



This discussion paper is/has been under review for the journal Atmospheric Chemistry and Physics (ACP). Please refer to the corresponding final paper in ACP if available.

Increasing concentrations of dichloromethane, CH₂Cl₂, inferred from CARIBIC air samples collected 1998–2012

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Received: 18 June 2014 – Accepted: 30 July 2014 – Published: 12 August 2014

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Published by Copernicus Publications on behalf of the European Geosciences Union.

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Abstract

Dichloromethane, CH₂Cl₂, is a short-lived chlorocarbon of predominantly anthropogenic origin. Increasing industrial usage and associated emissions resulted in an increasing atmospheric burden throughout the 1900s. Atmospheric abundance peaked around 1990 and was followed by a decline in the early part of the 21st century. Despite the importance of ongoing monitoring and reporting of atmospheric CH₂Cl₂ (it is a regulated toxic air pollutant and a contributor to stratospheric ozone depletion) no time series has been discussed in detail since 2006. The CARIBIC project (Civil Aircraft for the Regular Investigation of the atmosphere Based on an Instrument Container) has analysed the halocarbon content of whole air samples collected at altitudes of between ~10–12 km via a custom-built container installed on commercial passenger aircraft since 1998, providing a long-term record of CH₂Cl₂ observations. In this paper we present this unique CH₂Cl₂ time series, discussing key flight routes which have been traversed at various times over the past 15 years. Between 1998 and 2012 increases were seen in all northern hemispheric regions and at different altitudes, ranging from ~7–9 ppt in background air to ~12–15 ppt in regions with stronger emissions (equating to a 38–69% increase). Of particular interest is the rising importance of India as a source of atmospheric CH₂Cl₂: based on CARIBIC data we provide regional emission estimates for the Indian subcontinent and show that regional emissions have increased from 3–15 Gg yr⁻¹ (1998–2000) to 16–25 Gg yr⁻¹ (2008). Potential causes of the increasing atmospheric burden of CH₂Cl₂ are discussed. One possible source is the increased use of CH₂Cl₂ as a feedstock for the production of HFC-32, a chemical used predominantly as a replacement for ozone-depleting substances in a variety of applications including air conditioners and refrigeration.

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1 Introduction

Dichloromethane, CH₂Cl₂, is a short-lived chlorocarbon of mainly (up to 90 %) industrial origin. Its main applications include use in paint strippers, degreasers and solvents; in foam production and blowing applications; and as an agricultural fumigant (Montzka et al., 2011b). The most recent estimation (Simmonds et al., 2006) for its natural component suggests it is comprised of a 10 % combined biomass burning and marine source. With an atmospheric lifetime of around 5 months (Montzka et al., 2011b), CH₂Cl₂ displays significant atmospheric spatial variations and temporal trends. Its seasonal cycle is mainly due to reaction with the OH radical, with maxima in late winter/early spring and corresponding minima in late summer or early autumn (Cox et al., 2003). There are no discernible seasonal variations in emissions or end uses (Gentner et al., 2010; McCulloch and Midgley, 1996). Significantly higher concentrations are observed in the Northern Hemisphere (NH, Southern Hemisphere = SH) due to the predominant industrial source. A NH : SH mole fraction ratio of 2.7 has been reported (Simmonds et al., 2006).

Long-term trends in atmospheric CH₂Cl₂ have been observed since measurements began in the 1970s. In the SH, Advanced Global Atmospheric Gases Experiment (AGAGE) atmospheric measurements began at Cape Grim in 1998 and reported mean CH₂Cl₂ concentrations for 1998–2000 of 8.9 (±0.2) ppt (Cox et al., 2003). These early measurements were supported by firn records which indicated that SH CH₂Cl₂ concentrations increased from 1–2 ppt at the beginning of the record (pre-1940) to 9 ppt around 1990 (Trudinger et al., 2004). The earliest reported NH atmospheric measurements of CH₂Cl₂ were made in the mid-1970s and observed concentrations of 35 ± 19 ppt (Cox et al., 1976). A range of global measurements in the 1980s and 1990s (many of which will be discussed further throughout this manuscript and are included in Table 2) showed a consistent picture of peaking concentrations, with an average of ~ 30–40 ppt around 1990, followed by a decline linked to decreasing industrial use of CH₂Cl₂ (McCulloch et al., 1999). Measurements made between 1989 and 1996 at Alert, Canadian

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Arctic, observed a decline of around -4% (-1.8 ppt) per year (Gautrois et al., 2003). An in-depth discussion of many of the observations during this period can be found in Simmonds et al. (2006).

Short-lived chlorocarbons, including CH_2Cl_2 , contribute to stratospheric chlorine and its cycling with stratospheric ozone. Current and projected decreases of longer-lived anthropogenic chlorocarbons (such as CH_3CCl_3 , CCl_4 , halons and CFCs) could mean a greater relative importance of shorter-lived chlorocarbons with respect to stratospheric chlorine, especially if their atmospheric abundances were to increase. At 10–12 km, the contribution of very short-lived chlorocarbons to total atmospheric chlorine is dominated by CH_2Cl_2 and CHCl_3 . Whilst CHCl_3 is of predominantly natural origin, CH_2Cl_2 has a mainly anthropogenic source and so is susceptible to changes in industrial emissions. Montzka et al. (2011b) reported increasing CH_2Cl_2 concentrations in recent years, with an increase of around 8% between 2007 and 2008, based on updated AGAGE data from Simmonds et al. (2006). There was no corresponding increase in CHCl_3 , 7% of which is believed to be of natural origin (Worton et al., 2006). The increase was also noted in Montzka et al. (2011a, see their Supplement) whose time series of CH_2Cl_2 between 1995 to 2009 shows increasing atmospheric concentrations in recent years. CH_2Cl_2 is also of concern as it is also a toxic air pollutant and carcinogen and, as such, is regulated by national and EU law. Altogether, an improved understanding of the recent changes in atmospheric CH_2Cl_2 is desirable. For this reason we present and discuss measurements of air samples obtained by the CARIBIC (Civil Aircraft for the Regular Investigation of the atmosphere Based on an Instrument Container) observatory.

2 Methods

CARIBIC centres on a large air-freight container accommodating a range of scientific equipment which is deployed monthly aboard a commercial passenger aircraft departing from Germany for up to four consecutive long-haul flights. Details of both

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CARIBIC phases can be found on our website, caribic-atmospheric.com. CARIBIC phase 1 (CARIBIC1) operated between 1997 and 2002 aboard a Boeing 767 departing for several global destinations from either Düsseldorf or Munich airport. Whole air samples were collected using 12 21 L stainless steel tanks pressurised to 17 bar. Details of CARIBIC1, including the range of other measurements made, can be found in Brenninkmeijer et al. (1999). Halocarbon data are available for 1998–2002.

Between 2003 and 2005 a new container was developed and this system was deployed aboard a Lufthansa Airbus 340–600 departing from Frankfurt Airport. CARIBIC phase 2 (CARIBIC2) began in 2005 and, at the time of writing, is still in operation. Samples are taken en route to destinations across the globe with flights occurring approximately monthly. Two whole air samplers consisting of 14 2.7 L glass flasks collect 28 air samples for halocarbon, non-methane hydrocarbon (NMHCs) and greenhouse gas measurements at pre-determined intervals during the flight; mainly within cruising altitudes of around 10–12 km. Filling times are between 30–90 s, averaging 45 s or 10 km of the flight path. Further air sampler information can be found in Baker et al. (2010) and Schuck et al. (2009). The fully-automated CARIBIC2 system contains a range of other sampling equipment, including, but not limited to, equipment for the in-situ or post-flight analysis of ozone (O_3), carbon monoxide (CO), aerosols and water vapour. Further information can be found in Brenninkmeijer et al. (2007).

During both CARIBIC1 and CARIBIC2 air samples were sent to the University of East Anglia (UEA, UK) for halocarbon analysis via gas chromatography mass spectrometry (GCMS). During CARIBIC1 subsamples were removed from the main canisters into electropolished stainless steel cans and sent to UEA. For CARIBIC2 the whole air sampling units were sent directly to UEA for analysis. During CARIBIC1 two separate GCMS systems were used. The first was an Agilent/HP 5890A GC coupled to a double-focusing, tri-sector mass spectrometer (V.G./Micromass Autospec). Each 200 mL air sample was dried by passing through magnesium perchlorate (MPC) before being trapped in a previously evacuated stainless steel loop filled with 100 μ m glass beads and immersed in liquid argon (-186°C). The bulk of the air passed into an

of the quadrupole analyser is more stable than the Autospec, the working standard is analysed less frequently.

To assist with the transition between the V.G. Autospec and the Entech system parallel analysis was conducted for two flights in July 2000. Agreement between the two systems was excellent. Of the 24 samples analysed on both systems all but 5 had a difference of less than ± 1 ppt (corresponding to a difference of $< 3\%$ standard deviation, σ , or less than the precision of these instruments). For the 5 remaining samples the difference was less than ± 2 ppt. The CH_2Cl_2 samples that were analysed on both systems were treated in the following manner. If the difference was less than ± 1 ppt ($3\% \sigma$) the values were averaged and the variation between the two measurements incorporated into error bars plotted with these values. Where the difference was greater than ± 1 ppt the V.G. Autospec value was selected based on the better precision of this instrument.

To provide additional support to the CARIBIC2 dataset, three flights, one per year between 2009–2011, were also analysed on a highly sensitive Waters Autospec magnetic sector GCMS. This system is the direct replacement of the V.G. Autospec described above and, whilst a number of minor modifications have been made to the analytical procedure (see Laube et al., 2010), the system is essentially very similar. Where the Entech and Autospec values agreed within $\pm 1\sigma$ (based on replicate Autospec measurements) the values were combined. As with CARIBIC1, these values all agreed within ± 1 ppt. For the remaining samples the values from the higher precision Autospec system have been used.

For CH_2Cl_2 the UEA calibration is tied to the 2003 GCMS gravimetric scale of the Global Monitoring Division of the Earth System Research Laboratory of the National Oceanic and Atmospheric Administration (NOAA-ESRL-GMD) in Boulder, CO, USA. A number of calibrated, high pressure whole-air samples collected at Niwot Ridge (a remote site near Boulder) were acquired between 1994 and 2009. These were used for the propagation of mixing ratios to all CARIBIC measurements. Further details on this procedure can be found in the Supplementary Material. In Sect. 3.1, CARIBIC data are

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gate the concentration of CH_2Cl_2 in air masses with the potential to enter the tropical tropopause layer (TTL). Further details of these five case studies can be found in Table 1 and they are highlighted in Fig. 1. Throughout the manuscript mean values prefaced by \pm refer to the 1 sigma standard error associated with this average.

3.1 Europe

The CH_2Cl_2 time series of European CARIBIC and NOAA Mace Head data can be seen in Fig. 2a. A fairly consistent seasonal cycle is observed in the boundary layer air samples from Mace Head whereas the CARIBIC data show greater variability. This variability in the CARIBIC data is mainly as a result of the fact that these samples represent a wide variety of air masses sampled over a large area (Fig. 2c). Analysis of back trajectories indicates that air sampled by CARIBIC over Europe originates from a large NH geographical region, including industrial areas where high emission “pollution” events may occur as well as contrasting regions where pristine tropospheric air masses are sampled. In contrast, the Mace Head site commonly samples clean sector air. Although previous studies have observed seasonality in data collected by aircraft at an average altitude of 4 km (Miller et al., 2012) we do not see a strong seasonal pattern at 10–12 km in our more sporadic dataset. Further analysis, discussed in subsequent sections, will highlight the importance of strong source regions (e.g. India and Southeast Asia) on observed CH_2Cl_2 concentrations in the mid and upper troposphere.

The trend in European observations of CH_2Cl_2 is shown in Fig. 2b. Error bars represent the 1σ variation associated with the mean of all tropospheric samples taken within each year. As seen in Fig. 2a, only a small number of NOAA samples were collected in the first few years of the dataset. Due to this small sample size, biases, for example the influence of seasonality, could be introduced. This adds an additional, unquantified uncertainty to these averages. To account for this, data from Mace Head can be compared to data collected at other NOAA NH sites such as Barrow, Alaska. Data from Barrow show a very similar pattern to those from Mace Head and support the trend seen at Mace Head (data not shown but available

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coverage, and so cannot be used to estimate a NH:SH ratio, they do show an increasing latitudinal variation indicative of increasing NH industrial activity with respect to CH₂Cl₂.

Mean tropospheric values measured in years for which data are available are shown in Fig. 3b, separated into samples taken above and below 30° N. A statistically significant (Mann-Whitney Rank Sums Test, $p < 0.001$) increase in CH₂Cl₂ is seen between samples collected in 1998–2001 and those collected in 2009–2011. As the northern (above 30° N) segment of the flight route crosses Europe the increase seen over time, 12.5 (± 1.8) ppt or 58 %, is similar to that discussed in Sect. 3.1. The increase is slightly lower at latitudes $< 30^\circ$ N, a 44 % increase in CH₂Cl₂, although still statistically significant (Mann-Whitney test at $p < 0.001$). Further details of the concentrations observed above and below 30° N are provided in Table 2. Inferring year-on-year trends is difficult given the varying data coverage between years. However, the increase seen between 2009 and 2011 (Fig. 3b), along with the European dataset (Fig. 2), suggest that concentrations continue to increase into the 2010s.

Despite the importance of biomass burning with respect to atmospheric trace gases over Africa (Roberts et al., 2009) no correlation ($r = 0.14$, $p > 0.05$) was observed between CH₂Cl₂ and the common combustion tracer CO. Enhancements of CO, which commonly peak near the equator in CARIBIC data (Umezawa et al., 2014), are predominantly from biomass burning sources. In contrast to the latitudinal distribution of CO, CH₂Cl₂ decreased constantly from north to south (Fig. 3). Observations of CH₂Cl₂ along the CARIBIC flight track to South Africa appear to be dominated by a strong NH source and subsequent decline towards lower latitudes, with little impact from biomass burning. This observation fits with a recent study which saw no evidence for CH₂Cl₂ emissions in boreal biomass burning plumes, suggesting that previous calculations of CH₂Cl₂ emissions from biomass burning (e.g. Rudolph et al., 1995) were overestimates (Simpson et al., 2011). Whilst emissions from boreal and tropical forest fires may differ, recent analysis of air samples collected during flights over biomass burning events in

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sence of seasonality based on their analysis of data on the global industrial data us of CH_2Cl . Previous analyses of meteorological parameters during 2008 (Rauthe-Schöch et al., 2014; Schuck et al., 2010) have demonstrated that the monsoon anticyclone was present in July–September. For emission estimates we take all tropospheric samples where both CO and CH_2Cl_2 were measured and which were collected $< 40^\circ \text{N}$ (Baker et al., 2011; Schuck et al., 2010) in July–September 2008, a total of 35 samples.

Emission estimates are usually calculated using emission or enhancement ratios whereby the compound of interest is compared to a compound with which it correlates and for which emissions are quantified, in this case CO . Equation (1) shows the calculation of an emission estimate for CH_2Cl_2 , $E_{\text{CH}_2\text{Cl}_2}$, from a known emission of CO , E_{CO} , and an emission ratio. In this case, we use the slope of the linear regression from the correlation between CH_2Cl_2 and CO as our emission ratio, rather than a ratio based on our observed “excess” monsoon concentrations relative to a background mixing ratio, as discussed in Andreae and Merlet (2001) and Le Canut et al. (1996).

Before discussing our emission estimates we provide details of the assumptions and potential errors associated with this method and our treatment of these factors. Firstly, this method assumes that the two compounds share a common, dominant source and/or that emissions are co-located. Whilst the sources of CO and CH_2Cl_2 differ, as CO has a larger relative source from biofuel burning (Dickerson et al., 2002), CO still provides the best option for emission estimates in this region. Other, anthropogenic, compounds have their own limitations. Emission estimates are available for SF_6 but it does not correlate well with CH_2Cl_2 , possibly as it has large point sources and is not as widely and evenly distributed as sources of CH_2Cl_2 , CO and many other trace gases. For other compounds, e.g. HFC-32 (Sect. 3.6) or C_2Cl_4 , Indian emission estimates are not available. On the scale of a regional emission estimate, CO and CH_2Cl_2 sources are co-located: both show strong signatures from the Indian subcontinent where it is known that air-masses sampled within the monsoon anticyclone have likely originated from. Comparisons between CO and anthropogenic chlorocarbons, including CH_2Cl_2 ,

have also been conducted in several other studies including Gentner et al. (2010); Millet et al. (2009); Simmonds et al. (2006) and Palmer et al. (2003).

To further support the suitability of the CH_2Cl_2 -CO ratio for estimating CH_2Cl_2 emissions we describe two analyses which demonstrate that the variability we observe is due to recent emissions, as opposed to variations in transport time or route prior to sampling. Firstly, we compared the $\Delta\text{CH}_2\text{Cl}_2/\Delta\text{CO}$ value from our sample set ($n = 35$) to the $\Delta\text{CH}_2\text{Cl}_2/\Delta\text{CO}$ value obtained from a smaller dataset based on the method used in Baker et al. (2011). Baker et al., when performing emission estimates for the same CARIBIC 2008 monsoon dataset, minimised the influence of variability with respect to processing or other transport effects by selecting a dataset that included only those samples whose back trajectories indicated low level (pressure > 600 hPa) contact within the monsoon anticyclone in the previous 5 days ($n = 15$). Comparing this subset of samples to our full sample set gives a very similar $\Delta\text{CH}_2\text{Cl}_2/\Delta\text{CO}$ and r value. The $\Delta\text{CH}_2\text{Cl}_2/\Delta\text{CO}$ and r values for our dataset are shown in Table 3 and for the Baker et al. subset are 4.39×10^{-4} (age corrected slope) and $r = 0.67$. This suggests that the correlation observed in our dataset (Fig. 5) is influenced by local emissions and not differences in transport times or source regions. This is supported by a second method in which we compared the CH_2Cl_2 vs. CO correlation for the ASM samples with the correlation calculated for samples taken within the same 14–40° N latitudinal band but along flight routes to Africa (Sect. 3.2) and across the Atlantic (Sect. 3.4). The correlation for the Africa and Atlantic flights are much weaker and do not show the same dynamical range as the correlation for the India data. This supports our assumption that samples taken during the ASM provide a unique correlation that represents local emissions due to the rapid convection and isolation that occurs within the monsoon system.

$$E_{\text{CH}_2\text{Cl}_2} = E_{\text{CO}} \times \left(\frac{\Delta\text{CH}_2\text{Cl}_2}{\Delta\text{CO}} \right) \quad (1)$$

Secondly, there are errors and assumptions associated with the measured emission ratio. This includes the assumption that the emission ratio measured by CARIBIC is

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similar to that at the source, i.e. it has not been affected by dilution and/or photochemical/chemical loss processes. We believe this assumption to be valid with respect to dilution as previous studies on both the CARIBIC dataset (Rauthe-Schöch et al., 2014; Baker et al., 2011) and independent studies from the same region (Scheeren et al., 2002) have shown that air within the monsoon anticyclone is isolated and mixing in of background air is minimal. One can compare our $\Delta\text{CH}_2\text{Cl}_2/\Delta\text{CO}$ value with those from previous studies (Table 3). Our ratio is within $\sim 20\%$ of the ratio calculated for a ground-based station nearer to emission sources (Gentner et al., 2010). This is an excellent agreement considering that the (lower) Gentner et al. ratio is from 2005 data whereas our ratio is from 2008 and the ratio is increasing over time (changes to the ratio over time are discussed further below). Dilution is also considered in our discussion of the validity of our ratio with respect to the transport time between emission source and sampling (previous paragraph), assuming variations in transport time lead to variations in the degree of mixing, and in the bootstrapping error analysis of the $\Delta\text{CH}_2\text{Cl}_2/\Delta\text{CO}$ regression line (subsequent paragraph). With respect to photochemical loss processes we assume transport times from the boundary layer to our sampling altitude of around 4 days based on Baker et al. (2011). Within this time, CH_2Cl_2 , with a lifetime of around 5 months, does not experience large losses. However, changes to CO are likely to have occurred during this time and so we age-correct our emission ratios with respect to CO using Eq. (2), a method used by both Baker et al. (2011) and Scheeren et al. (2002). Here, the emission ratio at time 0, ER_0 , is related to the emission ratio at time t , ER_t , by accounting for the change in time, Δt (4 days), the reaction rates, k , of CO and CH_2Cl_2 with OH at 298 K and the average concentration of OH. Both k_{CO} at $2.1 \times 10^{-13} \text{ cm}^3 \text{ molec}^{-1} \text{ s}^{-1}$ and $\langle[\text{OH}]\rangle$ at $2.48 \times 10^6 \text{ molec cm}^{-3}$ are taken from Baker et al. (2011, and refs. within) and $k_{\text{CH}_2\text{Cl}_2}$ at $1.1 \times 10^{-13} \text{ cm}^3 \text{ molec}^{-1} \text{ s}^{-1}$ from Villenave et al. (1997). This procedure leads to a correction in our emission ratio for the 2008 monsoon season of around -8% .

$$\text{ER}_0 = \text{ER}_t e^{(k_{\text{CO}} - k_{\text{CH}_2\text{Cl}_2}) \langle[\text{OH}]\rangle \Delta t} \quad (2)$$

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Also associated with the emission ratio are errors arising during the calculation of the $\Delta\text{CH}_2\text{Cl}_2/\Delta\text{CO}$ slope. These errors arise from two sources: (1) uncertainties in the analytical measurements of both CH_2Cl_2 and CO and (2) uncertainties associated with using a slope calculated from a discrete set of samples to calculate a regional emission estimate. The errors associated with (1) are small compared to those associated with (2) and so we use (2) to set bounds on our emission estimates. To investigate possible variations in $\Delta\text{CH}_2\text{Cl}_2/\Delta\text{CO}$ a bootstrapping procedure was applied. Using the Wood (2003) procedure we resampled, with replacement, our CH_2Cl_2 and CO datasets 10 000 times, each time calculating $\Delta\text{CH}_2\text{Cl}_2/\Delta\text{CO}$. The output from the resampling procedure provides a probability distribution for the slope of $\text{CH}_2\text{Cl}_2/\text{CO}$, allowing us to understand how dependent $\Delta\text{CH}_2\text{Cl}_2/\Delta\text{CO}$ may be on the sampled data and allowing us to provide an idea of the potential variation in $\Delta\text{CH}_2\text{Cl}_2/\Delta\text{CO}$. The bootstrapping procedure has been used to calculate a possible range of emission values to aid the comparison between years. In the following text this $\pm 1\sigma$ range is given in brackets following each emission estimate.

CO emissions for the Indian region are taken from the Emission Database for Global Atmospheric Research (EDGAR) v. 4.2 (JRC & PBL, 2009). We include emissions from the following countries; Bangladesh, Bhutan, India, Sri Lanka, Maldives, Nepal and Pakistan. EDGAR emissions are provided per year and are split into categories including various industrial and domestic processes, transport and biomass and bio-fuel burning. Baker et al. (2012) used the Global Fire Emission Database (GFED, v.3.2, van der Werf et al., 2010) to show that CO emitted from biomass burning was greatly reduced during the monsoon, accounting for around 0.5 % of total annual CO fire emissions during 2008. To account for this reduction, the EDGAR emissions were split into two categories: (1) biomass burning related emissions and (2) all other, mainly anthropogenic, emissions. Category (2) was assumed to remain constant throughout the year and the annual emissions provided by EDGAR for CO sources within this category were divided equally between all months. Category (1) emissions were corrected for the effect of the monsoon using the GFED data and the method in Baker

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et al. (2012). This method gave an average monthly emission during the 2008 monsoon of $4.2 \text{ Tg CO month}^{-1}$. Maximum errors on the EDGAR CO database are given as up to $\pm 50\%$ (Olivier et al., 1999), hopefully reduced by our additional use of the GFED database. We do not consider the given error on the EDGAR data further as the main objective of these emission estimates is to provide a comparison of CH_2Cl_2 emissions over time and we assume this error remains constant throughout the EDGAR database.

The CH_2Cl_2 vs. CO correlation within the monsoon ($<40^\circ \text{ N}$, July–September 2008) can be seen in Fig. 5a. The correlation has a statistically significant (Pearson's correlation coefficient, $p < 0.05$) r value of 0.62. Correlations for individual months are also shown in Fig. 5a. No statistical difference (Fisher's z test with a z -crit. value of 0.05) exists between the slopes for individual months, allowing us to use the slope of the correlation for the whole ASM period for our emission estimate. Table 3 includes $\Delta\text{CH}_2\text{Cl}_2/\Delta\text{CO}$ values from this study as well as a range of published values from a variety of sources, including; biomass burning (Simmonds et al., 2006; Rudolph et al., 1995), urban emissions (Gentner et al., 2010) and aircraft data (Scheeren et al., 2002). Whilst the Simmonds et al. (2006) ratio, observed at the remote Cape Grim site, is extremely low, the rest of the ratios show a consistent increase over time due to increased emissions of CH_2Cl_2 .

Using this ratio and the EDGAR CO emissions of $4.2 \text{ Tg CO month}^{-1}$ gives an emission estimate of $1.7 (1.3\text{--}2.1) \text{ Gg CH}_2\text{Cl}_2 \text{ month}^{-1}$ from the Indian region. As industrial sources of CH_2Cl_2 have no seasonality we assume this emission rate is constant over the year and so estimate that $20.3 (15.8\text{--}24.8) \text{ Gg of CH}_2\text{Cl}_2$ were emitted from the Indian region in 2008. The most recent estimate of global emissions is $515 \pm 22 \text{ Gg yr}^{-1}$ given in Montzka et al. (2011b) which is based on top down estimates from Simmonds et al. (2006) from data collected between 1999 and 2003. Considering the caveat that global emissions are likely to have increased since this figure was published, our estimate for emissions from the Indian region in 2008 is roughly 4% of the global total. These estimates are discussed further in Sect. 3.6.

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Venezuela). Figure 6 shows the distribution of CH_2Cl_2 against both latitude and longitude sampled on flights to these destinations. The gradient along the flight tracks, shown in Fig. 6 as average values binned for every 5° latitude and 10° longitude, show very little variation in the early years of the dataset (2001–2002). The latitudinal gradient increases over time and can be seen in the binned mean values for 2009–2011. However, it is still less pronounced than the latitudinal gradients observed en route to Africa or India. A slight decrease in CH_2Cl_2 is observed from north to south or moving west across the Atlantic away from Europe, although the number of samples taken to the west and south are less than to the north and east.

Average values for tropospheric samples taken within each year (Fig. 6c) show an increase in time, as discussed in previous sections. The increase between the 2000–2002 and 2009–2011 periods (the latter chosen to match with the years covered in previous sections) was from $23.2 (\pm 0.4)$ ppt to $32.0 (\pm 0.6)$ ppt, an increase of $8.8 (\pm 0.7)$ ppt or 38 %. The magnitude of this increase is smaller than that observed for Europe and India and closer to that observed for the southern section of the flights to Africa. This is most likely due to the influence of cleaner air masses over the Atlantic and Central America. Similarly low concentrations were seen over the Suriname rainforest in 1998 by Scheeren et al. (2003b) who reported a mean concentration of 16.3 ± 4.4 ppt at similar altitudes to CARIBIC (Table 2).

Quasi-vertical profiles of CH_2Cl_2 along this route can also provide information on changes in CH_2Cl_2 over time. In Fig. 7 profiles of CH_2Cl_2 are plotted as a function of O_3 . A detailed discussion of O_3 as a chemical marker for the structure of the UTLS was provided in Zahn and Brenninkmeijer (2003) and Sprung and Zahn (2010). Briefly, the extratropical O_3 chemical tropopause is observed around 100 ppbv O_3 and can be seen in changes in the relationship between O_3 and tropospheric tracers such as CO and acetone. Above the chemical tropopause, a compact “mixing line” between O_3 and, for example, CO, denotes the mixing of tropospheric and stratospheric air in the extratropical tropopause layer (ExTL). The ExTL extends up to a maximum of 400–500 ppbv O_3 , above which lies the lowermost stratosphere (LMS).

an increase in high concentration “pollution” events (e.g. Figs. 4 and 8). The increase must be linked to increasing industrial use of CH_2Cl_2 , such as its use as a feedstock for the production of HFC-32.

Our results show that CH_2Cl_2 emissions from the Indian subcontinent have increased two- to fourfold in a decade (1998–2008). The annual emissions from the Indian region in 2008, at ~ 20 Gg, are similar to those estimated for the USA in 2005. Other Asian regions may also prove to be emitting large quantities of CH_2Cl_2 : from a limited dataset we suggest that Southeast Asia may be an important source region. However, there is little in situ data available for this region and further investigations are warranted.

Increases in CH_2Cl_2 in UT air masses with the potential to enter the TTL can also be observed in the CARIBIC database. Whilst CH_2Cl_2 is only a minor contributor to stratospheric ozone depletion many other chlorocarbons have stable (e.g. CH_3Cl , see Umezawa et al., 2014) or decreasing (e.g. CFCs, CH_3CCl_3) atmospheric abundances. As our data suggest that CH_2Cl_2 may still be increasing in the atmosphere its relative importance may increase if this trend continues.

**The Supplement related to this article is available online at
doi:10.5194/acpd-14-20721-2014-supplement.**

Acknowledgements. The authors would like to thank the CARIBIC team and associated partner institutions, Lufthansa Airlines, Lufthansa Technik and Fraport for their work and support that has led to the success of CARIBIC over many years. In particular we thank Andreas Zahn and colleagues for the O_3 data, Dieter Scharffe for the CO data and Peter van Velthoven for the back trajectory information. Thanks to Claus Koeppel, Dieter Scharffe, Stefan Weber and Martin Körner for facilitating the operations of the container and TRAC samplers before, during and after flights. Early CARIBIC samples were partly analysed at UEA by Georgina Sturrock. J. Laube acknowledges the support of a NERC fellowship (number NE/I021918/1). The data available in the Supplement would not be possible without the staff at the Cape Grim station and at CSIRO GASLAB Aspendale for collecting and maintaining the Cape Grim air archive and preparing the UEA flask and sub-samples. We also acknowledge

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CSIRO and the Bureau of Meteorology for funding these activities.

The service charges for this open access publication have been covered by the Max Planck Society.

References

- Andreae, M. O. and Merlet, P.: Emission of trace gases and aerosols from biomass burning, *Global Biogeochem. Cy.*, 15, 955–966, doi:10.1029/2000GB001382, 2001.
- Baker, A. K., Slemr, F., and Brenninkmeijer, C. A. M.: Analysis of non-methane hydrocarbons in air samples collected aboard the CARIBIC passenger aircraft, *Atmos. Meas. Tech.*, 3, 311–321, doi:10.5194/amt-3-311-2010, 2010.
- Baker, A. K., Schuck, T. J., Slemr, F., van Velthoven, P., Zahn, A., and Brenninkmeijer, C. A. M.: Characterization of non-methane hydrocarbons in Asian summer monsoon outflow observed by the CARIBIC aircraft, *Atmos. Chem. Phys.*, 11, 503–518, doi:10.5194/acp-11-503-2011, 2011.
- Baker, A. K., Schuck, T. J., Brenninkmeijer, C. A. M., Rauthe-Schöch, A., Slemr, F., Van Velthoven, P. F. J., and Lelieveld, J.: Estimating the contribution of monsoon-related biogenic production to methane emissions from South Asia using CARIBIC observations, *Geophys. Res. Lett.*, 39, L10813, doi:10.1029/2012GL051756, 2012.
- Barletta, B., Meinardi, S., Simpson, I. J., Rowland, F. S., Chan, C.-Y., Wang, X., Zou, S., Chan, L. Y., and Blake, D. R.: Ambient halocarbon mixing ratios in 45 Chinese cities, *Atmos. Environ.*, 40, 7706–7719, doi:10.1016/j.atmosenv.2006.08.039, 2006.
- Bin Babar, Z. and Shareefdeen, Z.: Management and control of air emissions from electronic industries, *Clean Technol. Environ.*, 16, 69–77, doi:10.1007/s10098-013-0594-6, 2014.
- Brenninkmeijer, C. A. M., Crutzen, P. J., Fischer, H., Güsten, H., Hans, W., Heinrich, G., Heintzenberg, J., Hermann, M., Immelmann, T., Kersting, D., Maiss, M., Nolle, M., Pitscheider, A., Pohlkamp, H., Scharffe, D., Specht, K., and Wiedensohler, A.: CARIBIC – civil aircraft for global measurement of trace gases and aerosols in the tropopause region, *J. Atmos. Ocean. Tech.*, 16, 1373–1383, doi:10.1175/1520-0426(1999)016<1373:CCAFGM>2.0.CO;2, 1999.

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Brenninkmeijer, C. A. M., Crutzen, P., Boumard, F., Dauer, T., Dix, B., Ebinghaus, R., Filippi, D., Fischer, H., Franke, H., Frieß, U., Heintzenberg, J., Helleis, F., Hermann, M., Kock, H. H., Koepfel, C., Lelieveld, J., Leuenberger, M., Martinsson, B. G., Miemczyk, S., Moret, H. P., Nguyen, H. N., Nyfeler, P., Oram, D., O'Sullivan, D., Penkett, S., Platt, U., Pucek, M., Ramonet, M., Randa, B., Reichelt, M., Rhee, T. S., Rohwer, J., Rosenfeld, K., Scharffe, D., Schlager, H., Schumann, U., Slemr, F., Sprung, D., Stock, P., Thaler, R., Valentino, F., van Velthoven, P., Waibel, A., Wandel, A., Waschitschek, K., Wiedensohler, A., Xueref-Remy, I., Zahn, A., Zech, U., and Ziereis, H.: Civil Aircraft for the regular investigation of the atmosphere based on an instrumented container: The new CARIBIC system, *Atmos. Chem. Phys.*, 7, 4953–4976, doi:10.5194/acp-7-4953-2007, 2007.

Cox, M. L., Sturrock, G. A., Fraser, P. J., Siems, S. T., Krummer, P. B., and O'Doherty, S.: Regional sources of methyl chloride, chloroform and dichloromethane identified from AGAGE observations at Cape Grim, Tasmania, 1998–2000, *J. Atmos. Chem.*, 45, 79–99, doi:10.1023/A:1024022320985, 2003.

Cox, R. A., Derwent, R. G., and Eggleton, E. J.: Photochemical oxidation of halocarbons in the troposphere, *Atmos. Environ.*, 10, 305–308, doi:10.1016/0004-6981(76)90170-0, 1976.

Daikin: World's First Commercialization of Air Conditioning Equipment Using Next-Generation Refrigerant HFC-32 by Daikin, available at: http://www.daikinindia.com/media-room/daikin-announcements.html?view=detail&n_id=10 (last access: 11 April 2014), 2012.

Dalvi, M., Beig, G., Patil, U., Kaginalkar, A., Sharma, C., and Mitra, A. P.: A GIS based methodology for gridding of large-scale emission inventories: application to carbon-monoxide emissions over Indian region, *Atmos. Environ.*, 40, 2995–3007, doi:10.1016/j.atmosenv.2006.01.013, 2006.

Dickerson, R. R., Andreae, M. O., Campos, T., Mayol-Bracero, O. L., Neusuess, C., and Streets, D. G.: Analysis of black carbon and carbon monoxide observed over the Indian Ocean: implications for emissions and photochemistry, *J. Geophys. Res.*, 107, 8017, doi:10.1029/2001JD000501, 2002.

Fortems-Cheiney, A., Chevallier, F., Pison, I., Bousquet, P., Szopa, S., Deeter, M. N., and Clerbaux, C.: Ten years of CO emissions as seen from Measurements of Pollution in the Troposphere (MOPITT), *J. Geophys. Res.*, 116, D05304, doi:10.1029/2010JD014416, 2011.

Gautrois, M., Brauers, T., Koppmann, R., Rohrer, F., Stein, O., and Rudolph, J.: Seasonal variability and trends of volatile organic compounds in the lower polar troposphere, *J. Geophys. Res.*, 108, 4393, doi:10.1029/2002JD002765, 2003.

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- Gentner, D. R., Miller, A. M., and Goldstein, A. H.: Seasonal variability in anthropogenic halo-carbon emissions, *Environ. Sci. Technol.*, 44, 5377–5382, doi:10.1021/es1005362, 2010.
- IHS: Chlorinated Methanes, available from: <http://www.ihs.com/products/chemical/planning/ceh/chlorinatedmethanes.aspx> (last access: 13 May 2014), 2014.
- 5 JRC & PBL (European Commission Joint Research Centre, JRC, and the Netherlands Environmental Assessment Agency, PBL): Emission Database for Global Atmospheric Research (EDGAR), release version 4.0, available at: <http://edgar.jrc.ec.europa.eu/> (last access: 8 May 2014), 2009.
- Koppmann, R., Johnen, F. J., Plass-Dulmer, C., and Rudolph, J.: Distribution of methyl chloride, dichloromethane, trichloroethene and tetrachloroethene over the North and South Atlantic, *J. Geophys. Res.*, 98, 517–526, doi:10.1029/93JD01864, 1993.
- 10 Kowalska, J. and Gierczak, T.: Qualitative and quantitative analyses of the halogenated volatile organic compounds emitted from the office equipment items, *Indoor Built Environ.*, 22, 920–931, doi:10.1177/1420326X12458299, 2012.
- 15 Laube, J. C., Martinerie, P., Witrant, E., Blunier, T., Schwander, J., Brenninkmeijer, C. A. M., Schuck, T. J., Bolder, M., Röckmann, T., van der Veen, C., Bönisch, H., Engel, A., Mills, G. P., Newland, M. J., Oram, D. E., Reeves, C. E., and Sturges, W. T.: Accelerating growth of HFC-227ea (1,1,1,2,3,3,3-heptafluoropropane) in the atmosphere, *Atmos. Chem. Phys.*, 10, 5903–5910, doi:10.5194/acp-10-5903-2010, 2010.
- 20 Law, K. S., Sturges, W. T., Blake, D. R., Blake, N. J., Burkholder, J. B., Butler, J. H., Cox, R. A., Haynes, P. H., Ko, M. K. W., Kreher, K., Mari, C., Pfeilsticker, K., Plane, J. M. C., Salawitch, R. J., Schiller, C., Sinnhuber, B.-M., von Glasow, R., Warwick, N. J., Weubbles, D. J., Yvon-Lewis, S. A., Butz, A., Considine, D. B., Dorf, M., Froidevaux, L., Kovalenko, L. J., Livesey, N. J., Nassar, R., Sioris, C. E., Weisenstein, D. K.: Halogenated very short-lived substances, Chapter 2, in: *Scientific Assessment of Ozone Depletion: 2006, Global Ozone Research and Monitoring Project Report No. 50*, World Meteorological Organization, Geneva, Switzerland, 2007.
- 25 Le Canut, P., Andreae, M. O., Harris, G. W., Wienhold, F. G., and Zenker, T.: Airborne studies of emissions from savannah fires in southern Africa. 1. Aerosol emissions measured with a laser optical particle counter, *J. Geophys. Res.*, 101, 23615–23630, doi:10.1029/95JD02610, 1996.
- 30

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Majumdar, D. and Srivastava, A.: Volatile organic compound emissions from municipal solid waste disposal sites: a case study of Mumbai, India, *JAPCA J. Air Waste Ma.*, 62, 398–407, doi:10.1080/10473289.2012.655405, 2012

McCulloch, A.: Determination of comparative HCFC and HFC emission profiles for the foam and refrigeration sectors until 2015. Part 3: Total emissions and global atmospheric concentrations, Marbury Technical Consulting and University of Bristol, 2004.

McCulloch, A. and Midgley, P. M.: The production and global distribution of emissions of trichloroethene, tetrachloroethene and dichloromethane over the period 1988–1992, *Atmos. Environ.*, 30, 601–608, doi:10.1016/1352-2310(09)50032-5, 1996.

McCulloch, A., Aucott, M. L., Graedel, E., Kleinman, G., Midgley, P. M., and Li, Y.-F.: Industrial emissions of trichloroethene, tetrachloroethene and dichloromethane: reactive chlorine emissions inventory, *J. Geophys. Res.*, 104, 8417–8427, doi:10.1029/1999JD900011, 1999.

Miller, J. B., Lehman, S. J., Montzka, S. A., Sweeney, C., Miller, B. R., Karion, A., Wolak, C., Dlugokencky, E. J., Southon, J., Turnbull, J. C., and Tans, P. P.: Linking emissions of fossil fuel CO₂ and other anthropogenic trace gases using atmospheric ¹⁴CO₂, *J. Geophys. Res.*, 117, D08302, doi:10.1029/2011JD017048, 2012.

Millet, D. B., Atlas, E. L., Blake, D. R., Blake, N. J., Diskin, G. S., Holloway, J. S., Hudman, R. C., Meinardi, S., Ryerson, T. B., and Sachse, G. W.: Halocarbon emissions from the United States and Mexico and their global warming potential, *Environ. Sci. Technol.*, 43, 1055–1060, doi:10.1021/es802146j, 2009.

Montzka, S. A., Krol, M., Blugokencky, E., Hall, B., Jöckel, P., and Lelieveld, J.: Small interannual variability of global atmospheric hydroxyl, *Science*, 331, 67–69, doi:10.1126/science.1197640, 2011a.

Montzka, S. A., Reimann, S., Engel, A., Kruger, K., O'Doherty, S., Sturges, W. T., Blake, D., Dorf, M., Fraser, P., Friodevaux, L., Jucks, K., Kreher, K., Kurylo, M. J., Miller, J., Neilson, O.-J., Orkin, V. L., Prinn, R. G., Rhew, R., Santee, M. L., Stohl, A., and Verdonik, D.: Ozone-depleting substances (ODSs) and related chemicals, Chapter 1, in: *Scientific Assessment of Ozone Depletion: 2010*, Global Ozone Research and Monitoring Project-Report No. 52, World Meteorological Organization, Geneva, Switzerland, 2011b.

NRDC: Cooling India with Less Warming: The Business Case for Phasing Down HFCs in Room and Vehicle Air Conditioners, available at: <http://www.nrdc.org/international/india/files/air-conditioner-efficiency-IP.pdf> (last access: 11 April 2014), National Resources Defense Council, 2013.

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- O'Doherty, S., Rigby, M., Mühle, J., Ivy, D. J., Miller, B. R., Young, D., Simmonds, P. G., Reimann, S., Vollmer, M. K., Krummel, P. B., Fraser, P. J., Steele, L. P., Dunse, B., Salameh, P. K., Harth, C. M., Arnold, T., Weiss, R. F., Kim, J., Park, S., Li, S., Lunder, C., Hermansen, O., Schmidbauer, N., Zhou, L. X., Yao, B., Wang, R. H. J., Manning, A. J., and Prinn, R. G.: Global emissions of HFC-143a (CH_3CF_3) and HFC-32 (CH_2F_2) from in situ and air archive atmospheric observations, *Atmos. Chem. Phys. Discuss.*, 14, 6471–6500, doi:10.5194/acpd-14-6471-2014, 2014.
- Olivier, J. G. J., Bloos, J. P. J., Berdowski, J. J. M., Visschedijk, A. J. H., and Bouwman, A. F.: A 1990 global emission inventory of anthropogenic sources of carbon monoxide on $1^\circ \times 1^\circ$ developed in the framework of EDGAR/GEIA, *Chemosphere*, 1, 1–17, doi:10.1016/S1465-9972(99)00019-7, 1999.
- Palmer, P. I., Jacob, D. J., Mickley, L. J., Blake, D. R., Sachse, G. W., Fuelberg, H. E., and Kiley, C. M.: Eastern Asian emissions of anthropogenic halocarbons deduced from aircraft concentration data, *J. Geophys. Res.-Atmos.*, 108, 4753, doi:10.1029/2003JD003591, 2003.
- Ramanathan, R., Anand, R., Jain, A., and Rao, J. M.: Process for the Production of Difluoromethane, United States Patent no. US 6,723,887 B2, 2004.
- Rauthe-Schöch, A., Baker, A. K., Schuck, T., Brenninkmeijer, C. A. M., Zahn, A., Herrmann, M., Stratmann, G., Ziereis, H., and van Velthoven, P.: Pollution transport from the South Asian Monsoon anticyclone, *Atmos. Chem. Phys. Discuss.*, to be submitted, 2014.
- Roberts, G., Wooster, M. J., and Lagoudakis, E.: Annual and diurnal african biomass burning temporal dynamics, *Biogeosciences*, 6, 849–866, doi:10.5194/bg-6-849-2009, 2009.
- Rudolph, J., Khedim, A., Koppmann, R., and Bonsang, B.: Field study of the emissions of methyl chloride and other halocarbons from biomass burning in western Africa, *J. Atmos. Chem.*, 22, 67–80, doi:10.1007/BF00708182, 1995.
- Scharffe, D., Slemr, F., Brenninkmeijer, C. A. M., and Zahn, A.: Carbon monoxide measurements onboard the CARIBIC passenger aircraft using UV resonance fluorescence, *Atmos. Meas. Tech.*, 5, 1753–1760, doi:10.5194/amt-5-1753-2012, 2012.
- Schauffler, S. M., Heidt, L. E., Pollock, W. H., Gilpin, T. M., Vedder, J. F., Solomon, S., Lueb, R. A., and Atlas, E. L.: Measurements of halogenated organic compounds near the tropical tropopause, *Geophys. Res. Lett.*, 20, 2567–2570, doi:10.1029/93GL02840, 1993.
- Scheele, M. P., Siegmund, P. C., and Van Velthoven, P. F. J.: Sensitivity of trajectories to data resolution and its dependence on the starting point: in or outside a tropopause fold, *Meteorol. Appl.*, 3, 267–273, doi:10.1002/met.5060030308, 1996.

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Scheeren, H., Lelieveld, J., De Gouw, J. A., Van der Veen, C., and Fischer, H.: Methyl chloride and other chlorocarbons in polluted air during INDOEX, *J. Geophys. Res.*, 107, 8015, doi:10.1029/2001JD001121, 2002.

Scheeren, H. A., Lelieveld, J., Roelofs, G. J., Williams, J., Fischer, H., de Reus, M., de Gouw, J. A., Warneke, C., Holzinger, R., Schlager, H., Klüpfel, T., Bolder, M., van der Veen, C., and Lawrence, M.: The impact of monsoon outflow from India and South-east Asia in the upper troposphere over the eastern Mediterranean, *Atmos. Chem. Phys.*, 3, 1589–1608, doi:10.5194/acp-3-1589-2003, 2003a.

Scheeren, H. A., Lelieveld, J., Williams, J., Fischer, H., and Warneke, C.: Measurements of reactive chlorocarbons over the Surinam tropical rain forest: indications for strong biogenic emissions, *Atmos. Chem. Phys. Discuss.*, 3, 5469–5512, doi:10.5194/acpd-3-5469-2003, 2003b.

Schuck, T. J., Brenninkmeijer, C. A. M., Slemr, F., Xueref-Remy, I., and Zahn, A.: Greenhouse gas analysis of air samples collected onboard the CARIBIC passenger aircraft, *Atmos. Meas. Tech.*, 2, 449–464, doi:10.5194/amt-2-449-2009, 2009.

Schuck, T. J., Brenninkmeijer, C. A. M., Baker, A. K., Slemr, F., von Velthoven, P. F. J., and Zahn, A.: Greenhouse gas relationships in the Indian summer monsoon plume measured by the CARIBIC passenger aircraft, *Atmos. Chem. Phys.*, 10, 3965–3984, doi:10.5194/acp-10-3965-2010, 2010.

Simmonds, P. G., Manning, A. J., Cunnold, D. M., McCulloch, A., O'Doherty, S., Derwent, R. G., Krummel, P. B., Fraser, P. J., Dunse, B., Porter, L. W., Wang, R. H. J., Grealley, B. R., Miller, B. R., Salameh, P., Weiss, R. F., and Prinn, R. G.: Global trends, seasonal cycles, and European emissions of dichloromethane, trichloroethene, and tetrachloroethene from the AGAGE observations at Mace Head, Ireland, and Cape Grim, Tasmania, *J. Geophys. Res.*, 111, D18304, doi:10.1029/2006JD007082, 2006.

Simpson, I. J., Akagi, S. K., Barletta, B., Blake, N. J., Choi, Y., Diskin, G. S., Fried, A., Furlberg, H. E., Meinardi, S., Rowland, F. S., Vay, S. A., Weinheimer, A. J., Wennberg, P. O., Wiebring, P., Wisthaler, A., Yang, M., Yokelson, R. J., and Blake, D. R.: Boreal forest fire emissions in fresh Canadian smoke plumes: C₁–C₁₀ volatile organic compounds (VOCs), CO₂, CO, NO₂, NO, HCN and CH₃CN, *Atmos. Chem. Phys.*, 11, 6445–6463, doi:10.5194/acp-11-6445-2011, 2011.

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Sprung, D. and Zahn, A.: Acetone in the upper troposphere/lowermost stratosphere measured by the CARIBIC passenger aircraft: distribution, seasonal cycle, and variability, *J. Geophys. Res.*, 115, D16301, doi:10.1029/2009JD012099, 2010.

Streets, D. G., Bond, T. C., Carmichael, R., Fernandes, S. D., Fu, Q., He, D., Klimont, Z., Nelson, S. M., Tsai, N. Y., Wang, M. Q., Woo, J.-H., and Yarber, K. F.: An inventory of gaseous and primary aerosol emissions in Asia in the year 2000, *J. Geophys. Res.*, 108, 8809, doi:10.1029/2002JD003093, 2003.

Trudinger, C. M., Etheridge, D. M., Sturrock, G. A., Fraser, P. J., Krummer, P. B., and McCulloch, A.: Atmospheric histories of halocarbons from analysis of Antarctic firn air: methyl bromide, methyl chloride, chloroform, and dichloromethane, *J. Geophys. Res.*, 109, D22310, doi:10.1029/2004JD004932, 2004.

Umezawa, T., Baker, A. K., Oram, D., Sauvage, C., O'Sullivan, D., Rauthe-Schöch, A., Montzka, S. A., Zahn, A., and Brenninkmeijer, C. A. M.: Methyl chloride in the upper troposphere observed by the CARIBIC passenger aircraft observatory: large-scale distributions and Asian summer monsoon outflow, *J. Geophys. Res.-Atmos.*, 119, 5542–5558, doi:10.1002/2013JD021396, 2014.

UNEP CAP: Fact Sheet no. 26. Alternatives for Carbon Tetrachloride – CCl₄ (CTC) in laboratory and analytical uses, United Nations Environment Programme Compliance Assistance Programme, 2009.

Velders, G. J. M., Fahey, D. W., Daniel, J. S., McFarland, M., and Andersen, S. O.: The large contribution of projected HFC emissions to future climate forcing, *P. Natl. Acad. Sci. USA*, 106, 10949–54, doi:10.1073/pnas.0902817106, 2009.

Villeneuve, E., Orkin, V. L., Huie, R. E., and Kurylo, M. J.: Rate Constant for the reaction of OH radicals with dichloromethane, *J. Phys. Chem.*, 101, 8513–8517, doi:10.1021/jp9721614, 1997.

van der Werf, G. R., Randerson, J. T., Giglio, L., Collatz, G. J., Mu, M., Kasibhatla, P. S., Morton, D. C., DeFries, R. S., Jin, Y., and van Leeuwen, T. T.: Global fire emissions and the contribution of deforestation, savanna, forest, agricultural, and peat fires (1997–2009), *Atmos. Chem. Phys.*, 10, 11707–11735, doi:10.5194/acp-10-11707-2010, 2010.

Wofsy, S. C., Daube, B. C., Jimenez, R., Kort, E., Pittman, J. V., Park, S., Commane, R., Xiang, B., Santoni, G., Jacob, D., Fisher, J., Pickett-Heaps, C., Wang, H., Wecht, K., Wang, Q.-Q., Stephens, B. B., Shertz, S., Watt, A. S., Romashkin, P., Campos, T., Haggerty, J., Cooper, W. A., Rogers, D., Beaton, S., Hendershot, R., Elkins, J. W., Fahey, D. W., Gao, R. S.,

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Moore, F., Montzka, S. A., Schwarz, J. P., Perring, A. E., Hurst, D., Miller, B. R., Sweeney, C., Oltmans, S., Nance, D., Hintsa, E., Dutton, G., Watts, L. A., Spackman, J. R., Rosenlof, K. H., Ray, E. A., Hall, B., Zondlo, M. A., Diao, M., Keeling, R., Bent, J., Atlas, E. L., Lueb, R., and Mahoney, M. J.: HIPPO Combined Discrete Flask and GC Sample GHG, Halo-, Hydrocarbon Data (R_20121129), available at: doi:10.3334/CDIAC/hippo_012 (Release 20121129), file HIPPO_noaa_flask_allparams_merge_insitu_20121129 (last access: 6 April 2014), Carbon Dioxide Information Analysis Center, Oak Ridge National Laboratory, Oak Ridge, Tennessee, USA, 2012.

Wood, M.: Resample Procedure for Bootstrap Confidence Intervals, available at: http://woodm.myweb.port.ac.uk/programs.htm (last access: 8 April 2014), 2003.

Worton, D. R., Sturges, W. T., Schwander, J., Mulvaney, R., Barnola, J.-M., and Chappel-laz, J.: 20th century trends and budget implications of chloroform and related tri-and dihalomethanes inferred from firn air, *Atmos. Chem. Phys.*, 6, 2847–2863, doi:10.5194/acp-6-2847-2006, 2006.

Zahn, A. and Brenninkmeijer, C. A. M.: New directions: a chemical tropopause defined, *Atmos. Environ.*, 37, 439–440, doi:10.1016/S1352-2310(02)00901-9, 2003.

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Table 1. A summary of CARIBIC data used in this study.

Region	Annual	Temporal data coverage		Number of tropospheric samples (*)
			Monthly	
Europe	Data collected 1998–2002 and 2005–2012 Min. coverage: $n = 1$ in 1999 Max. coverage: $n = 16$ in both 2009 and 2011	All months covered	123	
Africa	2000, 2009–2011. Individual flights:	Mainly NH winter	140	
	2000	May, Jul, Dec	32	
	2009	Mar, Oct	47	
	2010	Nov, Dec	22	
	2011	Jan, Feb, Mar	39	
India	Data collected 1998–2001, 2008, 2011 and 2012. The summer monsoon (Jul, Aug, Sep) was sampled in 1998, 1999, 2000 and 2008. In other years samples were only taken outside of the monsoon season.		295	
	Monsoon	Jul, Aug, Sep	105	
	Non-monsoon	Rest of year	190	
North and Central Atlantic	Data collected 2001–2002, 2007, 2009–2012	All months covered	282 (108)	
Tropics	All data within $\pm 25^\circ$ of equator. All years included.	All months covered	539 (34)	

* Stratospheric samples have been excluded from the analysis of most regions, see Sect. 2. Where stratospheric data are used (flights across the North and Central Atlantic and in the tropics) the number of stratospheric samples is shown in brackets.

Table 2. CH₂Cl₂ descriptive statistics for regions included in this study and a comparison to data from the existing literature.

Study and Region	Time period	Values ^a /ppt
This paper		
Europe		
CARIBIC data	1998–2001	$\bar{x} = 24.9 \pm 1.1, n = 16$
	2009–2012	$\bar{x} = 38.6 \pm 1.2, n = 49$
	Increase	13.7 ± 1.6
Mace Head NOAA data	1998–2001	$\bar{x} = 32.1 \pm 1.6, n = 15$
	2009–2012	$\bar{x} = 45.7 \pm 0.7, n = 88$
	Increase	13.6 ± 1.7
Africa		
Above 30° N	2000	$\bar{x} = 21.7 \pm 0.8, n = 3$
	2009–2011	$\bar{x} = 34.2 \pm 1.6, n = 23$
	Increase	12.5 ± 1.8
Below 30° N	2000	$\bar{x} = 15.8 \pm 0.6, n = 29$
	2009–2011	$\bar{x} = 22.7 \pm 0.6, n = 85$
	Increase	6.9 ± 0.8
India		
Summer monsoon (Jul–Sep)	1998–2000	$\bar{x} = 21.5 \pm 0.5, n = 56$
	2008	$\bar{x} = 36.4 \pm 1.3, n = 50$
	Increase	14.9 ± 1.4
Non-monsoon months	1998–2000	$\bar{x} = 20.1 \pm 0.5, n = 81$
	2008	$\bar{x} = 30.4 \pm 1.2, n = 62$
	Increase	10.3 ± 1.3
North and Central Atlantic	2000–2002	$\bar{x} = 23.2 (\pm 0.4), n = 89$
	2009–2011	$\bar{x} = 32.0 (\pm 0.6), n = 180$
	Increase	$8.8 (\pm 0.7)$
Other aircraft studies		
TTL (Schauffler et al., 1993)	1991–1992	$\bar{x} = 14.9, \sigma = 1.1, n = 12$
Suriname rainforest (Scheeren et al., 2003b)	1998	
< 2 km altitude		$\bar{x} = 26.0 \pm 2.9, \bar{x} = 26.2, n = 42$
3–7 km altitude		$\bar{x} = 16.4 \pm 0.3, \bar{x} = 16.4, n = 12$
7–12 km altitude		$\bar{x} = 16.3 \pm 4.4, \bar{x} = 17.5, n = 26$
Tropical Indian Ocean 1.2–12.5 km altitude (Scheeren et al., 2002)	1999	$\bar{x} = 29, SD = 12, n = 71$
ASM outflow, E. Mediterranean 6–13 km alt. (Scheeren et al., 2003a)	2001	$\bar{x} = 23, \sigma = 3$
Canada & Greenland, up to 12 km alt. (Simpson et al., 2011)	2008	$\bar{x} = 35.8 \pm 2.9$
Tropics (0 ± 25°), 345–350 K θ band (HIPPO, Wofsy et al., 2012)	2009–2011	$\bar{x} = 26.3, R = 15.87\text{--}49.83, n = 20$
Ground-based		
Atlantic cruise 45° N–30° S (Koppmann et al., 1993)	1989	
SH		$\bar{x} = 18 \pm 4$
NH		$\bar{x} = 36 \pm 6$ ppt
Alert, Canada (Gautrois et al., 2003)	1989–1996	$\bar{x} = 47.2 \pm 2, \bar{x} = 45.8, R = 24.2\text{--}71.6$
Cape Grim, Tasmania (Cox et al., 2003)	1998–2000	$\bar{x} = 8.9 \pm 0.2$
Chinese cities (Barletta et al., 2006)	2001	
Background		$\bar{x} = 28, \sigma = 4$
Urban		$\bar{x} = 226, \sigma = 232$

^a \bar{x} = mean, \bar{x} = median, σ = standard deviation, R = range and n = number of samples. For our study \pm refers to the standard error.

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Table 3. A comparison of enhancement ratios from this study (airmass age corrected) and existing literature.

Source	$\Delta\text{CH}_2\text{Cl}_2/\Delta\text{CO}/\text{mol mol}^{-1}$
India, summer monsoon period (this study)	
1998	1.00×10^{-4} , $r = 0.69$
1999	1.58×10^{-4} , $r = 0.53$
2000	2.49×10^{-4} , $r = 0.86$
2008	4.00×10^{-4} , $r = 0.62$
Other studies ^a	
Biomass burning, Africa savanna, 1991 ^b	2.48×10^{-5} (error = 26 %), $r = 0.65$
Wildfires, Cape Grim, 1998–2004 ^c	$< 1-6 \times 10^{-7}$
Asian pollution outflow, Bay of Bengal, 1999 ^d	4.4×10^{-5} ($\pm 4.7 \times 10^{-5}$), $r = 0.39$
Urban, California, 2005 ^e	$3.10 \times 10^{-4} \pm 3.0 \times 10^{-5}$, $r = 0.58-0.66$
Urban, USA, 2004 ^f	2.39×10^{-4} ($1.78 \times 10^{-4}-2.9 \times 10^{-4}$), $r = 0.56-0.83$
Urban, Mexico, 2006 ^f	1.93×10^{-4} ($1.09 \times 10^{-4}-2.92 \times 10^{-4}$), $r = 0.43-0.81$

^a Described by: emission source, location, year of study.

^b Rudolph et al. (1995).

^c Simmonds et al. (2006).

^d Scheeren et al. (2002).

^e Gentner et al. (2010).

^f Millet et al. (2009).

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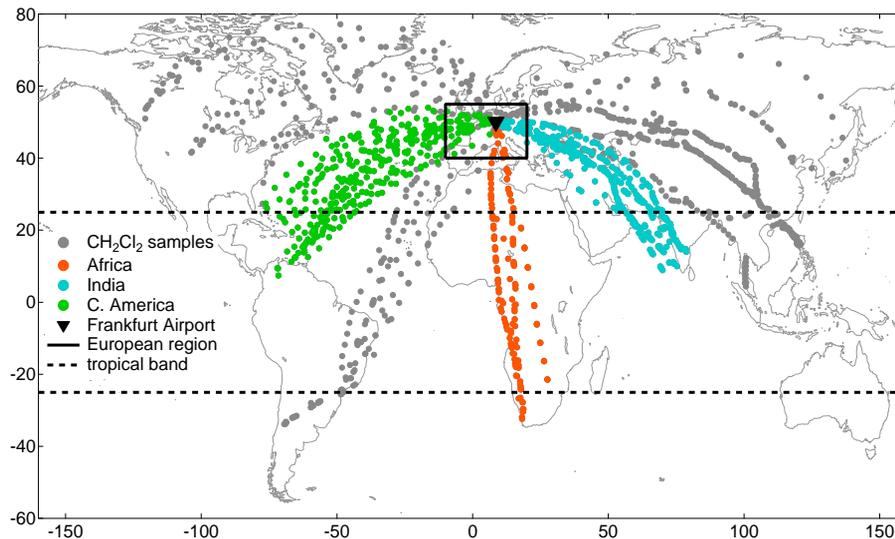


Figure 1. CARIBIC whole air samples analysed for CH_2Cl_2 between 1998–2002 and 2005–2012. Flight routes used in this study are highlighted, see inset legend.

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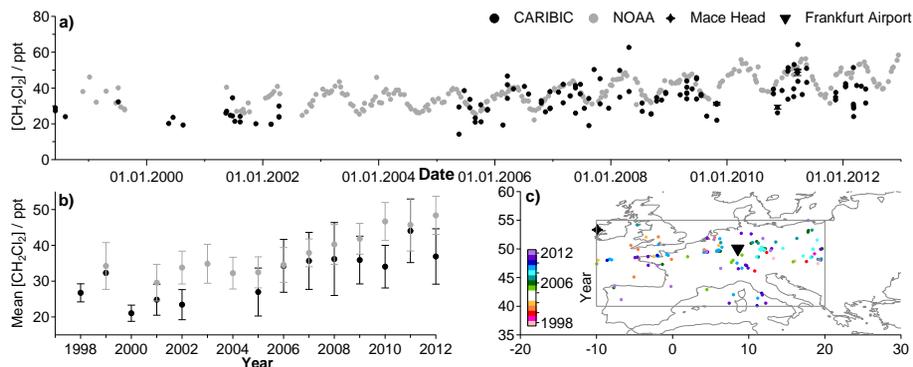


Figure 2. (a) European CH_2Cl_2 time series from June 1998 to December 2012. Error bars given for combined value based on multiple measurements, see Sect. 2. (b) Annual mean CH_2Cl_2 values for all CARIBIC and NOAA Mace Head data, error bars are 1σ . (c) Geographical distribution of CARIBIC samples coloured by year (see inset colour bar) within a box spanning 40–55° N and -10–20° E and in relation to Frankfurt Airport (CARIBIC2 base) and NOAA sampling site at Mace Head.

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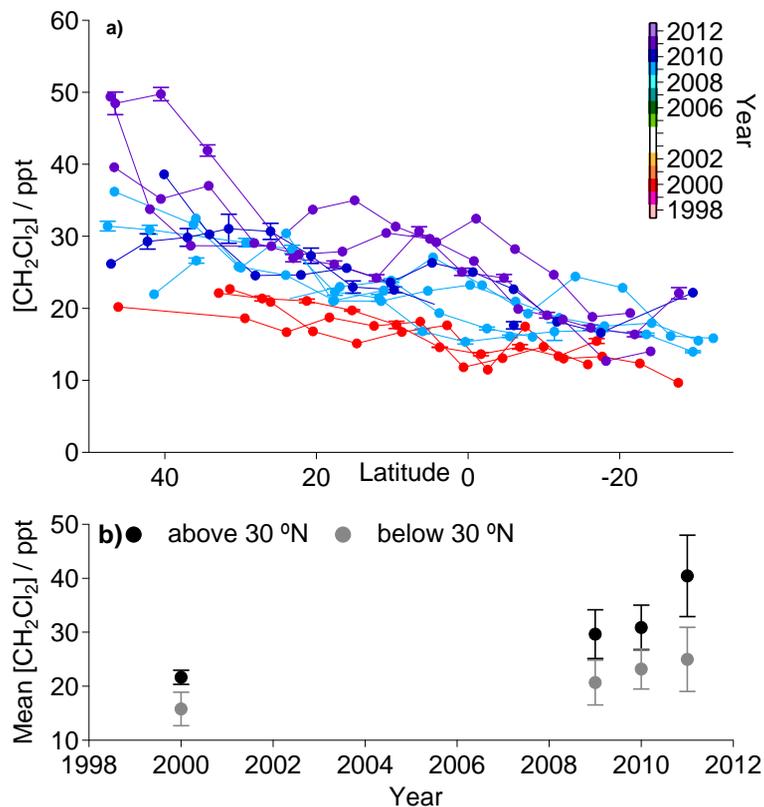


Figure 3. (a) Latitudinal distributions of CH_2Cl_2 observed during flights to South Africa, colour = year (see inset colour bar). Error bars given for combined value based on multiple measurements, see Sect. 2. (b) Annual mean tropospheric CH_2Cl_2 values for data shown in panel a, split into above and below 30°N , error bars are 1σ .

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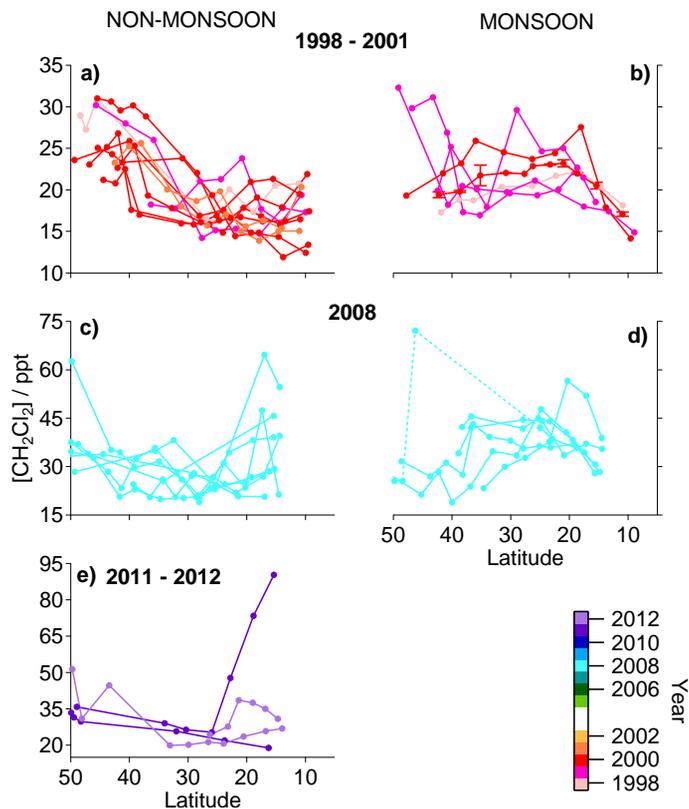


Figure 4. Latitudinal distributions of CH_2Cl_2 observed during flights to India for non-monsoon months on the left and monsoon months (July, August, September) on the right. Colour = year, see inset colour bar. Error bars given for combined value based on multiple measurements, see Sect. 2.

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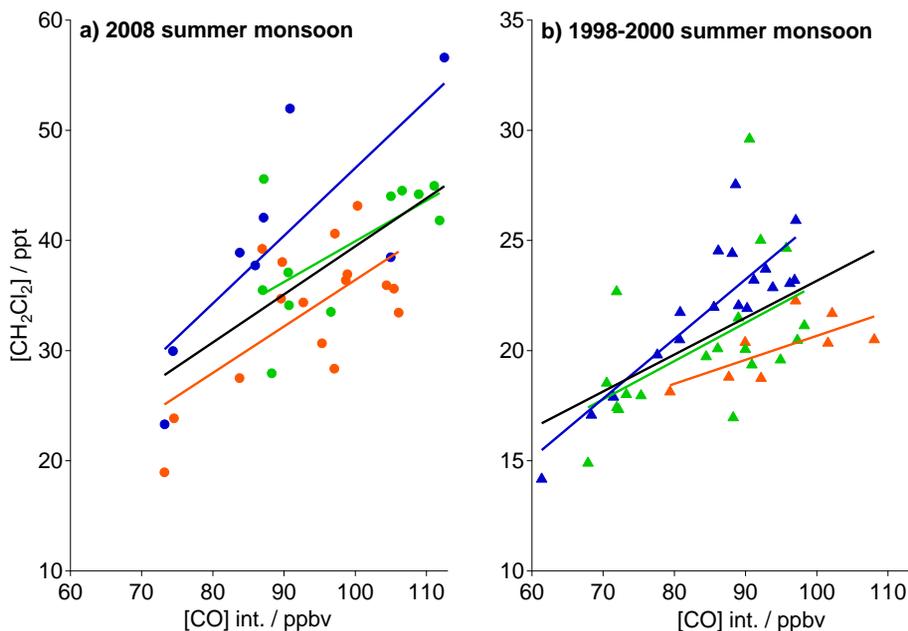


Figure 5. Correlation plots of CH_2Cl_2 and CO for **(a)** the 2008 summer monsoon period (coloured by individual months) and **(b)** the early years of the CARIBIC India dataset (coloured by individual years). Further details in Table 3.

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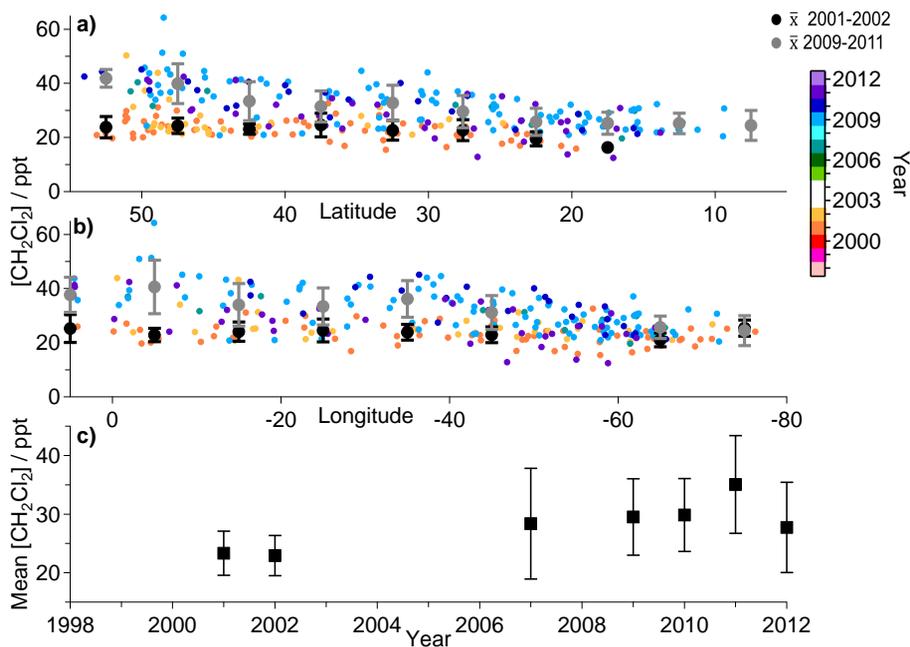


Figure 6. (a) Latitudinal and (b) longitudinal distributions of CH_2Cl_2 along flight routes across the North and Central Atlantic to Central America. Colour = year (see inset colour bar). Average, \bar{x} , values for 5° latitude and 10° longitude bins are shown for 2001–2002 and 2009–2011 (see Sect. 3.4), error bars are 1σ variation within these bands. (c) Annual mean CH_2Cl_2 concentrations for all samples taken along these routes, error bars are 1σ variation.

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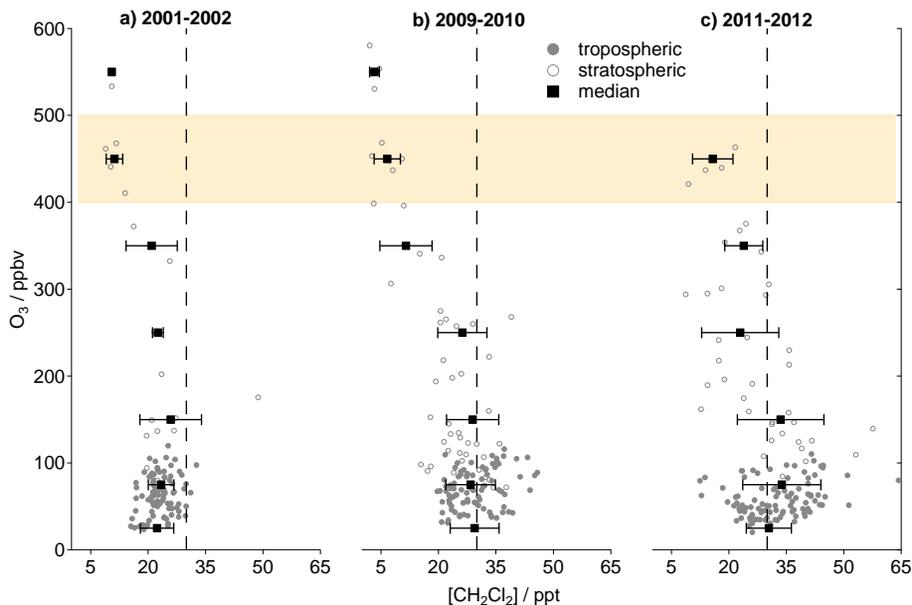


Figure 7. Profiles of CH_2Cl_2 relative to O_3 from samples collected on flights across the North and Central Atlantic to Central America. Median values are for 50 ppbv O_3 bins between 0–100 ppbv and 100 ppbv O_3 bins above this, error bars are 1σ . The coloured band highlights the region between 400–500 ppbv O_3 discussed in Sect. 3.4. The dashed line represents 30 ppt CH_2Cl_2 (see Sect. 1), provided as a visual marker to illustrate the shift over time to higher concentrations of CH_2Cl_2 .

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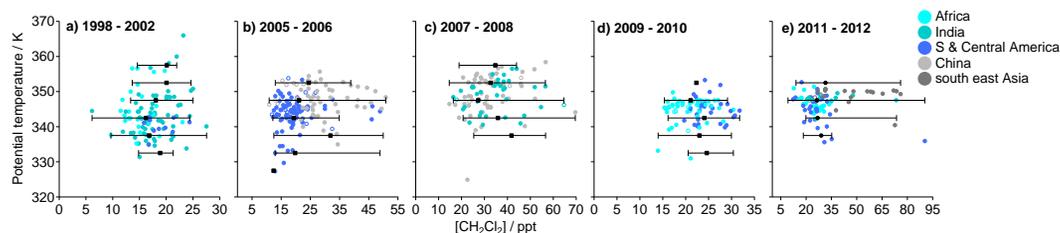


Figure 8. Profiles of CH₂Cl₂ relative to potential temperature for samples taken within the latitude range $0^\circ \pm 25^\circ$. Median (error bars are range) values for 5 K bins are overlaid in black (error bars = 1σ). Colour represents flight route, see inset colour bar.

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