normalized contributions of the 50 species in each "source" is equal to 1 in the FP plots. Figures 5a through 5g present the diurnal profiles based on mean hourly concentrations (in normalized units) of each PMF factor with standard deviations explaining the variability. The interpretation of the individual FPs is discussed below (in Sects. 4.2–4.8). Molar

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emission factor (EF) of tracers with respect to (w.r.t) one another can be derived for each FP. These EFs can then be compared to those from previous source-specific and apportionment studies (Table 2 through 5). The ratio of PMF-derived total CH₄ enhancement to the input measured CH₄ enhancement ranges from 0.90 to 0.95 through the whole time series except outliers with really high values (> 500 ppb). For N₂O, the ratio is somewhat lower (0.82–0.92) and this is reflected in the PMF-derived uncertainties. The apportionment of some N₂O mass into a statistically weak and time-varying factor is discussed in Sect. 4.5. The general assessment is that PMF analysis is able to reconstruct a majority of the measured enhancements for both CH₄ and N₂O.

4.1 Time trends of measured CH₄, CO₂, CO, and N₂O

The time series of CH₄, CO₂, CO, and N₂O mixing ratios have been plotted in Fig. 6a through d while the diurnal variations have been plotted in Fig. 6e–h, respectively. The color markers in each plot indicate the median wind direction. The daily minima for the three GHGs and CO occur during the late afternoon period when daytime heating, mixing and subsequent dilution occurs rapidly. The daily minimum values of CH₄ and N₂O were larger than that observed at National Oceanic and Atmospheric Administration’s (NOAA) Mauna Loa station (Dlugokencky et al., 2014) by at least 70 and 0.5 ppb, respectively, for this period. This indicates that there are large regional sources of these two GHGs that keep the mixing ratio levels high. Winds during the highest temperature period between noon and evening (12:00–20:00 LT) almost always arrive through the urban core in the northwest. Any PMF factor whose dominant source direction is northwest is likely to contain contributions from VOCs emitted from urban sources, regional sources further upwind or contain contributions from secondary tracers generated from photochemical processing during the day. The three GHGs show a sharp increase during the nighttime when the inversion layer builds up and traps primary emissions close to the ground. For CO, measured concentrations show two distinct peaks in the diurnal plot (Fig. 6g). The observed early morning peak in the concentration is a combination of decreased dilution and fresh emissions from the morning motor vehicle traffic. The

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late evening peak in CO concentrations is not coincident with rush hour and is a result of build-up of evening emissions in the boundary layer that is getting shallower as the night progresses. Figure 6a indicates CH₄ enhancements of 500 ppb or more on almost every night with peak mixing ratios exceeding 3000 ppb on several occasions indicating an active methane source(s) in the region. Figure 6d shows that peak N₂O mixing ratios rise above 330 ppb on almost every night suggesting large sources in the region. Huge enhancements of CH₄, CO₂ and N₂O (on DOY 157, 164, and 165) (in Fig. 6a, b and d, respectively) may appear well-correlated to each other due to regional sources emitting into the inversion layer. However, the shapes of the diurnal cycles differ indicating different emission distributions, with the early morning maximum in CH₄ occurring before the maxima for CO₂ and N₂O, and the morning maximum for CO occurring slightly later. These differences in timing allow PMF analysis to differentiate their contributions into separate factors.

4.2 Factor 1: evaporative and fugitive emissions

Factor 1 has a chemical signature indicative of evaporative and fugitive losses of VOCs. The FP of this source is dominated by C₃ to C₆ straight-chain and branched alkanes and some cycloalkanes (Fig. 4). The average diurnal cycle of Factor 1 (Fig. 5a) shows a broad peak during late night and early morning hours after which the concentrations begin to decrease as the day proceeds reaching a minimum at sunset before beginning to rise again. This is strong indication of a source containing primary emissions that build up in the shallow pronounced nighttime inversions of southern SJV. The subsequent dilution of primary emissions as the mixed layer expands leads to low concentrations during the daytime.

Most of the propane, n-butane and pentanes signal is apportioned to this factor, but not the typical vehicle emission tracers like isooctane or CO or any of the alkenes or aromatics. Absence of these tracers in the FP suggests this factor is not related to vehicular exhaust and is a combination of non-tailpipe emissions and fugitive losses from petroleum operations. None of the CH₄ signal at the SJV site is apportioned to this fac-

tor, but almost all of the small straight-chain alkanes, exclusively apportion to this factor. This is in agreement with Gentner et al. (2014a) which concluded that VOC emissions from petroleum operations are due to fugitive losses of associated gas from condensate tanks following separation from CH₄. Table 2 compares EFs derived from this PMF study for the non-tailpipe (evaporative) and fugitive petroleum operation source factor with those from the Gentner et al. (2014a) study done on the same CalNex dataset using an independent source receptor model with chemical mass balancing and effective variance weighting method, and also to, reports of fugitive emissions from the oil and natural gas sources (Pétron et al., 2012; Gilman et al., 2013) and similar factors produced by other PMF studies (Buzcu and Fraser, 2006; Leuchner and Rappenglück, 2010; Bon et al., 2011). Good agreement of Factor 1 VOC EFs with those from the mentioned studies confirms petroleum operations in Kern County as the major source contributing to this factor. The PMF apportionment indicates that this source factor does not contribute to CH₄ enhancements observed at the SJV site (Fig. 7a) and thus most of the “associated” CH₄ is likely separated from the condensate prior to emission. As mentioned before, a tiny fraction (~ 5 %) of the total input CH₄ enhancement is not resolved into source-apportioned contributions. There could be a minor contribution to CH₄ signal from this source, which is unresolved within the framework of uncertainties in the PMF analysis.

4.3 Factor 2: motor vehicle emissions

Factor 2 has a chemical signature consistent with the tailpipe exhausts of gasoline and diesel motor vehicles. This source factor includes the combustion tracer CO, and other vehicular emissions tracers, such as isooctane (Fig. 4). Alkenes are a product of incomplete fuel combustion in motor vehicles, and almost all of the propene and a significant portion of the isobutene signal are attributed to this source factor. The diurnal variation of Factor 2 shows two distinctive peaks (Fig. 5b). The first peak occurs in the morning between 06:00 and 07:00 local time and is influenced by morning rush hour traffic, with suppressed mixing allowing vehicle emissions to build up. As the day pro-

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ceeds, accelerated mixing and dilution (and perhaps chemical processing of reactive VOCs) reduce the enhancements to a minimum by late afternoon. The evening peak mainly occurs as the dilution process slows down after sunset and emissions build up. The increased motor vehicle traffic in the evening adds more emissions to the shrinking boundary layer. This build-up reaches a peak around 22:00 in the night. The occasional high wind events from the northwest (unstable conditions) and the reasonably lesser number of vehicles operating on the roads during the late nighttime hours contribute to the relatively lower levels of enhancements as compared to the peaks on either side of this nighttime period.

Table 3 compares selective PMF derived EFs from vehicle emissions factor with the measured gasoline composition collected during CalNex in Bakersfield (Gentner et al., 2012), analysis of gasoline samples from Riverside in Los Angeles basin (Gentner et al., 2009) and ambient VOC emission ratios measured during CalNex at the Pasadena supersite (Borbon et al., 2013). Although, the two Bakersfield studies employ different source apportionment techniques (and so do the studies conducted in the Los Angeles basin), we observe a broad agreement of relative emission rates of vehicular emission tracers. This agreement validates our assertion that Factor 2 represents a broad suite of vehicular tailpipe emissions.

The PMF derived CH_4/CO EF in Factor 2 is $0.58 \text{ (mol mol}^{-1}\text{)}$ and is significantly higher than the range of $0.03\text{--}0.08 \text{ (mol mol}^{-1}\text{)}$ calculated from results of a vehicle dynamometer study of 30 different cars and trucks (Nam et al., 2004) and an EF of $0.014 \text{ (mol mol}^{-1}\text{)}$ calculated for SJV district during summer of 2010 using EMFAC, which is ARB's model for estimating emissions from on-road vehicles operating in California (EMFAC, 2011). In spite of the non-negligible proportion of CH_4 in the Factor 2 source profile, the contribution of the factor to CH_4 enhancements (Fig. 7a) at Bakersfield is negligible relative to the dairy and livestock factor.

The state GHG inventory attributes about 18 % of the 2010 statewide N_2O emissions to the on-road transportation sector (CARB 2012). Our PMF analysis shows essentially a negligible enhancement of N_2O associated with the vehicle emission Factor 2 with

a PMF derived $\text{N}_2\text{O}/\text{CO}$ EF of 0.00015 (mol mol^{-1}). The EMFAC generated $\text{N}_2\text{O}/\text{CO}$ EF in SJV during summer of 2010 is more than 20 times higher at 0.0034 (mol mol^{-1}). The PMF derived “vehicle emissions” contribution to N_2O is in stark contrast to the inventory and is an important outcome suggesting a significant error in the statewide inventory for N_2O .

4.4 Factor 3: dairy and livestock emissions

Factor 3 has a chemical signature indicative of emissions from dairy operations. This source factor is the largest contributor to CH_4 enhancements (Fig. 7a) and a significant portion of the N_2O signal (Fig. 7c). The FP also has major contributions from methanol (MeOH) and ethanol (EtOH), with minor contributions from aldehydes and ketones (Fig. 4). A separate PMF analysis with a broader set of VOC measurements at the same site showed that most of the acetic acid (CH_3COOH) and some formaldehyde (HCHO) signal attributed to this factor as well (Allen Goldstein, personal communication, 2014). All the above-mentioned VOCs are emitted in significant quantities from dairy operations and cattle feedlots (Filipy et al., 2006; Shaw et al., 2007; Ngwabie et al., 2008; Chung et al., 2010). About 70–90 % of the diurnal CH_4 signal is attributed to this factor (Fig. 7a) depending on the time of day. From propagation of errors, an uncertainty of 29 % is determined in the diurnal CH_4 enhancements in Factor 3. This source factor contributes about 60–70 % of the total N_2O daily enhancements as seen in Fig. 7c with an uncertainty of 33 %.

Comparing the Factor 3 profile to dairy source profiles from various studies is challenging. A dairy is, in essence, a collection of area sources with distinct emission pathways and chemical characteristics. Hence, a lot of dairy studies do not look at facility-wide emissions instead focusing on specific area sources within the facility. In contrast, PMF captures the covariance of CH_4 , N_2O , and VOCs emitted from the ensemble source as downwind plumes from dairies arrive at the site. Table 4 compares the PMF derived EFs of CH_4 w.r.t MeOH and EtOH with those from other studies. Pre-

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viously, cow chamber experiments (Shaw et al., 2007; Sun et al., 2008) have measured emissions from ruminants and their fresh manure; emissions have also been studied in a German cowshed (Ngwabie et al., 2008) and EFs have been derived from SJV dairy plumes sampled from aircrafts (Gentner et al., 2014a; Guha et al., 2014). Since enteric fermentation and waste manure is the predominant CH₄ source in dairies, CH₄ emission rates calculated by Shaw et al. (2007) are representative of a whole facility. However, their MeOH/CH₄ ratios are lower than those measured by PMF and aircraft studies. Animal feed and silage are the dominant source of many VOCs including MeOH and EtOH (Alanis et al., 2010; Howard et al., 2010) and the ratios in (Shaw et al., 2007) do not reflect these emissions. In (Ngwabie et al., 2008), experiments were performed in cold winter conditions (−2 to 8°C) when temperature dependent VOC emissions from silage and feed are at a minimum. The authors comment that MeOH emissions from California dairies is likely higher, as the alfalfa-based feed are a big source of MeOH owing to its high pectin content (Galbally and Kirstine, 2002). These observations explain why MeOH/CH₄ ratios in these studies are lower than PMF derived ratios. The PMF range for EtOH/CH₄ EF for Factor 3 agrees with the slope derived from ground-site data (Gentner et al., 2014a) and is similar, but somewhat larger than the German dairy study (Ngwabie et al., 2008). Miller and Varel (2001) and Filipuy et al. (2006) did not measure CH₄ emission rates so a direct derivation of EF w.r.t CH₄ is not possible. These studies, however, reported EtOH emission rates (from dairies and feedlots in United States) which are used to derive EFs w.r.t to CH₄ using an averaged CH₄ emission rate from (Shaw et al., 2007). Using this method, we get EFs that are comparable to PMF derived EF of CH₄/EtOH (Table 4). Hence, we demonstrate within reasonable terms that the relative fractions of masses in Factor 3 are consistent with CH₄ and VOC emissions from dairies.

Enteric fermentation is a part of the normal digestive process of livestock such as cows, and is a large source of CH₄ while the storage and management of animal manure in lagoons or holding tanks is also a major source of CH₄. According to the state GHG inventory (CARB, 2013), ~ 58% of the statewide CH₄ emissions results

factor has uncertainties greater than those for other factors (Fig. S4). This is potentially because not all crops emit the same combination of VOCs nor are all agricultural fields fertilized at the same time. The existence of this statistically weak factor is confirmed by bootstrapping runs (Sect. S3) and numerous PMF trials all of which produce a distinct factor with N₂O as a dominant contributor along with certain biogenic VOCs, though often in varying proportions. CO₂ is not included in the PMF analysis reported in the paper, but PMF runs involving CO₂ indicate that most of the CO₂ is apportioned to this factor. Plant and soil respiration (especially during the night) is a major source of CO₂ and the apportionment of CO₂ to Factor 4 confirms the nature of this source. The temporal correlation between CO₂ and N₂O is also evident in their average diurnal cycles (Fig. 6f and h), which have a coincident early morning peak. The absence of monoterpenes from the FP of this factor can be explained by their shorter atmospheric lifetimes compared to VOCs like acetone and MeOH and the rapid daytime mixing which dilutes the terpenoid emissions arriving at the site during the day. At nights, when the atmospheric dilution has been reduced to a low, monoterpene emissions from agriculture are more likely to get apportioned into a separate source factor dominant during nighttime, when temperature-sensitive biogenic emissions of MeOH and acetone can be expected to be a minor constituent in the FP (see Sect. 4.8).

Factor 4 is a significant source of GHGs contributing about 20–25 % of the total N₂O enhancements in the diurnal cycle (Fig. 7c) with a relatively larger uncertainty of 70 %. Kern County is one of the premier agricultural counties of California accounting for USD 4.2 billion (about 18 %) of the total agricultural revenue from fruits and nuts, vegetables and field crops (KernAg, 2011; CASR, 2013) and is also the biggest consumer of synthetic fertilizers. Agricultural soil management accounts for about 60 % of the statewide N₂O emission inventory (CARB, 2013). Our assessment of diurnal source distribution of N₂O emissions from the agriculture source factor (Fig. 7c) is consistent with the inventory estimates from agricultural and soil management sector.

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sector. Our results emphasize the significance of this sector in the SJV although we do not derive total emission estimates to compare directly with the inventory.

The contribution of fugitive emissions from the oil and gas industry in Bakersfield to CH₄ emissions is found to be negligible especially in the presence of the much larger dairy source. The PMF analysis, though, clearly establishes an evaporative and fugitive source that contributes to emissions of lighter hydrocarbons. This supports the conclusion that the majority of the CH₄ is being separated at the point of extraction from the “associated gas” and is not released with fugitive emissions (Gentner et al., 2014a). Kern County produces 75 % of all the oil produced in California (~ 6% of US production) and has 81 % of the state’s 60 000+ active oil wells (CDC, 2013). There is, however, a surprising scarcity of measured data to quantify the estimates of fugitive CH₄ from the prolific oil fields in the county and validate the bottom-up, activity data-based inventory. Currently, fugitive emissions from fossil fuel extraction and distribution contribute ~ 5 % to the county’s CH₄ emissions inventory (KernGHG, 2012). Nationwide, a number of recent studies have reported significantly higher emissions of fugitive CH₄ from oil and gas operations in other regions (Karion et al., 2013; Miller et al., 2013; Pétron et al., 2012). The PMF apportionment in this study is consistent with the fraction of fugitive CH₄ emissions in the inventory (< 5%) but the PMF method in itself is limited in accurate partitioning of minor sources.

We find that the vehicle emissions source factor identified in this study makes no detectable contribution to observed N₂O enhancements. Our findings do not agree with the significant contribution (~ 18 %) of the transportation sector to the state’s N₂O emission inventory (CARB, 2013). Vehicle dynamometer studies have indicated rapidly declining N₂O EFs with advancement in catalyst technologies, declining sulfur content in fuel and newer technology vehicles (Huai et al., 2004). N₂O emissions from California vehicles, required to meet progressively stringent emission standards, are expected to decline and should have a minimal contribution to the CARB inventory in this decade. However, it seems the updates to the mobile N₂O emissions inventory is not keeping in pace with the improvements in vehicle catalyst technologies and corresponding de-

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cline in tailpipe N₂O emissions. Bakersfield is a fairly large population urban region (~ 500 000) and the essentially non-existent contribution of the PMF vehicle emissions source to the N₂O apportionment and large divergence of the PMF derived N₂O/CO EF from the state inventory EF for motor vehicles is a significant outcome pointing to overestimation of N₂O from motor vehicles in the inventory.

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Table 1. PMF dataset with total samples (N) and mixing ratio range (in pptv).

Class	Compound	N	1st percentile	99th percentile	Background	
GHG	CH ₄ ^{a, c}	619	1855.0	3400.8	1813.6	
	CO ₂ ^{b, c}	619	390.8	468.3	390.0	
	N ₂ O ^{a, d}	490	323.3	339.5	323.2	
combustion tracer	CO ^{a, d}	653	118.9	330.6	102.1	
	propane	592	580.8	30 839.0	455.5	
	straight chain alkanes					
	<i>n</i> -butane	587	96.4	12 649.0	73.6	
	<i>n</i> -pentane	647	93.2	3805.4	64.4	
	<i>n</i> -hexane	647	23.1	960.5	17.2	
	dodecane	643	1.56	54.3	0	
branched alkanes	isopentane	646	165.4	7490.5	100.4	
	2,3-dimethylbutane	650	52.5	1747.7	41.1	
	2,5-dimethylhexane	651	2.37	145.8	0	
	isooctane	647	16.6	476.9	12.3	
	4-ethylheptane	651	1.45	52.6	0	
	dimethyl undecane	643	0.46	24.9	0	
cyclo alkanes	methylcyclopentane	647	23.3	1329.6	20.3	
	methylcyclohexane	649	8.10	813.9	0	
	ethylcyclohexane	651	1.78	169.1	0	
alkenes	propene	592	34.7	3299.9	28.6	
	isobutene	595	16.7	422.1	10.7	
aromatics	toluene	647	48.8	1749.5	33.1	
	ethylbenzene	647	5.83	282.0	0	
	<i>m</i> , <i>p</i> -xylene	647	21.8	1127.1	21.8	
	<i>o</i> -xylene	647	4.31	405.0	0	
	cumene	640	0.55	22.8	0	
	1-ethyl-3,4-methylbenzene	651	2.22	358.6	0	
	<i>p</i> -cymene	649	0.84	93.9	0	
	indane	647	0.45	27.9	0	
	1,3-dimethyl-4-ethylbenzene	635	0.46	23.9	0	
	naphthalene	654	0.44	19.9	0	
	unsaturated aldehyde	methacrolein	573	14.2	337.0	0
	alcohol	methanol	429	2636.81	88 691.8	1085.2
		ethanol	598	1021.93	65 759.8	1021.9
isopropyl alcohol		583	25.7	2001.0	25.7	
ketone	acetone	663	142.9	3505.8	142.9	
	methyl ethyl ketone	605	8.55	1111.2	0	
	methyl isobutyl ketone	629	2.03	71.9	0	
aldehyde	propanal	636	3.68	140.8	0	
	butanal	589	1.72	35.1	0	
biogenics	isoprene	651	9.70	310.0	0	
	alpha-pinene	740	1.67	525.8	0	
	<i>d</i> -limonene	641	1.10	357.1	0	
	nopinone	614	0.78	89.5	0	
	alpha-thujene	591	0.52	23.8	0	
	camphene	645	0.72	100.3	0	
	chloroalkanes	chloroform	647	34.1	209.3	31.6
	tetrachloroethylene	641	3.41	120.9	0	
	1,2-dichloroethane	640	20.6	103.8	20.6	
	1,2-dichloropropane	627	2.40	28.4	0	
sulfides	carbon disulfide	610	7.84	133.7	0	
	thiol	491	4.54	685.8	0	

^a parts per billion volume (pptv).

^b parts per million (ppmv).

^c measured using LGR Fast Green House Gas Analyzer.

^d measured using LGR N₂O/CO analyzer.

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Table 3. Comparison of hydrocarbon ratios to toluene ($\text{gC}(\text{gC})^{-1}$) from PMF vehicle emission factor with similar ratios from other California specific studies.

Study	Bakersfield PMF vehicle emissions factor ^a	Bakersfield gasoline source profile ^{b, c}	Riverside liquid gasoline profile ^e	CalNex Los Angeles ambient emission ratios ^g
Source	This study	Gentner et al. (2014)	Gentner et al. (2009)	Borbon et al. (2013)
CH ₄	8.1 ± 2.1	NA	NA	NA
CO	14.0 ± 0.4	NA	NA	22.26
toluene	1	1	1	1
isopentane	0.69 ± 0.01	0.77 ± 0.04	0.64–0.84	1.95
isooctane	0.29 ± 0.03	0.34 ± 0.02	0.64–0.80	NA
<i>n</i> -dodecane	0.03 ± 0.001	(0.04 ± 0.004) ^d	NA	NA
methylcyclopentane	0.24 ± 0.01	0.32 ± 0.02	NA	NA
ethyl benzene	0.17 ± 0.01	0.14 ± 0.01	NA	0.2
<i>m/p</i> -xylene	0.65 ± 0.01	0.65 ± 0.03	(0.45–0.52) ^f	0.64
<i>o</i> -xylene	0.22 ± 0.01	0.23 ± 0.01	NA	0.24

^a Errors are SD of 12 unique PMF solutions between FPEAK = −1.6 to +0.4; see Sect. S2.

^b Derived from liquid gasoline fuel speciation profile (Table S9; Gentner et al., 2012).

^c Errors bars derived from propagation of uncertainties.

^d Derived by combining diesel fuel and gasoline speciation profile (Tables S9 and S10; Gentner et al., 2012).

^e Summer data.

^f Only *m*-xylene.

^g Derived from Linear Regression Fit slope of scatterplot from CalNex Pasadena supersite samples.

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Table 4. Comparison of PMF dairy and livestock emission factors (mmol mol^{-1}) with previous studies.

Study	Source	Cow/manure type (if applicable)	methanol/methane EF avg. (range)	ethanol/methane EF avg. (range)
PMF analysis of regional measurements	This study		15–47	9–32.2
Environmental chamber with cows and/or manure	Shaw et al. (2008)	Dry	3.2 (0.6–7.4)	NA
Environmental chamber with cows and/or manure	Sun et al. (2008)	Lactating	1.9 (0.8–3.6)	NA
		Dry	13.4 (4–25)	14.4 (11–19)
Cowshed with regular dairy operations (winter)	Ngwabie et al. (2008)	Lactating	19.2 (15–25)	24.2 (18–32)
Cow stall area with regular dairy operations (summer)	Filipy et al. (2006)		2.0 (1.6–2.4)	9.3 (4–16)
Manure from cattle feedlot	Miller and Varel (2001)	Fresh (< 24 h)	NA	14 ^b
		Aged (> 24 h)		118 ^b
Measured slope of regression (CalNex 2010)	Gentner et al. (2014)		7.4 (7–16) ^c	18 ^d
Sampling of dairy plumes from aircraft (CABERNET 2011)	Guha et al. (2014)		9.6 (9–30) ^c	NA

^a Calculated based on CH_4 emission rate of $4160 \mu\text{g cow}^{-1} \text{s}^{-1}$ for mid-lactating cows (Shaw et al., 2007).

^b Calculated based on CH_4 emission rate of $4160 \mu\text{g cow}^{-1} \text{s}^{-1}$ for mid-lactating cows (Shaw et al., 2007) and ethanol emission rate for fresh and aged manure of 175 and $1223 \mu\text{g cow}^{-1} \text{s}^{-1}$, respectively, derived by Filipy et al. (2006).

^c slope of regression with range of measured slopes (in parentheses) from sampling of dairy plumes by aircraft.

^d ground site data; lower limit of slope of non-vehicular ethanol vs. methane.

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Table 5. Comparison of PMF agricultural and soil management emission factor for acetone vs. methanol ($\text{gC}(\text{gC})^{-1}$) with ratios of basal emission factors generated for major crops grown in the Kern County. Errors denote SDs computed by propagation of uncertainty.

Bakersfield PMF agricultural and soil management factor	Almond greenhouse summer 2008	Table grape greenhouse summer 2008	Pistachio greenhouse summer 2008	Navel oranges greenhouse summer 2008*	Valencia oranges greenhouse summer 2008
This study 0.58 ± 0.37	Gentner et al. (2014b) 0.14 ± 0.2	Gentner et al. (2014b) 0.04 ± 0.02	Gentner et al. (2014b) 0.5 ± 0.6	Fares et al. (2011) 0.57 ± 0.1	Fares et al. (2012) 0.5 ± 0.3

* branch with flowers not removed.

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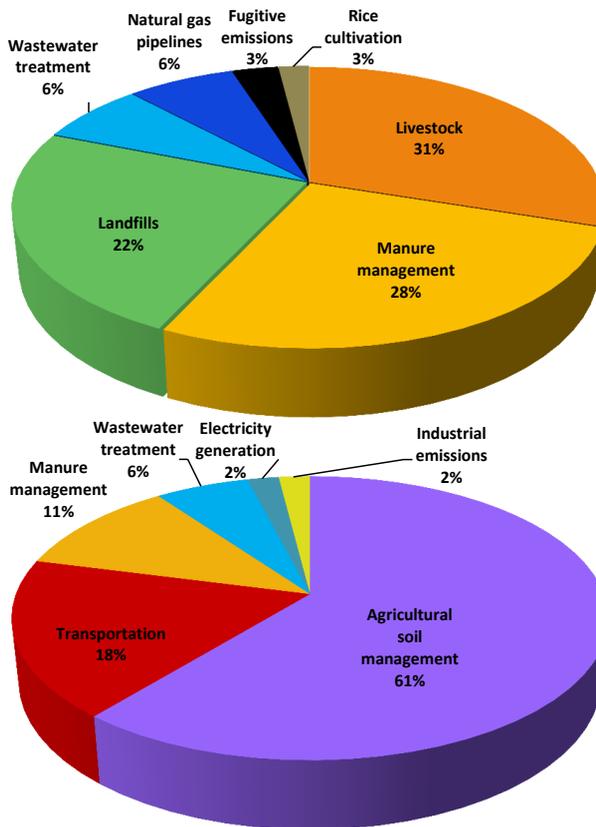


Figure 1. 2011 California emission inventory for (top) methane (CH₄) – 32.5 million tCO₂ eq at GWP = 25; and (bottom) nitrous oxide (N₂O) – 13.4 million tCO₂ eq at GWP = 298. (Source: CARB GHG Inventory Tool, August 2013).

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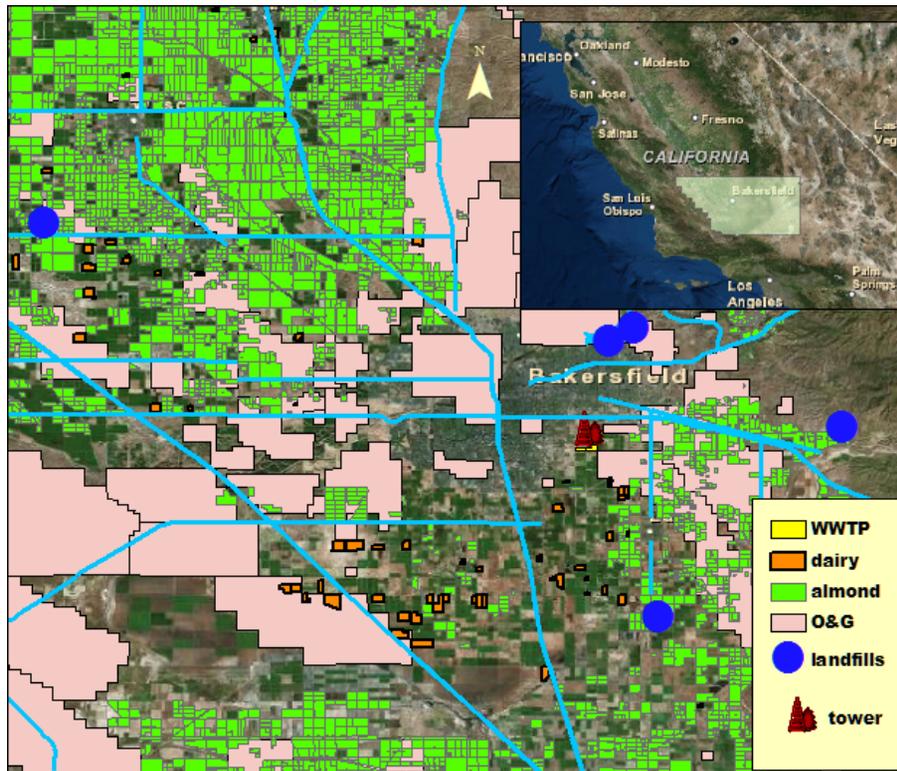


Figure 2. Map of potential sources of methane and nitrous oxide in/around the city of Bakersfield and the surrounding parts of the valley. The inset map is a zoomed out image of the southern part of San Joaquin Valley (SVJ) with location of Kern County superimposed. The light blue lines mark the highways, WWTP stands for waste water treatment plant, and O&G stands for oil and gas fields. The location of the CalNex experiment site is marked by the “tower” symbol.

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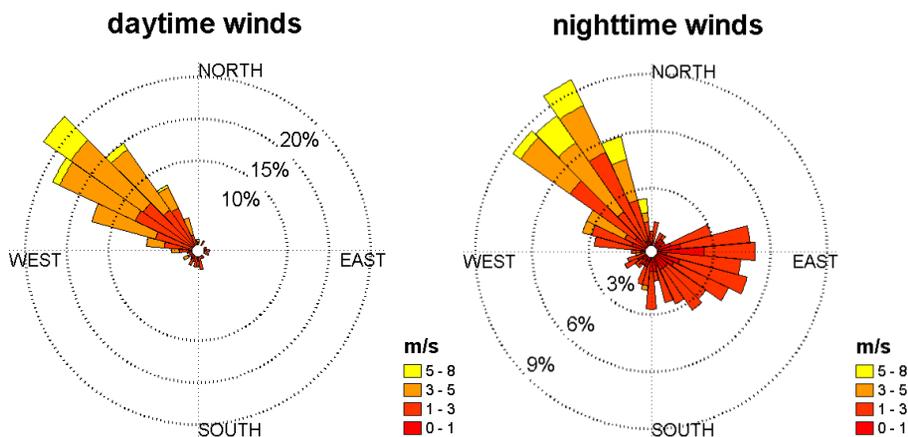


Figure 3. Wind rose plots showing mean wind direction measured at the site during (left) day time (07:00–16:00 h), and (right) nighttime (17:00–06:00 h). The concentric circles represent the percentage of total observations; each colored pie represents a range of 10° while the colors denote different wind speed ranges.

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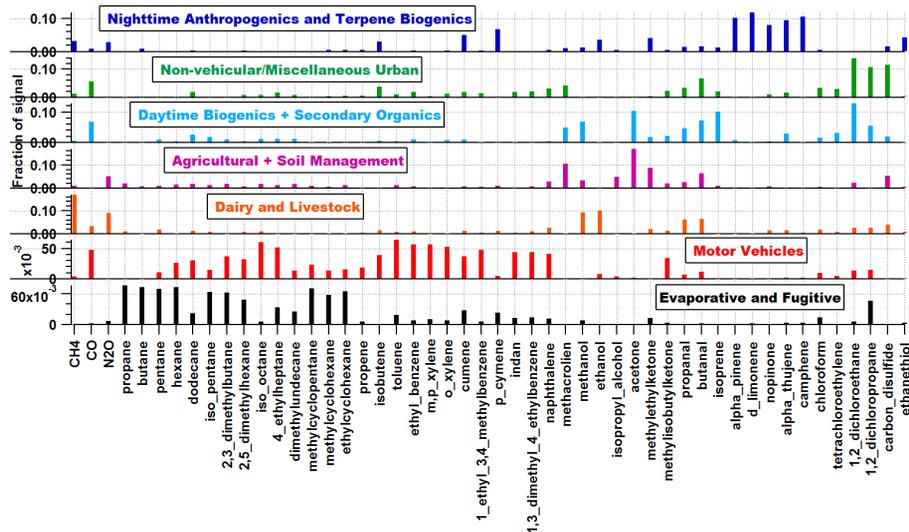


Figure 4. Source profile of the seven factors derived using PMF. The source factors are evaporative and fugitive, motor vehicles, dairy and livestock, agricultural + soil management, daytime biogenics + secondary organics, urban, and nighttime anthropogenics + terpene biogenics. The x axis represents the normalized fraction of mass in each source factor, while the y axis lists all the chemical species included in the PMF analysis.

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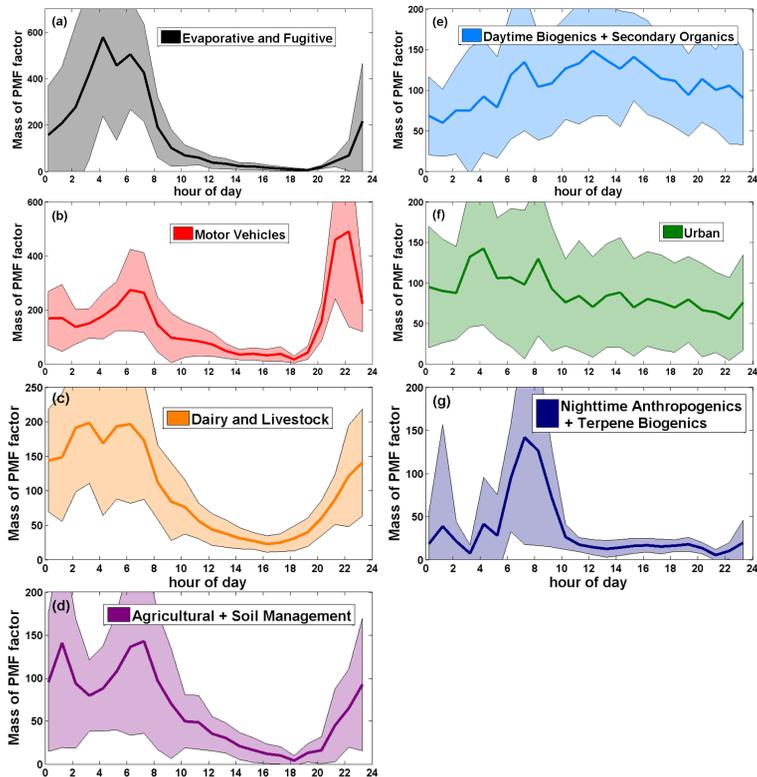


Figure 5. Mean hourly diurnal plots of PMF source factor concentration enhancements for (a) evaporative and fugitive, (b) motor vehicles, (c) dairy and livestock, (d) agricultural + soil management, (e) daytime biogenics and secondary organics, (f) non-vehicular/miscellaneous urban and (g) nighttime anthropogenics + terpene biogenics. The x axis represents sum of normalized mass concentrations from all tracers contributing to the factor. The y axis is hour of day (local time). The solid lines represent the mean and the shaded area represents the standard deviation (variability) at each hour.

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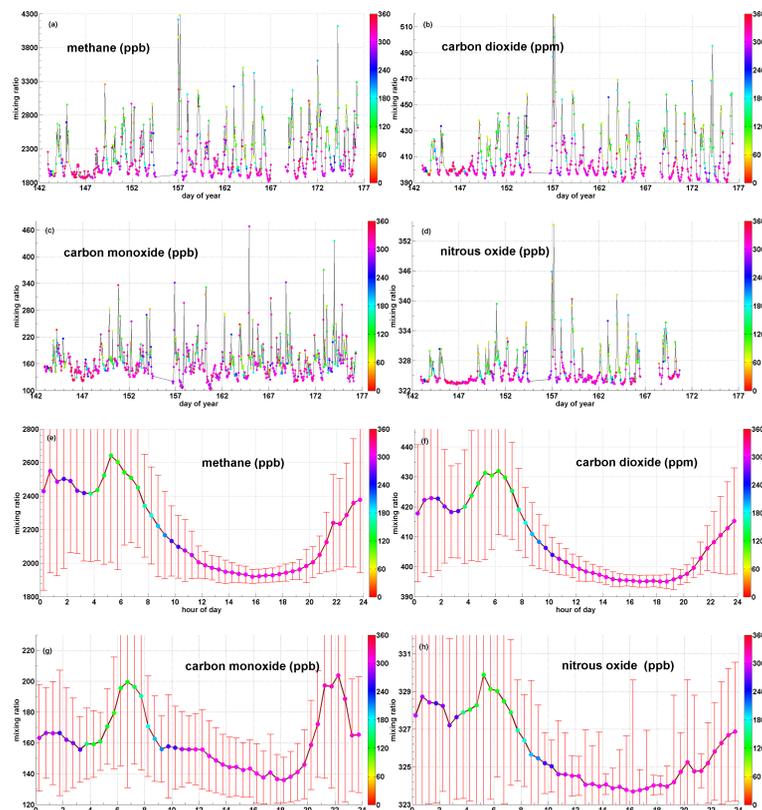


Figure 6. Time series of (a) CH_4 , (b) CO_2 , (c) CO , and (d) N_2O obtained from 30 min averages over the entire sampling period. The color bar indicates the average wind direction during each 30 min period. Mixing ratios plotted as average diurnal cycles for (e) CH_4 , (f) CO_2 , (g) CO and (h) N_2O along with wind direction. The curve and the red whiskers represent the mean and the standard deviations about the mean, respectively.

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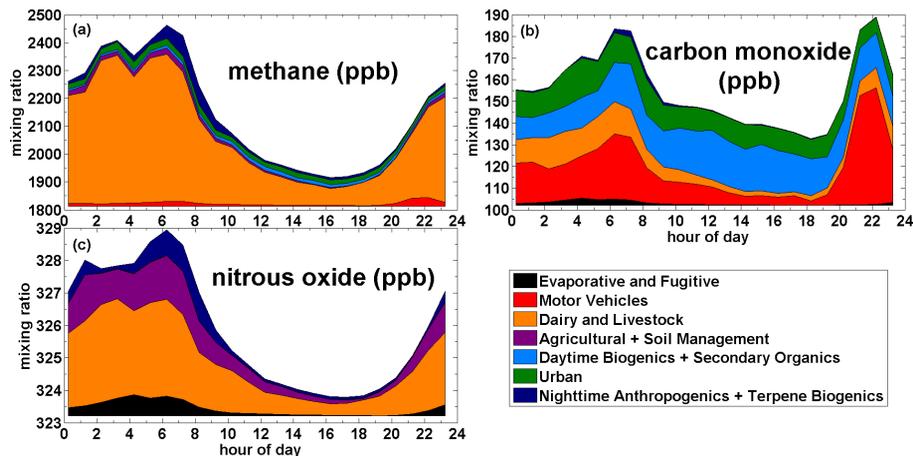


Figure 7. Diurnal plot of PMF derived (a) CH₄, (b) CO, and (c) N₂O concentrations sorted by PMF source category. The legend on the bottom right shows the names of the PMF source factor which each color represents. The PMF derived enhancements from each source have been added to the background concentrations.